#### List of changes

We marked our changes in the text by green color.

#### <u>Page 1</u> M1

All the names are correct.

We would like to add affiliations department names for Alexander V. Knyazev (Department of Solid State Chemistry) and Alexey V. Markin (Department of Physical Chemistry). We corrected some abbreviations of author's names.

# M2

We would like to keep e-mail <u>melnikovanb@gmail.com</u>

# M3

Bold is not necessary in this case. We removed it in whole paper.

## **M4**

DSC is differential scanning calorimetry.

TG is thermogravimetric analysis.

# Page 2

M5

We defined DSC/TG earlier.

#### <u>Page 3</u> M6

We would like to move the Figure 10 to Supplementary Materials as Figure S1.

## Page 5

**M7** 

We would prefer not to insert this space.

## <u>Page 7</u>

**M8** 

"n" should not be italic.

# M9

We confirm the correction.

# Page 9

M10

We confirm the correction.

## **3.1 Materials**

Betulin was isolated from Betula Pendula bark using the methods in [25. Kuznetsova, S.A.; Skvortsova, G.P.; Maliar, I.N.; Skurydina, E.S.; Veselova, O.F. Extraction of betulin from birch bark and study of its physico-chemical and pharmacological properties. Russ. J. Bioorganic Chem. 2014, 40, 742–747]. Phosphorus oxytrichloride (Sigma Aldrich, Moscow, Russia), purified water (resistivity  $\geq 18 \text{ M}\Omega \cdot \text{cm}$ , Millipore, Merck, Darmstadt, Germany).

## M11

Shimadzu, Kyoto, Japan

# M12

Jeol Ltd., Tokyo, Japan

# M13

Netzsch-Gerätebau, Selb, Germany

# M14

We corrected the order of references.

# M15

We would prefer not to insert this space.

# M16

Netzsch Proteus Software (Selb, Germany)

# Page 10

M17

Phosphorus oxytrichloride (7.56 mL, 81.6 mmol; Sigma Aldrich, Moscow, Russia, 99%)

# M18

We confirm the correction.

#### <u>Page 11</u>

## M19

Italic is not necessary.

## M20

Ethical code number is correct.

## <u>Page 12</u>

We would like to change "Na-BDP $\cdot$ 8H<sub>2</sub>O" to "Na-BDP  $\times$  8H<sub>2</sub>O".

# M21

The following are available online at www.mdpi.com/xxx/s1, Figure S1: <sup>31</sup>P-NMR spectrum of BDP (DMSO-d<sub>6</sub>, standard Ph<sub>3</sub>P), Figure S2: <sup>31</sup>P-NMR spectra of BDP, DMSO-d<sub>6</sub>. a) BDP-1 sample, standard Met<sub>3</sub>P; b) BDP-1 in the presence of H<sub>3</sub>PO<sub>4</sub>, standard Ph<sub>3</sub>P ( $\delta$  = - 5.93 ppm) and H<sub>3</sub>PO<sub>4</sub> ( $\delta$  = -0.4 ppm); c) BDP-2 sample, standard Met<sub>3</sub>P, Figure S3: <sup>1</sup>H-NMR spectrum of BDP. DMSO-d<sub>6</sub>, standard TMS, 400 MHz, Figure S4: <sup>31</sup>P-NMR spectrum of Na-BDP. D<sub>2</sub>O, standard H<sub>3</sub>PO<sub>4</sub>.

We don't know the correct link of our Supplementary Materials. Please, add this link, if you don't mind.

## M22

We confirm the correction.

## M23

Funding: This research received no external funding.

## M24

We would like to remove section Acknowledgments, if it is possible.

# Page 13

M25 Patent RU 2,243,233, 27 December 2004.

**M26** Patent US 3,764,616, 9 October 1973.