



Supplementary information

Structural and Photophysical Properties of 2,1,3-Benzothiadiazole-Based Phosph(III)azane and Its Complexes

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Figure S1. Experimental (black) and simulated (red and blue) powder patterns of the compounds 2a·3C7H8 and 2b·2.5C7H8.



Figure S2. Representation of selected torsion angles in H₂L or L²⁻ fragments. Hydrogens are not shown.



Figure S3. Molecular structure of **1** showing attraction interaction Zn–N⁴ marked by dashed red line.



Figure S4. The d_{norm} Hirshfeld surface of the complexes in crystal structures **2a**·3C7H8 (left) and **2b**·2.5C7H8 (right). Area with intermolecular contacts closer than the sum of atoms van der Waals radii are red, longer contacts are blue, and the contacts around the sum of van der Waals radii are white.



Figure S5. Representation of a Cu–H₂L model with the geometry derived from XRD data for free H₂L (left; Cu atom was placed geometrically) and the corresponding fragment from XRD data for **3** (right). Red and Green dashed lines indicate Cu…H repulsion interactions.



Figure 6. Overlaid TD-DFT calculations results and experimental UV-Vis spectrum of H2L.

n	λ, nm	f	electronic states	contribution
1	452.1	0.0872	H→L	0.9438
2	436.8	0.0431	H→L+1	0.9520
3	402.3	0.0196	H-1→L	0.9563
4	398.5	0.0128	H-1→L+1	0.9460
5	322.3	0.0030	H-2→L	0.9845
6	317.6	0.0045	H-2→L+1	0.9818
7	305.6	0.0135	H→L+2	0.9825

Table S1. Calculated properties of the first singlet excited states $S_0 \rightarrow S_n$ of H₂L: transition wavelength (λ), oscillator strength (f). H is for HOMO, L is for LUMO abbreviation.

Identification code	H_2L	1	2a-3C7H8	2a.2.5C7H8	3·THF	4
Empirical formula	$C_{18}H_{13}N_6PS_2$	C50H38N12P2S4Zn2	$C_{57}H_{50}Cl_4N_{12}P_2S_4Zn_2$	C53.5H46Cl4N12P2S4Zn2	C22H21ClCuN6OPS2	C30H19N9P2S3
Formula weight	408.43	1127.84	1365.81	1319.74	579.53	663.66
Space group	$Pna2_1$	C2/c	C2/m	$P4_{2}2_{1}2$	C2/c	<i>P</i> –1
a/Å	17.8032(12)	19.7209(12)	14.349(3)	15.9225(8)	23.7977(11)	10.119(4)
b/Å	22.9545(17)	19.3027(11)	24.505(4)	15.9225(8)	15.1196(9)	11.337(4)
c/Å	4.3288(3)	13.0051(8)	10.6832(18)	23.1212(14)	16.4906(9)	13.556(5)
$\alpha /^{\circ}$	90	90	90	90	90	81.340(12)
β/°	90	103.554(2)	127.670(5)	90	123.116(2)	73.235(12)
γ/°	90	90	90	90	90	72.680(13)
Volume/Å ³	1769.0(2)	4812.7(5)	2973.4(9)	5861.8(7)	4969.7(5)	1418.0(10)
Z	4	4	2	4	8	2
$Qcalcg/cm^3$	1.534	1.557	1.526	1.495	1.549	1.554
µ/mm ⁻¹	0.409	1.288	1.231	1.246	1.247	0.416
F(000)	840.0	2304.0	1396.0	2692.0	2368.0	680.0
Crystal size/mm ³	$0.18 \times 0.14 \times 0.06$	$0.14 \times 0.14 \times 0.08$	$0.15 \times 0.12 \times 0.1$	$0.12\times0.09\times0.09$	$0.22\times0.14\times0.1$	$0.08 \times 0.06 \times 0.03$
2 Θ range for data collection/°	2.894 to 46.488	4.018 to 48.924	3.952 to 48.808	3.106 to 49.068	3.68 to 55.95	3.146 to 49.1
Index ranges	$-16 \le h \le 19, -21 \le k \le$	$-22 \le h \le 19, -22 \le k \le$	$-16 \le h \le 16, -28 \le k \le$	$-18 \le h \le 18, -18 \le k \le$	$-31 \le h \le 30, -19 \le k \le$	$-11 \le h \le 11, -13 \le k \le$
	25,-4 ≤ 1 ≤ 3	21,-14 ≤ 1 ≤ 15	25,-11 ≤ 1 ≤ 12	18,-26 ≤ 1 ≤ 22	19,-21 ≤ 1 ≤ 21	13, -14 ≤ 1 ≤ 15
Reflections collected	4262	11184	7651	63517	17435	15306
Independent reflections	2043	3952	2515	4898	5963	4672
	[Kint = 0.0755, $P_{\text{c}} = 0.08061$	[Kint = 0.0310, $\text{R}_{\pm} = 0.04221$	$[K_{int} = 0.0549,$	[Kint = 0.0723, $P_{\text{c}} = 0.02571$	[Kint = 0.0285,	$[K_{int} = 0.1012,$
	Rsigma – 0.0000]	Ksigma – 0.0455]	$R_{sigma} = 0.0742$	$R_{sigma} = 0.0337$	Rsigma – 0.0502]	$R_{sigma} = 0.1075$
Restraints/parameters	3/251	0/317	43/168	254/396	2/314	96/397
Goodness-of-fit on F ²	1.029	1.028	1.007	1.029	1.038	1.034
Final R indexes $[I \ge 2\sigma (I)]$	$R_1 = 0.0446, wR_2 =$	$R_1 = 0.0338$, $wR_2 =$	$R_1 = 0.0495$, $wR_2 =$	$R_1 = 0.0338$, $wR_2 =$	$R_1 = 0.0334$, $wR_2 =$	$R_1 = 0.0887, wR_2 =$
	0.0956	0.0794	0.1182	0.0770	0.0777	0.2166
Final R indexes [all data]	$R_1 = 0.0649$, $wR_2 =$	$R_1 = 0.0485$, $wR_2 =$	$R_1 = 0.0830$, $wR_2 =$	$R_1 = 0.0465$, $wR_2 =$	$R_1 = 0.0449$, $wR_2 =$	$R_1 = 0.1238$, $wR_2 =$
Final K indexes [an data]	0.1042	0.0860	0.1320	0.0825	0.0829	0.2399

Table S2. Crystal data and structure refinement for H₂L, 1–4.

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Largest diff. peak/hole / e A-3	0.36/-0.31	0.58/-0.41	0.81/-0.89	0.60/-0.48	0.43/-0.36	0.87/-0.79
Flack parameter	0.39(18)			0.009(7)		

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