

Article

Scalable Green Approach Toward Fragrant Acetates

Eva Puchl'ová and Peter Szolcsányi *

Department of Organic Chemistry, Slovak University of Technology, Radlinského 9, 81237 Bratislava, Slovakia; eva.puchlova@stuba.sk

* Correspondence: peter.szolcsanyi@stuba.sk

Received: 18 June 2020; Accepted: 13 July 2020; Published: date

Contents

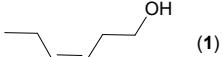
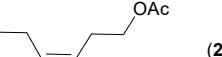
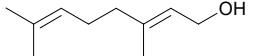
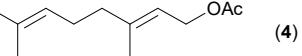
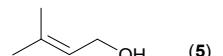
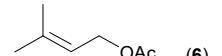
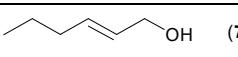
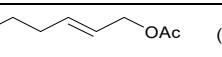
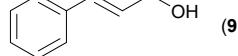
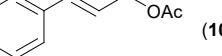
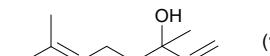
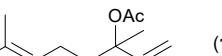
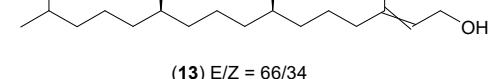
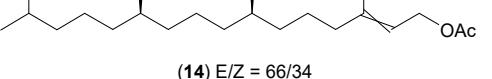
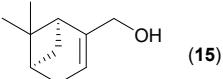
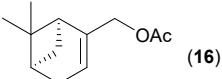
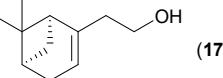
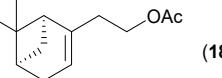
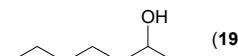
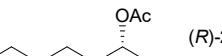
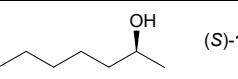
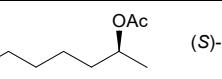
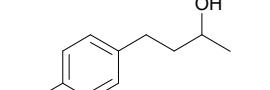
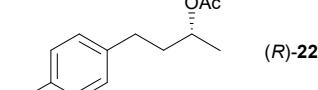
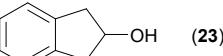
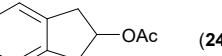
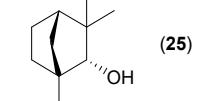
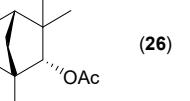
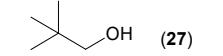
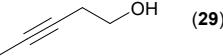
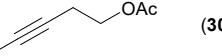
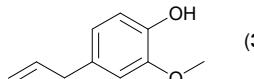
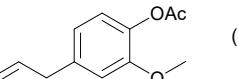
Optimisation screening of enzymatic acetylation of (<i>Z</i>)-hex-3-en-1-ol	Page 1
GC-FID retention times of respective alcohols and corresponding acetates	Pages 2-3
Time-dependent GC-FID ratios of competitive enzymatic acetylations	Pages 3-4
Chiral GC-FID analyses of heptan-2-ol and heptan-2-yl acetate	Page 5
Chiral HPLC analysis of rhododendrol	Page 6
Copies of NMR spectra of isolated compounds	Pages 6-8

Table S1. Optimisation screening of enzymatic acetylation of (*Z*)-hex-3-en-1-ol (1).

Entry	Lipozyme 435 [% wt]	EGDA [equiv]	Temperature [°C]	Cosolvent (c) [mol. L ⁻¹]	Reaction time [h]	GC-FID ratio (1)/(2) ^a [%]
1	10	4.5	40	-	5.5	3/97
2	1	4.5	40	-	18	3/97
3	1	3	40	-	22	4/96
4	1	1.5	40	-	24	9/91
5	1	1.5	45	-	17	9/91
6	2	1.5	40	-	24	9/91
7	2	2	40	-	19	6/94
8	2	1	40	-	16	15/85
9	2	0.5	40	-	16	29/71
10	1	0.5	40	MTBE (2.0)	44	32/68
11	1	1	40	MTBE (2.0)	44	15/85
12	1	1	40	MTBE (4.0)	44	14/86
13	1	1	40	Hexane (4.0)	44	12/88

a) The compositions were obtained by GC-FID analysis by comparing peak areas of alcohol **1** vs. acetate **2**.

Table S2. GC-FID retention times of respective alcohols and corresponding acetates.

Entry	Alcohol	Retention time [min]	Acetate	Retention time [min]
1	 (1)	7.849	 (2)	5.996
2	 (3)	10.956	 (4)	11.935
3	 (5)	4.129	 (6)	5.051
4	 (7)	6.950	 (8)	6.026
5	 (9)	14.790	 (10)	15.982
6	 (11)	13.797	 (12)	13.521
7	 (13) E/Z = 66/34	21.781/22.453	 (14) E/Z = 66/34	24.843/26.046
8	 (15)	11.952	 (16)	11.243
9	 (17)	11.358	 (18)	10.253
10	 (19)	5.900	 (R)-20	5.202
11	 (S)-19	5.900	 (S)-20	5.202
12 ^b	 (21)	15.795	 (R)-22	13.797
13	 (23)	14.674	 (24)	13.609
14	 (25)	9.043	 (26)	8.921
15	 (27)	3.556	 (28)	2.729
16	 (29)	7.538	 (30)	7.128
17	 (31)	14.930	 (32)	n.a.

18		(33)	3.445		(34)	3.095
19		(35)	2.300		(36)	2.098
20		(37)	2.139		(38)	n.a.

Table S3. Time-dependent GC-FID ratios of competitive enzymatic acetylation of 1-pentanol (39) *vs.* 3-pentanol (33).

Reaction time [h]	Reaction temperature [°C]	Pentan-1-ol (39) tr = 5.051 min	Pentan-1-yl acetate (40) tr = 4.129 min	Pentan-3-ol (33) tr = 3.445 min	Pentan-3-yl acetate (34) tr = 3.095 min
1	28	26.8	26.8	46.3	0.1
2	40	11.6	43.8	44.2	0.4
5	40	4.0	52.8	41.9	1.3
7	40	3.4	53.8	41.0	1.8
24	40	3.4	53.8	37.9	4.9
48	40	3.4	53.8	33.6	9.2

GC-FID: column DB-Wax (30 m x 0.25 mm x 0.15 µm), injection 0.01 µL, split 50:1, temperature gradient 40 °C (0 min) → 10 °C/min → 200 °C (12 min), carrier gas H₂ (1.2 mL/min). The compositions were obtained by comparing peak areas of both alcohols and acetates.

Table S4. Time-dependent GC-FID ratios of competitive enzymatic acetylation of cyclopentanol (41) *vs.* 3-pentanol (33).

Reaction time [h]	Reaction temperature [°C]	Cyclopentanol (41) tr = 5.667 min	Cyclopentyl acetate (42) tr = 5.036 min	Pentan-3-ol (33) tr = 3.445 min	Pentan-3-yl acetate (34) tr = 3.095 min
1	28	43.6	7.7	48.0	0.7 0.0
2	40	39.3	13.8	45.4	1.5
5	40	30.2	23.7	43.2	2.9
7	40	27.9	26.8	41.9	3.4
24	40	16.5	38.6	38.8	6.1
48	40	11.0	43.9	36.4	8.7

GC-FID: column DB-Wax (30 m x 0.25 mm x 0.15 µm), injection 0.01 µL, split 50:1, temperature gradient 40 °C (0 min) → 10 °C/min → 200 °C (12 min), carrier gas H₂ (1.2 mL/min). The compositions were obtained by comparing peak areas of both alcohols and acetates.

Table S5. Time-dependent GC-FID ratios of competitive enzymatic acetylation of prenol (5) *vs.* divinylcarbinol (43).

Reaction time [h]	Reaction temperature [°C]	Prenol (5) $t_R = 5.911\text{ min}$	Prenyl acetate (6) $t_R = 5.051\text{ min}$	1,4-Pentadien-3-ol (43) $t_R = 4.888\text{ min}$	1,4-Pentadien-3-yl acetate (44) $t_R = 3.871\text{ min}$
1	28	40.1	17.2	42.1	0.6
2	40	28.6	27.5	42.7	1.2
5	40	12.4	43.8	41.0	2.8
7	40	9.0	47.0	40.5	3.5
24	40	5.8	52.2	33.9	8.1
48	40	5.2	52.7	29.3	12.8

GC-FID: column DB-Wax (30 m × 0.25 mm × 0.15 μm), injection 0.01 μL, split 50:1, temperature gradient 40 °C (0 min) → 10 °C/min → 200 °C (12 min), carrier gas H₂ (1.2 mL/min). The compositions were obtained by comparing peak areas of both alcohols and acetates.

Table S6. Time-dependent GC-FID ratios of competitive enzymatic acetylation of 3-pentanol (33) *vs.* divinylcarbinol (43).

Reaction time [h]	Reaction temperature [°C]	Pentan-3-ol (33) $t_R = 3.445\text{ min}$	Pentan-3-yl acetate (34) $t_R = 3.095\text{ min}$	1,4-Pentadien-3-ol (43) $t_R = 4.888\text{ min}$	1,4-Pentadien-3-yl acetate (44) $t_R = 3.871\text{ min}$
2.5	28.5	50.0	2.0	44.0	4.0
3	40	48.9	3.2	41.9	6.0
5.5	40	46.4	4.6	40.7	8.3
23	40	39.9	9.9	33.6	16.6
56	40	30.7	20.0	27.1	22.2

GC-FID: column DB-Wax (30 m × 0.25 mm × 0.15 μm), injection 0.01 μL, split 50:1, temperature gradient 40 °C (0 min) → 10 °C/min → 200 °C (12 min), carrier gas H₂ (1.2 mL/min). The compositions were obtained by comparing peak areas of both alcohols and acetates.

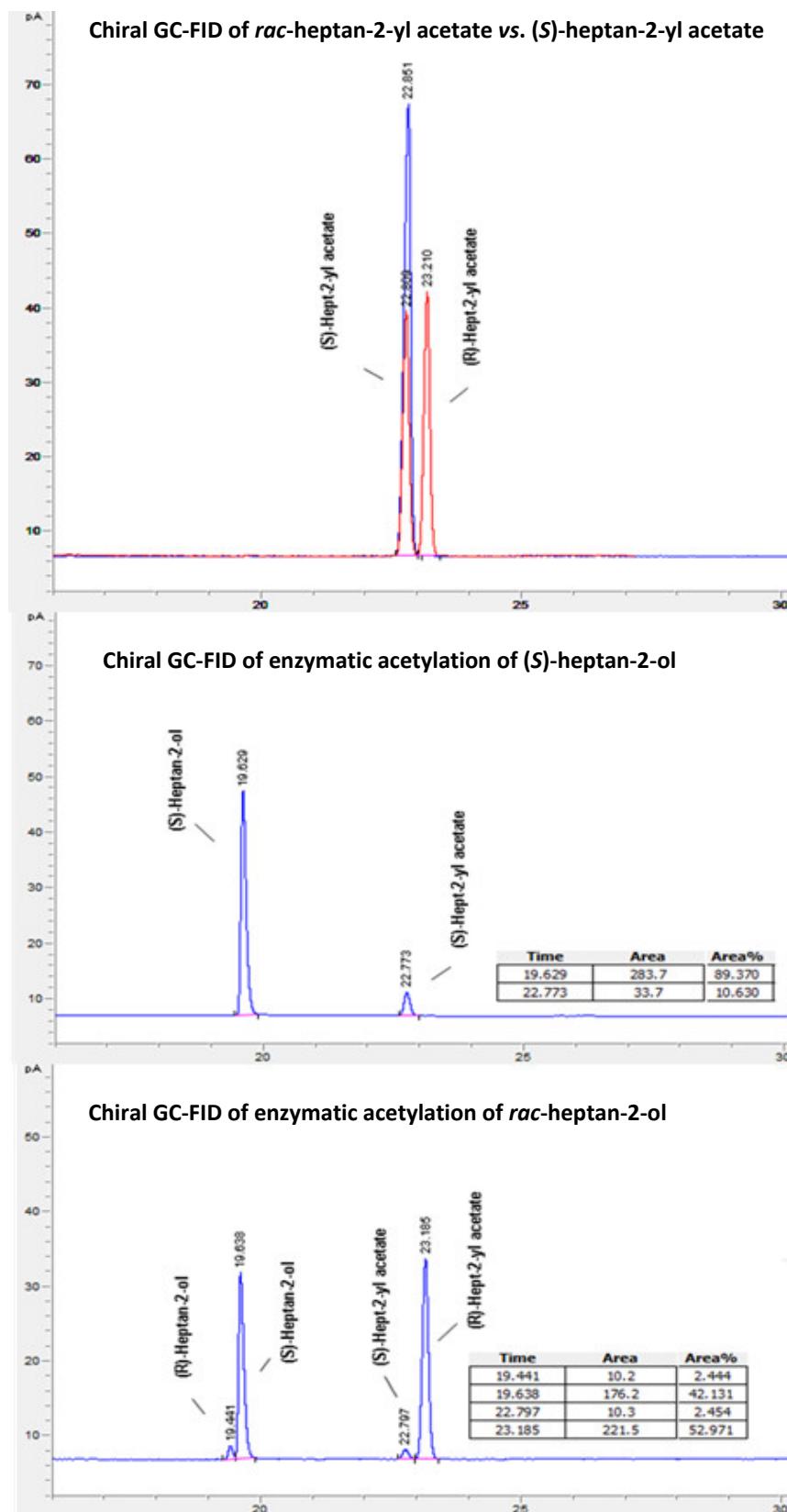


Figure S1. Chiral GC-FID analyses of heptan-2-ol (19) and heptan-2-yl acetate (20).

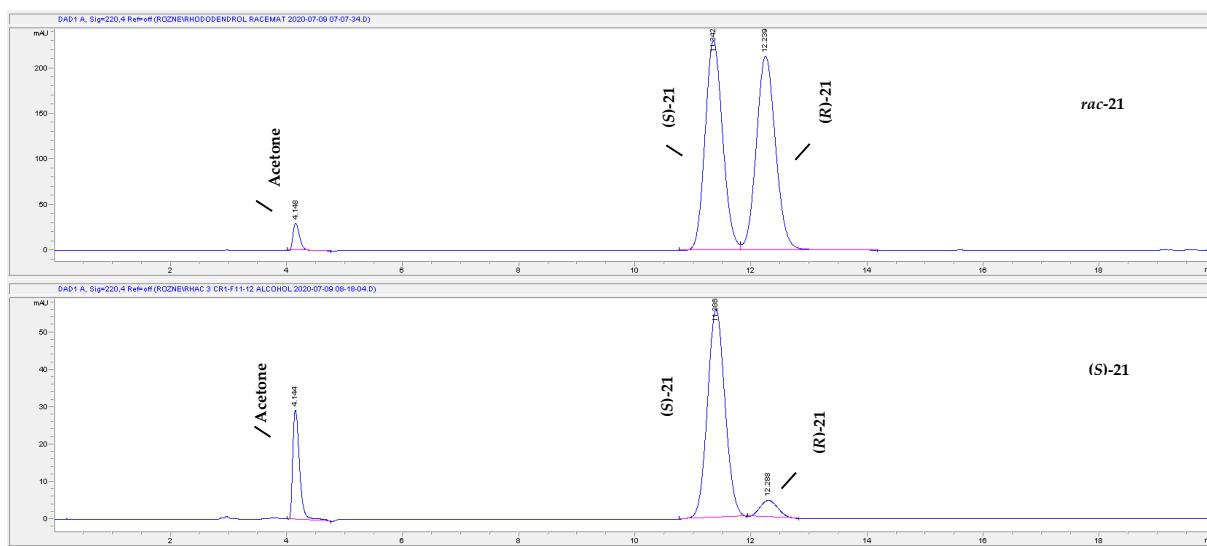
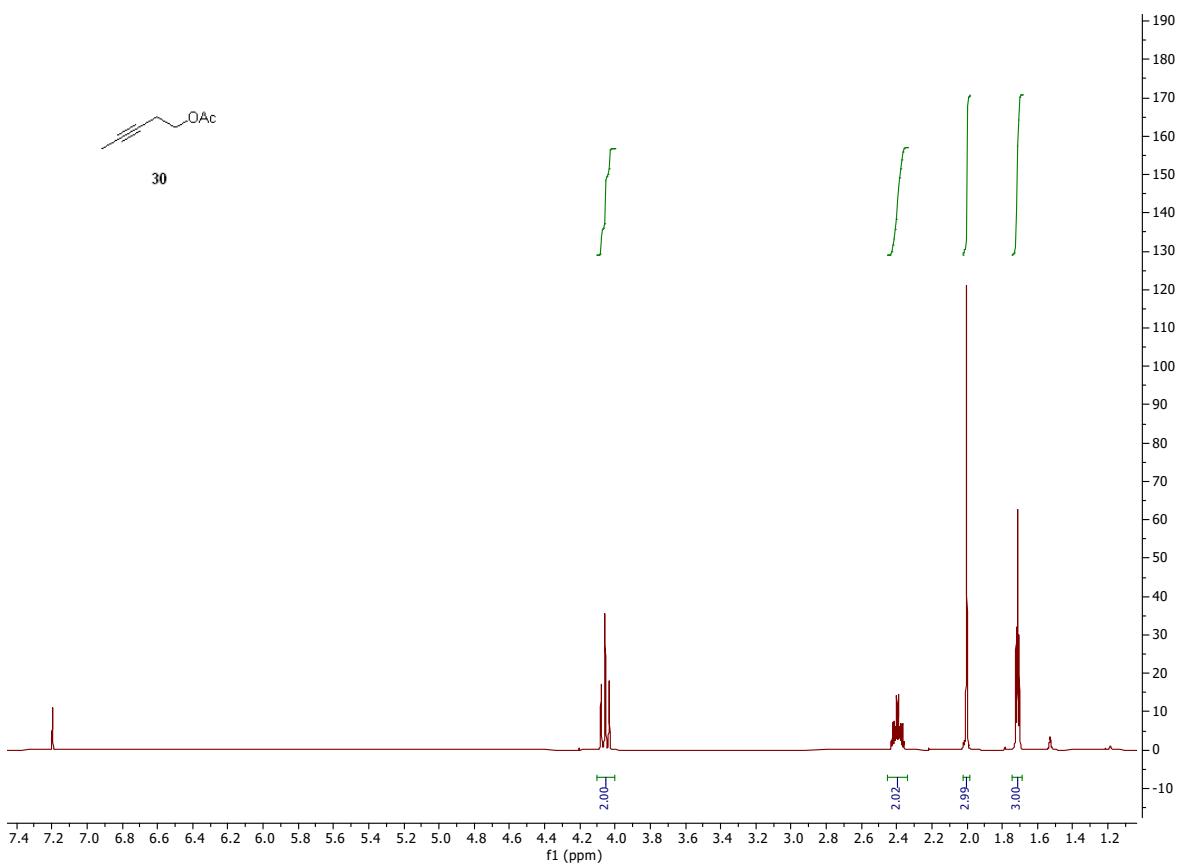
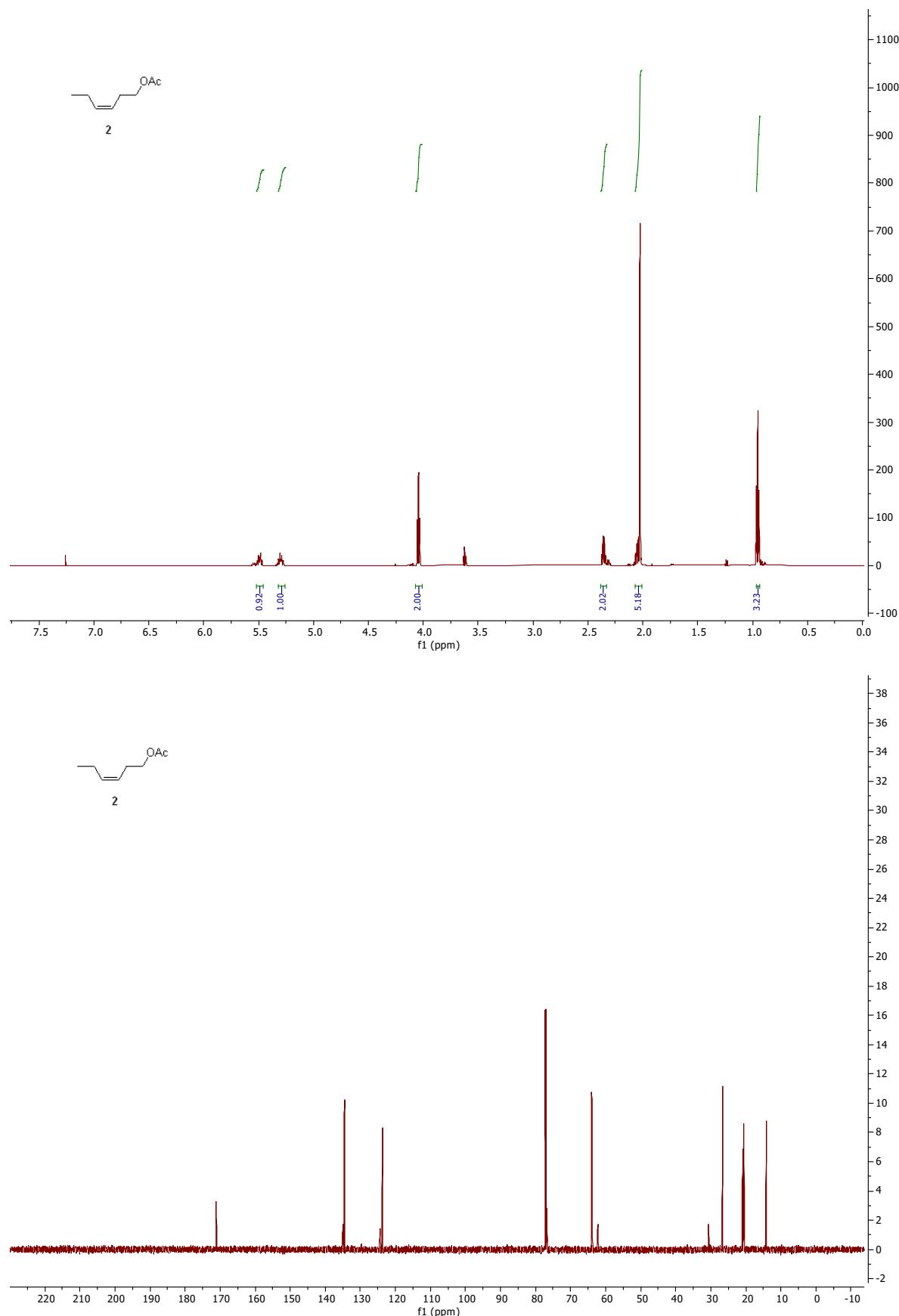


Figure S2. Chiral HPLC analyses of *rac*-rhododendrol *rac*-21 and enantioenriched (*S*)-rhododendrol (*S*-21).





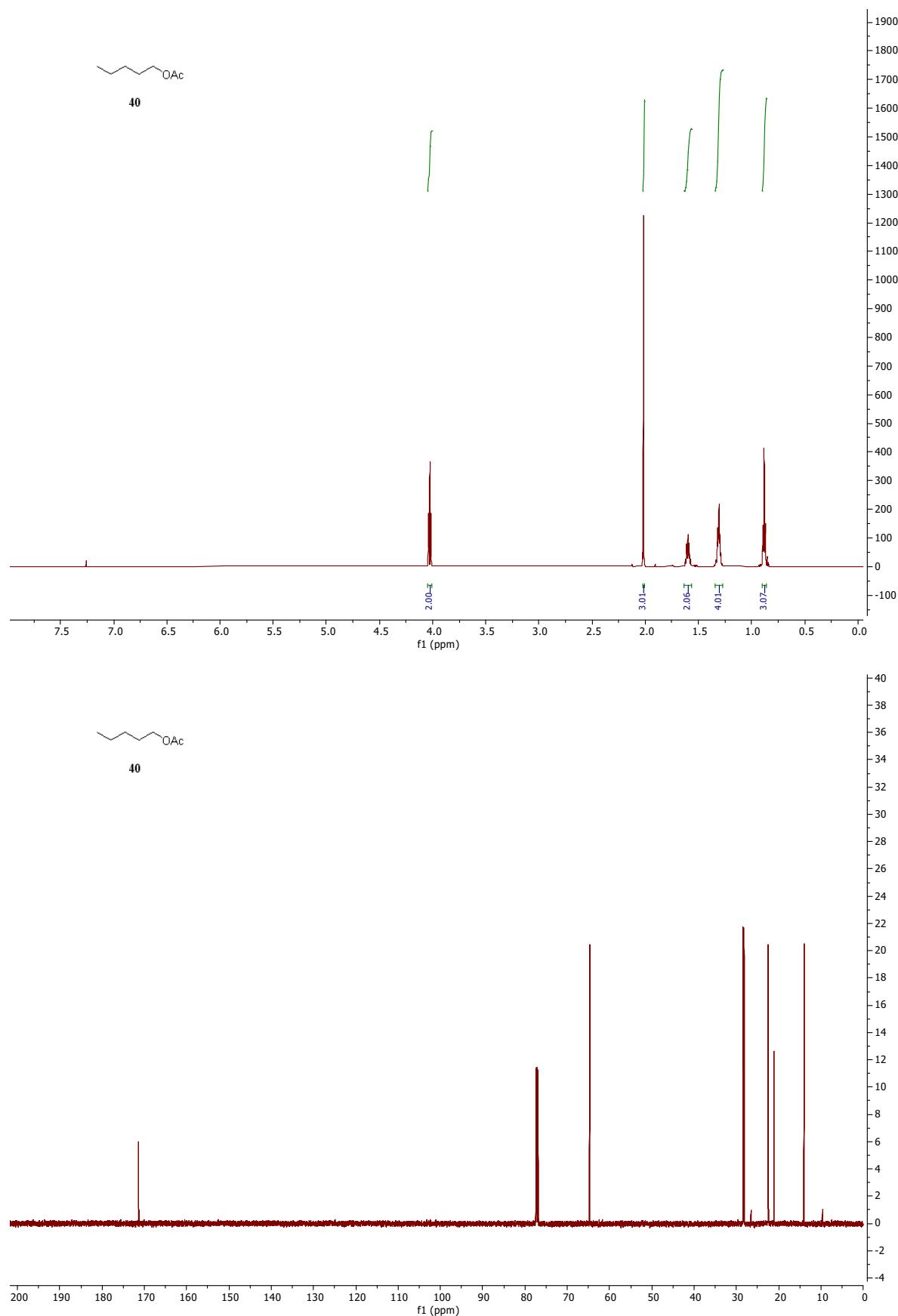


Figure S3. Copies of NMR spectra of isolated compounds.