

**Supporting Information**

**Synthesis, Characterization and Protonation behaviors of  
Quinoxaline-Fused Porphycenes**

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## Supporting information

Table 1. Crystal data and structure refinement for **2a**.

Empirical formula	C36 H40 N4 S2
Formula weight	592.84
Temperature	90 K
Wavelength	0.71073 Å
Crystal system	Monoclinic
Space group	<i>C</i> 2/ <i>c</i>
Unit cell dimensions	<i>a</i> = 37.389(4) Å <i>b</i> = 5.1476(5) Å $\beta$ = 93.283(2) $^\circ$ <i>c</i> = 32.260(3) Å
Volume	6198.7(11) Å <sup>3</sup>
<i>Z</i>	8
Density (calculated)	1.270 g/cm <sup>3</sup>
Absorption coefficient	0.204 mm <sup>-1</sup>
<i>F</i> (000)	2528
Crystal size	0.10 x 0.05 x 0.05 mm <sup>3</sup>
Theta range for data collection	1.62 to 26.92 $^\circ$
Index ranges	-47 $\leq$ <i>h</i> $\leq$ 45, -4 $\leq$ <i>k</i> $\leq$ 6, -31 $\leq$ <i>l</i> $\leq$ 41
Reflections collected	18346
Independent reflections	6704 [ <i>R</i> (int) = 0.0495]
Completeness to theta = 26.92 $^\circ$	99.7%
Absorption correction	Empirical
Max. and min. transmission	0.9899 and 0.9799
Refinement method	Full-matrix least-squares on <i>F</i> <sup>2</sup>
Data / restraints / parameters	6704 / 0 / 410
Goodness-of-fit on <i>F</i> <sup>2</sup>	1.009
Final R indices [ <i>I</i> > 2 <i>σ</i> ( <i>I</i> )]	<i>R</i> <sub>1</sub> = 0.0428, <i>wR</i> <sub>2</sub> = 0.0919
<i>R</i> indices (all data)	<i>R</i> <sub>1</sub> = 0.0739, <i>wR</i> <sub>2</sub> = 0.1088
Largest diff. peak and hole	0.439 and -0.267 e.Å <sup>-3</sup>
CCDC No.	1543212

Table 2. Crystal data and structure refinement for **2b**.

Empirical formula	C32 H28 N4 O4	
Formula weight	532.58	
Temperature	90 K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	<i>P</i> 2 <sub>1</sub> /c	
Unit cell dimensions	<i>a</i> = 11.750(2) Å	
	<i>b</i> = 10.982(2) Å	<i>β</i> = 104.032(4)°
	<i>c</i> = 21.089(5) Å	
Volume	2640.3(9) Å <sup>3</sup>	
<i>Z</i>	4	
Density (calculated)	1.340 g/cm <sup>3</sup>	
Absorption coefficient	0.090 mm <sup>-1</sup>	
<i>F</i> (000)	1120	
Crystal size	0.30 x 0.30 x 0.02 mm <sup>3</sup>	
Theta range for data collection	1.79 to 24.00°	
Index ranges	-13 ≤ <i>h</i> ≤ 11, -12 ≤ <i>k</i> ≤ 12, -14 ≤ <i>l</i> ≤ 24	
Reflections collected	12564	
Independent reflections	4160 [ <i>R</i> (int) = 0.0943]	
Completeness to theta = 24.00°	100.0%	
Absorption correction	Empirical	
Max. and min. transmission	0.9982 and 0.9735	
Refinement method	Full-matrix least-squares on <i>F</i> <sup>2</sup>	
Data / restraints / parameters	4160 / 0 / 365	
Goodness-of-fit on <i>F</i> <sup>2</sup>	1.095	
Final <i>R</i> indices [ <i>I</i> > 2σ( <i>I</i> )]	<i>R</i> <sub>1</sub> = 0.0727, <i>wR</i> <sub>2</sub> = 0.1849	
<i>R</i> indices (all data)	<i>R</i> <sub>1</sub> = 0.1405, <i>wR</i> <sub>2</sub> = 0.2184	
Largest diff. peak and hole	1.470 and -0.276 e.Å <sup>-3</sup>	
CCDC No.	1543211	

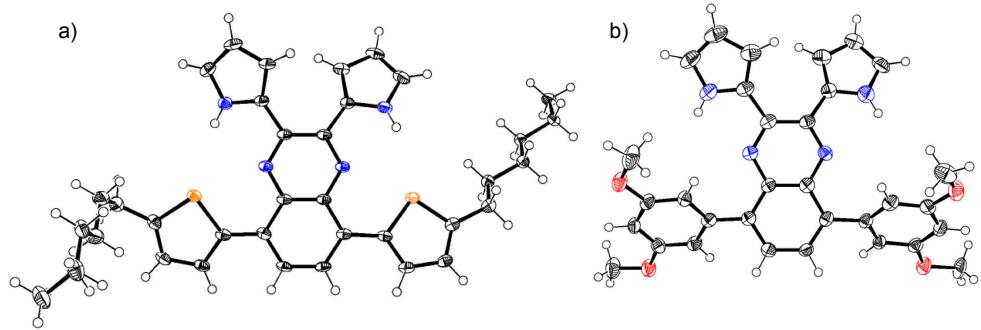


Figure S1. Crystal structures of **2a** and **2b**. Thermal ellipsoids represent for 50% probability.

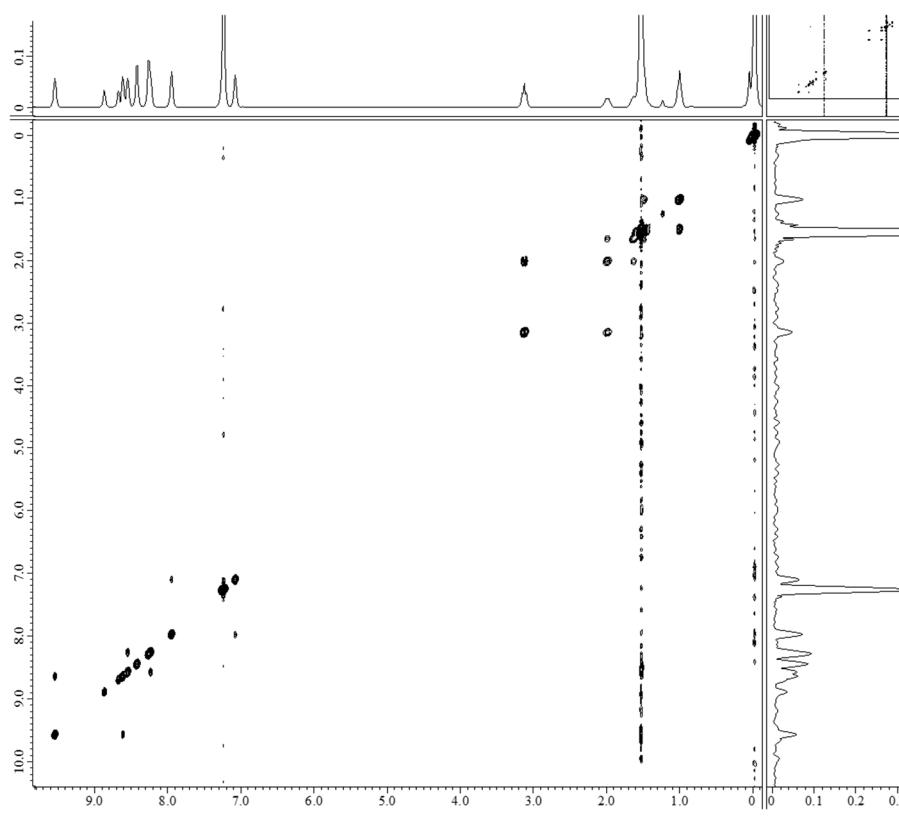


Figure S2. H-H-COSY spectrum of **1a**-H<sub>2</sub> in CDCl<sub>3</sub>.

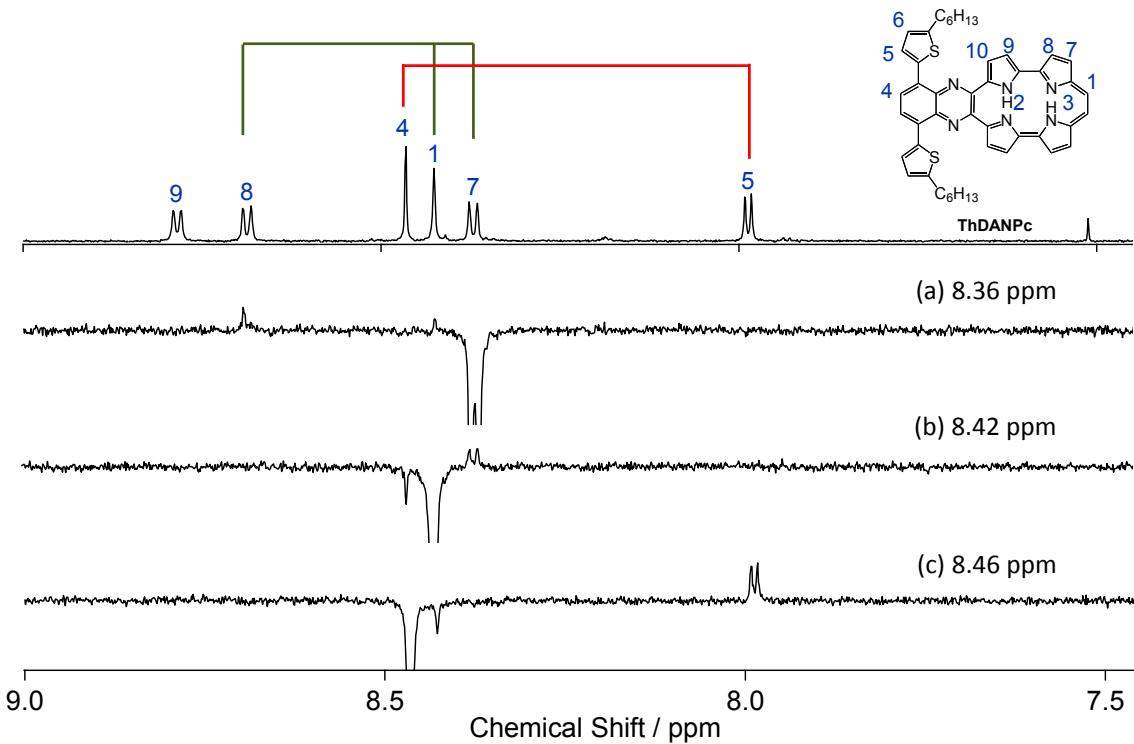


Figure S3. NOE spectra of **1a**-H<sub>2</sub> in CDCl<sub>3</sub>.

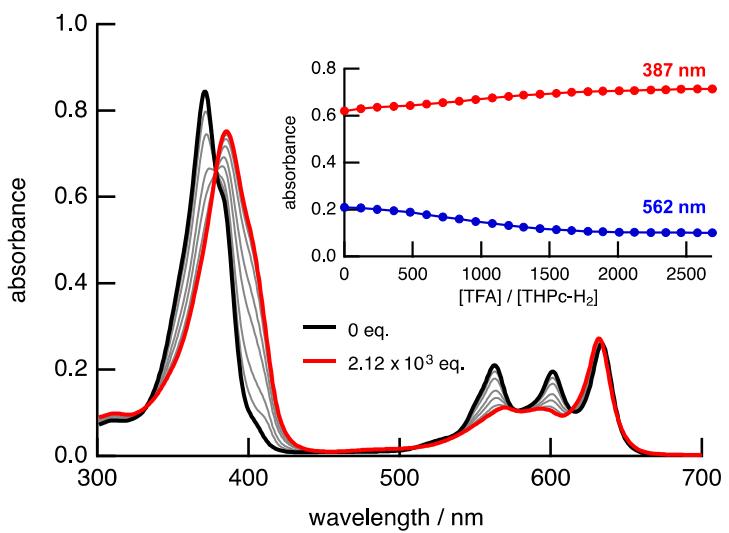


Figure S4. Changing of the absorption spectra of **THPc-H<sub>2</sub>** upon addition of the TFA.

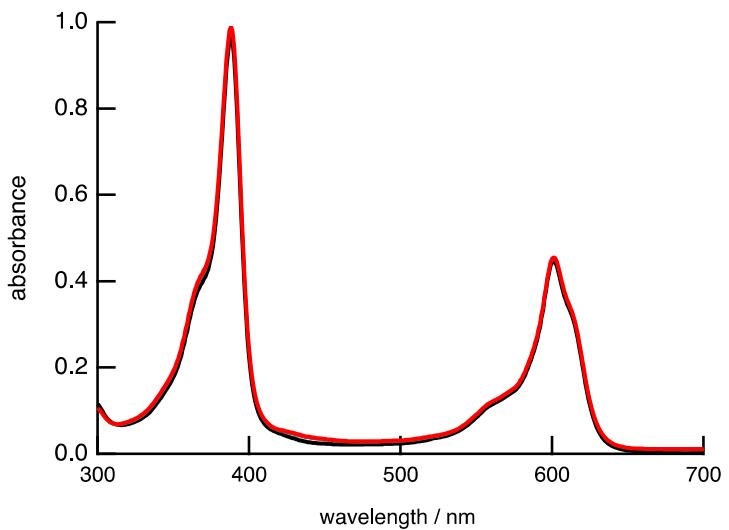


Figure S5. Changing of the absorption spectra of **THPc-Ni** upon addition of the TFA.