

Supporting information for

Water-soluble Manganese and Iron *meso*-tetrakis(carboxyl)porphyrin: DNA Binding, Oxidative Cleavage, and Cytotoxic Activities

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1. Characterization of compounds

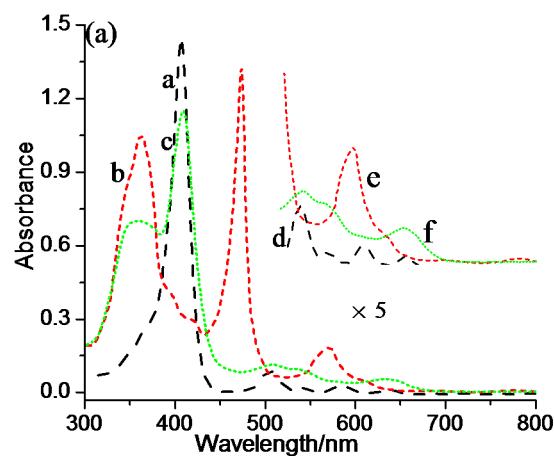


Figure S1. UV-Vis absorption spectra of **1**, **1-Mn** and **1-Fe** in dichloromethane.

a: **1**-Soret band; **b:** **1-Mn**-Soret band; **c:** **1-Fe**-Soret band; **d:** **1**-Q band; **e:** **1-Mn**-Q band; **f:** **1-Fe**-Q band

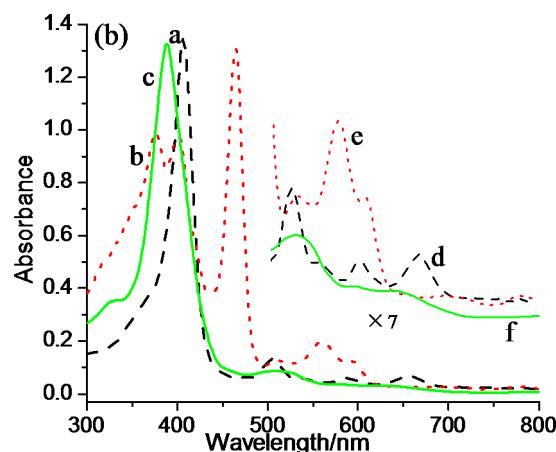


Figure S2. UV-Vis absorption spectra of **2**, **2-Mn** and **2-Fe** in 5 mM Tris-HCl/50 mM NaCl buffer.

a: **2**-Soret band; **b:** **2-Mn**-Soret band; **c:** **2-Fe**-Soret band; **d:** **2**-Q band; **e:** **2-Mn**-Q band; **f:** **2-Fe**-Q band

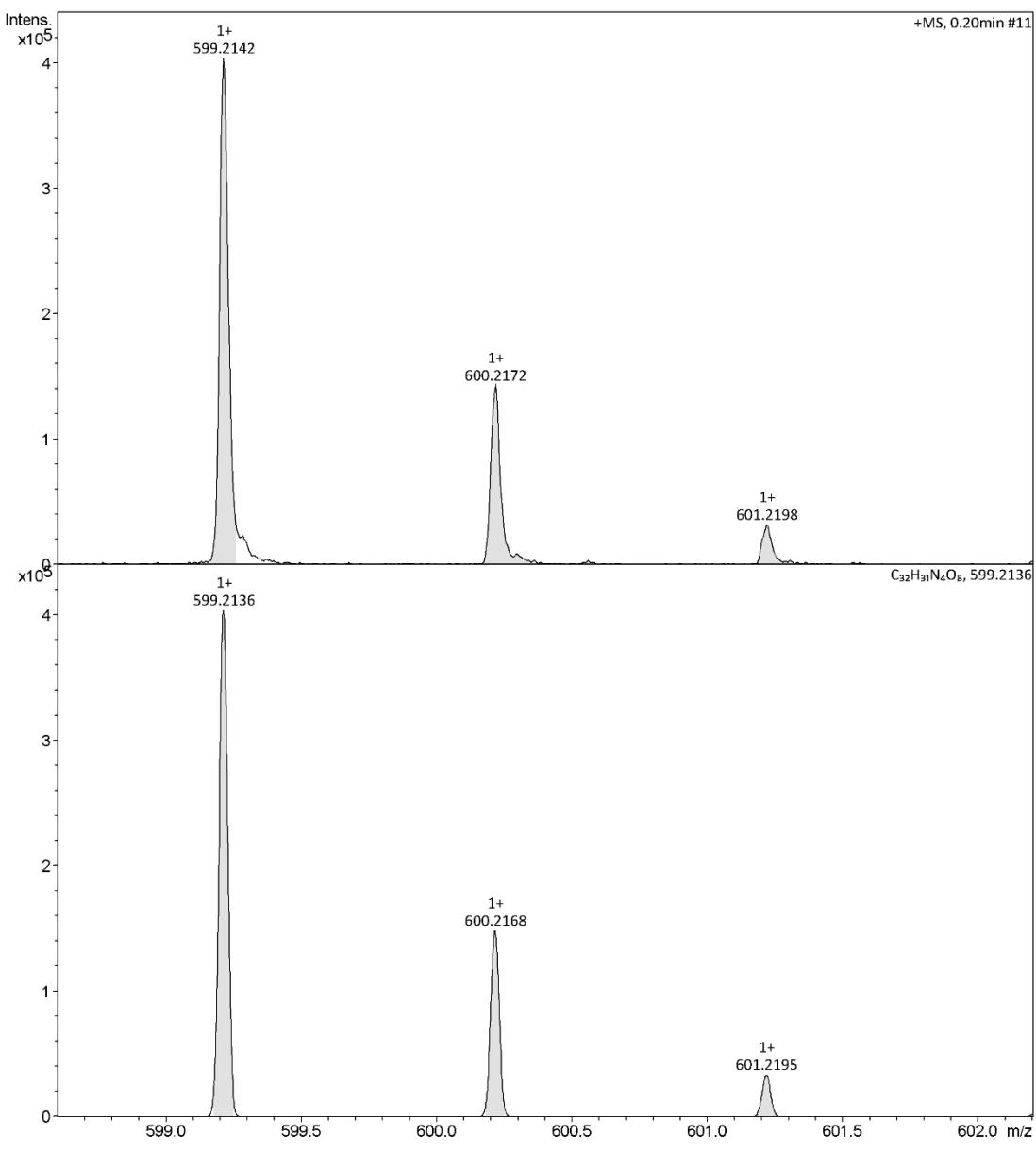


Figure S3. HR-MS of 5, 10, 15, 20-tetrakis(ethoxycarbonyl)porphyrin (**1**)

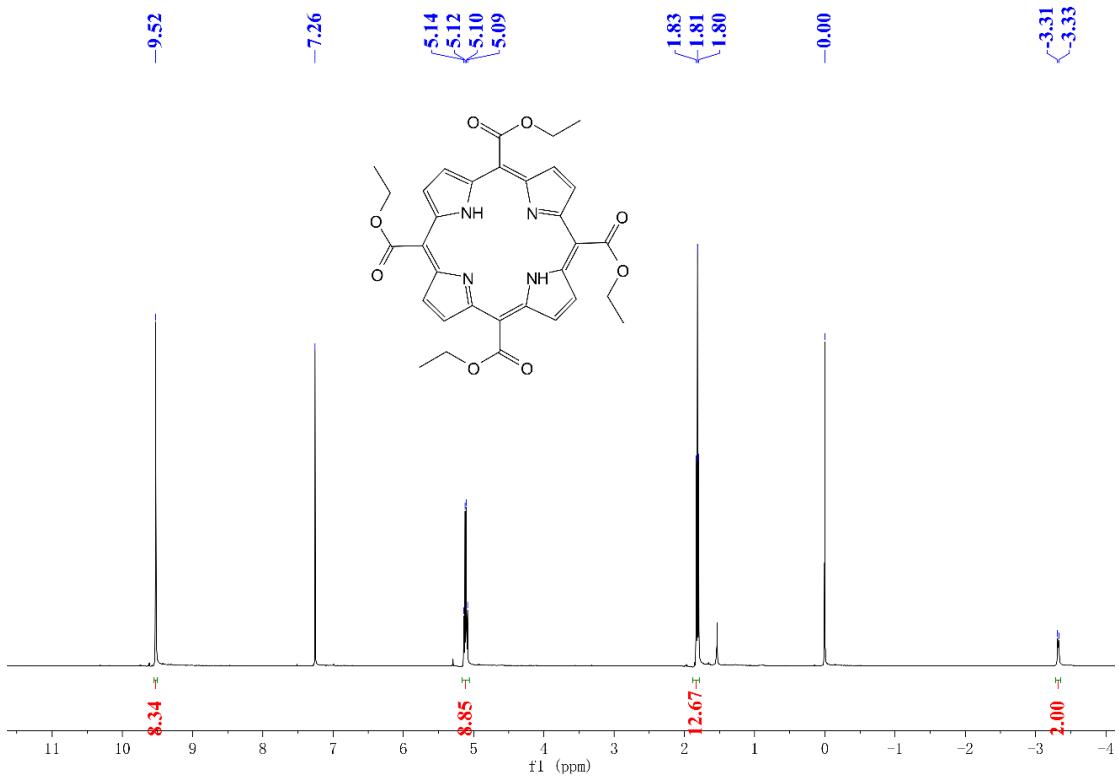


Figure S4. ¹H NMR of 5, 10, 15, 20-tetrakis(ethoxycarbonyl)porphyrin (**1**).

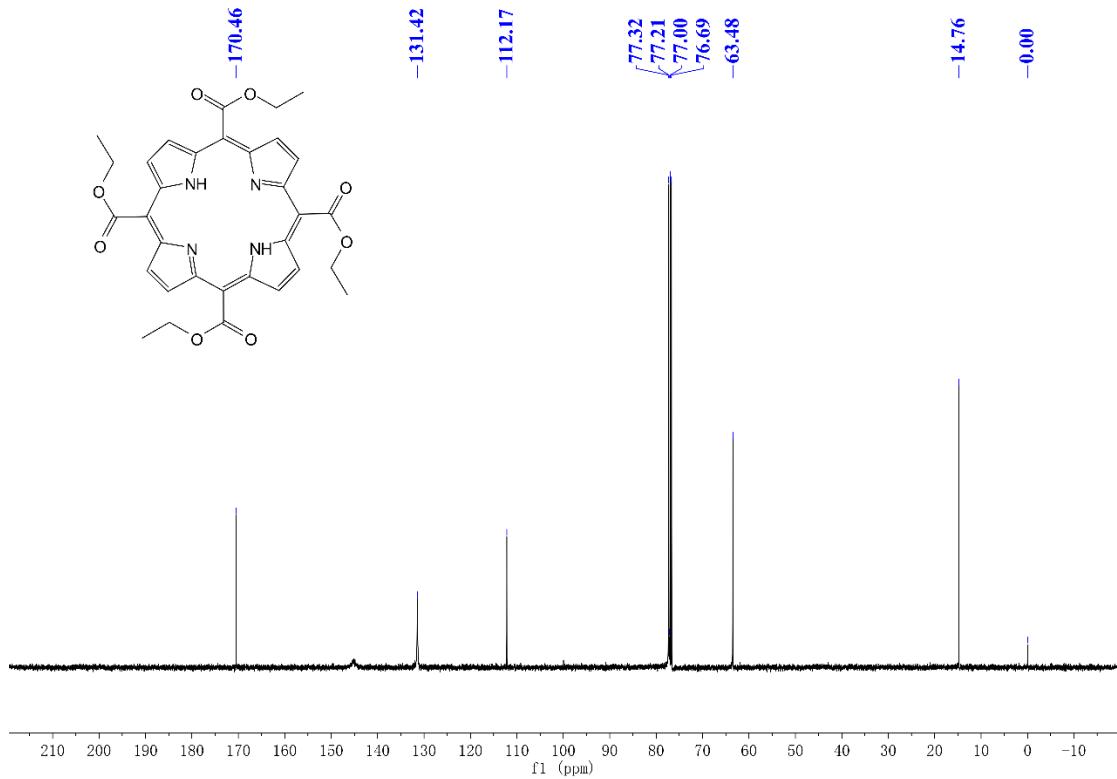


Figure S5. ¹³C NMR of 5, 10, 15, 20-tetrakis(ethoxycarbonyl)porphyrin (**1**).

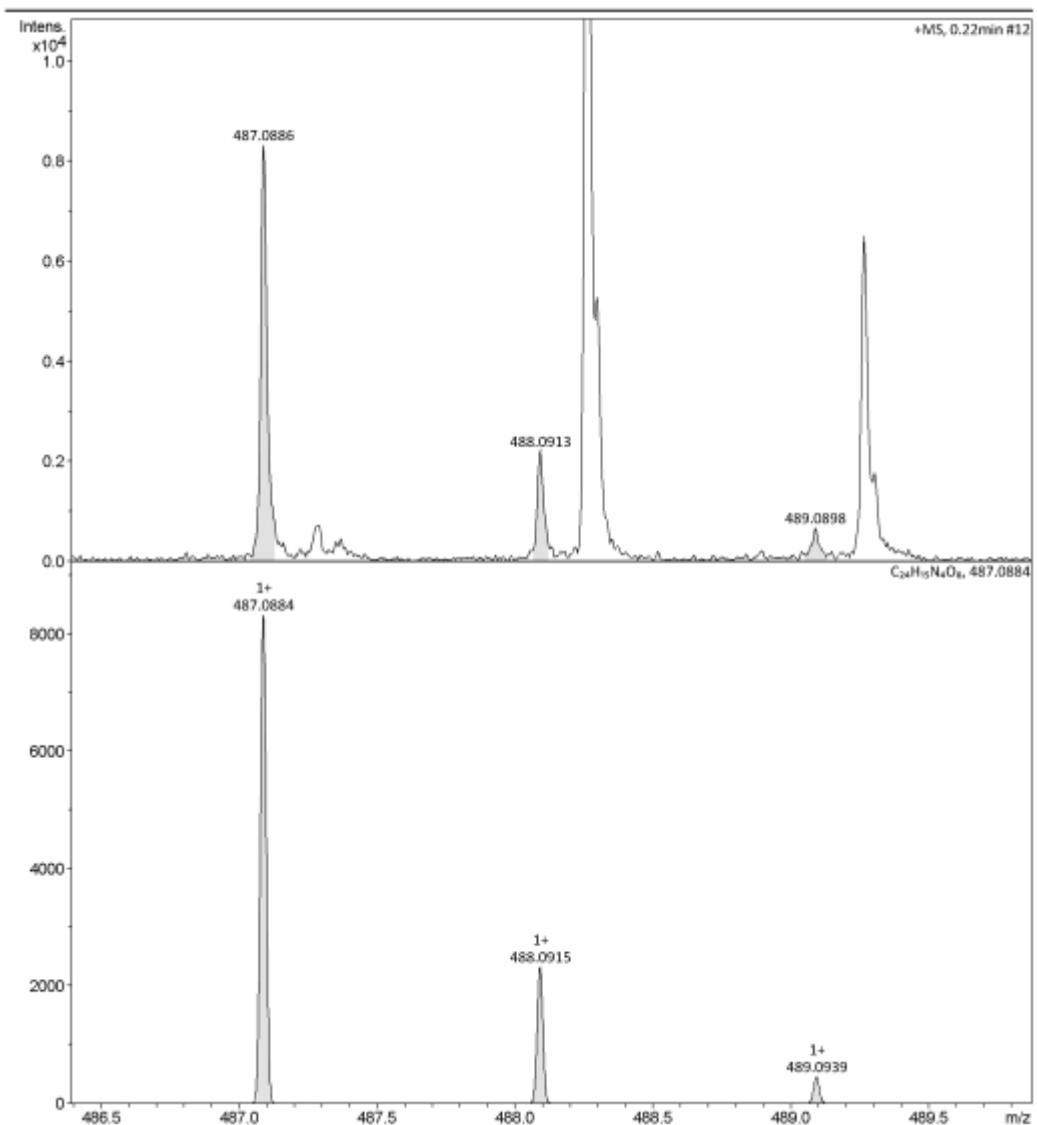


Figure S6. HR-MS of 5, 10, 15, 20-tetrakis (carboxyl) porphyrin (**2**)

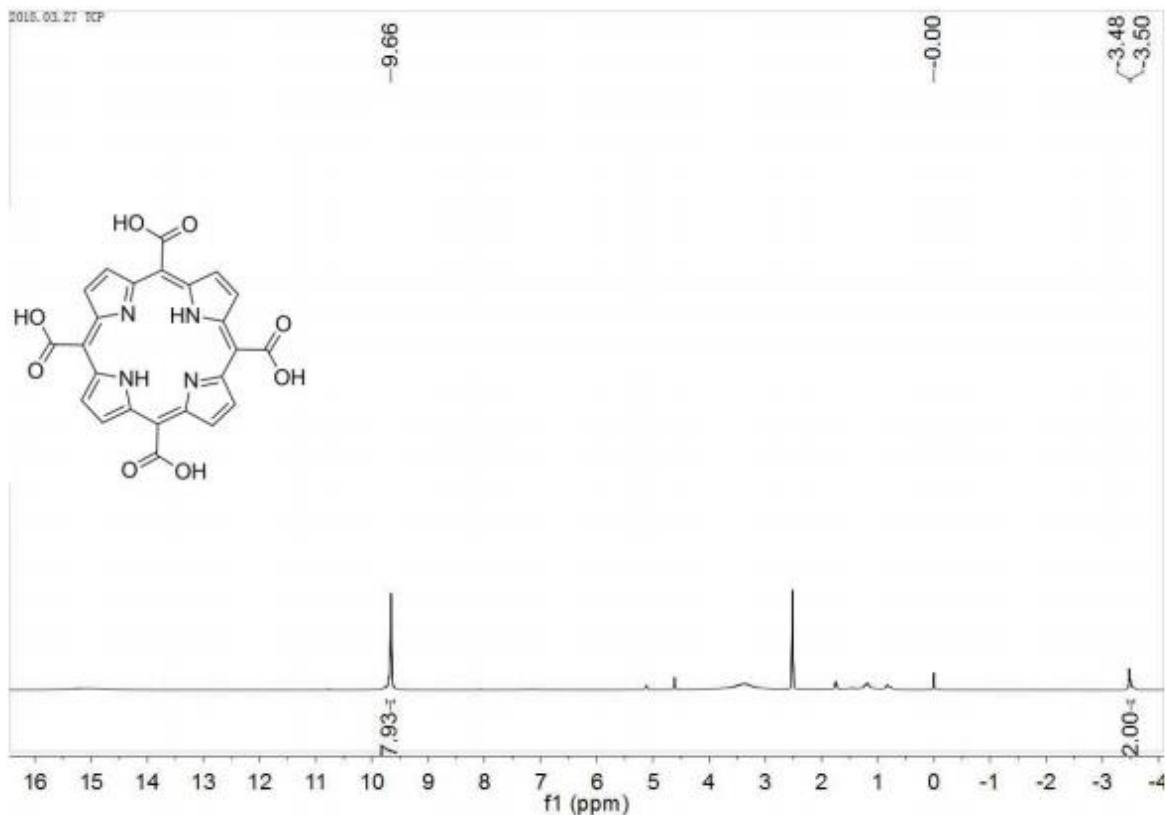


Figure S7. ^1H NMR of 5, 10, 15, 20- tetrakis (carboxyl) porphyrin (**2**)

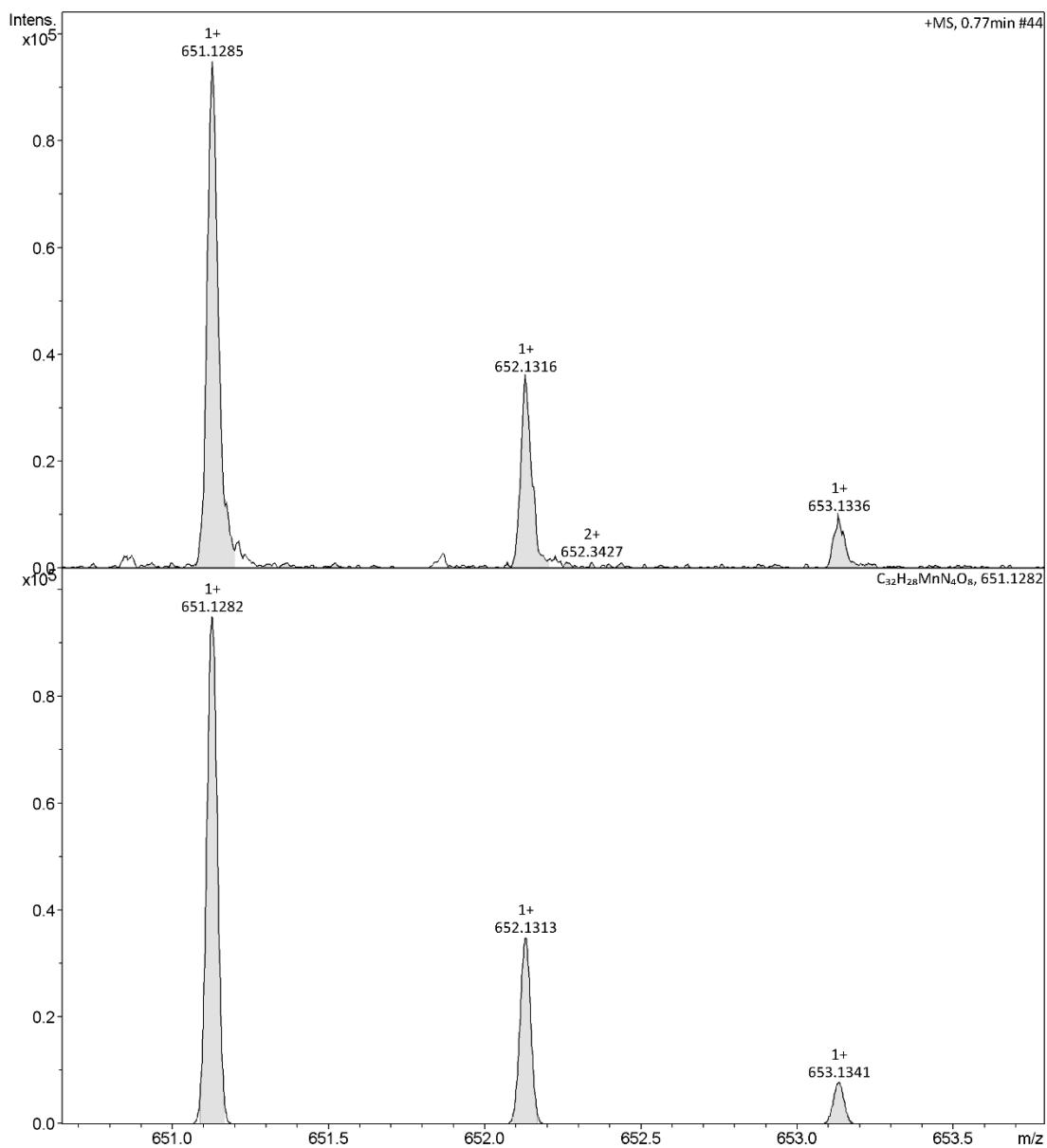


Figure S8. HR-MS of 5, 10, 15, 20-tetrakis(ethoxycarbonyl)porphyrin manganese (III) (**1-Mn**)

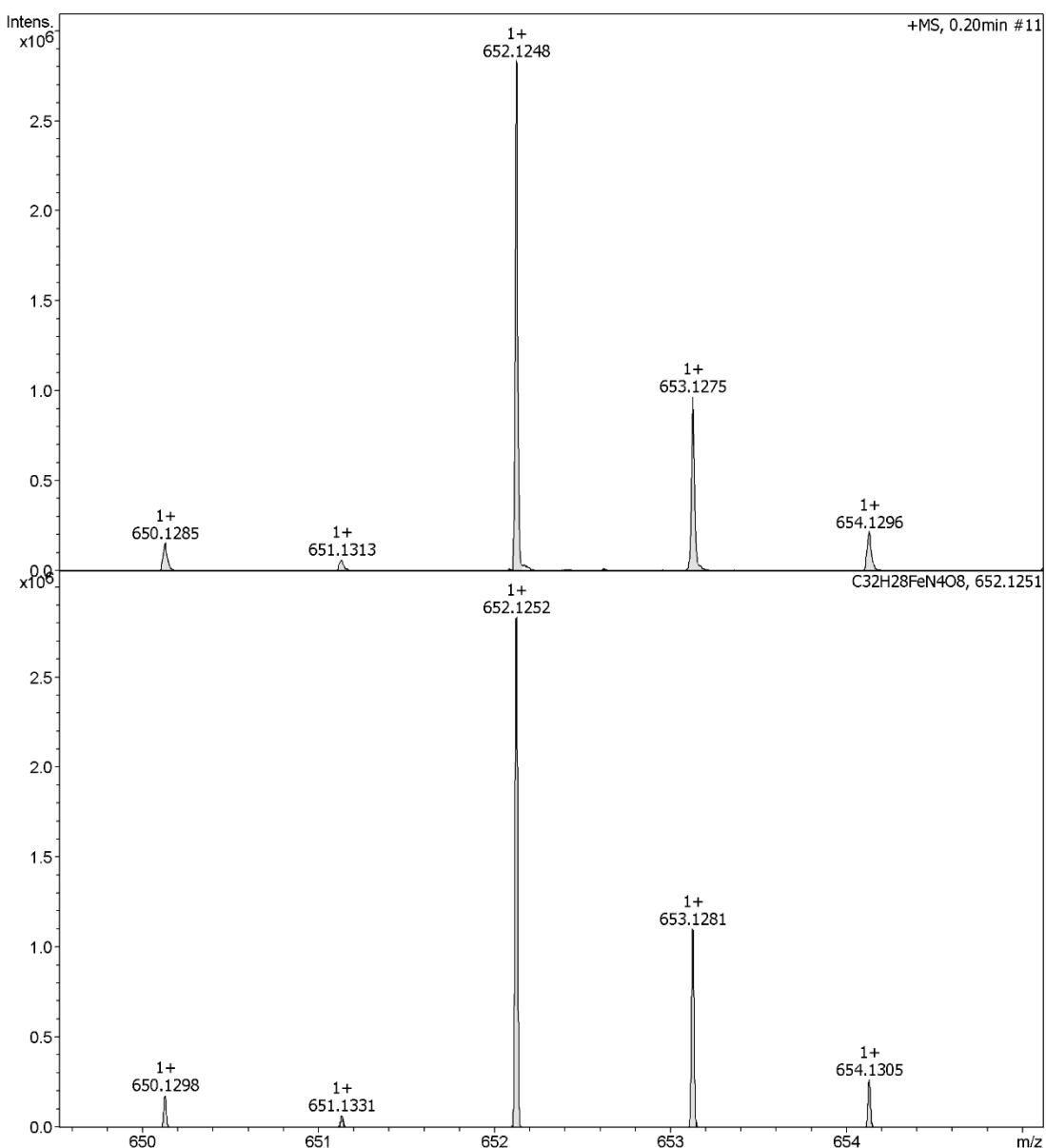


Figure S9. HR-MS of 5, 10, 15, 20-tetrakis(ethoxycarbonyl)porphyrin iron (III) (**1-Fe**)

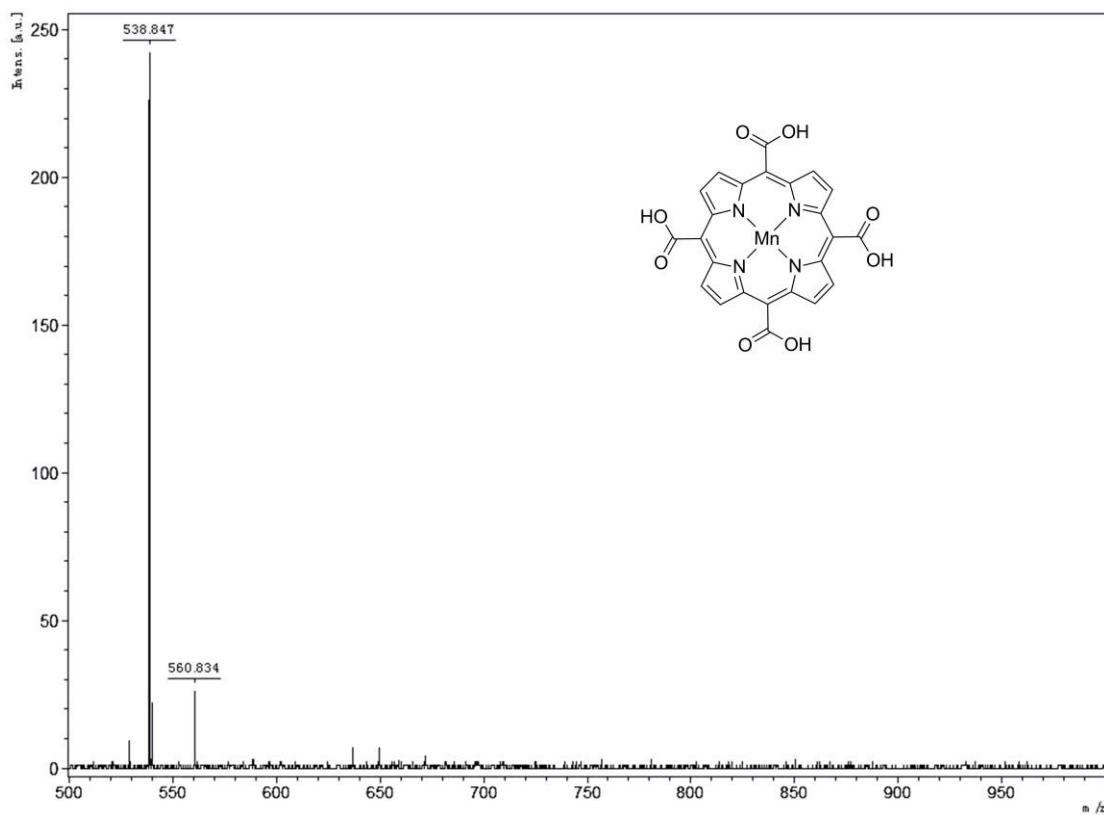


Figure S10. MALDI-TOF MS of 5, 10, 15, 20-tetrakis(carboxyl)porphyrin manganese (III) (**2-Mn**)

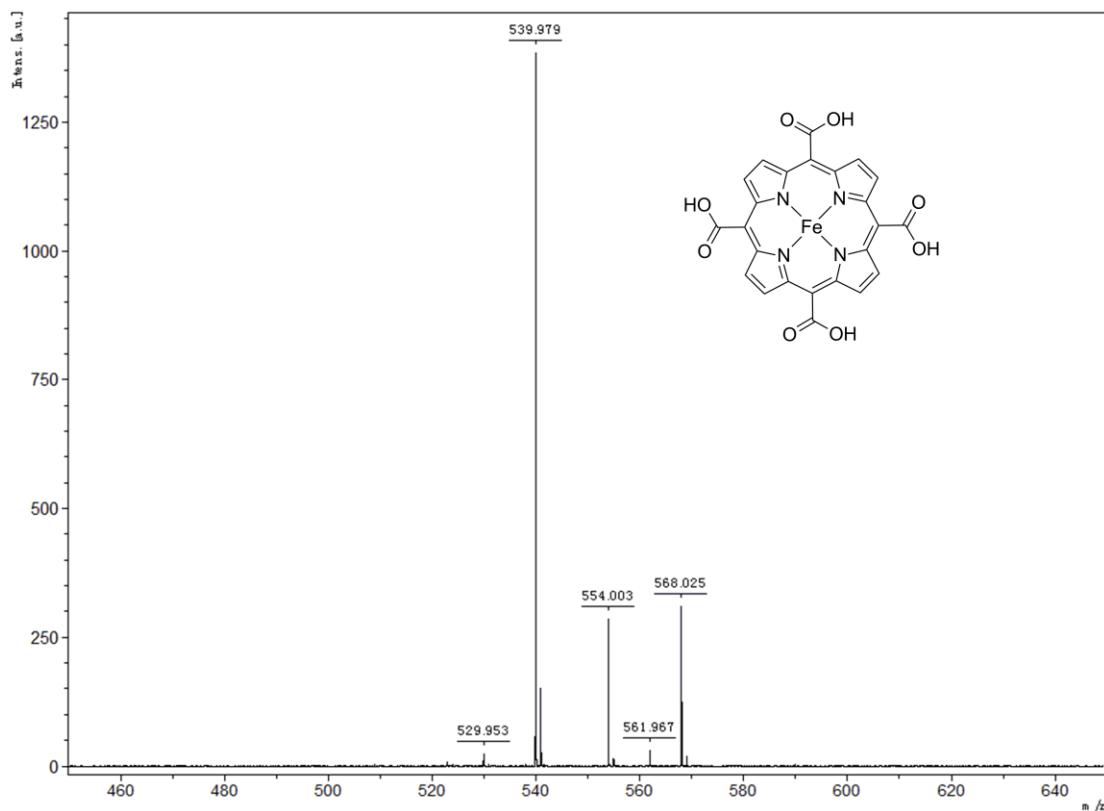
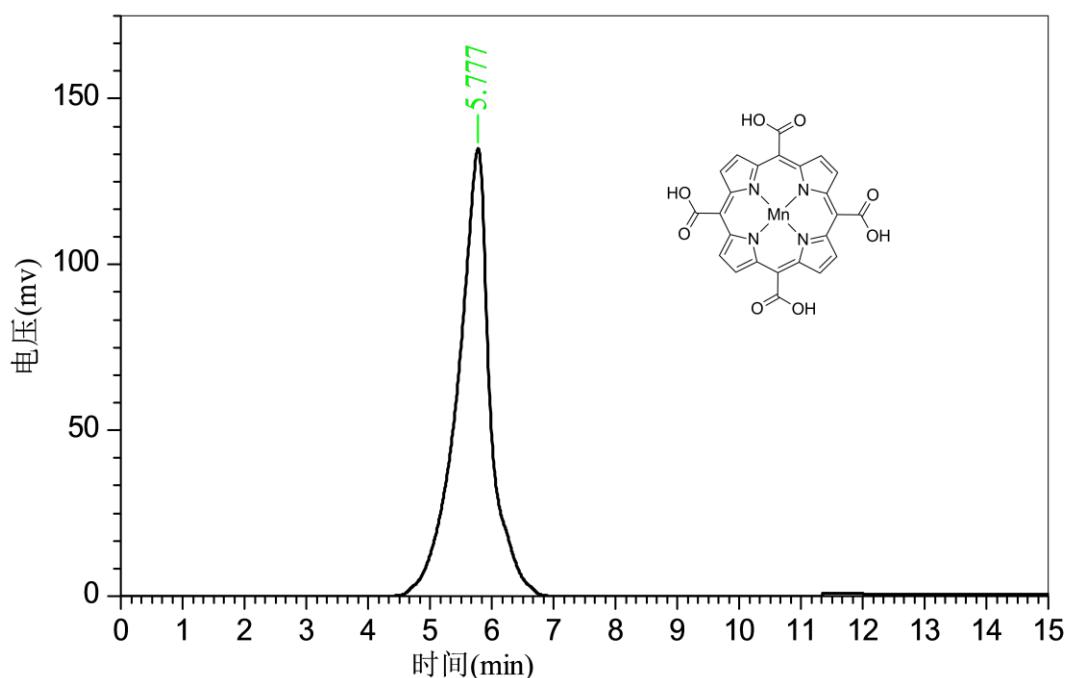
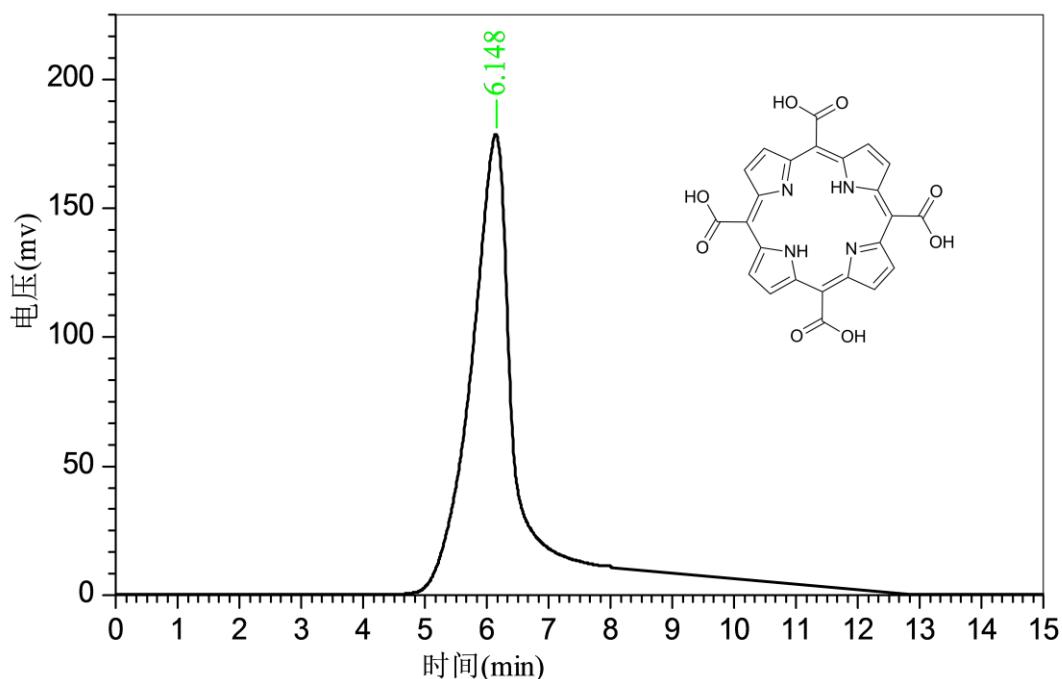


Figure S11. MALDI-TOF MS of 5, 10, 15, 20-tetrakis(carboxyl)porphyriniron (III) (**2-Fe**)

. HPLC trace of porphyrins 2, 2-Mn and 2-Fe

Reversed-phase HPLC analysis: The samples were analyzed on AT. ChromC¹⁸ column (4.6×250mm).



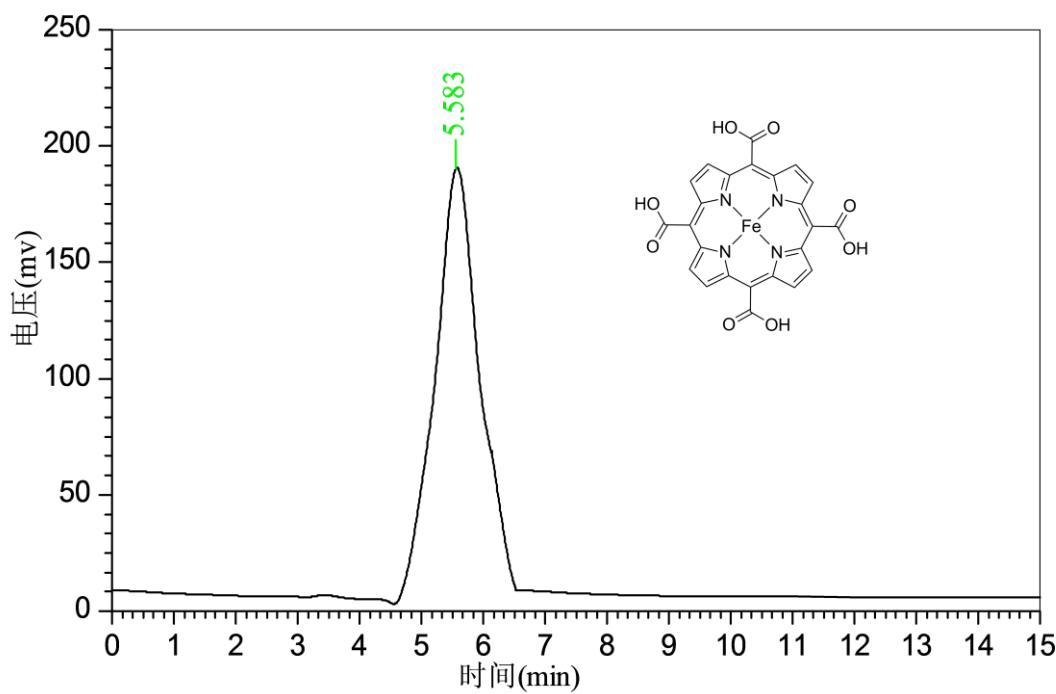


Figure S12. Porphyrin **2**, **2-Mn**, **2-Fe**, detection at 400 nm, $\text{CH}_3\text{OH} : \text{H}_2\text{O}$ 95 : 5, 0.3 mL/min, retention time is 6.148, 5.777, 5.583min, respectively.

3. The UV-Vis spectra changes of **2-Mn** (a) and **2-Fe** (b) upon the addition H_2O_2

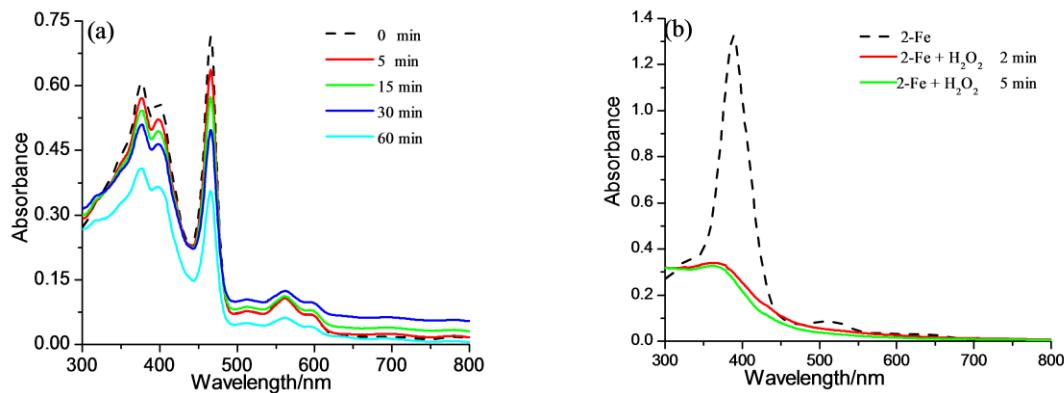


Figure S13. The UV-Vis spectra changes of **2-Mn** (a) and **2-Fe** (b) upon the addition H_2O_2 at different time. The arrow shows the absorbance along with the change of time. $[\text{2-Mn}] = [\text{2-Fe}] = 30 \mu\text{M}$, $[\text{H}_2\text{O}_2] = 20 \text{ mM}$.