## Dendrimer-functionalized hybrid materials based on silica as novel carriers of bioactive acids

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## **Supplementary Information**

Section A: The ESI-MS spectra of the synthesized dendrimers and their complexes with bioactive compounds studied



Figure S1. The ESI-MS positive spectrum of ester intermediate.





**Figure S2.** The ESI-MS positive spectra of the synthesized PAMAM dendrimers: (**a**) EDA, (**b**) TETA, (**c**) TREN and (**d**) TRI-OXA.







**Figure S3.** The ESI-MS spectra (positive – top; negative – bottom) of exemplary TREN poly(amidoamine) dendrimer complexes with the studied biomolecules: (**a**) salicylic acid, (**b**) nicotinic acid, (**c**) folic acid.

Section B: The supplement of the conducted adsorption experiments



**Figure S4.** The Langmuir isotherm model fitted to the experimental data of the adsorption processes. For some points SDs are smaller than the plotted symbols.





log c<sub>eq</sub> [-]



**Figure S5.** The Freundlich isotherm model fitted to the experimental data of the adsorption processes. For some points SDs are smaller than the plotted symbols.





ln c<sub>eq</sub> [-]



**Figure S6.** The Temkin isotherm model fitted to the experimental data of the adsorption processes. For some points SDs are smaller than the plotted symbols.



**Figure S7.** The Dubinin-Radushkevich isotherm model fitted to the experimental data of the adsorption processes. For some points SDs are smaller than the plotted symbols.

	Adsorbent	Temkin Isotherm			Dubinin-Raduschkevich Isotherm		
Biomolecule		B [J mol <sup>-1</sup> ]	R <sup>2</sup>	$\chi^2$	E [kJ mol <sup>-1</sup> ]	<b>R</b> <sup>2</sup>	$\chi^2$
Folic Acid	SiO2-epoxy	1.54	0.8608	0.010	$0.209 \pm 0.075$	0.6052	43.592
	SiO <sub>2</sub> -EDA	16.75	0.7564	16.250	$0.260 \pm 0.106$	0.5474	9.514
	SiO <sub>2</sub> -TETA	19.83	0.8038	7.689	$0.296 \pm 0.115$	0.5717	9.091
	SiO <sub>2</sub> -TREN	17.07	0.7635	12.862	$0.273 \pm 0.110$	0.5508	9.160
	SiO2-TRI-OXA	11.58	0.7532	9.980	$0.242\pm0.099$	0.5440	10.392
Salicylic Acid	SiO2-epoxy	0.25	0.9490	0.512	$0.406 \pm 0.108$	0.7376	6.140
	SiO <sub>2</sub> -EDA	9.47	$0.7628$ 27.602 $0.526 \pm 0.185$		$0.526 \pm 0.185$	0.6173	12.798
	SiO <sub>2</sub> -TETA	3.31	0.8354	8.626	$0.462\pm0.156$	0.6366	14.136
	SiO <sub>2</sub> -TREN	5.73	0.9098	12.498	$2.448 \pm 0.927$	0.5824	6.369
	SiO2-TRI-OXA	2.82	0.8713	19.115	$0.496 \pm 0.166$	0.6426	12.631
Nicotinic Acid	SiO2-epoxy	0.22	0.9299	0.636	$0.020\pm0.005$	0.7540	61.718
	SiO <sub>2</sub> -EDA	2.99	0.8557	6.049	$0.871 \pm 0.373$	0.5207	8.278
	SiO <sub>2</sub> -TETA	0.84	0.9587	0.473	$0.813 \pm 0.257$	0.6676	6.779
	SiO <sub>2</sub> -TREN	2.90	0.9294	6.128	$0.997 \pm 0.314$	0.6593	7.170
	SiO2-TRI-OXA	1.18	0.9221	2.068	$0.620\pm0.219$	0.6359	11.258

Table S1. Fitting of the experimental data to the Tekmin and the Dubinin-Radushkevich isothermal models.



**Figure S8.** The thermodynamic plots of the biomolecules adsorption processes corresponding the van't Hoff equation. For some points SDs are smaller than the plotted symbols.

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Section C	: The supp	lement of the	conducted	driig-rele	ease experiments
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		Zero-Order	Model	Hixson-Crowe	Hixson-Crowell Model		
Biomolecule	Adsorbent	$\mathbf{k}_1$	R <sup>2</sup>	<b>k</b> н-с	R <sup>2</sup>		
		[mg h-1]	(χ <sup>2</sup> )	[mg <sup>1/3</sup> h <sup>-1</sup> ]	(χ <sup>2</sup> )		
	SiO2-EDA	$0.006 \pm 0.002$	0.5433	$0.008 \pm 0.004$	0.4265		
			(0.085)		(0.078)		
	SO TETA	$0.005 \pm 0.002$	0.5477	$0.007 \pm 0.003$	0.4596		
Folic Acid	5102-1E1A		(0.058)	$0.007 \pm 0.003$	(0.073)		
Folic Acia	SiO2-TREN	$0.005 \pm 0.002$	0.5698	$0.007 \pm 0.004$	0.4599		
			(0.061)	$0.007 \pm 0.004$	(0.063)		
	SiO2-TRI-OXA	$0.004 \pm 0.001$	0.5860	$0.008 \pm 0.004$	0.4649		
			(0.059)		(0.072)		
	SiO <sub>2</sub> -EDA	$0.007\pm0.004$	0.3547	$0.003 \pm 0.002$	0.3371		
			(0.042)		(0.111)		
	SiO <sub>2</sub> -TETA	$0.006 \pm 0.002$	0.4885	$0.006 \pm 0.003$	0.4280		
Salicylic Acid			(0.066)		(0.039)		
Sancyne Acia	SiO <sub>2</sub> -TREN	$0.006 \pm 0.004$	0.3611	$0.003 \pm 0.002$	0.3416		
			(0.032)	0.000 ± 0.002	(0.071)		
	SiO2-TRLOXA	$0.005 \pm 0.003$	0.3809	$0.005 \pm 0.003$	0.3374		
	5102-TRI-OAA		(0.069)		(0.056)		
	SiO2-ED A	$0.004 \pm 0.001$	0.6156	$0.004 \pm 0.001$	0.5821		
	0102 1011	0.001 ± 0.001	(0.013)	0.001 ± 0.001	(0.043)		
	SiO2-TETA	$0.004 \pm 0.001$	0.5863	$0.006 \pm 0.002$	0.5729		
Nicotinic Acid	5102 1111		(0.045)	0.000 ± 0.002	(0.032)		
	SiO2-TRFN	$0.004 \pm 0.002$	0.5655	$0.001 \pm 0.002$	0.0592		
	0102-11\LIN		(0.011)	$0.001 \pm 0.002$	(0.156)		
	SiO2-TRI-OXA	$0.008 \pm 0.004$	0.4099	$0.008 \pm 0.005$	0.3438		
	5102-11XI-0AA		(0.159)	0.000 ± 0.000	(0.235)		

**Table S2.** The drug release parameters calculated for the fitting of experimental data tothe zero-order and the Hixson-Crowell release models.