Supplementary materials

Synthesis, Characterization and Solubility Determination of 6-Phenyl-pyridazin-3(2H)-one

in Different Pharmaceutical Solvents

Faiyaz Shakeel^{*1}, Mohd. Imran², Nazrul Haq¹, Sultan Alshehri¹ and Md. Khalid Anwer³

¹Department of Pharmaceutics, College of Pharmacy, King Saud University, P.O. Box 2457,

Riyadh 11451, Saudi Arabia

²Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Northern Border University,

P.O. Box 840, Rafha 919111, Saudi Arabia

³Department of Pharmaceutics, College of Pharmacy, Prince Sattam bin Abdulaziz University,

Al-Kharj, Saudi Arabia

*Author for correspondence Dr. Faiyaz Shakeel, Associate Professor, Department of Pharmaceutics, College of Pharmacy, King Saud University, P.O. Box 2457, Riyadh 11451, Saudi Arabia Email: faiyazs@fastmail.fm



Figure S1. FT-IR spectra of compound PPD showing some characteristic peaks at different wave numbers.



Figure S2: Proton NMR spectra of compound PPD showing some characteristic peaks at

different δ values



Figure S3: ¹³C NMR spectra of compound PPD showing some characteristic peaks at different δ values



Figure S4: Mass spectra of compound PPD showing some characteristic peaks at different m/z values



Figure S5: Correlation of experimental natural logarithmic solubilities ($\ln x_e$) of PPD with "van't Hoff model" in various neat solvents as a function of 1/T; symbols represent the experimental $\ln x_e$ values of PPD and the solid lines represent the $\ln x^{van't}$ values calculated by "van't Hoff model"



Figure S6: Scheme for the synthesis of compound PPD

Table S1: Elemental analysis of	compound PPD in	terms of C,	H and N

Compound	Mass fraction of C (%)		Mass fraction of H (%)		Mass fraction of N (%)	
PPD	Calculated	Found	Calculated	Found	Calculated	Found
	69.76	69.75	4.68	4.66	16.27	16.25

Components	Hansen solubility parameters					
	$\delta_{\rm d}/{\rm MPa}^{1/2}$	$\delta_{\rm p}/{ m MPa}^{1/2}$	$\delta_{\rm h}/{\rm MPa}^{1/2}$	$\delta/\mathrm{MPa}^{1/2}$		
PPD	20.00	12.90	6.50	24.70		
Water	15.50	16.00	42.30	47.80		
Ethanol	16.20	8.40	17.60	25.40		
Methanol	17.40	10.60	22.40	30.30		
PG	17.40	9.10	21.70	29.20		
PEG-400	14.60	7.50	9.40	18.90		
Transcutol	16.30	7.20	11.90	21.40		
EG	18.00	11.10	23.40	31.60		
IPA	15.80	6.60	14.30	22.30		
1-Butanol	15.90	6.30	15.20	22.90		
2-Butanol	15.80	5.40	12.40	20.80		
EA	15.70	5.60	7.00	18.10		
DMSO	17.40	14.20	7.30	23.60		

Table S2: The δ value of PPD and different neat solvents at T = 298.2 K calculated by HSPiP software

Material	Molecular formula	Molar mass (g mol ⁻¹)	CAS Registry no.	Purification method	Mass fraction purity	Analysis method	Source
PPD	$C_{10}H_8N_2O$	172.18	2166-31-6	Recrystallization	>0.97	HPLC	Synthesized
Methanol	CH ₃ OH	32.04	67-56-1	None	>0.99	GC	Sigma Aldrich
Ethanol	C ₂ H ₅ OH	46.07	64-17-5	None	>0.99	GC	Sigma Aldrich
EG	$C_2H_6O_2$	62.07	107-21-1	None	>0.99	GC	Fluka Chemica
Transcutol	$C_{6}H_{14}O_{3}$	134.17	111-90-0	None	>0.99	GC	Gattefosse
PG	$C_3H_8O_2$	76.09	57-55-6	None	>0.99	GC	Fluka Chemica
PEG-400	H(OCH ₂ CH ₂) _n OH	400	25322-68-3	None	>0.99	HPLC	Fluka Chemica
IPA	C_3H_8O	60.10	67-63-0	None	>0.99	GC	Sigma Aldrich
1-Butanol	$C_4H_{10}O$	74.12	71-36-3	None	>0.99	GC	Sigma Aldrich
2-Butanol	$C_4H_{10}O$	74.12	78-92-2	None	>0.99	GC	Sigma Aldrich
DMSO	C_2H_6OS	78.13	67-68-5	None	>0.99	GC	Fluka Chemica
EA	$C_4H_8O_2$	88.11	141-78-6	None	>0.99	GC	Fluka Chemica
Water	H ₂ O	18.07	7732-18-5	None	-	-	Milli-Q

Table S3: Information about materials used in the experiment

Both the analysis method and purity were provided by supplier except in case of PPD which was synthesized