

Supplementary Materials

Catalysis of Organic Pollutants Abatement Based on Pt-Decorated Ag@Cu₂O Heterostructures

Xiaolong Zhang ^{1,2}, Bingbing Han ^{1,3}, Yaxin Wang ^{1,2}, Yang Liu ^{1,2}, Lei Chen ^{1,3,*} and Yongjun Zhang ^{1,2,*}

¹ Key Laboratory of Functional Materials Physics and Chemistry of the Ministry of Education, Jilin Normal University, Changchun 130103, China

² National Demonstration Center for Experimental Physics Education, Jilin Normal University, Siping 136000, China.

³ College of Chemistry, Jilin Normal University, Siping 136000, China

* Correspondence: chenlei@jlnu.edu.cn (L. C.); yjzhang@jlnu.edu.cn (Y.Z.); Tel.: +86-0434-3294566 (L.C.); +86-0434-3294566 (Y.Z.)

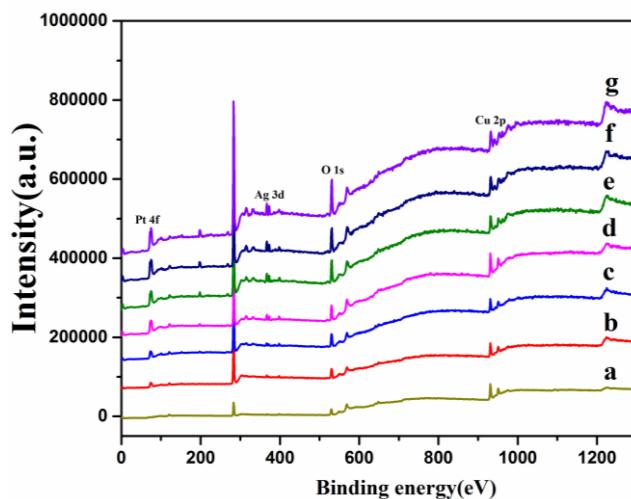


Figure S1. XPS spectra survey spectra of Ag@Cu₂O (a) and Ag@Cu₂O-Pt (b: 0.95×10^{-4} , c: 1.26×10^{-4} , d: 1.43×10^{-4} , e: 1.52×10^{-4} , f: 1.58×10^{-4} , and g: 1.63×10^{-4} mol/L).

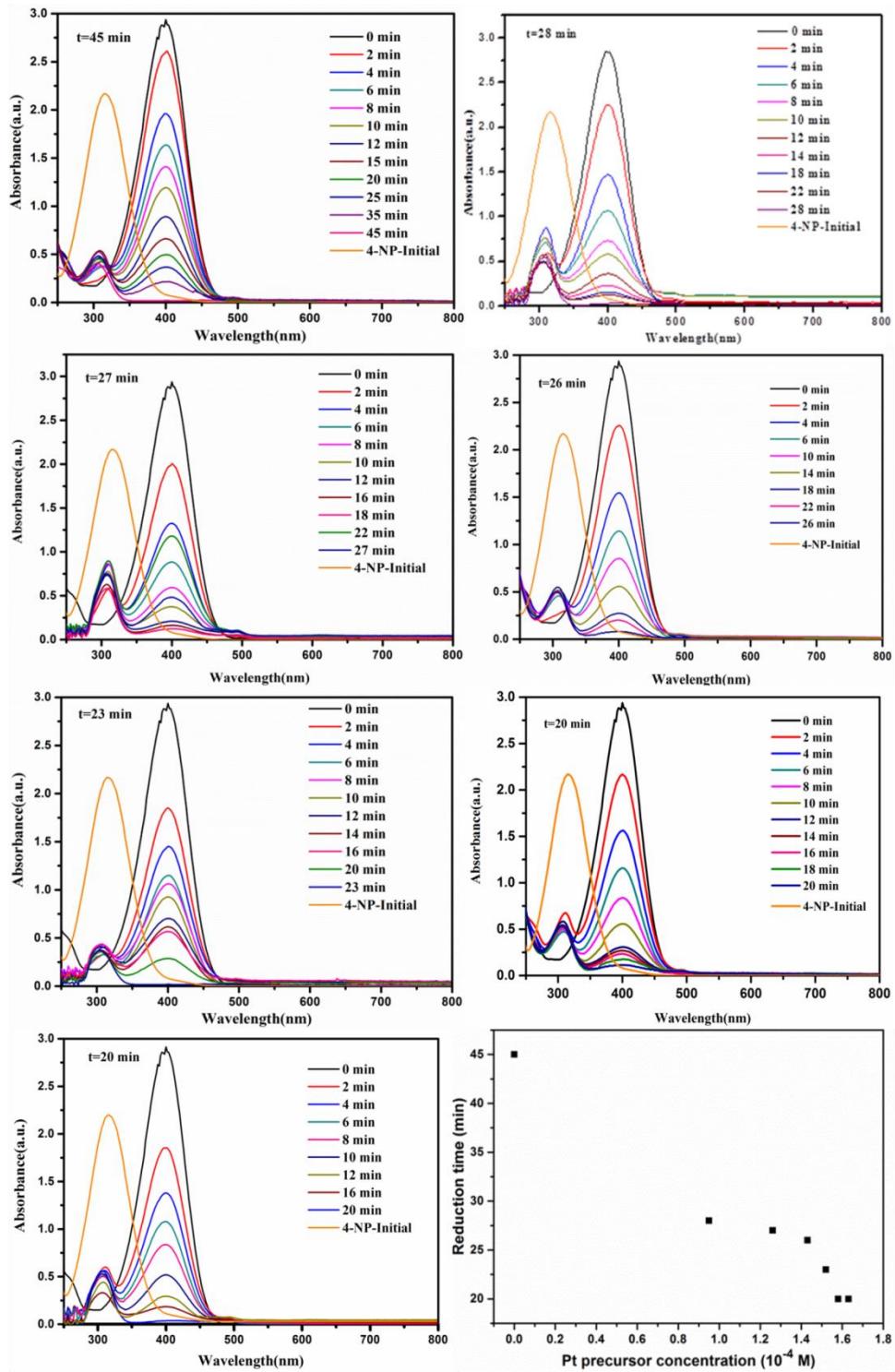


Figure S2. UV-Vis absorbance spectra of the reduction of 4-NP by NaBH₄ in the presence of different Ag@Cu₂O-Pt (the Pt precursor concentrations were 0, 0.95×10^{-4} , 1.26×10^{-4} , 1.43×10^{-4} , 1.52×10^{-4} , 1.58×10^{-4} , and 1.63×10^{-4} mol/L), and Catalytic reduction time vs. different Ag@Cu₂O-Pt (Pt precursor concentrations are 0, 0.95×10^{-4} , 1.26×10^{-4} , 1.43×10^{-4} , 1.52×10^{-4} , 1.58×10^{-4} , and 1.63×10^{-4} mol/L).