Aromatic S_N^F-Approach to Fluorinated Phenyl *tert*-Butyl Nitroxides

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Content:

NMR, IR and electronic spectroscopy data	S 2
ESR spectroscopy data	S19
DSC and TG data for complexes [Cu(hfac) ₂ ($3a$) ₂], [Cu(hfac) ₂ ($3b$) ₂]	S20
CVA data for nitroxides 3a,b and complexes [Cu(hfac) ₂ (3a) ₂], [Cu(hfac) ₂ (3b) ₂]	S21
Crystallographic data for amine 2b and complex [Cu(hfac) ₂ (2b) ₂]	S22







Figure S2. ¹³C NMR spectrum of 2a (125.75 MHz, CDCl₃).

S3



Figure S3. ¹⁹F NMR spectrum of 2a (282.37 MHz, CDCl₃).



Figure S4. IR spectrum of 2a (KBr).



Figure S5. ¹H NMR spectrum of **2b** (400.13 MHz, CDCl₃).

S6



Figure S6. ¹³C NMR spectrum of 2b (100.62 MHz, CDCl₃).





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Figure S8. IR spectrum of 2b (KBr).



Figure S9. IR spectrum of 3a (KBr).



Figure S10. UV spectrum of 3a (EtOH).



Figure S11. IR spectrum of 3b (neat).



Figure S12. UV spectrum of 3b (EtOH).



Figure S13. IR spectrum of [Cu(hfac)₂(3a)₂] (KBr).



Figure S14. IR spectrum of [Cu(hfac)₂(3a)₂] (KBr) after sublimation.



Figure S15. IR spectrum of [Cu(hfac)₂(3b)₂] (KBr).



Figure S16. IR spectrum of [Cu(hfac)₂(3b)₂] (KBr) after sublimation.



Figure S17. IR spectrum of [Cu(hfac)₂(2b)₂] (KBr).

ESR spectroscopy data



Figure S18. Experimental (black curve) and simulated (red curve) ESR spectrum for 3b.



DSC and TG data for complexes [Cu(hfac)₂(3a)₂], [Cu(hfac)₂(3b)₂]

Figure S19. DSC and TG curves for [Cu(hfac)₂(3a)₂].



Figure S20. DSC and TG curves for [Cu(hfac)₂(3b)₂].



CVA data for nitroxides 3a,b and complexes [Cu(hfac)₂(3a)₂], [Cu(hfac)₂(3b)₂]

Figure S21. Cyclic voltammograms of nitroxides **3a,b** and complexes [Cu(hfac)₂(**3a**)₂], [Cu(hfac)₂(**3b**)₂] in CH₂Cl₂ solution.

Crystallographic data for amine 2b and complex [Cu(hfac)₂(2b)₂]

Crystallographic data for [Cu(hfac)₂(**2b**)₂]: C₃₂H₂₂CuF₂₀N₄O₄, *M* 970.08, triclinic, P-1, *a* 9.7317(4), *b* 10.5505(4), *c* 10.6014(4) Å; *a* 95.728(2), *β* 101.163(2), *γ* 111.330(2)°; *V* 977.34(7) Å³, *Z* 1, *D*_{calcd} 1.648 g·cm⁻³, μ (Mo-*K* α) 0.696 mm⁻¹, F(000) 483, (θ 2.11–30.09°), completeness (θ 50°) 99.9%, *T* = 296(2) K, green, (0.73 × 0.60 × 0.10) mm³, transmission 0.5079–0.6042, 22970 measured reflections in index range -13 ≤ h ≤13, -14 ≤ k ≤14, -14 ≤ 1 ≤14, 5708 independent (*R*_{int} 0.041), 277 parameters, *R*₁ 0.0586 (for 4614 observed *I*> 2 σ (*I*)), *wR*₂ 0.1978 (all data), GOOF 1.09, largest diff. peak and hole 0.850 and -0.502 e·A⁻³

Crystallographic data for **2b**: C₁₁H₁₀N₂F₄, *M* 246.21, monoclinic C2/c, *a* 22.849(1), *b* 7.8244(4), *c* 14.4219(6) Å; β 120.058(1)°; *V* 2231.6(2) Å³, *Z* 8, *D*_{calcd} 1.466 g·cm⁻³, μ (Mo-*K* α) 0.135 mm⁻¹, F(000) 1008, (θ 2.8–30.2°, completeness (θ 50°) 98.2%), *T* = 200(2) K, red, (1.0 × 0.7 × 0.2) mm³, transmission 0.8129–0.8622, 11034 measured reflections in index range $-30 \le h \le 29$, $-11 \le k \le 10$, $-16 \le l \le 19$, 2780 independent (*R*_{int} 0.0308), 160 parameters, *R*₁ 0.0465 (for 2415 observed *I*> 2 σ (*I*)), *wR*₂ 0.1219 (all data), GOOF 1.05, largest diff. peak and hole 0.336 and $-0.210 e \cdot A^{-3}$.



Figure S22. The molecular structure and atom-labelling (left), the fragment of crystal packing of compound **2b** (right) (displacement ellipsoids are drawn at the 50% probability level).

Analysis of the crystal packing of amine **2b** revealed molecular chains along the axis *c* (Figure S21) formed by $C \equiv N...\pi$ interactions with $N...C_g$ and D_{pln} distances equaling to 3.508(2) and 3.411 Å respectively, and also $\pi^F...\pi^F$ interactions with $C_g...C_g$ and D_{pln} distances being equal to 3.5321(9) and 3.3863(7) Å accordingly. The chains are combined into layers parallel with plane (*b*, *c*) due to weak hydrogen bonds N1-H1N...F13 with H...F, N...F and N–H...F parameters equaling to 2.50(2), 3.354(2) Å and 163(2)° correspondingly.