# Activity and Thermal Stability of Cobalt(II)-Based Olefin Polymerization Catalysts Adorned with Sterically Hindered Dibenzocycloheptyl Groups 

Muhammad Zada, ${ }^{1,2}$ Liwei Guo, ${ }^{1,3}$ Yanping Ma, ${ }^{1}$ Wenjuan Zhang, ${ }^{3,{ }^{*}}$ Zygmunt Flisak, ${ }^{1,4, *}$ Yang Sun, ${ }^{1}$ Wen-Hua Sun ${ }^{1,2, *}$
1 Key Laboratory of Engineering Plastics and Beijing National Laboratory for Molecular Science, Institute of Chemistry, Chinese Academy of Sciences, Beijing 100190, China. mzada17@iccas.ac.cn (M.Z.); mrliwei_guo@163.com (L.G.); myanping@ iccas.ac.cn (Y.M.); sy0471103 iccas.ac.cn (Y.S.)
${ }^{2}$ CAS Research/Education Center for Excellence in Molecular Sciences, University of Chinese Academy of Sciences, Beijing 100049, China.
3 Beijing Key Laboratory of Clothing Materials R\&D and Assessment, Beijing Engineering Research Center of Textile Nanofiber, School of Materials Science and Engineering, Beijing institute of Fashion Technology, Beijing 100029, China. zhangwj@bift.edu.cn (W.Z.)
4 Faculty of Chemistry, University of Opole, Oleska 48, 45-052 Opole, Poland. zgf@uni.opole.pl (Z.F.)

* Correspondence: zhangwj@bift.edu.cn (W.Z.); zgf@uni.opole.pl (Z.F.); whsun@iccas.ac.cn (W.-H.S.); Tel.: +86-10-6255-7955 (W.-H.S.)
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Figure $\mathbf{S 1}{ }^{1} \mathrm{H}$ NMR spectrum of $\mathbf{C o 1}$ in $\mathrm{CD}_{2} \mathrm{Cl}_{2}$ at room temperature.


Figure $\mathbf{S 2}{ }^{1} \mathrm{H}$ NMR spectrum of $\mathbf{C o 2}$ in $\mathrm{CD}_{2} \mathrm{Cl}_{2}$ at room temperature.


Figure $\mathbf{S 3}{ }^{1} \mathrm{H}$ NMR spectrum of $\mathbf{C o 3}$ in $\mathrm{CD}_{2} \mathrm{Cl}_{2}$ at room temperature.


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Figure S6 GPC curves of the obtained polyethylene (a); activity and $M_{\mathrm{w}}$ as a function of reaction temperature (b) for the Co1/MMAO system (Table 5, entries 1 - 7).


Figure $\mathbf{S 7}$ GPC curves of the obtained polyethylene (a); activity and $M_{\mathrm{w}}$ as a function of Al/Co ratio (b) for the Co1/MMAO system (Table 5, entries 3 and 8 -13).


Figure S8 GPC curves of the obtained polyethylene (a); activity and $M_{\mathrm{w}}$ as a function of run time (b) for the Co1/MMAO system (Table 5, entries 10 and 14 -17).


Figure S9 GPC curves of the obtained polyethylene (a); activity and $M_{\mathrm{w}}$ for different precatalysts (b) at the optimized reaction conditions with MMAO as cocatalyst (Table 6, entries $1-5$ ).


Figure S10 GPC curves of the obtained polyethylene (a); activity and $M_{\mathrm{w}}$ as a function of ethylene pressure (b) at the optimized reaction conditions for the Co1/MMAO system (Table 5, entries 10, 18 and 19).

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\mathrm{H}_{3} \mathrm{C}\left(\mathrm{CH}_{2}\right)_{\mathrm{n}} \mathrm{CH}_{3}
$$



Figure S11 The ${ }^{1} \mathrm{H}$ NMR spectrum of the polyethylene obtained with Co1/MMAO (Table 5, entry 10).


Figure S12 The ${ }^{13}$ C NMR spectrum of the polyethylene obtained with Co1/MMAO (Table 5, entry 10).

Table S1 The selected bond lengths ( $(\AA)$ and angles $\left({ }^{\circ}\right)$ for the $B$ molecules of Co3 and Co4

|  | Co3 | Co4 |
| :--- | :--- | :--- |
| Lengths $(\AA)$ | Molecule B | Molecule $B$ |
| $\mathrm{Co}(1)-\mathrm{N}(1)$ | $2.062(4)$ | $2.059(7)$ |
| $\mathrm{Co}(1)-\mathrm{N}(2)$ | $2.216(4)$ | $2.177(7)$ |
| $\mathrm{Co}(1)-\mathrm{N}(3)$ | $2.230(5)$ | $2.212(7)$ |
| $\mathrm{Co}(1)-\mathrm{Cl}(1)$ | $2.2572(16)$ | $2.265(2)$ |
| $\mathrm{Co}(1)-\mathrm{Cl}(2)$ | $2.2957(16)$ | $2.298(2)$ |
| $\mathrm{N}(2)-\mathrm{C}(10)$ | $1.441(7)$ | $1.437(10)$ |
| $\mathrm{N}(3)-\mathrm{C}(47)$ | $1.448(8)$ | $1.435(12)$ |
| $\mathrm{N}(1)-\mathrm{C}(3)$ | $1.349(7)$ | $1.335(13)$ |
| $\mathrm{N}(1)-\mathrm{C}(7)$ | $1.320(7)$ | $1.317(12)$ |
| $\mathrm{N}(2)-\mathrm{C}(8)$ | $1.288(7)$ | $1.272(10)$ |
| $\mathrm{N}(3)-\mathrm{C}(2)$ | $1.295(8)$ | $1.289(12)$ |
| $\mathrm{Bond} \mathrm{Angles}\left({ }^{\circ}\right)$ |  |  |
| $\mathrm{N}(1)-\mathrm{Co}(1)-\mathrm{N}(2)$ | $73.47(17)$ | $74.40(3)$ |
| $\mathrm{N}(1)-\mathrm{Co}(1)-\mathrm{N}(3)$ | $74.46(18)$ | $74.40(3)$ |
| $\mathrm{N}(2)-\mathrm{Co}(1)-\mathrm{N}(3)$ | $141.15(17)$ | $142.70(3)$ |
| $\mathrm{N}(1)-\mathrm{Co}(1)-\mathrm{Cl}(1)$ | $154.04(13)$ | $150.70(2)$ |
| $\mathrm{N}(2)-\mathrm{Co}(1)-\mathrm{Cl}(1)$ | $99.00(12)$ | $98.34(17)$ |
| $\mathrm{N}(3)-\mathrm{Co}(1)-\mathrm{Cl}(1)$ | $100.26(15)$ | $98.90(2)$ |
| $\mathrm{N}(2)-\mathrm{Co}(1)-\mathrm{Cl}(2)$ | $102.35(12)$ | $100.44(18)$ |
| $\mathrm{N}(3)-\mathrm{Co}(1)-\mathrm{Cl}(2)$ | $101.25(15)$ | $102.23(19)$ |
| $\mathrm{Cl}(1)-\mathrm{Co}(1)-\mathrm{Cl}(2)$ | $111.86(6)$ | $114.16(9)$ |
| $\mathrm{N}(1)-\mathrm{Co}(1)-\mathrm{Cl}(2)$ | $94.08(13)$ | $95.10(2)$ |
| $\mathrm{C}(10)-\mathrm{N}(2)-\mathrm{Co}(1)$ | $124.20(3)$ | $125.40(5)$ |
| $\mathrm{C}(47)-\mathrm{N}(3)-\mathrm{Co}(1)$ | $125.20(4)$ | $123.70(6)$ |

