

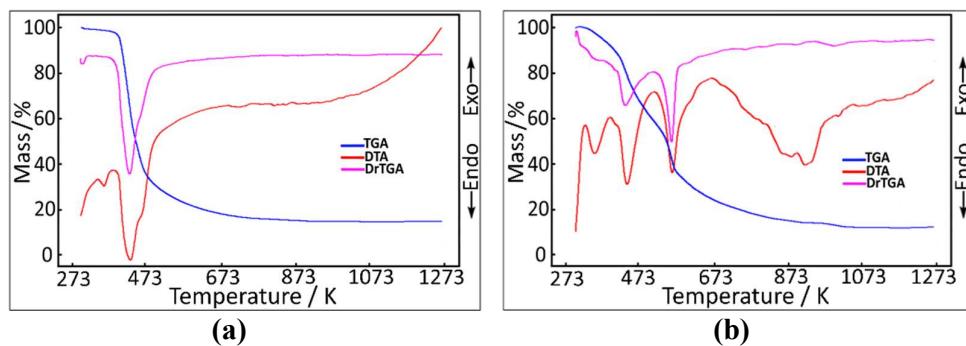
SUPPLEMENTARY FILE

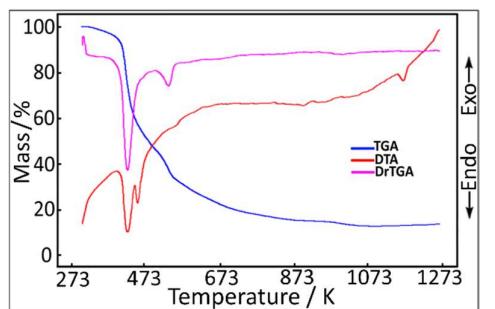
Novel Non-Metal Cations (NMCs) Pentaborate Salts with Some Aminoacids

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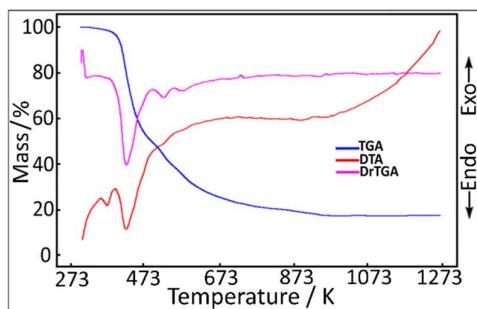
SI Table 1. ^{11}B -NMR analysis values of pentaborate salts containing organic compound as a complementary cation.

| Pentaborate salts | Boric acid | Triborate anion | Pentaborate anion |
|---|---|--|--|
| $[\text{C}_5\text{H}_{11}\text{NO}_2][\text{B}_5\text{O}_6(\text{OH})_4]\text{H}_2\text{O}$ $\text{C}_5\text{H}_{17}\text{B}_5\text{NO}_{13}$ | $\begin{array}{c} \text{OH} \\ \\ \text{HO}-\text{B}-\text{OH} \end{array}$ | $\begin{array}{c} \text{HO}-\text{B}-\text{O}-\text{B}-\text{OH} \\ \\ \text{O}^--\text{B}-\text{O} \\ \\ \text{HO}-\text{OH} \end{array}$ | $\begin{array}{c} \text{HO}-\text{B}-\text{O}-\text{B}-\text{OH} \\ \\ \text{O}^--\text{B}-\text{O} \\ \\ \text{O}-\text{B}-\text{O} \\ \\ \text{HO}-\text{B}-\text{O}-\text{B}-\text{OH} \end{array}$ |
| $[\text{C}_6\text{H}_{13}\text{NO}_2][\text{B}_5\text{O}_6(\text{OH})_4]1/2\text{H}_2\text{O}$ $\text{C}_6\text{H}_{18}\text{B}_5\text{NO}_{12.5}$ | | | |
| $[\text{C}_6\text{H}_{13}\text{NO}_2][\text{B}_5\text{O}_6(\text{OH})_4]\text{H}_2\text{O}$ $\text{C}_6\text{H}_{29}\text{B}_5\text{NO}_{13}$ | | | |
| $[\text{C}_4\text{H}_9\text{NO}_3][\text{B}_5\text{O}_6(\text{OH})_4]\text{H}_2\text{O}$ $\text{C}_4\text{H}_{15}\text{B}_5\text{NO}_{14}$ | | | |





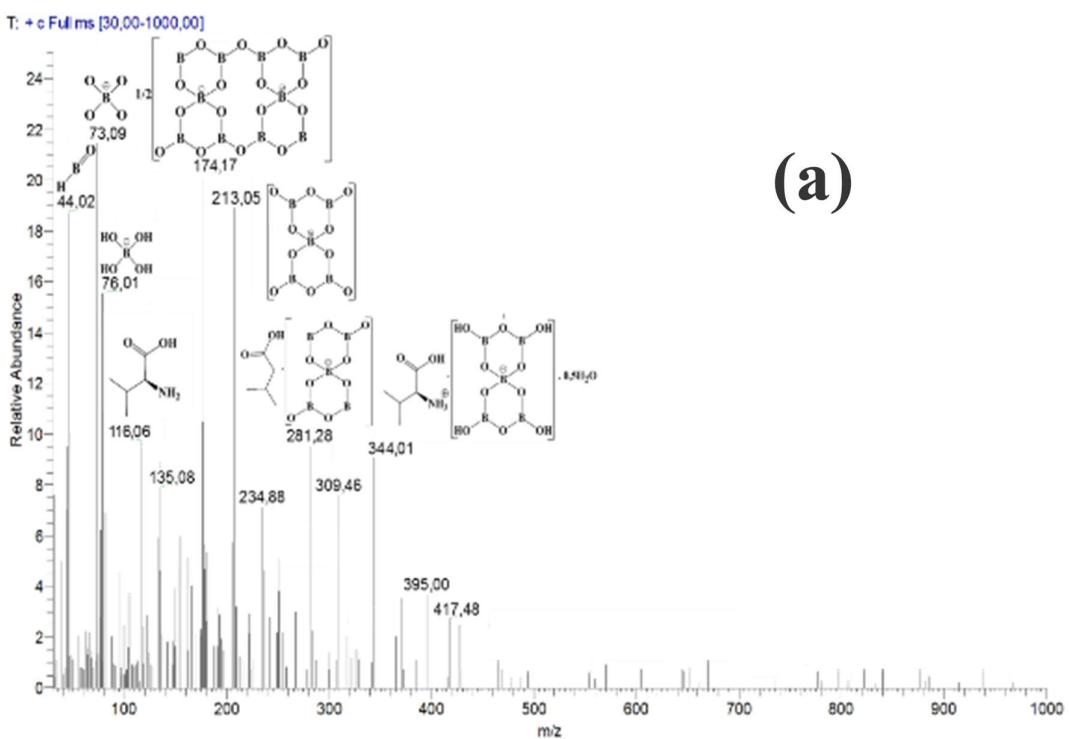
(c)



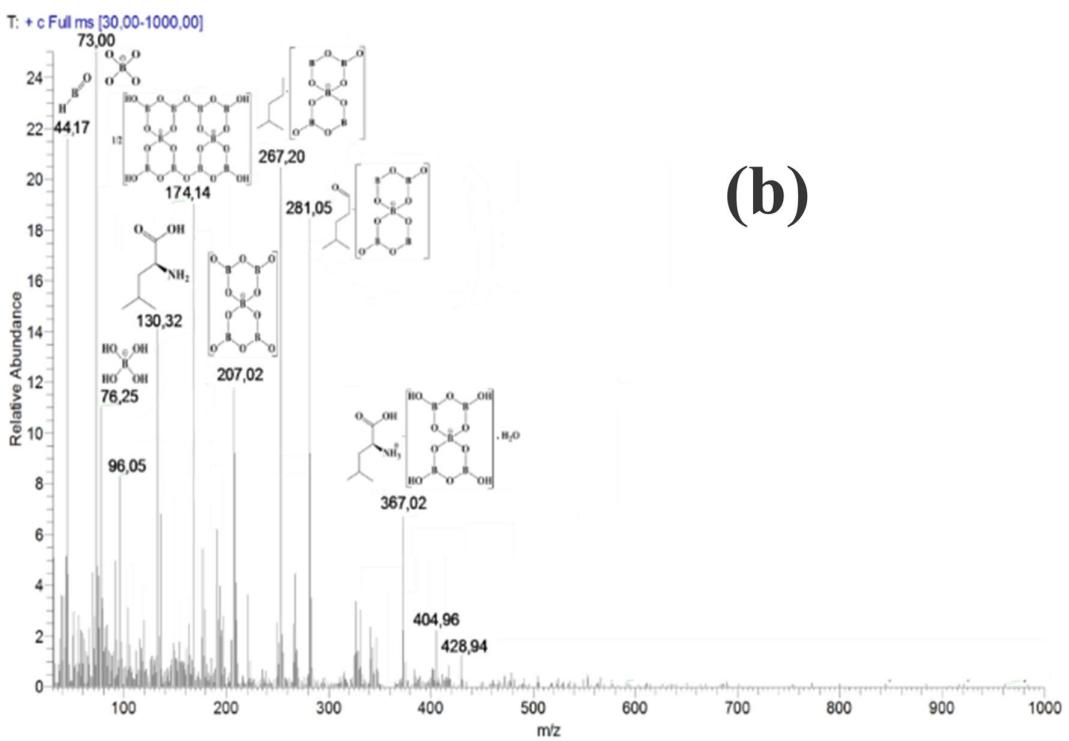
(d)

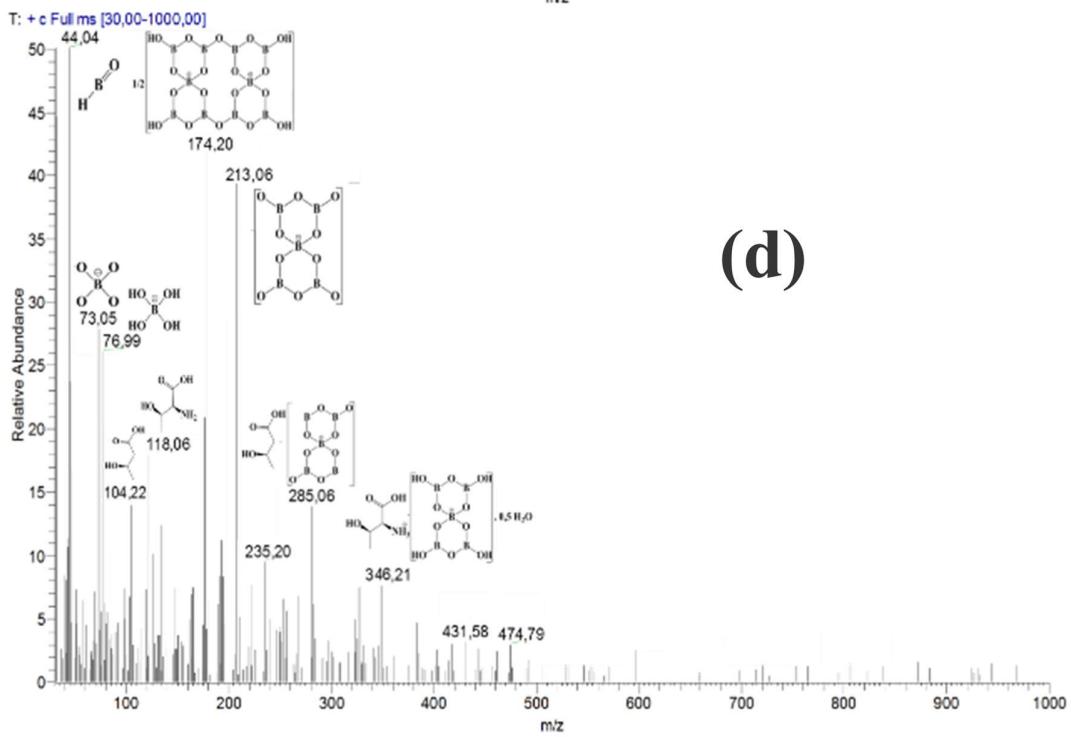
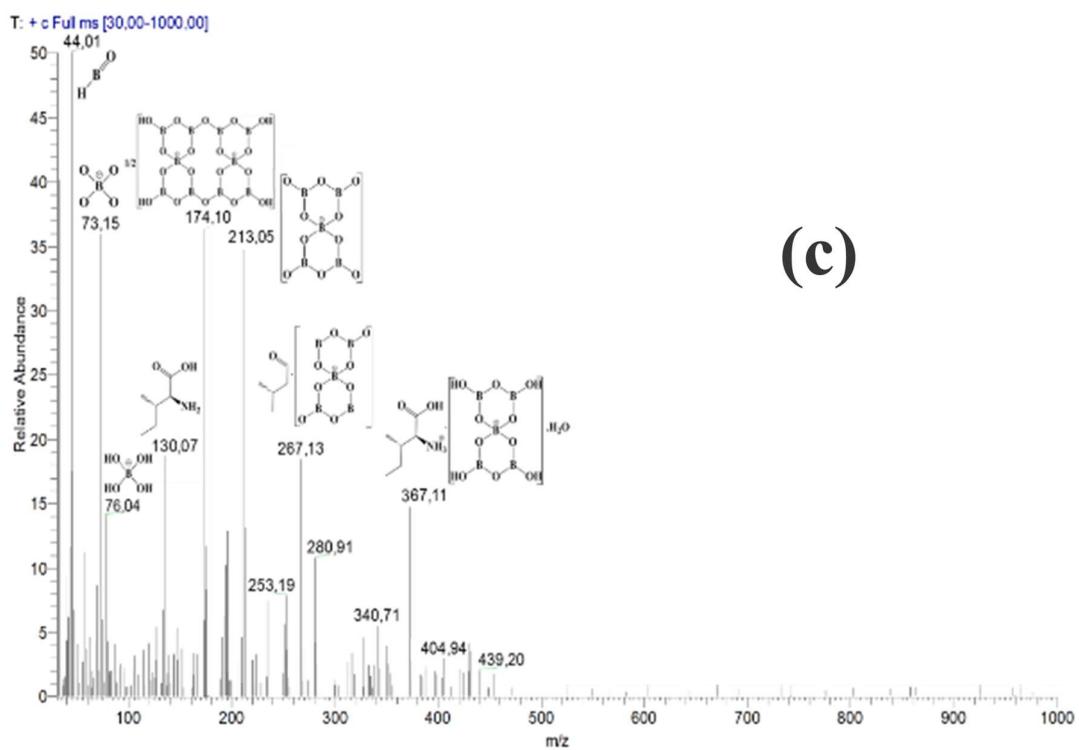
SI Figure 1. Thermal analysis curves of aminoacid pentaborate salts. **(a)** valine pentaborate, **(b)** leucine pentaborate, **(c)** isoleucine pentaborate, **(d)** threonine pentaborate.

(a)



(b)





SI Figure 2. Mass spectroscopy analysis of aminoacid pentaborate salts. **(a)** valine pentaborate, **(b)** leucine pentaborate, **(c)** isoleucine pentaborate, **(d)** threonine pentaborate.