## SUPPLEMENTARY MATERIAL

## DEGRADATION OF METHYL 2-AMINOBENZOATE (METHYL ANTHRANILATE) BY $H_2O_2/UV$ : EFFECT OF INORGANIC ANIONS AND DERIVED RADICALS

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## Direct photolysis quantum yield of methyl anthranilate (MA)

The value of the direct photolysis quantum yield ( $\Phi_{MA}$ ) was measured upon monochromatic and polychromatic irradiation of 0.2 mM MA solutions placed inside spectrophotometer cuvettes. The initial MA concentration was chosen to obtain absorbance values in the range of 0.5 to 0.7 at the studied wavelengths. Monochromatic radiation at 254 and 325 nm was obtained by a suitable combination of Horiba xenon lamps and diffraction-grid monochromators, and irradiation was carried out till 15% disappearance of the initial MA. Polychromatic irradiation at  $\lambda > 220$  nm was obtained by using an Atlas Suntest® apparatus (unfiltered xenon lamp emission). The spectral photon flux density emitted by the radiation sources was measured with a calibrated Ocean Optics USB 2000 CCD camera.

The photolysis quantum yield  $\Phi_{MA}$  was obtained as follows:

$$\Phi_{MA} = \frac{R_{MA}}{2.303b[MA] \int_{\lambda} p^{\circ}(\lambda) \varepsilon_{MA}(\lambda) d\lambda}$$
(SM1)

where  $R_{\text{MA}}$  is the initial photodegradation rate of MA (initial concentration [MA] = 0.2 mM), b = 1 cm the optical path length of the cuvette,  $\varepsilon_{\text{MA}}(\lambda)$  the molar absorption coefficient of MA, and  $p^{\circ}(\lambda)$  the measured spectral photon flux density emitted by the radiation source. In the case of monochromatic irradiation at a given wavelength (assuming  $\Delta \lambda = 1$  nm) it is  $\int_{\lambda} p^{\circ}(\lambda) \varepsilon_{MA}(\lambda) d\lambda = p^{\circ}(\lambda) \varepsilon_{MA}(\lambda)$ .

The following Table reports the reaction parameters as well as the photochemical data that were used to determine  $\Phi_{MA}$ .

**Table SM1.** Data used for the calculation of  $\Phi_{MA}$  under the different irradiation conditions. n/a = not applicable.

Irradiation conditions	$R_{MA}$ , mol $L^{-1}$ s <sup>-1</sup>	$A_{MA}(\lambda) = \varepsilon_{MA}(\lambda)$ b [MA]	$\int_{\lambda} p^{\circ}(\lambda) \varepsilon_{MA}(\lambda) d\lambda,$ ein L <sup>-1</sup> s <sup>-1</sup>	$\int_{\lambda} p^{\circ}(\lambda)  d\lambda,$ ein L <sup>-1</sup> s <sup>-1</sup>	$\mathbf{\Phi}_{\mathrm{MA}}$
254 nm	$6.0 \cdot 10^{-10}$	0.50	n/a	$1.4 \cdot 10^{-6}$	$3.8 \cdot 10^{-3}$
325 nm	$1.4 \cdot 10^{-9}$	0.53	n/a	$2.3 \cdot 10^{-7}$	$1.4 \cdot 10^{-3}$
Suntest	$1.8 \cdot 10^{-8}$	n/a	$2.33 \cdot 10^{-2}$	n/a	$1.7 \cdot 10^{-3}$