Supporting Information

Novel C7-Substituted Coumarins as Selective Monoamine Oxidase Inhibitors: Discovery, Synthesis and Theoretical Simulation

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1.1 Molecular docking studies.

Enlightened by the above-mentioned SARs, three compounds, M31, M32 and M43, were selected for docking simulations, aiming to provide a relatively reliable initial conformation in hMAO-A and hMAO-B for the next MD simulations. As illustrated in Figure S1, there were 7 different amino acids of the 24 residues in the two MAO subtypes. Among the 7 amino acids, it was worth noting that residue Ile199 acted as a gatekeeper between the entrance cavity and the substrate cavity with different conformations. Figure S2 showed the predicted binding poses of the 3 compounds in both hMAO-A and hMAO-B (PDB ID: 2Z5X and 2V61, respectively). As shown in Figure S2., these three coumarin derivatives which has different C7 substitutions bind to the active site with coumarin moiety ring towards the substrate cavity. It can be observed that these three C7 substituted coumarins all formed interactions with important residues, such as Phe208 and Gln215 in hMAO-A, Ile199 and Cys172 in hMAO-B. An extended ligand conformation of these 3 compounds was found after docking in hMAO-B. More specifically, the hydrogen bond between compound M32 and Ile199 could keep the gate open. However in hMAO-A, Ile199 residue was replaced by Phe208, which forced the ligand to adopt a more curved conformation. Starting from the optimal docked models, eight different simulation systems were therefore investigated to elucidate the selectivity mechanism, in which C18 was the co-crystallized ligand in 2V61 (MAO-A v.s. M31, MAO-A v.s. M32, MAO-A v.s. M43, MAO-A v.s. C18, MAO-B v.s. M31, MAO-B v.s. M32, MAO-B v.s. M43, MAO-B v.s. C18).



Figure S1. The residue differences of binding pockets between *h*MAO-A and *h*MAO-B. FAD cofactor is shown as space-filling model, MAO-A ligand as green sticks and MAO-B ligand as orange sticks.



Figure S2. Predicted binding modes of **M31**, **M32** and **M43** with *h*MAO-A and *h*MAO-B active site. FAD cofactor is shown as space-filling model and compound as green sticks. The H-bond receptor surfaces are shown in pink and the H-bond acceptor surfaces are shown in green.

1.2 Evaluation of the structural stability.

In MD simulations, RMSD analysis of the protein backbone was usually regarded as the first step for checking the systems stability. As depicted in Figure S3, the RMSD plots of the eight simulation systems were recorded with respect to the initial conformation at 0 ns over the 20 ns MD simulation. As clearly shown in the plots, only slight RMSD fluctuations were observed after 10 ns of simulation. This indicated that all the simulation systems achieved a relatively stable conformation during the equilibration period. Larger fluctuations were

observed more frequently in MAO-A systems than in MAO-B systems and this may due to the low resolution in the X-ray structure of the former protein. For each MAO subtypes, the complexes with the selective inhibitor have fewer fluctuations compared to the complexes with non-selective ligands. The stability of the selective complex systems was validated by the hydrogen bonds analysis between MAO subtypes and C7 substituted coumarins.

The regional conformational fluctuations of each residue were assessed by calculating the root-mean square fluctuation (RMSF). Similarly, RMSF profiles of main chains for eight simulation systems were depicted in Figure S4. In order to illustrate the different fluctuation trends of key regions in the binding site, three essential parts, the loop region, the α -helix and β -sheet were labeled in different colors in Figure S5. The residues with comparatively larger fluctuation suggested that they could adopt more flexible conformations during the whole MD simulations. As shown in Figure S4, the α -helix and the β -sheet regions of two subtypes exhibited small fluctuation patterns, indicating these two regions were more stable than loop regions. The decreased flexibility in the α -helix and the β -sheet region can be explained by stabilizing protein-ligand interactions between selective inhibitors and some key residues of MAOs (**M31** *v.s.* MAO-A, **M32** *v.s.* MAO-B). Moreover, **M31** *v.s.* MAO-A system had more fluctuations than **M43** systems in loop regions, since the C7 substituent of **M31** has higher number of rotatable bonds which cause the destabilization of the system. It could also be observed that selective systems had fewer larger fluctuations, which was consistent with the result of RMSD analysis.



Figure S3. RMSD plots of M31, M32, M43 and C18 with different hMAO subtypes.



Figure S4. Per-residue RMSF curves of the *h*MAO-A and *h*MAO-B systems.



Figure S5. The labeled key motifs of *h*MAO. Loop area is colored red. Alpha helix is colored magenta. Beta sheet is colored orange.

1.3 Hydrogen bonds Analysis.

The stability of hydrogen bonds interactions between *h*MAO and C7 substituted coumarins which are likely to contribute to the selectivity of MAOIs was investigated through the MD simulations. VMD 1.9.2 was used to analyze the hydrogen bonds in each of the eight simulation systems. The distance cutoff value was defined as 3.5 Å and the angle cutoff value was set as 35°. Only the occupancy over 5% during the equilibrium period was considered for analysis and the detailed results were listed in Table S1. Dual inhibitor **M43** formed H-bonds with residues in the α -helix region of two MAO subtypes, such as Asn181 of *h*MAO-A and Cys172 of *h*MAO-B. **M31** which is a selective inhibitor of MAO-A forms a dominant hydrogen bond interaction with residue Gln215 of α -helix region (occupancy = 95%), which may play a key role in *h*MAO-A selectivity. As for **M32** or **C18** against *h*MAO-B selective systems, H-bonds between Cys172 and Ile199 with low occupancy are observed. It was interesting that hydrogen

bonds formed by Tyr residues (Tyr326, Tyr398 and Tyr435), which caused the formation of a hydrophobic cage in *h*MAO-B binding site were stable during the MD simulations. These important hydrogen bond interactions stabilize a binding cavity and facilitating the binding of the inhibitors in *h*MAO-B receptor. The analysis suggested that a constant interaction with Gln215 from the α -helix would enhance the selectivity against *h*MAO-A. This observation is in agreement with the results of Giuseppe et al.[1]. Hydrogen bond between Cys172 and the inhibitor also contributed to the MAO-B selectivity since Asn181 in *h*MAO-A took the place of Cys172 in *h*MAO-B. Based on hydrogen bond analysis, a rational way to enhance MAO-subtype selectivity can be proposed. For instance, *h*MAO-A selectivity would be improved if the hydrogen bond interaction strength with Gln215 was enhanced. Structural modification to form a stable interaction with the hydrophobic cage could strengthen the activity against *h*MAO-B.

Svetome	Dopor	Acceptor	
		Acceptor	
M31 v.s. MAO-A	Gln215-Side	M31	95.1%
	Asn181-Side	M31	23.3%
	Ile207-Side	M31	10.4%
	Phe208-Side	M31	9.4%
M31 <i>v.s.</i> MAO-B	Tyr326-Side	M31	80.7%
	Phe343-Side	M31	9.4%
M32 <i>v.s.</i> MAO-A	Tyr197-Side	M32	77.7%
	Ile207-Side	M32	8.4%
	Gln215-Side	M32	53.5%
	M32	Asn181-Side	21.8%
M32 <i>v.s.</i> MAO-B	Tyr326-Side	M32	59.4%
	M32	Cys172-Main	5.5%
	M32	Ile199-Main	8.4%
M43 <i>v.s.</i> MAO-A	Tyr197-Side	M43	72.8%
	Asn181-Side	M43	11.4%
	Asn181-Main	M43	18.3%
	M43	Asn181-Side	11.4%
M43 <i>v.s.</i> MAO-B	M43	Cys172-Main	5.5%
	M43	Tyr435-Side	8.9%
C18 <i>v.s.</i> MAO-A	Phe208-Main	C18	40.1%
	C18	Asn181-Side	62.4%
-	C18	Asn181-Main	51.0%
	C18	Tyr444-Side	26.7%
-	C18	Thr336-Main	14.9%
	Ile207-Side	C18	19.3%
	C18	Phe208-Main	10.4%
	C18	Tyr407-Side	8.4%
С18 <i>v.s.</i> МАО-В	C18	Cys172-Main	21.8%
	C18	Tvr435-Side	22.8%

Table S1. Hydrogen bond analysis of four systems according to MD trajectories.

C	С18 Ту	r398-Side	20.3%
Ile199	9-Main	C18	5.5%

*Compound **M29-M34** and **M43** were obtained based on our BRS-3D VS protocol. **C18** was the selfcontained ligand in 2V61.

Table S2. Free energy decomposition results (kJ/mol) of *h*MAO-A and *h*MAO-B pocket residues.

Residues		MAO-A v.s. M43	МАО-В <i>v.s.</i> М43	MAO-A v.s. C18	МАО-В <i>v.s.</i> С18	
Tyr69	Tyr60	-2.1699 +/- 0.0485	-1.4899 +/- 0.0555	-2.2074 +/- 0.0487	-3.2926 +/- 0.0682	
Ala111	Pro102	1.2416 +/- 0.1103	2.2244 +/- 0.1224	1.0437 +/- 0.1108	2.4398 +/- 0.1487	
Phe112	Phe103	-0.7385 +/- 0.0290	0.3891 +/- 0.0531	1.0733 +/- 0.0554	0.8511 +/- 0.0559	
Pro113	Pro104	-0.1703 +/- 0.0073	-1.9962 +/- 0.1090	-0.1157 +/- 0.0595	-0.6065 +/- 0.0435	
Trp128	Trp119	0.0561 +/- 0.0050	-1.1615 +/- 0.0762	-0.1526 +/- 0.0090	-0.3703 +/- 0.0289	
Phe173	Leu164	-0.0873 +/- 0.0097	-1.9611 +/- 0.1391	-0.5923 +/- 0.0153	-1.8558 +/- 0.0540	
Leu176	Leu167	-0.0459 +/- 0.0127	-2.4919 +/- 0.0868	-0.8342 +/- 0.0254	-1.4404 +/- 0.0606	
Phe177	Phe168	0.3149 +/- 0.0327	-6.0771 +/- 0.1701	-2.8121 +/- 0.0751	-3.4035 +/- 0.0838	
Ile180	Leu171	-6.2327 +/- 0.0959	-6.1374 +/- 0.2283	-4.7420 +/- 0.1668	-9.9879 +/- 0.1790	
Asn181	Cys172	-1.7920 +/- 0.2895	-4.1876 +/- 0.1293	-19.2334 +/- 0.3718	-8.6903 +/- 0.2828	
Tyr197	Tyr188	-1.9959 +/- 0.2075	0.8642 +/- 0.0992	-2.1768 +/- 0.0831	-0.8742 +/- 0.1858	
Ile207	Ile198	-0.5426 +/- 0.1839	-1.7415 +/- 0.1249	-5.9627 +/- 0.1519	-3.5065 +/- 0.1354	
Phe208	Ile199	-3.4572 +/- 0.2106	-6.8623 +/- 0.2022	-6.7249 +/- 0.3277	-8.1047 +/- 0.2021	
Gln215	Gln206	-3.6756 +/- 0.1794	-2.5042 +/- 0.1704	-6.1142 +/- 0.2023	-6.0440 +/- 0.2239	
Cys323	Thr314	-0.8257 +/- 0.0808	-0.7217 +/- 0.0387	-0.2966 +/- 0.0893	-0.1698 +/- 0.0357	
Ile325	Ile316	-0.8940 +/- 0.0330	-6.0684 +/- 0.1105	-2.0462 +/- 0.1280	-4.1768 +/- 0.1016	
Ile335	Tyr326	-6.8937 +/- 0.1179	-5.5909 +/- 0.1882	-7.1344 +/- 0.1613	-2.2518 +/- 0.1711	
Thr336	Thr327	0.0826 +/- 0.0628	0.0767 +/- 0.0205	1.1541 +/- 0.1967	-0.6694 +/- 0.0298	
Leu337	Leu328	-3.8445 +/- 0.0960	-0.6418 +/- 0.0266	-3.4453 +/- 0.1016	-0.7503 +/- 0.0235	
Met350	Met341	-0.3784 +/- 0.0178	-0.1216 +/- 0.0091	-0.4051 +/- 0.0440	-0.1350 +/- 0.0266	
Phe352	Phe343	-0.9866 +/- 0.0386	-1.5778 +/- 0.0633	-0.5815 +/- 0.0494	-1.1044 +/- 0.0894	
Tyr407	Tyr398	-10.2894 +/- 0.1631	-6.7161 +/- 0.1788	-4.8024 +/- 0.2837	-8.0050 +/- 0.3251	
Tyr444	Tyr435	-3.9190 +/- 0.2187	-1.2072 +/- 0.1828	-5.3449 +/- 0.1905	-5.7731 +/- 0.2653	

Table S3. pIC₅₀ values of synthesized Esuprone derivatives.



Compds	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	MAO-A*	MAO-B*
5a		-(CH2)3-	Η	Н	C ₂ H ₅ SO ₂	CH ₃	0.011	2.3
5b	-	-(CH2)3-	Н	Н	p-CH ₃ C ₆ H ₄ SO ₂	CH ₃	0.012	4.9
5c		-(CH2)3-	Η	Н	<i>p</i> -NO ₂ C ₆ H ₄ SO ₂	CH ₃	< 0.005	0.59
7d	Н	Met	Н	Н	$C_2H_5SO_2$	Н	1.4	> 100
7e	Н	\$ 	Н	Н	<i>p</i> -CH ₃ C ₆ H ₄ SO ₂	Н	1.1	9.5
7 f	Н	₹	Н	Н	p-NO ₂ C ₆ H ₄ SO ₂	Н	0.095	9
7h	Н	(CH ₂) ₂ CH ₃	Н	Н	C ₂ H ₅ SO ₂	CH ₃	0.91	9.9

7i	Н	(CH ₂) ₂ CH ₃	Н	Н	<i>p</i> -CH ₃ C ₆ H ₄ SO ₂	CH ₃	0.17	4.1
7j	Н	$(CH_2)_2CH_3$	Н	Н	p-NO ₂ C ₆ H ₄ SO ₂	CH ₃	0.029	1.7

*: The unit of IC_{50} values is uM. IC_{50} value of Esuprone is equal to 10.7 nM. For Esuprone analogs, four compounds **5a-c** and **7j** had comparable activity with Esuprone. Compound **5c** was the most active MAO-A inhibitor with an IC_{50} value of ~5 nM. Comparing with **7d** and Esuprone, installation of cyclopropyl group at C4-position significantly decreased the MAO-A inhibition. However, flexible substitutes at C4-position can be accommodated for MAO-A inhibition (**7h-j**), which indicated large rigid substitutions were not suitable for MAO-A inhibitors It can also be observed that changing the substituent at C7 position generally influenced the inhibition activity towards MAO-A (**5a-c**).



Figure. S6. RMSD plots for 16 systems.

References

 Mangiatordi, G.F.; Alberga, D.; Pisani, L.; Gadaleta, D.; Trisciuzzi, D.; Farina, R.; Carotti, A.; Lattanzi, G.; Catto, M.; Nicolotti, O. A rational approach to elucidate human monoamine oxidase molecular selectivity. *Eur J Pharm Sci* 2017, *101*, 90-99, doi:10.1016/j.ejps.2017.02.008.





The ¹H NMR spectra of Compound 2





The ¹H NMR, ¹³C NMR spectra of Compound 3



The ¹H NMR spectra of Compound 4



The ¹H NMR spectra of Compound 6b







The ¹H NMR, ¹³C NMR spectra of Compound 7a





The ¹H NMR, ¹³C NMR spectra of Compound 7c

The ¹H NMR, ¹³C NMR spectra of Compound 7g



The ¹H NMR, ¹³C NMR spectra of Compound 7k



The ¹H NMR, ¹³C NMR spectra of Compound 5a



The ¹H NMR, ¹³C NMR spectra of Compound 5b



The ¹H NMR, ¹³C NMR spectra of Compound 5c



The ¹H NMR, ¹³C NMR spectra of Compound 7d



The ¹H NMR, ¹³C NMR spectra of Compound 7e



The ¹H NMR, ¹³C NMR spectra of Compound 7f



The ¹H NMR, ¹³C NMR spectra of Compound 7h



The ¹H NMR, ¹³C NMR spectra of Compound 7i



The ¹H NMR, ¹³C NMR spectra of Compound 7j

