

Supporting Information

Towards Covalent Photosensitizer-Polyoxometalate Dyads-Bipyridyl-Functionalized Polyoxometalates and Their Transition Metal Complexes

Andreas Winter ^{1,2}, Patrick Endres ^{1,2}, Erik Schröter ^{1,2}, Michael Jäger ^{1,2}, Helmar Görls ³, Christof Neumann ^{2,4}, Andrey Turchanin ^{2,4} and Ulrich S. Schubert ^{*,1,2}

¹ Laboratory of Organic and Macromolecular Chemistry (IOMC), Friedrich Schiller University Jena, Humboldtstr. 10, 07743 Jena, Germany; andreas.winter@uni-jena.de (A.W.); patrick.endres@uni-jena.de (P.E.); erik.schroeter@uni-jena.de (E.S.); michael.jaeger.iomc@uni-jena.de (M.J.)

² Center for Energy and Environmental Chemistry (CEEC) Jena, Friedrich Schiller University Jena, Philosophenweg 7a, 07743 Jena, Germany; christof.neumann@uni-jena.de (C.N.); andrey.turchanin@uni-jena.de (A.T.)

³ Institute for Inorganic and Analytical Chemistry (IAAC), Friedrich Schiller University Jena, Humboldtstr. 8, 07743 Jena, Germany; helmar.goerls@uni-jena.de

⁴ Institute of Physical Chemistry (ICP), Friedrich Schiller University Jena, Lessingstr. 10, 07743 Jena, Germany

* Correspondence: ulrich.schubert@uni-jena.de; Tel.: +49-3641-948201

Table of content:

• NMR spectroscopy	page 2
• Mass spectrometry	page 10
• X-Ray photoelectron spectroscopy	page 15
• Cyclic and square-wave voltammetry	page 19
• UV/vis absorption and emission spectroscopy	page 22

1. NMR spectroscopy

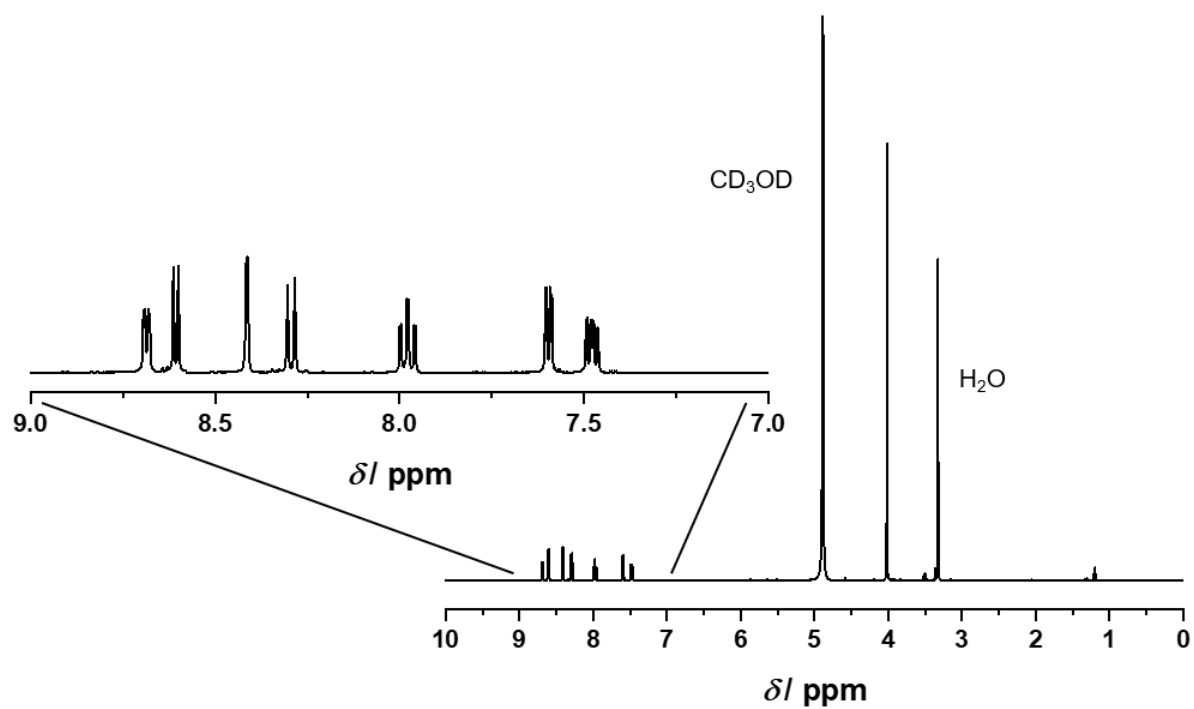


Figure SI- 1. ^1H NMR spectrum of **2** (400 MHz, CD_3OD).

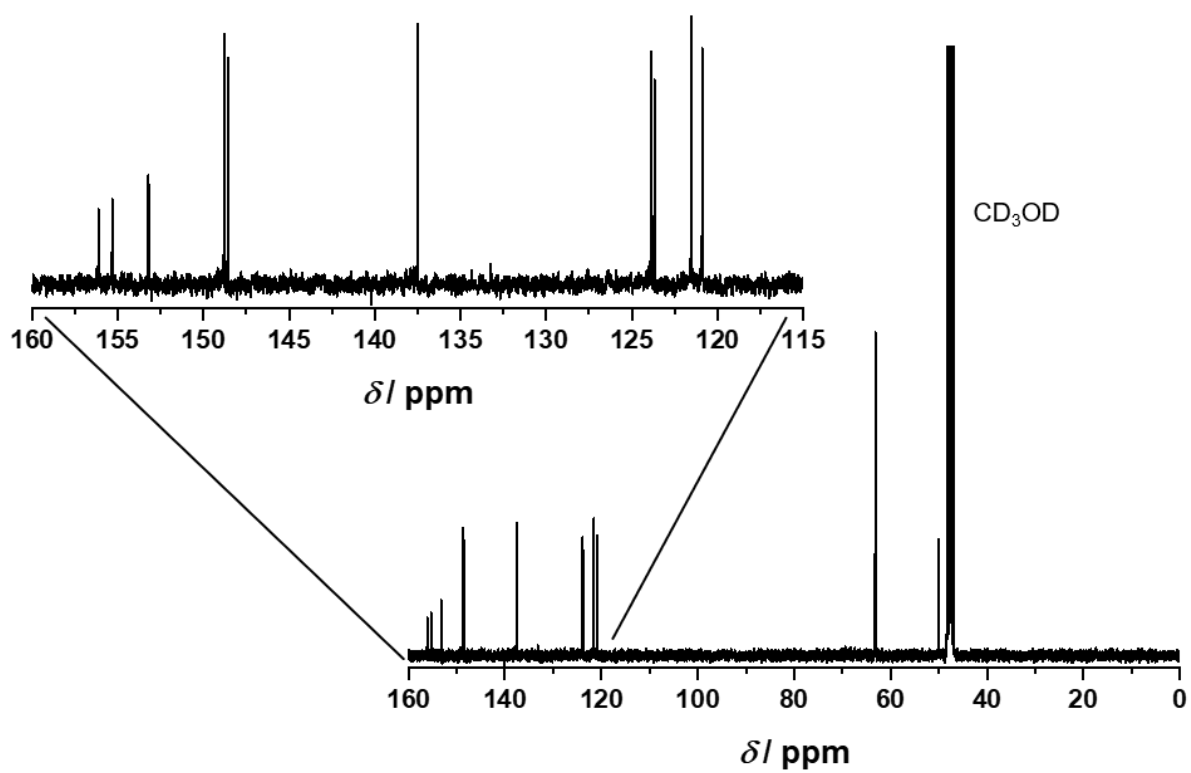


Figure SI- 2. ^{13}C NMR spectrum of **2** (101 MHz, CD_3OD).

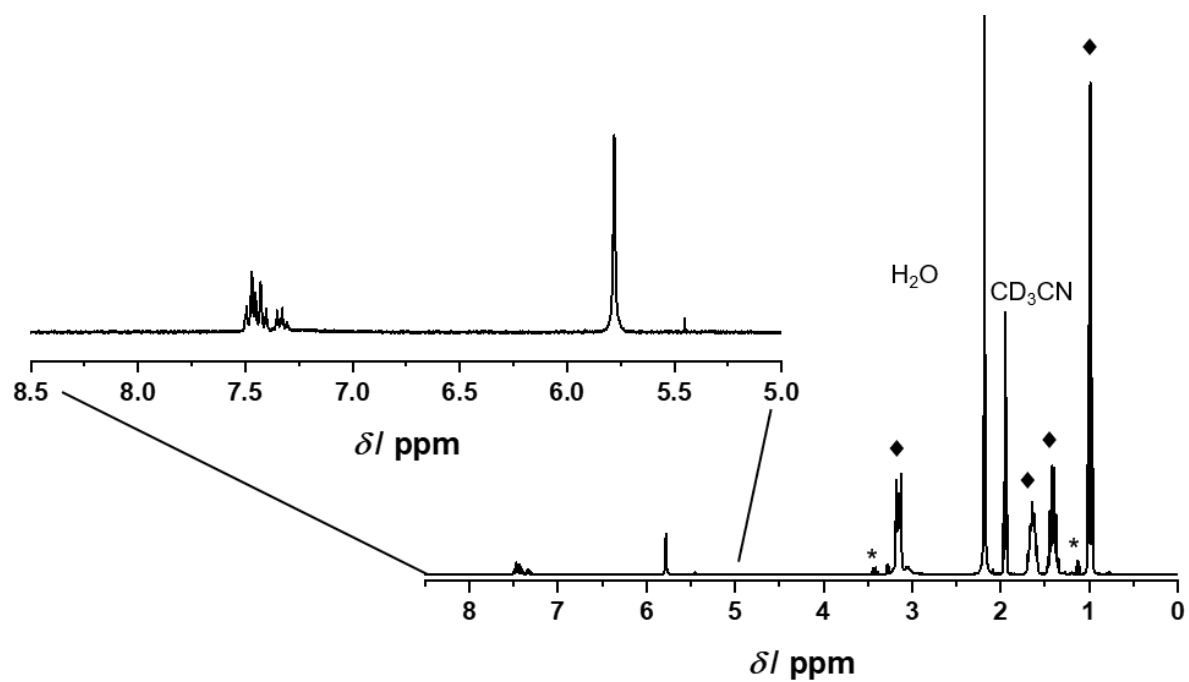


Figure SI- 3. ^1H NMR spectrum of **3** (400 MHz, CD_3CN). Residual diethyl ether (*) and TBA^+ signals (♦) are marked.

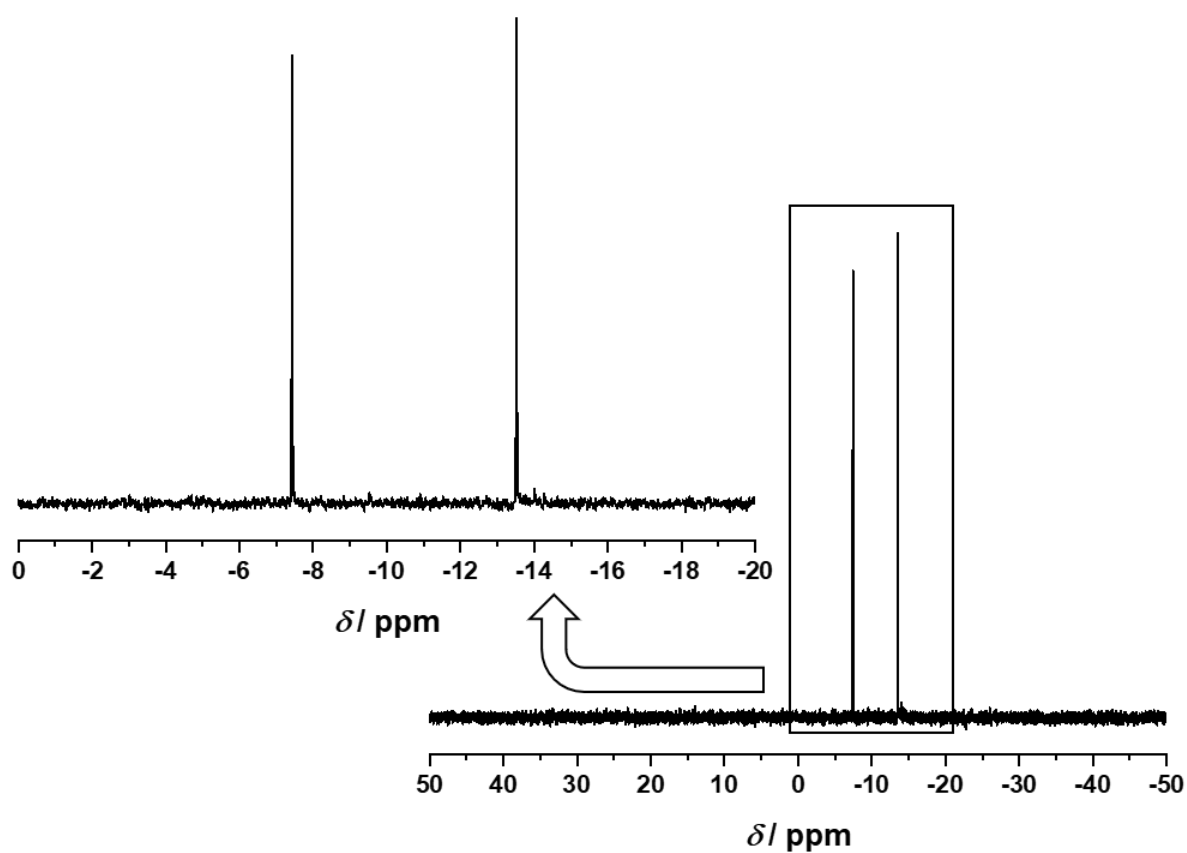


Figure SI- 4. ^{31}P NMR spectrum of **3** (162 MHz, CD_3CN).

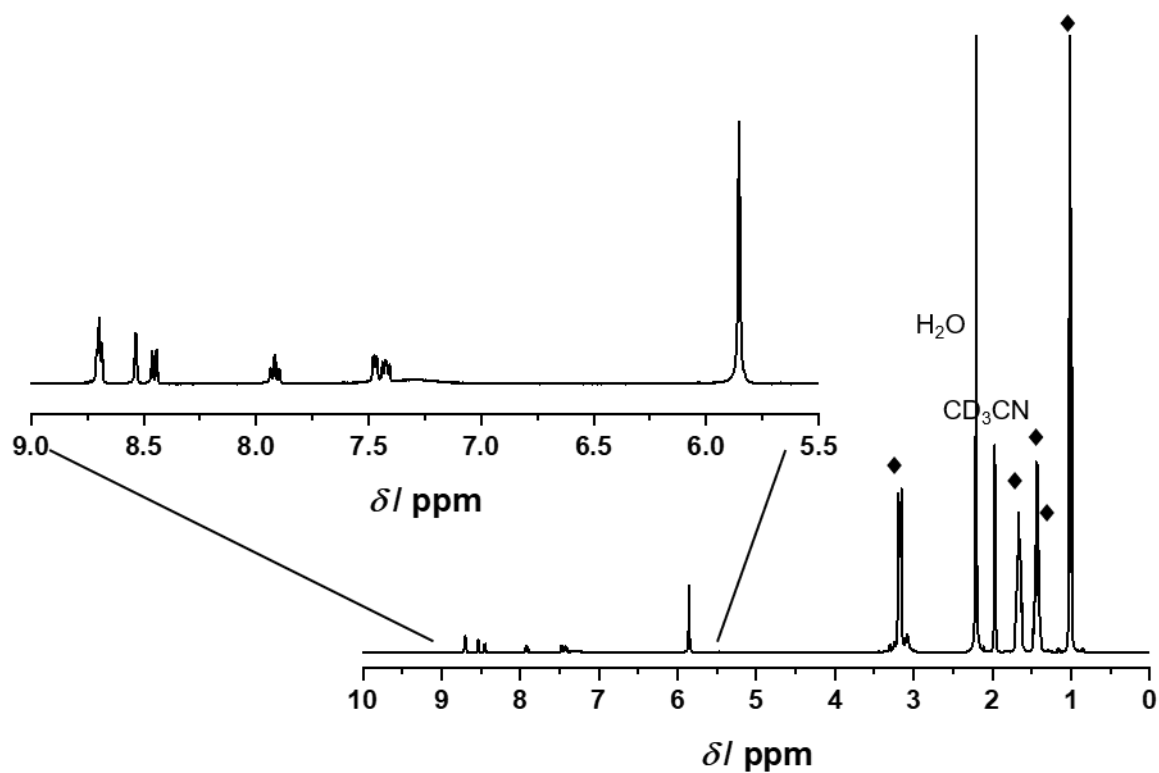


Figure SI- 5. ^1H NMR spectrum of **4** (400 MHz, CD_3CN). The TBA^+ signals (♦) are marked.

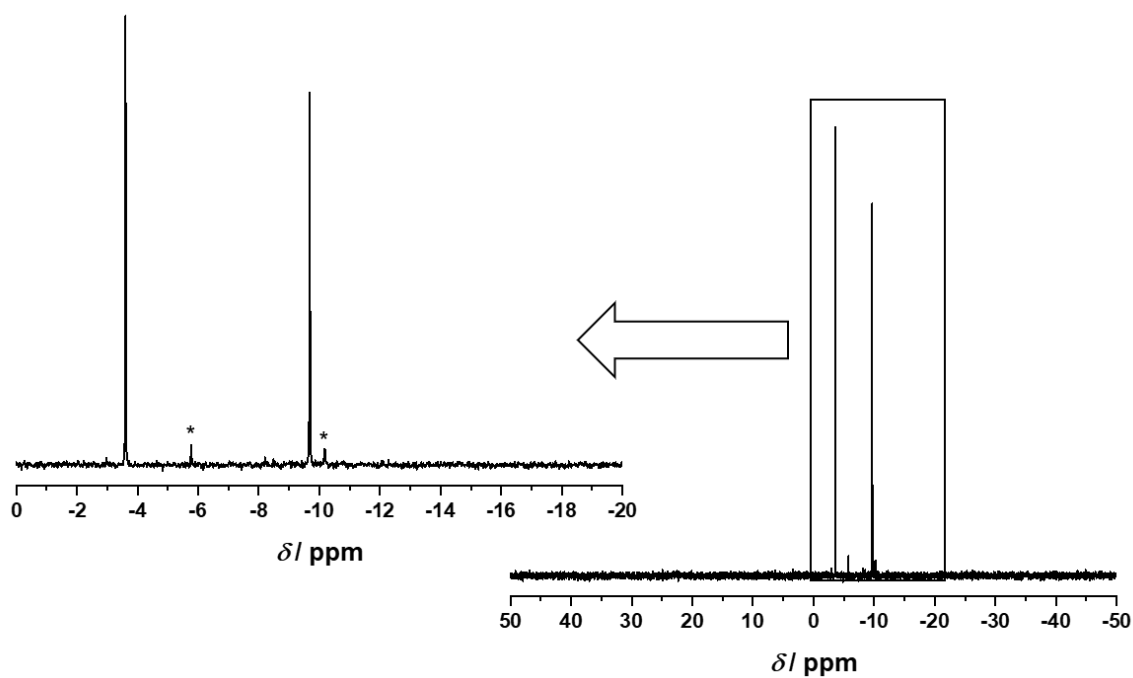


Figure SI- 6. ^{31}P NMR spectrum of **4** (162 MHz, CD_3CN). The signals arising from an unidentified cluster impurity are marked (*). Note, that this impurity could be removed after the subsequent complexation step (see Figures SI-8 and SI-9).

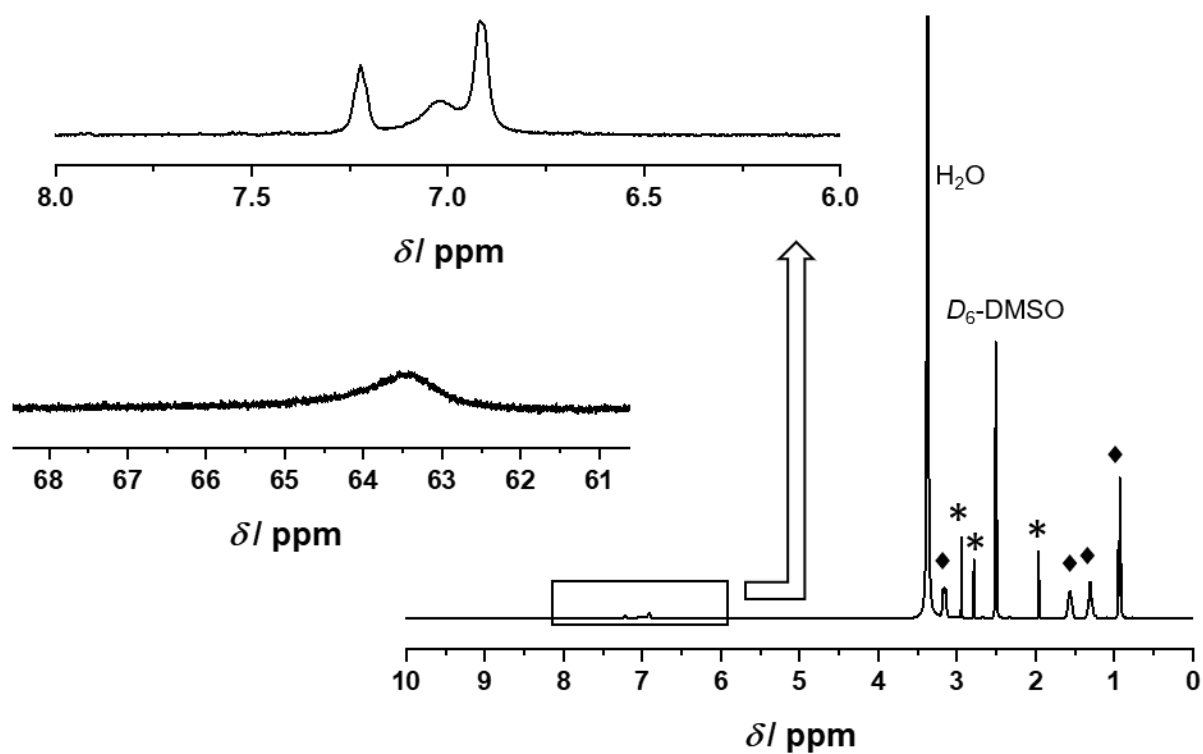
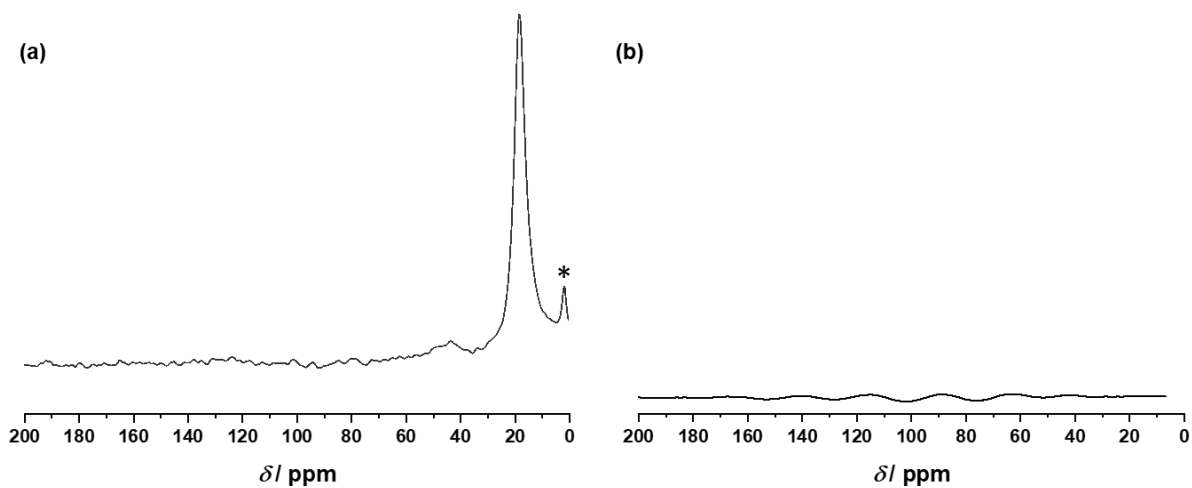


Figure SI- 7. ^1H NMR spectrum of **5** (400 MHz, D_6 -DMSO). Residual DMAC (*) and TBA^+ signals (•) are marked.



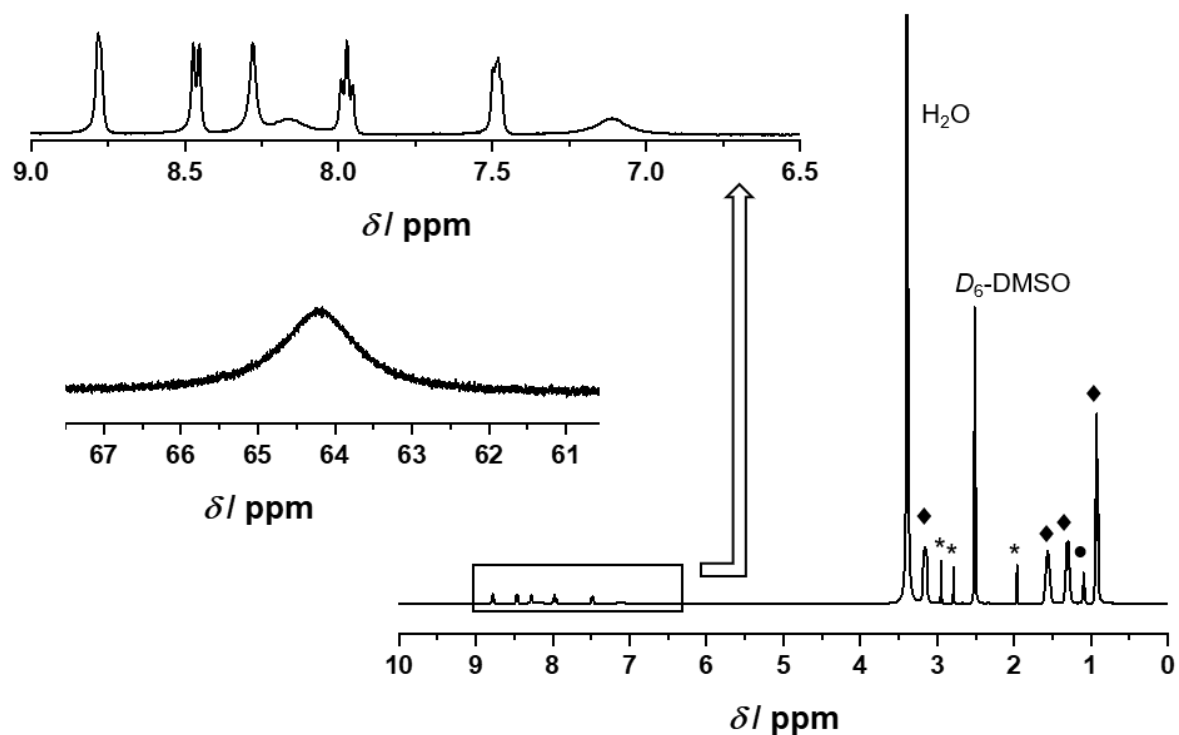


Figure SI- 9. ^1H NMR spectrum of **6** (400 MHz, D_6 -DMSO). Residual DMAC (*), Et₂O (•) and TBA⁺ signals (♦) are marked.

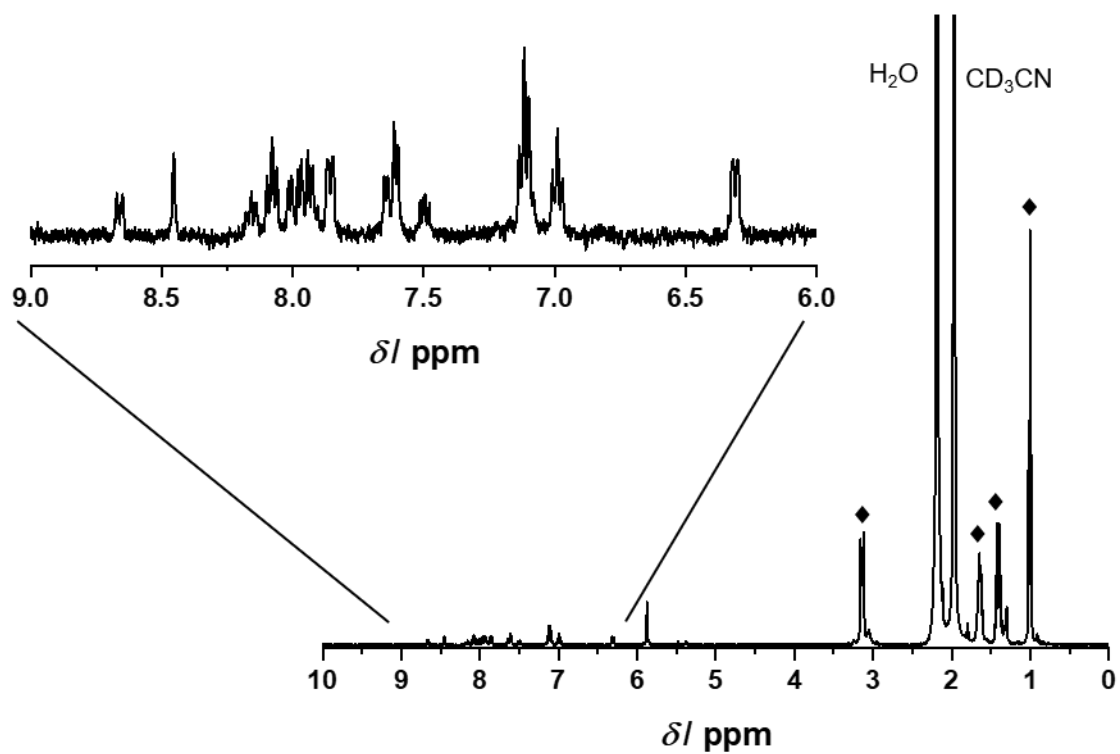


Figure SI- 10. ^1H NMR of **8** (400 MHz, CD_3CN). The TBA⁺ signals (♦) are marked.

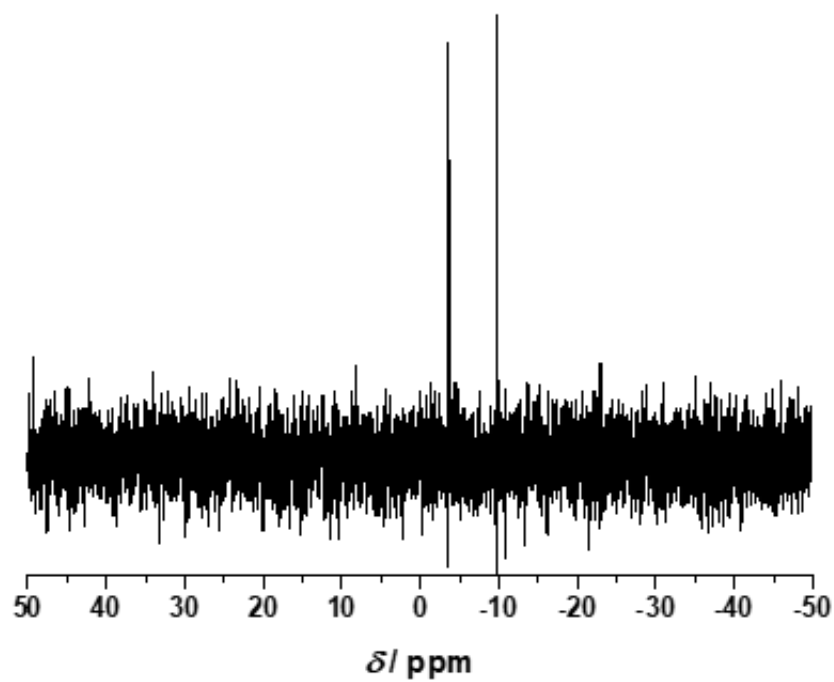


Figure SI- 11. ^{31}P NMR of **8** (162 MHz, CD_3CN).

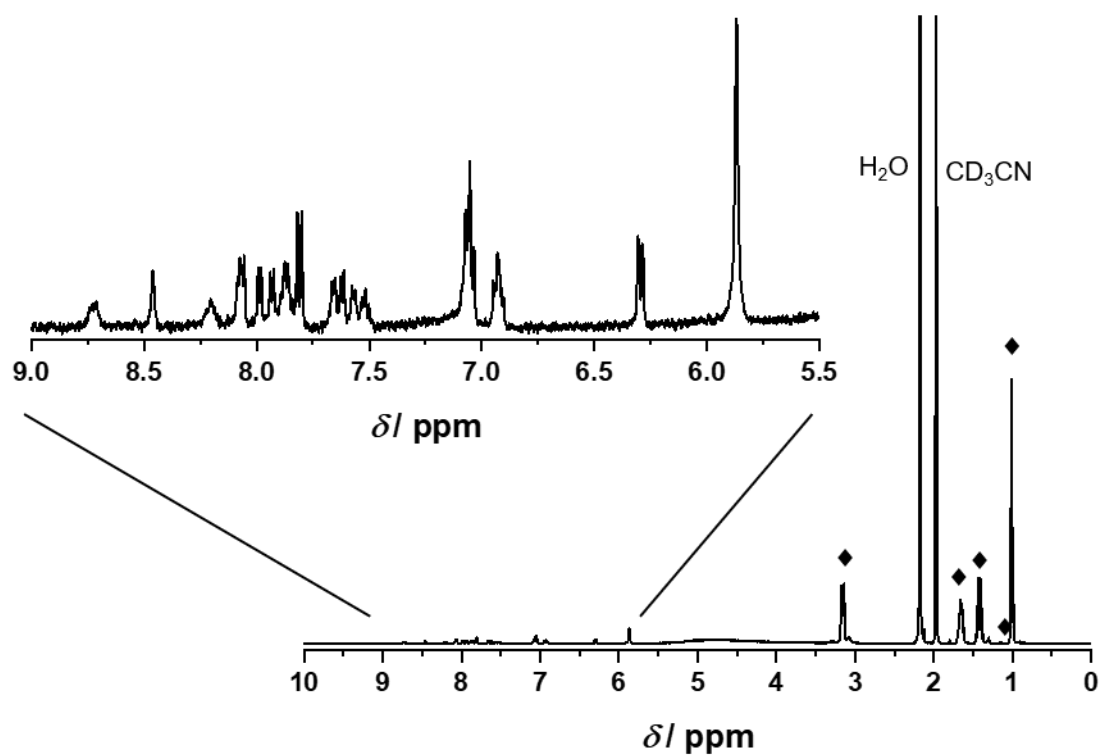


Figure SI- 12. ^1H NMR spectrum of **9** (400 MHz, CD_3CN). The TBA^+ signals (♦) are marked.

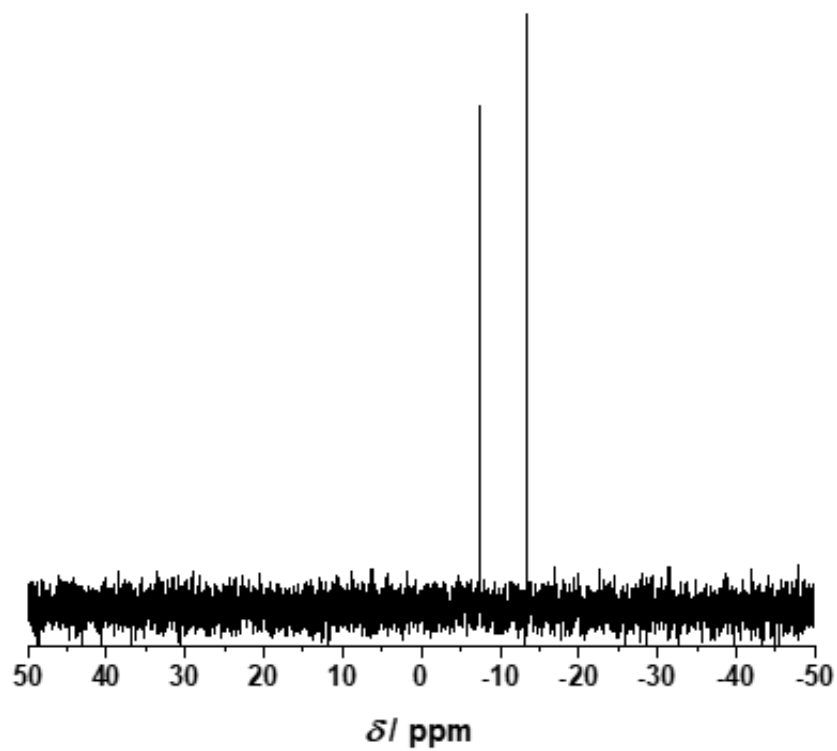


Figure SI- 13. ^{31}P NMR spectrum of **9** (162 MHz, CD_3CN).

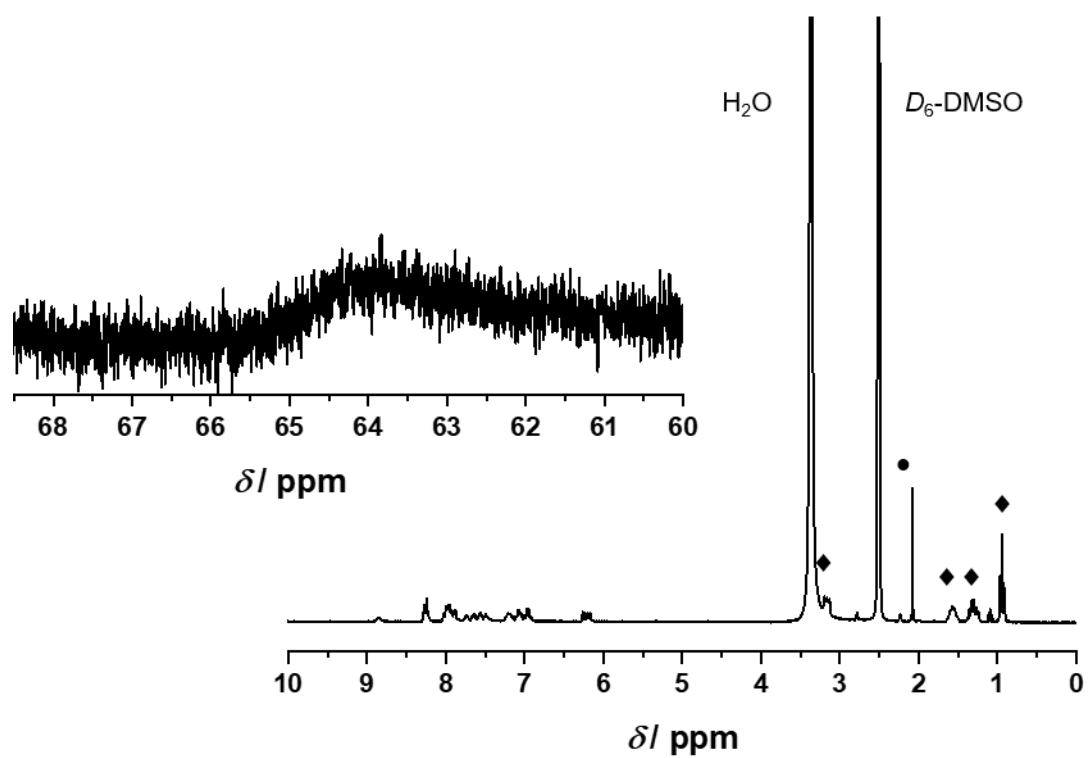


Figure SI- 14. ^1H NMR spectrum of **10** (400 MHz, D_6 -DMSO). Residual acetone (●) and TBA^+ signals (◆) are marked.

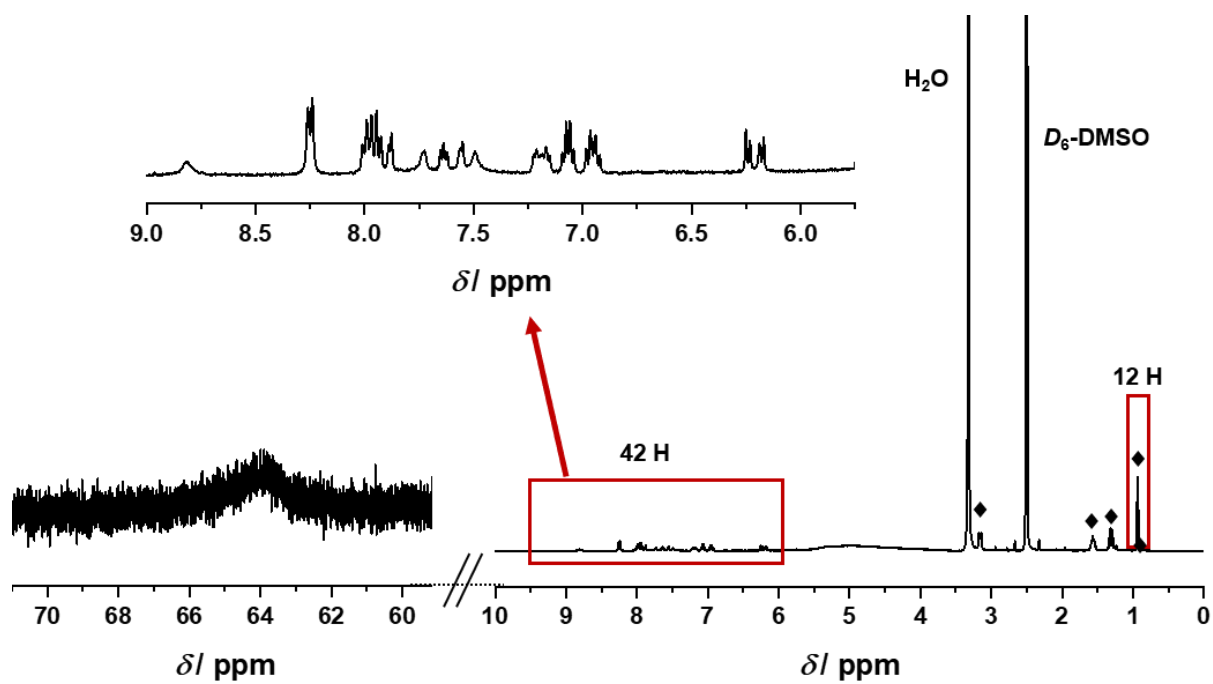


Figure SI- 15. ^1H NMR spectrum of **11** (400 MHz, D_6 -DMSO). The TBA⁺ signals (♦) are marked.

2. Mass spectrometry

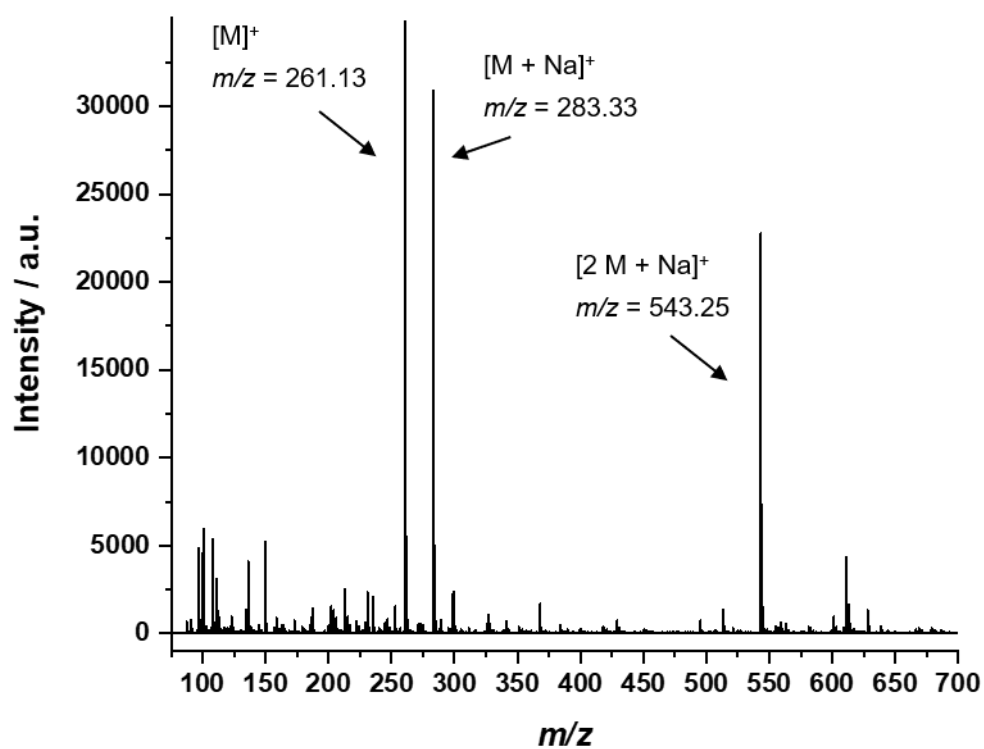


Figure SI- 16. ESI mass spectrum of 2 (positive mode).

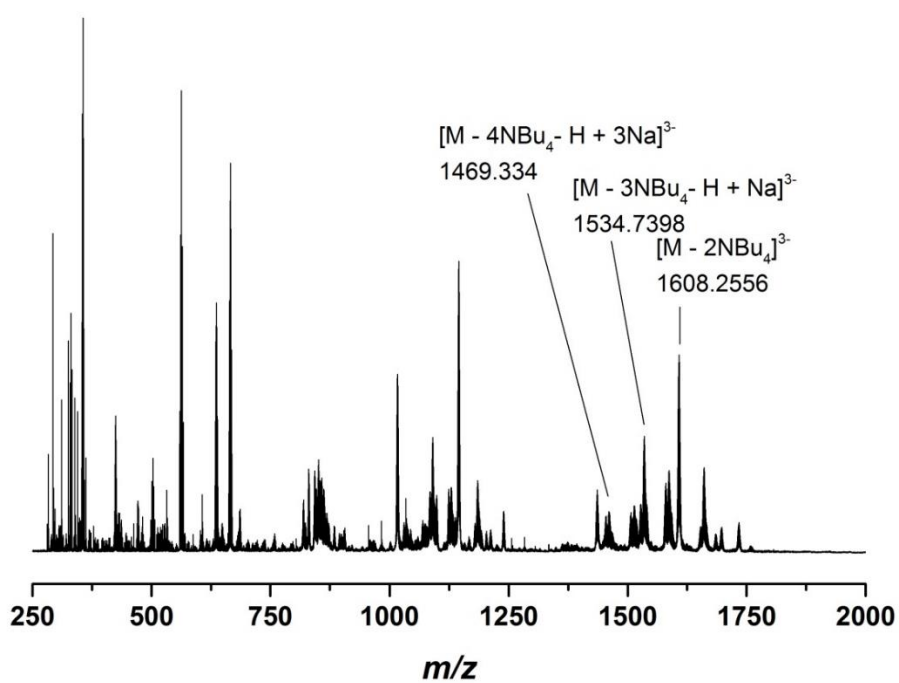


Figure SI- 17. ESI mass spectrum of 3 (negative mode).

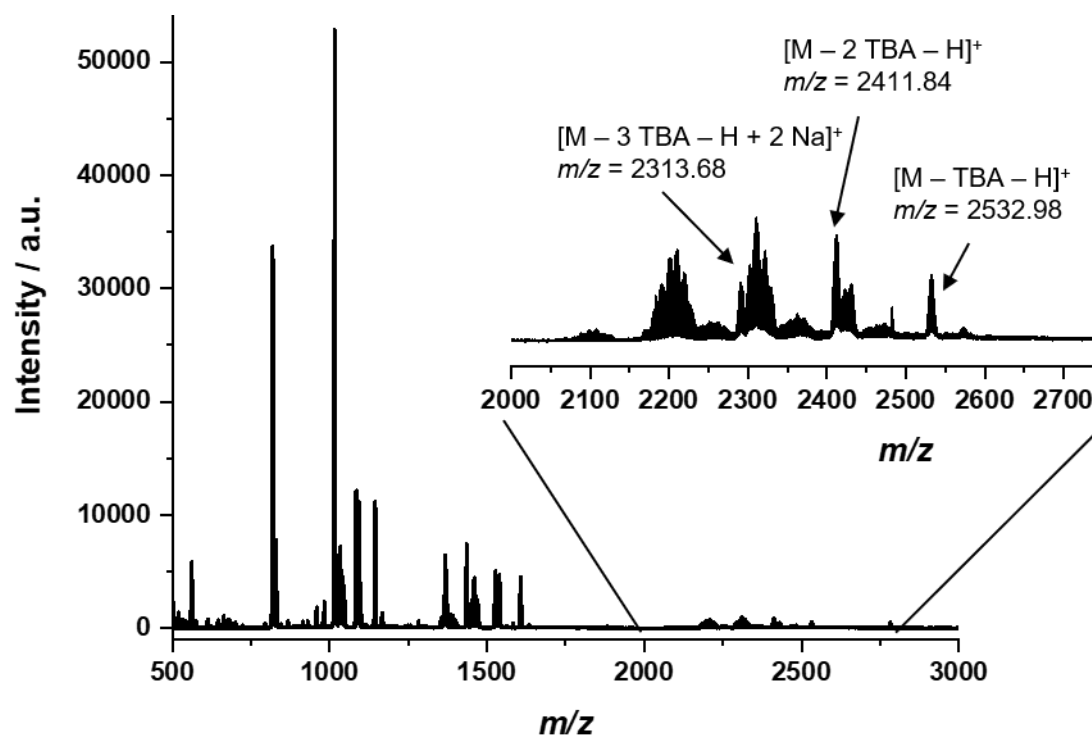


Figure SI- 18. MALDI-TOF mass spectrum of **3** (negative mode, DCTB as matrix, NaI as ionization salt).

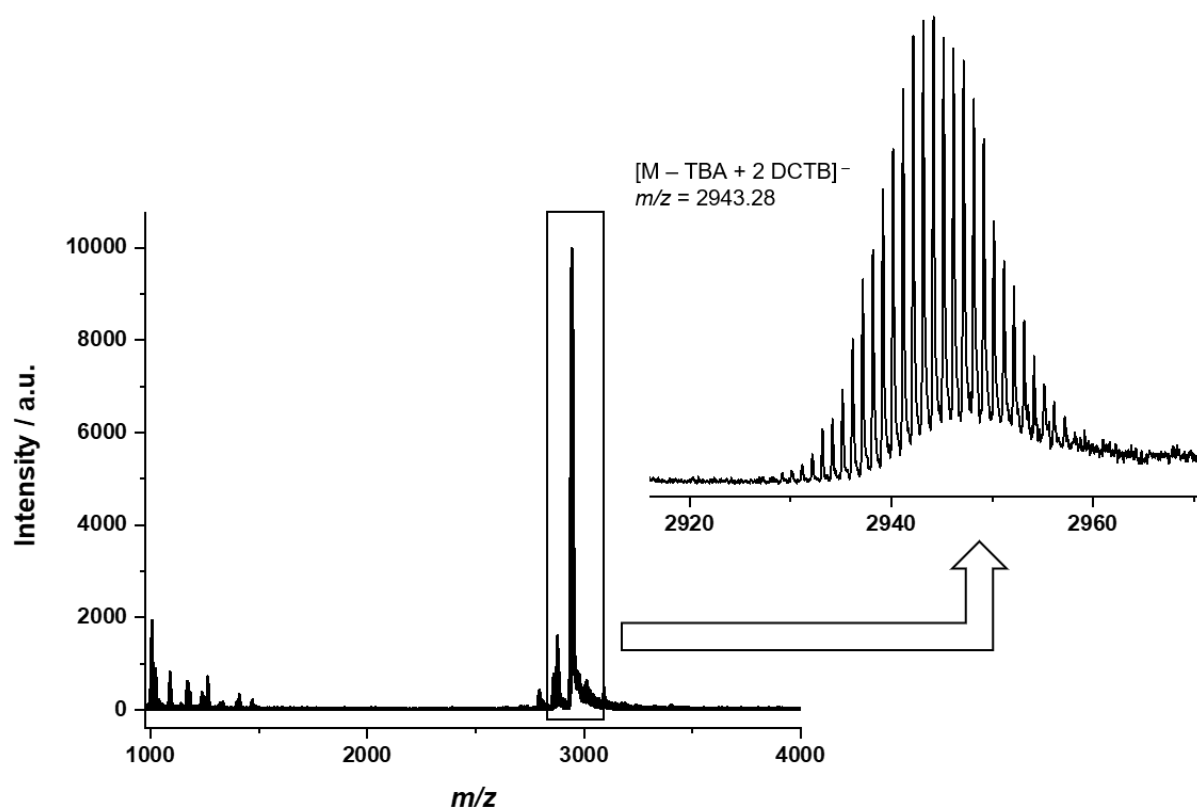


Figure SI- 19. MALDI-TOF mass spectrum of **4** (negative mode, DCTB as matrix, NaI as ionization salt).

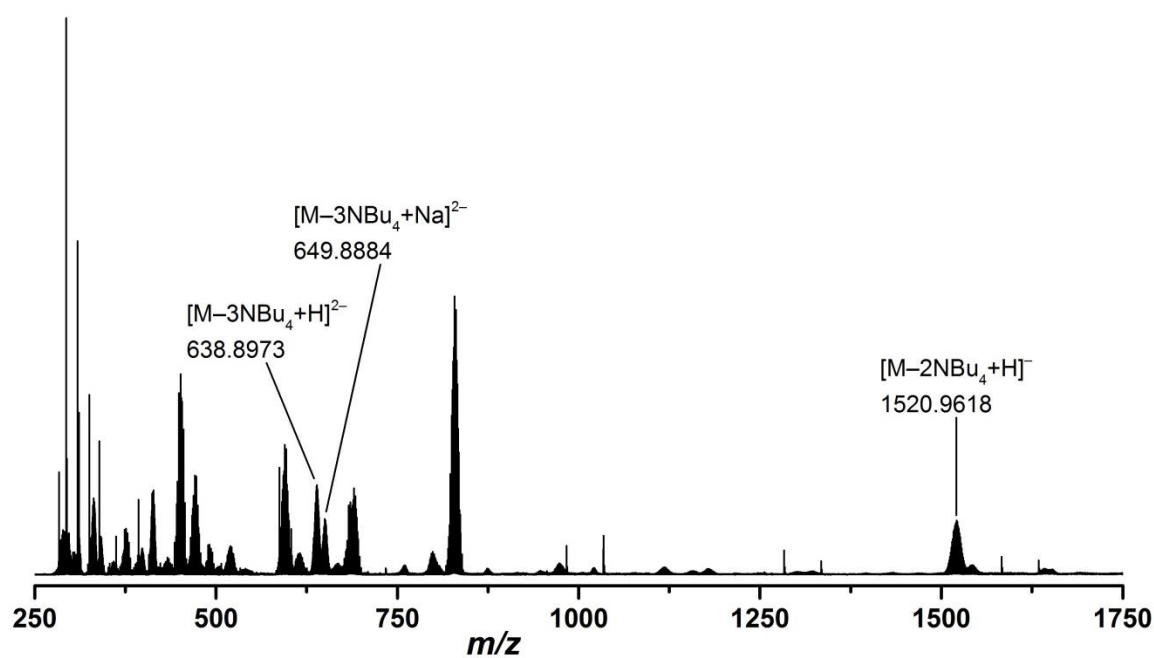


Figure SI- 20. ESI mass spectrum of 5 (negative mode).

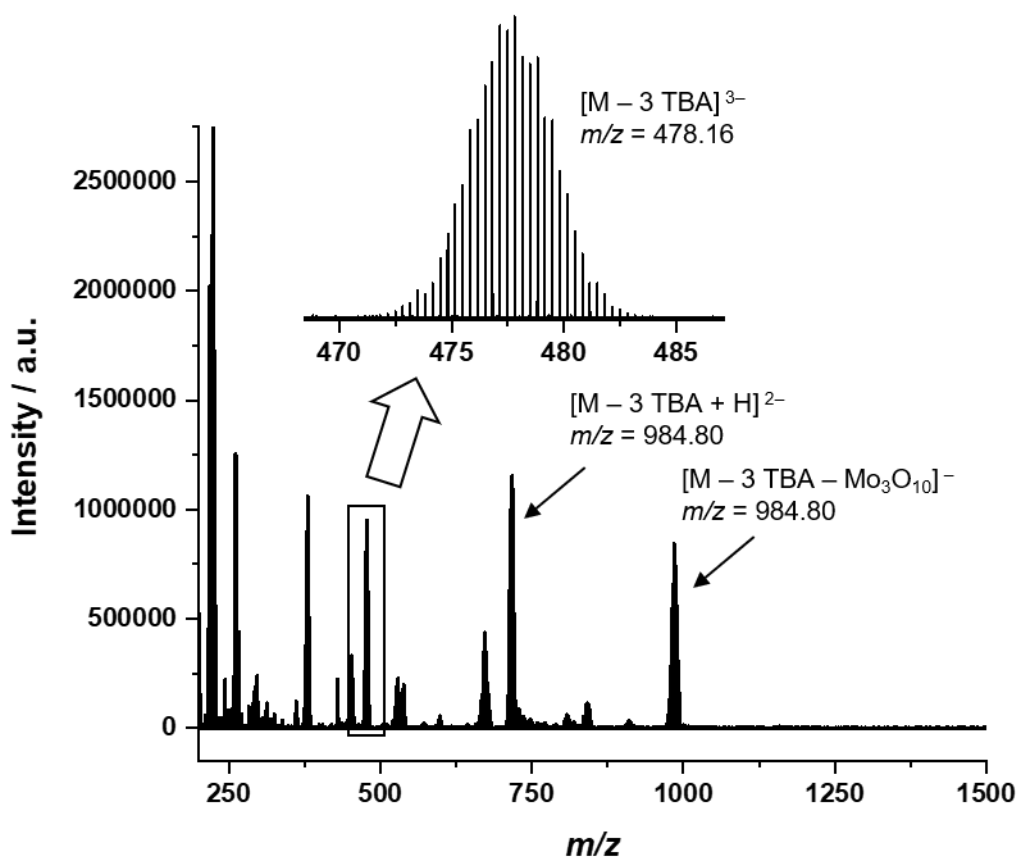


Figure SI- 21. ESI Mass spectrum of 6 (negative mode).

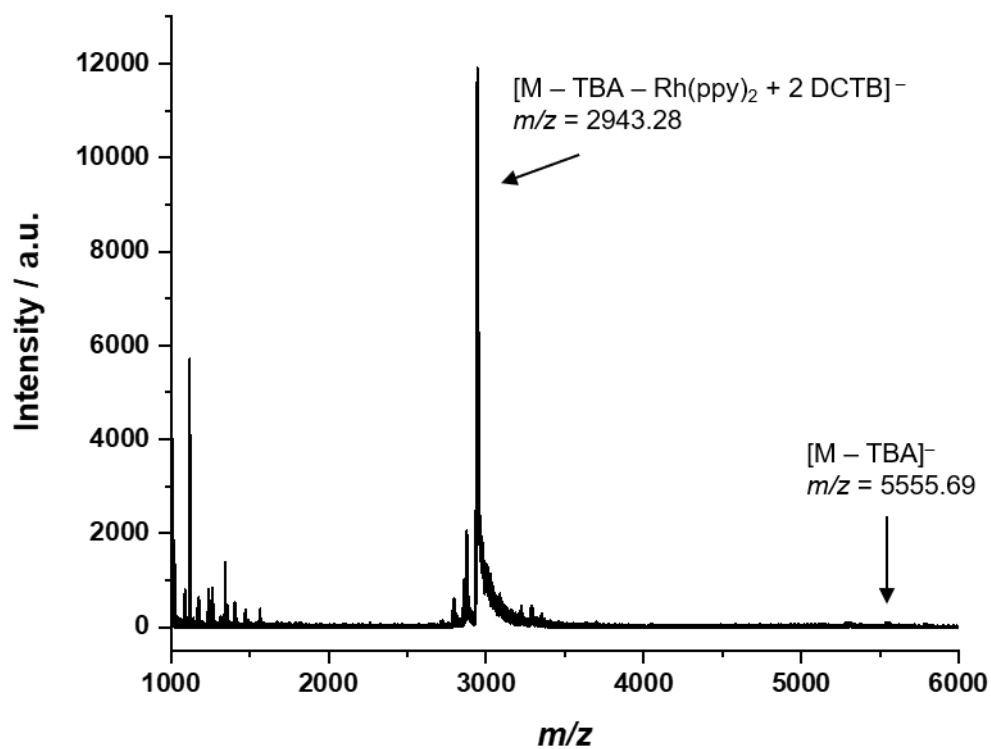


Figure SI- 22. MALDI-TOF mass spectrum of **8** (negative mode, DCTB as matrix, NaI as ionization salt).

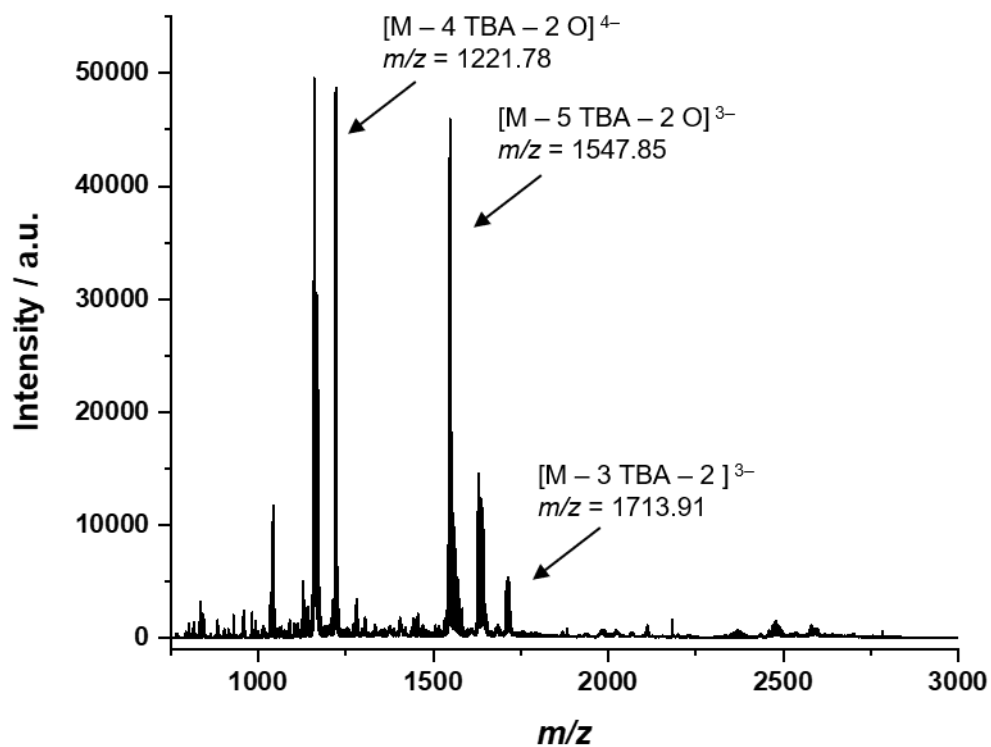


Figure SI- 23. ESI mass spectrum of **9** (negative mode).

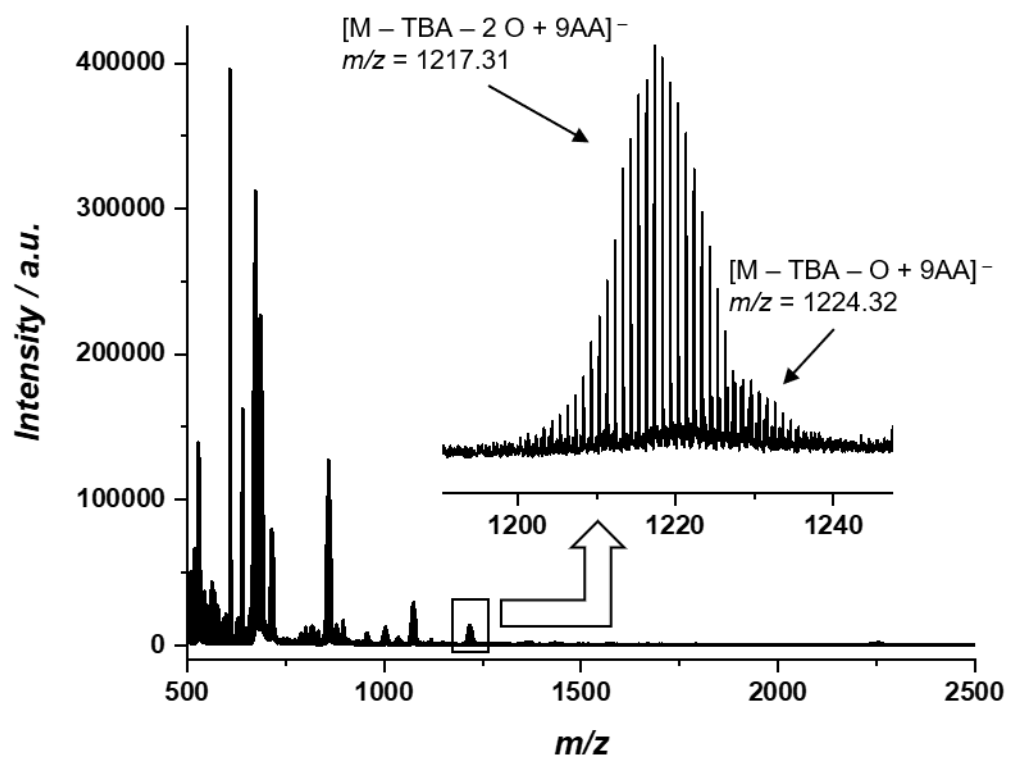


Figure SI- 24. MALDI-TOF mass spectrum of **10** (negative mode, 9-aminoacridine as matrix, NaI as ionization salt).

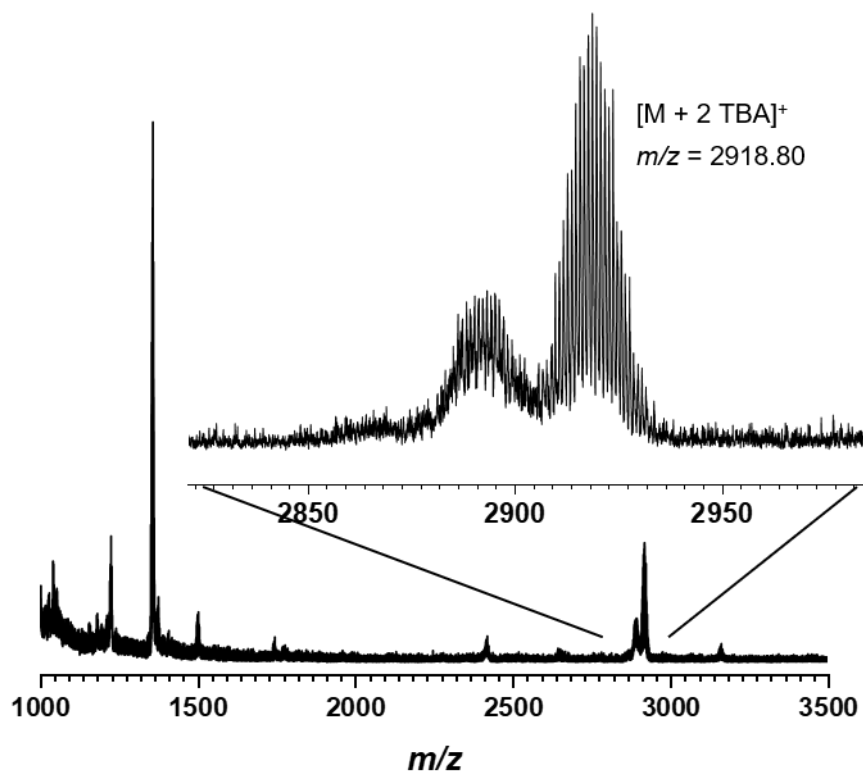


Figure SI- 25. MALDI-TOF mass spectrum of **11** (DCTB as matrix, KCl as ionization salt).

3. X-ray photoelectron spectroscopy

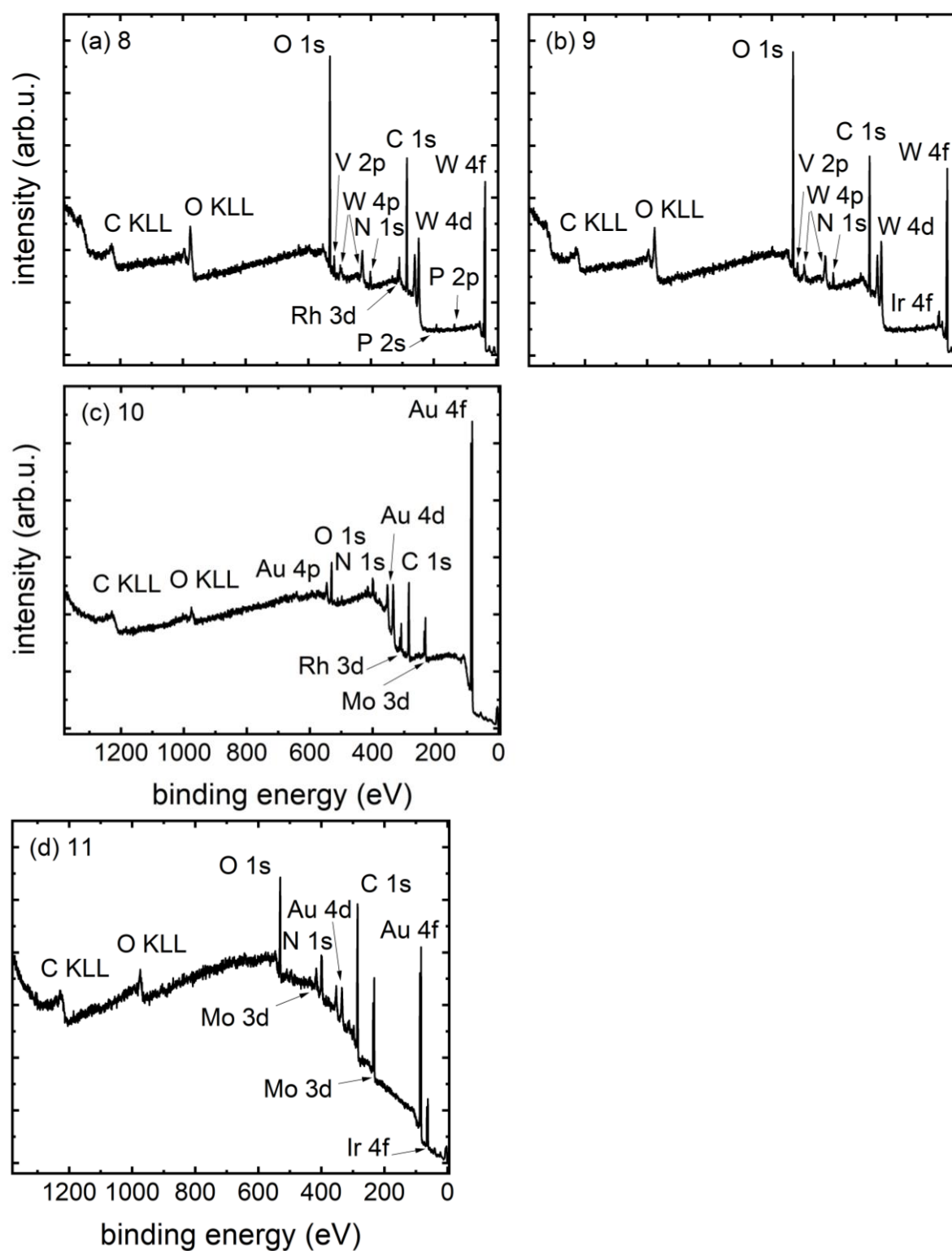


Figure SI- 26. XP overview spectra of the photosensitizer-POM dyads **8–11** with marked elements.

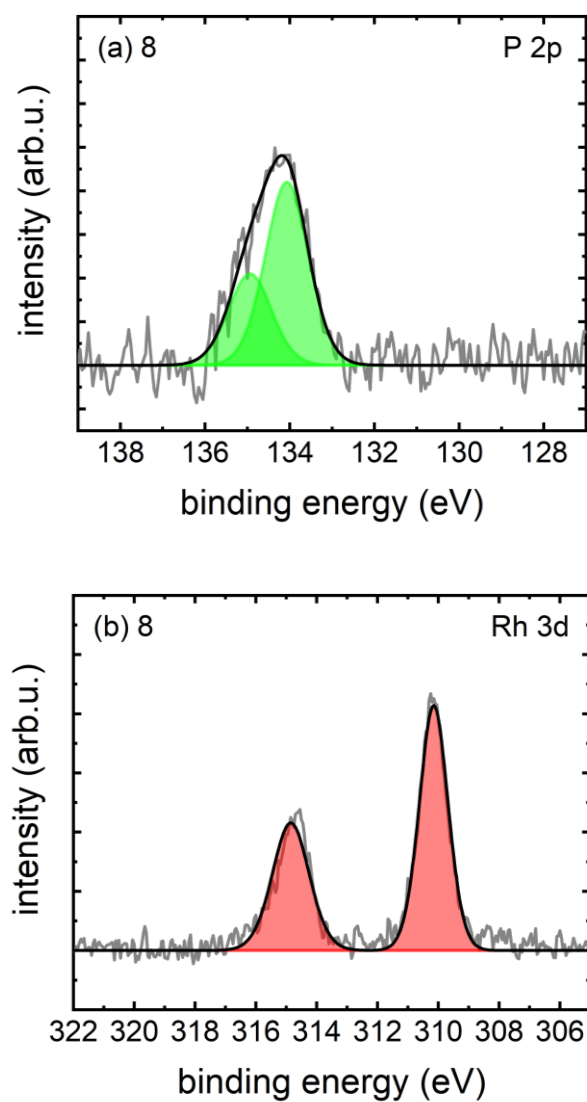


Figure SI- 27. High resolution XP P 2p and Rh 3d spectra of the photosensitizer-POM dyad 8.

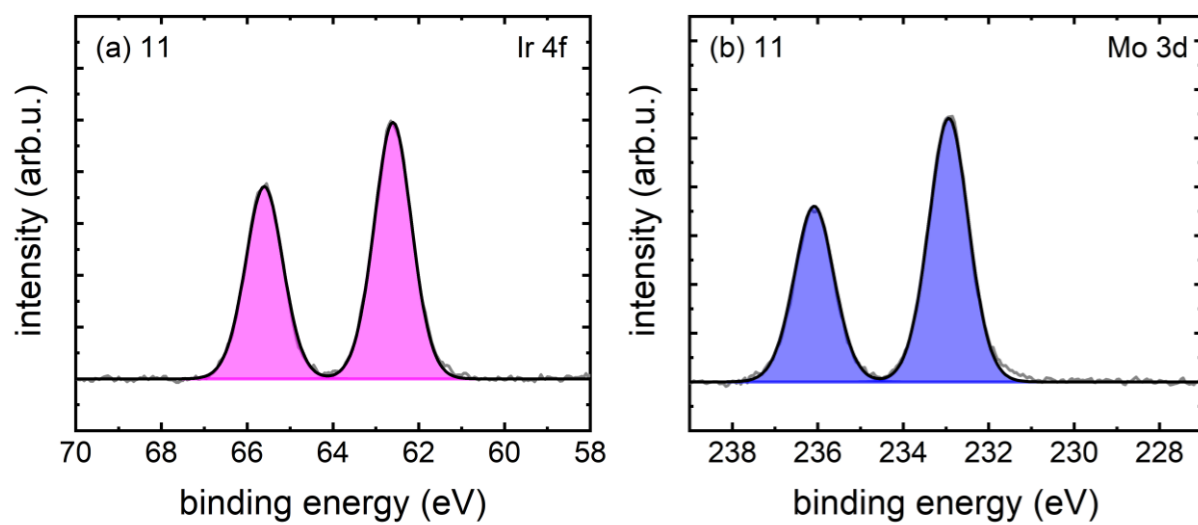


Figure SI- 28. High resolution XP Ir 4f and Mo 3d spectra of the photosensitizer-POM dyad 11.

Table SI-1. Quantitative analysis of the high-resolution XP spectra of the photosensitizer-POM dyads 8–11 including peak assignment, binding energies and full width at half maximum (FWHM) values obtained from the spectra deconvolution.

Peak assignment	Binding energy, eV	FWHM, eV
photosensitizer-POM dyad 8		
Rh 3d _{5/2}		
[(ppy) ₂ Rh] ⁺	310.2	1.1
P 2p _{3/2}		
P ₂ W ₁₅ V ₃ O ₆₂	134.1	1.2
W 4f _{7/2}		
P ₂ W ₁₅ V ₃ O ₆₂	36.1	1.0
photosensitizer-POM dyad 9		
Ir 4f _{7/2}		
[(ppy) ₂ Ir] ⁺	63.0	1.2
P 2p _{3/2}		
P ₂ W ₁₅ V ₃ O ₆₂	134.3	1.2
W 4f _{7/2}		
P ₂ W ₁₅ V ₃ O ₆₂	36.4	1.0
photosensitizer-POM dyad 10		
Rh 3d _{5/2}		
[(ppy) ₂ Rh] ⁺	309.8	1.0
Mo 3d _{5/2}		
MnMo ₆ O ₂₄	232.8	1.1
Mn 2p _{3/2}		
MnMo ₆ O ₂₄	642.5	5.0
photosensitizer-POM dyad 11		
Ir 4f _{7/2}		
[(ppy) ₂ Ir] ⁺	62.6	1.0
Mo 3d _{5/2}		
MnMo ₆ O ₂₄	232.9	1.1

Mn 2p _{3/2}		
MnMo ₆ O ₂₄	642.8	4.8

The peak fitting of the doublets was performed using fixed intensity ratios due to the spin-orbit coupling of the p, d and f photoelectrons, respectively. With respect to the determination of the elemental ratio (see the main manuscript), the following relative sensitivity factors (RSF) were used: 8.39 (Rh 3d_{5/2}), 0.79 (P 2p_{3/2}), 7.78 (Ir 4f_{7/2}), 5.62 (Mo 3d_{5/2}), 9.17 (Mn 2p_{3/2}) and 5.48 (W 4f_{7/2}).

4. Cyclic and square-wave voltammetry

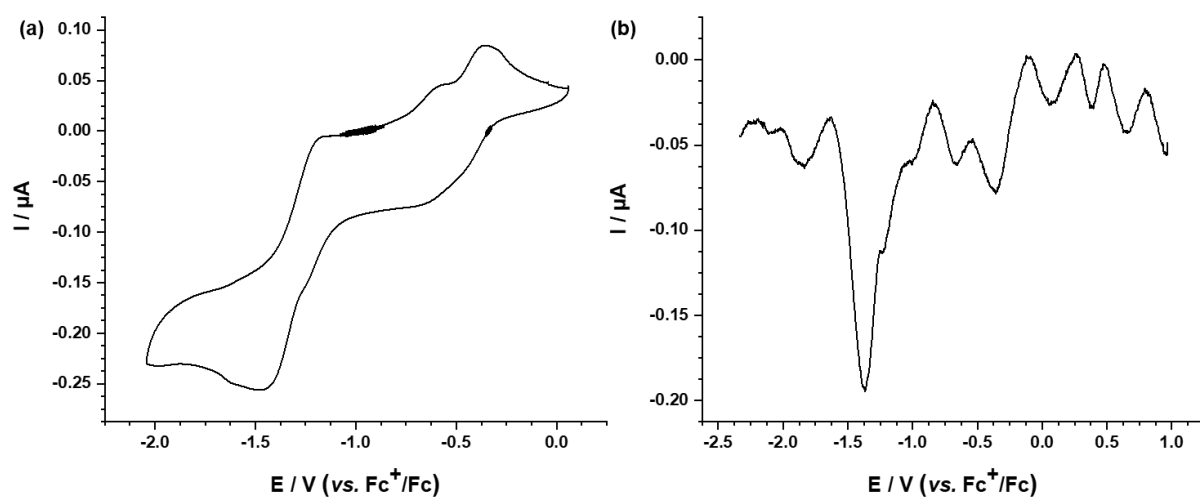


Figure SI- 29. Cyclic (a) and square-wave voltammograms (b) of **3**. CV and SWV were measured at room temperature in degassed CH_3CN containing 0.1 M $(\text{TBA})\text{PF}_6$ (scan rate of 100 mV/s).

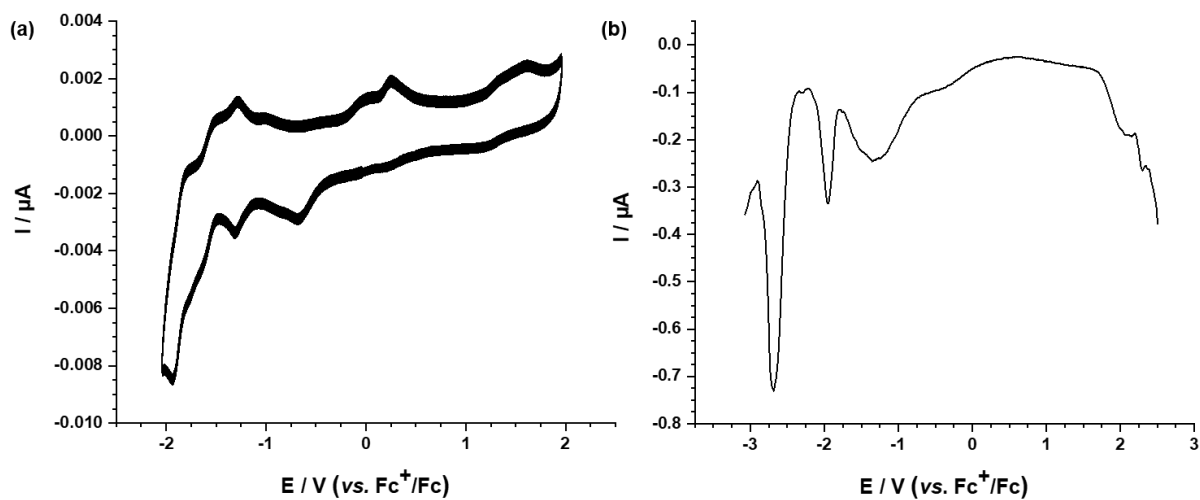


Figure SI- 30. Cyclic (a) and square-wave voltammograms (b) of **8**. CV and SWV were measured at room temperature in degassed CH_3CN containing 0.1 M $(\text{TBA})\text{PF}_6$ (scan rate of 200 mV/s).

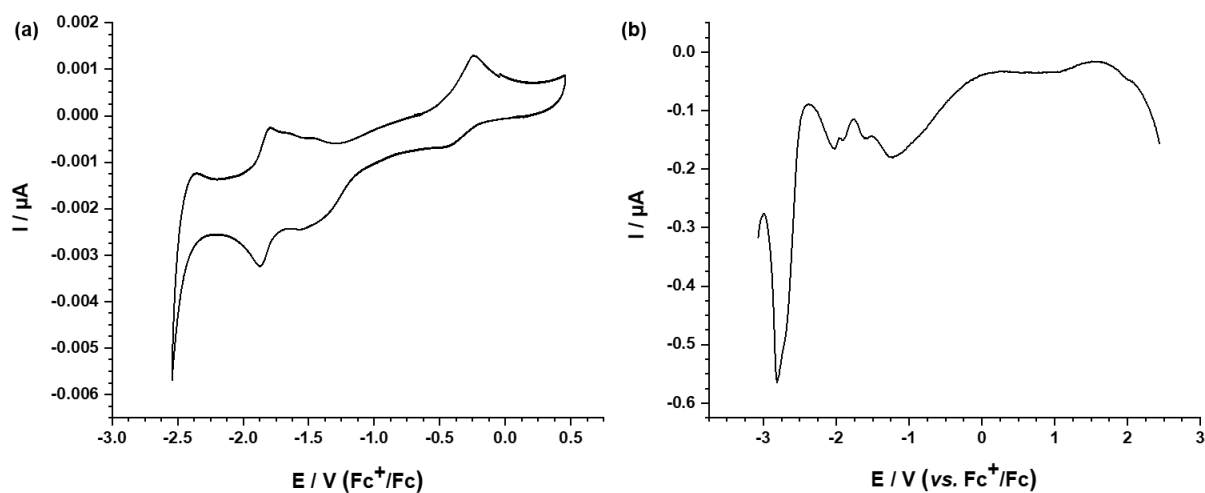


Figure SI- 31. Cyclic (a) and square-wave voltammograms (b) of **9**. CV and SWV were measured at room temperature in degassed CH₃CN containing 0.1 M (TBA)PF₆ (scan rate of 200 mV/s).

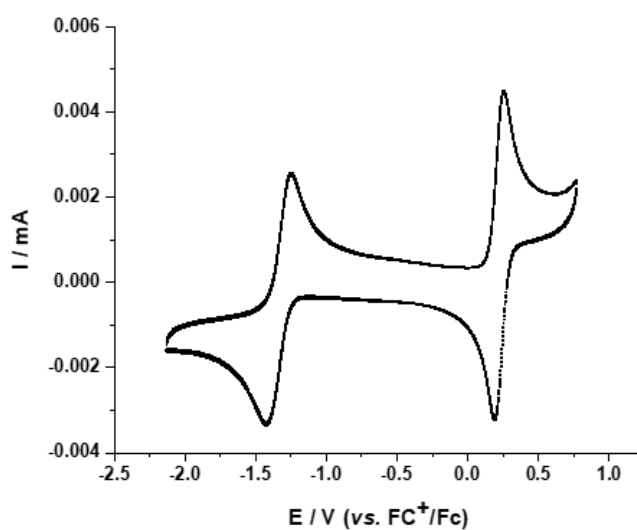


Figure SI- 32. Cyclic voltammogram of **5** measured at room temperature in degassed DMF containing 0.1 M (TBA)PF₆ (scan rates of 100 mV/s).

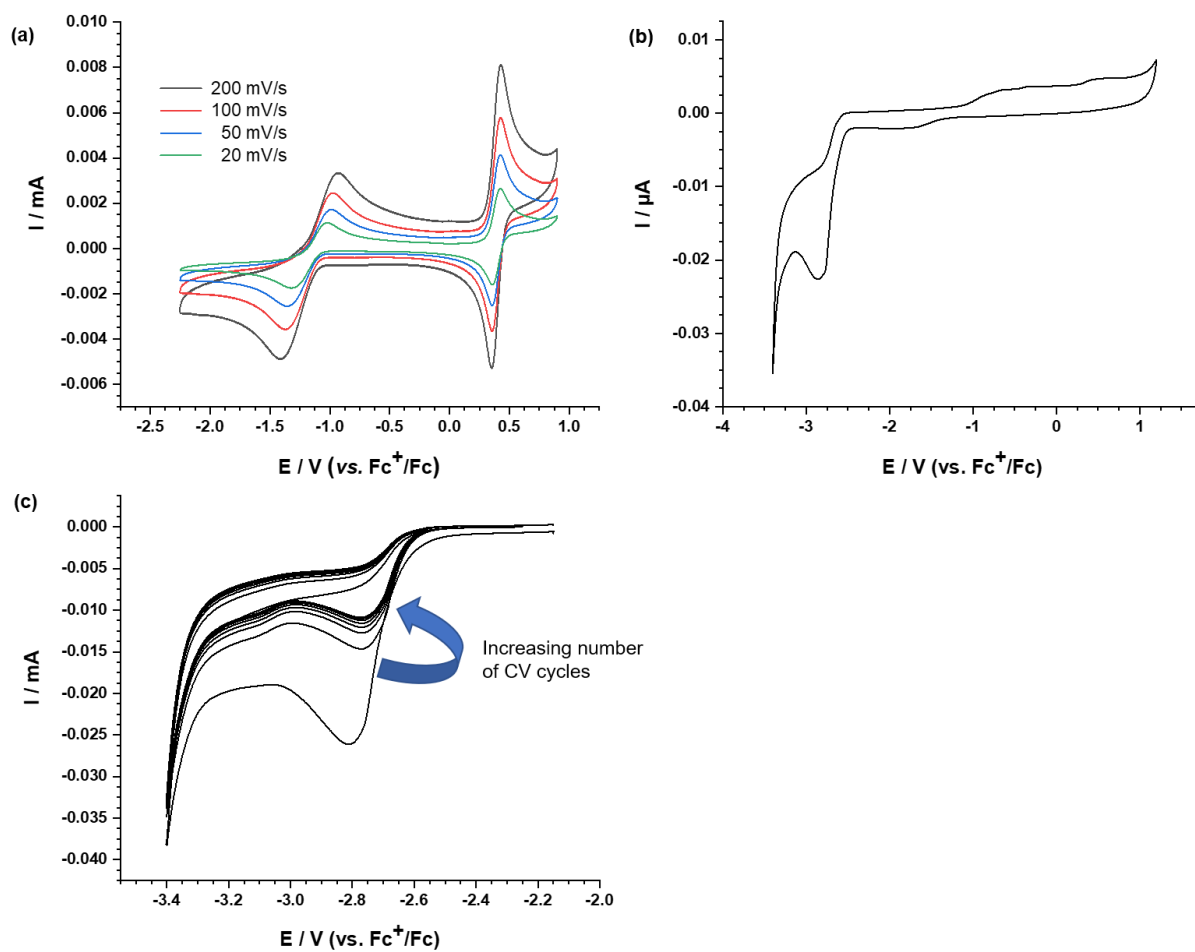


Figure SI- 33. Cyclic (a and c) and square-wave voltammograms (b) of 6. CV and SWV were measured at room temperature in degassed CH_3CN containing 0.1 M (TBA)PF₆. Different scan rates were used in the CV measurements in the potential range from -2.25 to 1 V.

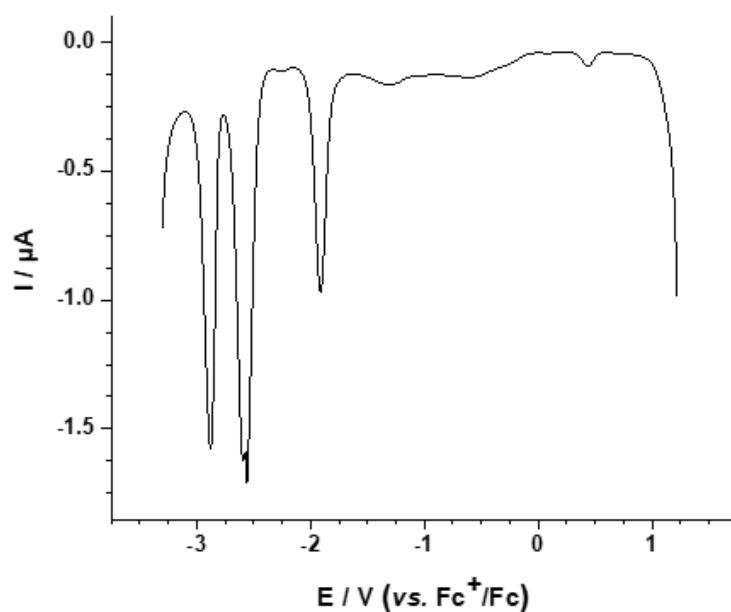


Figure SI- 34. Square-wave voltammogram of **10** measured at room temperature in degassed DMF containing 0.1 M TBPf₆ (scan rates of 100 mV/s).

5. UV/vis absorption and emission spectroscopy

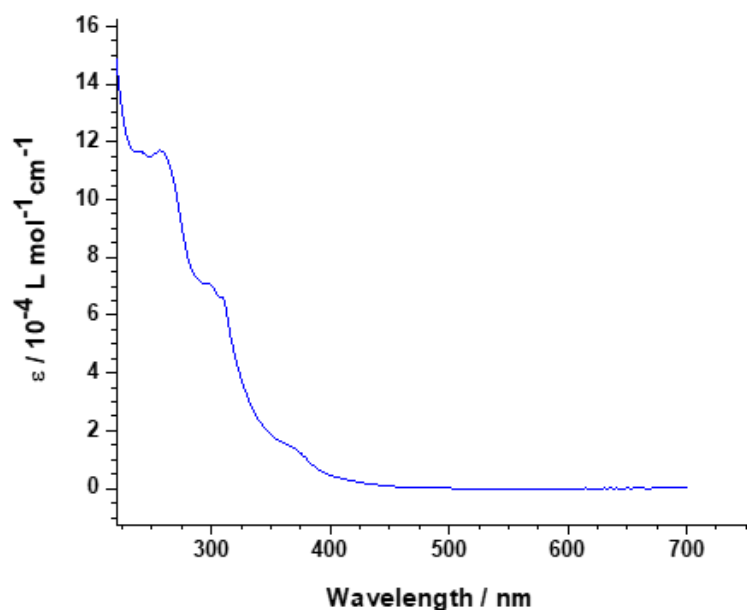


Figure SI- 35. UV/vis absorption spectrum of the Rh(III)-containing dyad **8**.