

Supplementary Materials: Effects of Substitution on Solid-State Fluorescence in 9-Aryl-9-methyl-9*H*-9-silafluorenes

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1. Absorption Spectra of 1–10 in *n*-Hexane

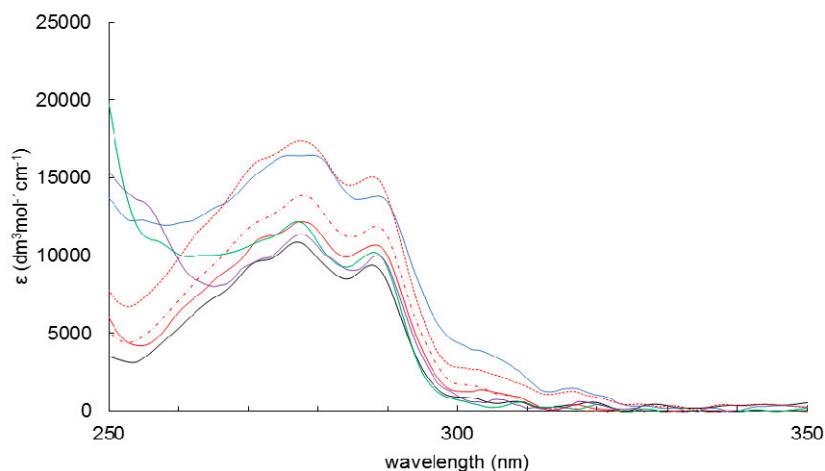


Figure S1. Absorption spectra of 1–7 in *n*-hexane.

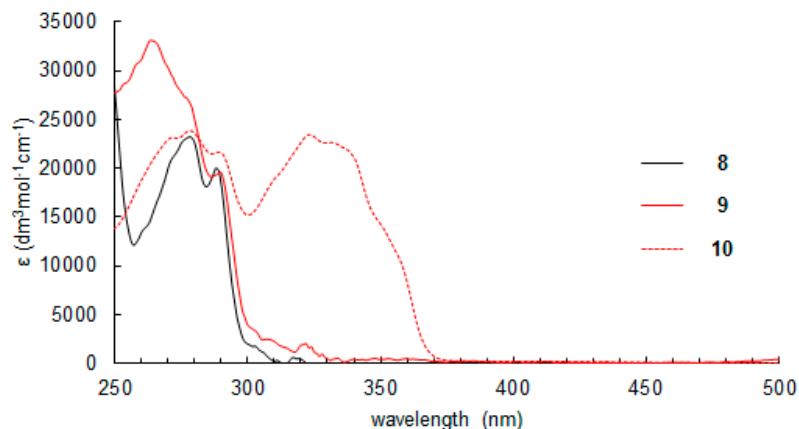


Figure S2. Absorption spectra of 8–10 in *n*-hexane.

2. Fluorescence Spectra of 1–10 in *n*-Hexane

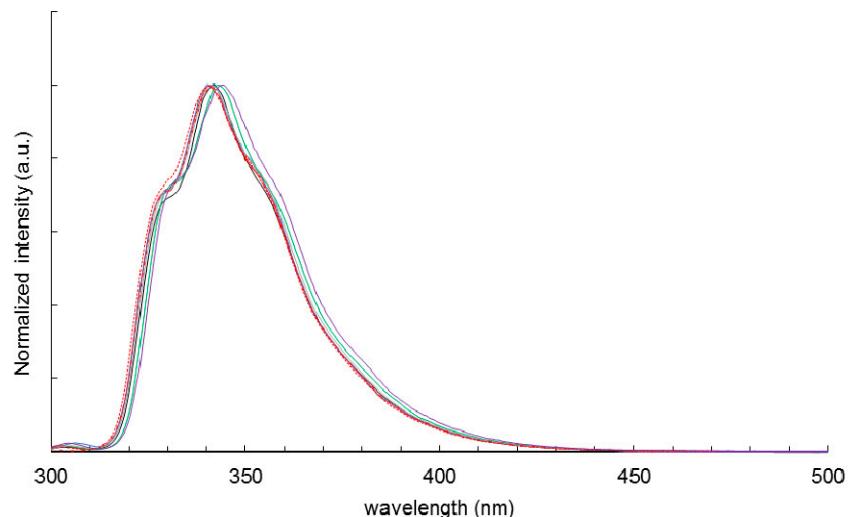


Figure S3. Fluorescence spectra of 1–7 in *n*-hexane.

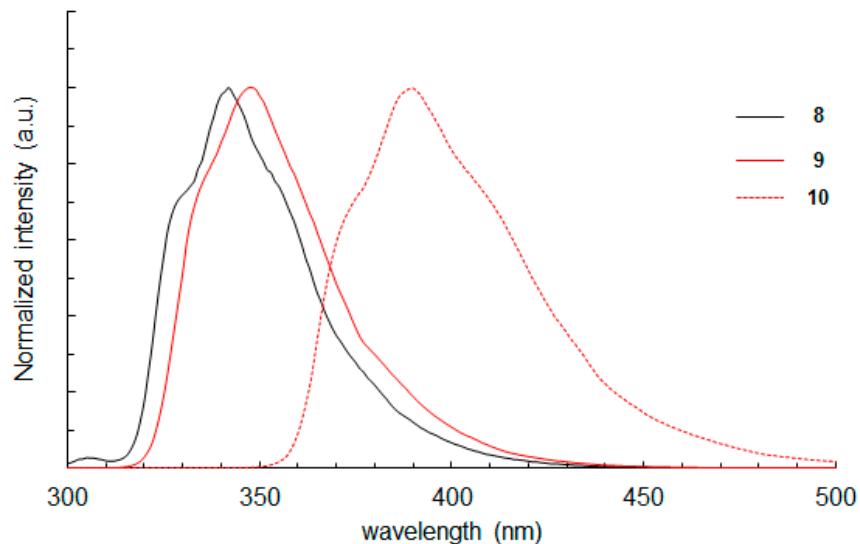


Figure S4. Fluorescence spectra of **8–10** in *n*-hexane.

3. Fluorescence Spectra of **9** and **10** under Various Conditions

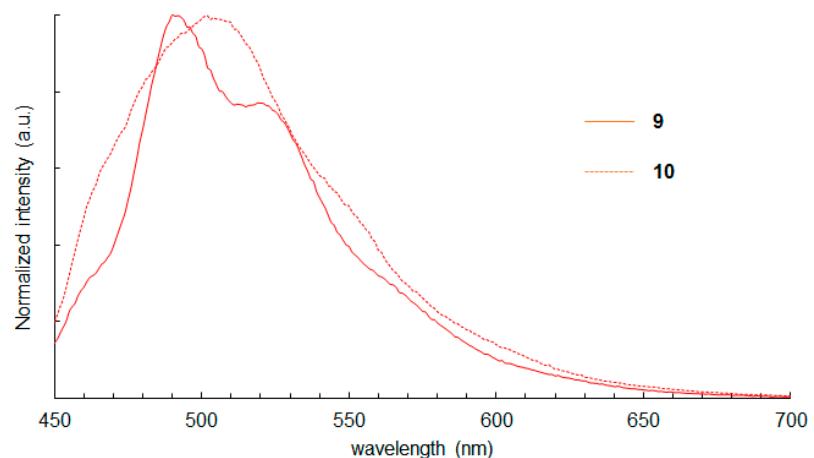


Figure S5. Fluorescence spectra of **9** (excited at 406 nm) and **10** (excited at 402 nm) in *n*-hexane.

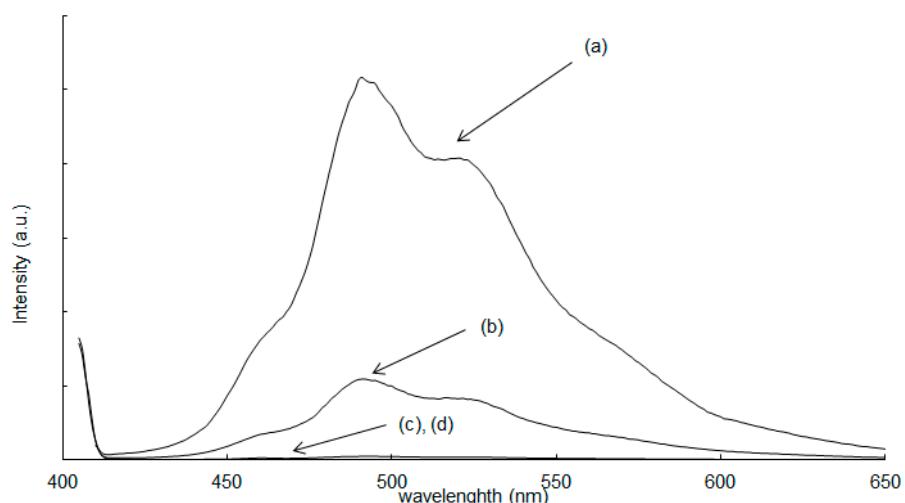


Figure S6. Fluorescence spectra of **9** (excited at 406 nm) in *n*-hexane under (a) 10^{-3} M, (b) 10^{-4} M, (c) 10^{-5} M and (d) 10^{-6} M.

4. Fluorescence Spectra of 1, 3, and 5–7 in the Solid State

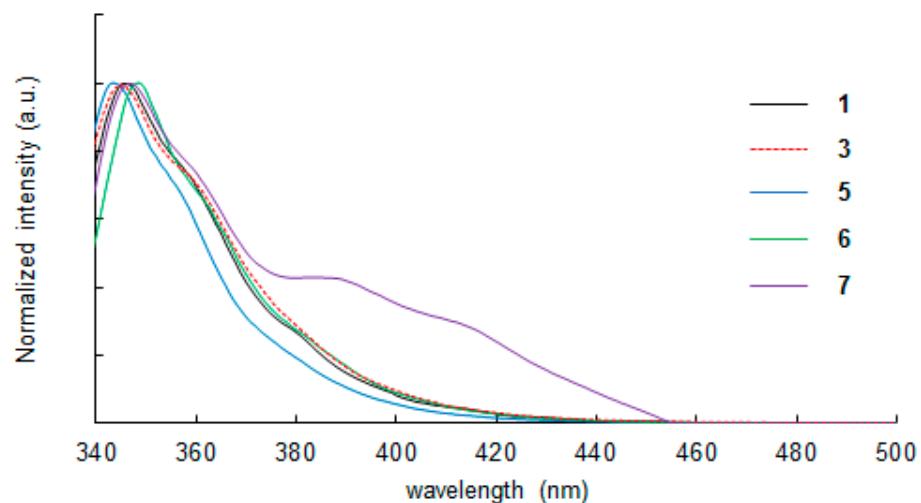


Figure S7. Fluorescence spectra of 1, 3, and 5–7 in the solid state.

5. TG-DTA Measurement of 7–10

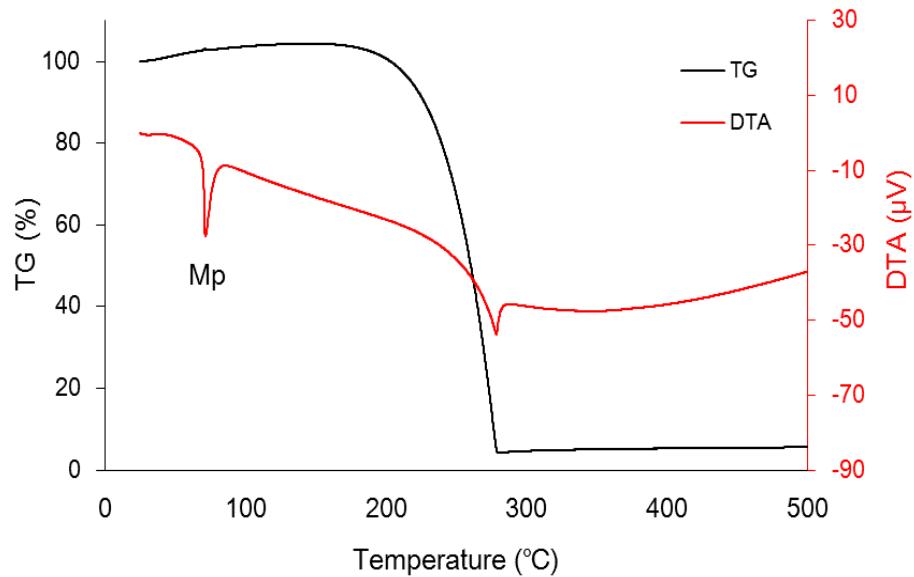
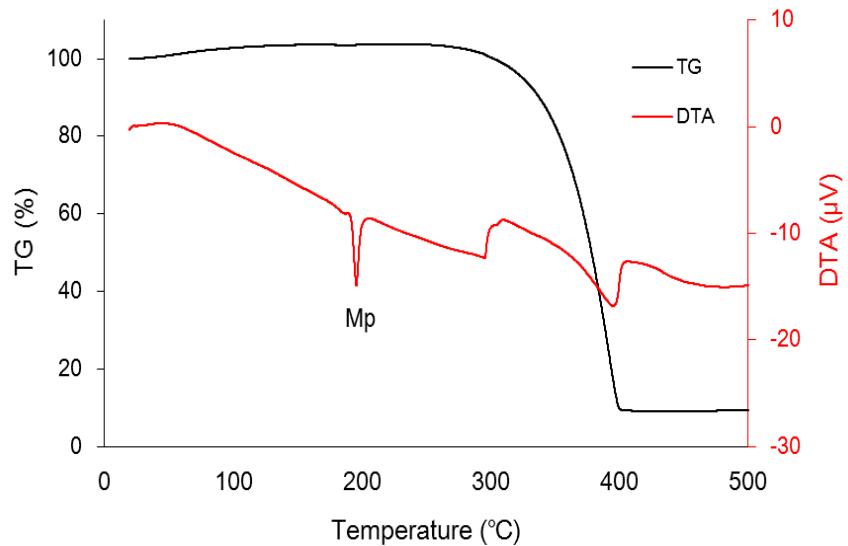
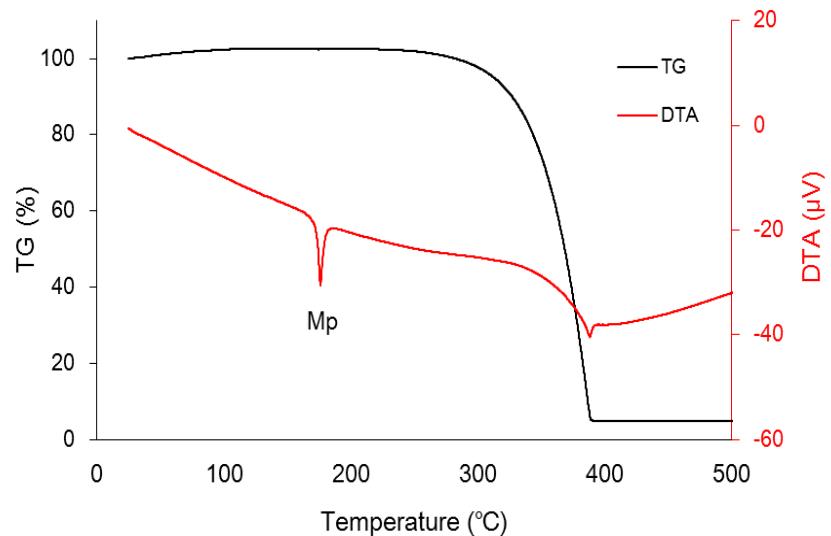
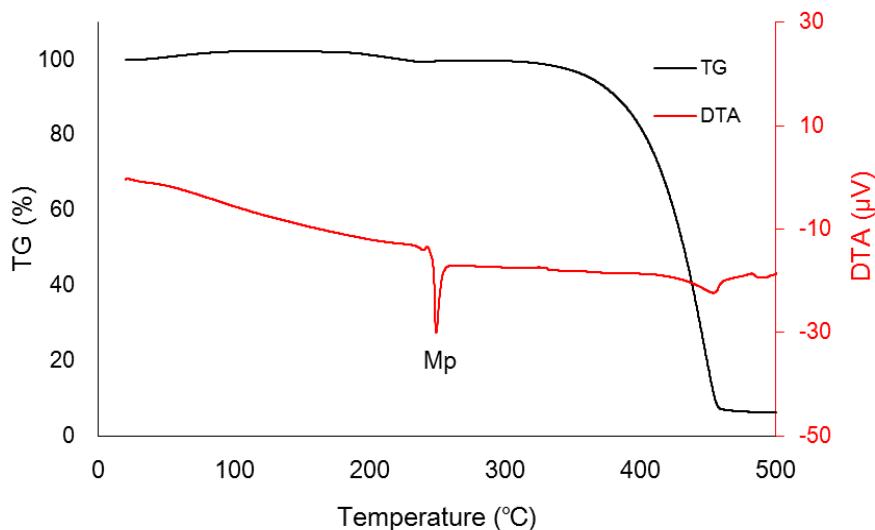


Figure S8. TG-DTA spectra of 7.

**Figure S9.** TG-DTA spectra of **8**.**Figure S10.** TG-DTA spectra of **9**.**Figure S11.** TG-DTA spectra of **10**.

6. Crystallographic Data of 8–10

Table S1. Selected crystallographic data of 8.

Empirical Formula	C ₃₂ H ₂₆ Si ₂
<i>F_w</i> /g·mol ⁻¹	466.73
Crystal system	monoclinic
Space group	<i>P</i> 2 ₁ / <i>c</i>
Crystal size/mm	0.250 × 0.170 × 0.060
Temperature/K	93
<i>a</i> /Å	13.373(7)
<i>b</i> /Å	11.094(5)
<i>c</i> /Å	8.642(4)
α/°	90
β/°	104.726(6)
γ/°	90
<i>V</i> /Å ³	1240.0(11)
<i>Z</i>	2
<i>D</i> _{calcd} /g·cm ⁻³	1.250
<i>λ</i> /Å	0.71075
<i>μ</i> /mm ⁻¹	0.1618
Reflections collected	8732
Independent reflections	2504
Parameters	154
<i>R</i> _{int}	0.0480
<i>R</i> ₁ (<i>I</i> > 2.00σ(<i>I</i>)) ^a	0.0486
<i>wR</i> ₂ (All reflections) ^b	0.1053
GoF ^c	1.119

^a $R_1 = \frac{\sum ||Fo| - |Fc||}{\sum |Fo|}$ ($I > 2\sigma(I)$). ^b $wR_2 = [\sum (w(Fo^2 - Fc^2)^2 / \sum w(Fo^2)^2)]^{1/2}$ ($I > 2\sigma(I)$). ^c GoF = $[\sum (w(Fo^2 - Fc^2)^2 / \sum (Nr - Np)^2)]$.

Table S2. Selected bond lengths, angles, and torsion angles of 8.

Bond lengths (Å)	
Si1–C2	1.865 (3)
Si1–C3	1.873 (3)
Si1–C8	1.8713 (19)
Si1–C13	1.861 (3)
Angles (°)	
C2–Si1–C3	91.55 (9)
C2–Si1–C8	112.02 (10)
C2–Si1–C13	113.78 (10)
C3–Si1–C8	115.66 (10)
C3–Si1–C13	112.05 (10)
C8–Si1–C13	110.66 (9)

Table S3. Selected crystallographic data of **9**.

Empirical Formula	C ₃₀ H ₂₄ SSi ₂
F _w /g·mol ⁻¹	472.75
Crystal system	triclinic
Space group	P-1
Crystal size/mm	0.110 × 0.061 × 0.030
Temperature/K	113
a/Å	8.717(3)
b/Å	12.385(4)
c/Å	12.634(4)
α/°	81.455(13)
β/°	76.448(12)
γ/°	69.996(10)
V/Å ³	1242.5(7)
Z	2
D _{calcd} /g·cm ⁻³	1.264
λ/Å	0.71070
μ/mm ⁻¹	0.243
Reflections collected	9020
Independent reflections	4770
Parameters	298
R _{int}	0.0335
R ₁ (I > 2.00σ(I)) ^a	0.0616
wR ₂ (All reflections) ^b	0.1634
GoF ^c	1.070

^a R₁ = Σ ||Fo| - |Fc|| / Σ |Fo| (I > 2σ(I)). ^b wR₂ = [Σ(w(Fo² - Fc²)² / Σ w(Fo²)²]^{1/2} (I > 2σ(I)). ^c GoF = [Σ(w(Fo² - Fc²)² / Σ(Nr - Np)²].

Table S4. Selected bond lengths, angles, and torsion angles of **9**.

Bond lengths (Å)	
Si2–C4	1.879(4)
Si2–C15	1.873(4)
Si2–C16	1.863(4)
Si2–C17	1.863(3)
Si3–C20	1.867(3)
Si3–C21	1.863(4)
Si3–C32	1.866(4)
Si3–C33	1.855(4)
Angles (°)	
C4–Si2–C15	91.53(16)
C4–Si2–C16	112.69(17)
C4–Si2–C17	113.69(17)
C15–Si2–C16	114.32(17)
C15–Si2–C17	114.98(14)
C16–Si2–C17	108.88(15)
C20–Si3–C21	112.45(17)
C20–Si3–C32	111.40(14)
C20–Si3–C33	109.03(15)
C21–Si3–C32	91.95(16)
C21–Si3–C33	114.85(16)
C32–Si3–C33	116.39(17)

Table S5. Selected crystallographic data of **10**.

Empirical Formula	C ₃₄ H ₂₆ S ₂ Si ₂
F _w /g·mol ⁻¹	554.87
Crystal system	monoclinic
Space group	P2 ₁ /c
Crystal size/mm	0.260 × 0.218 × 0.202
Temperature/K	113
a/Å	10.972(4)
b/Å	6.584(2)
c/Å	20.551(7)
α/°	90
β/°	102.513(4)
γ/°	90
V/Å ³	1449.2(8)
Z	2
D _{calcd} /g·cm ⁻³	1.271
λ/Å	0.71070
μ/mm ⁻¹	0.288
Reflections collected	10960
Independent reflections	3217
Parameters	172
R _{int}	0.0534
R ₁ (I > 2.00σ(I)) ^a	0.0540
wR ₂ (All reflections) ^b	0.1180
GoF ^c	1.094

^a R₁ = Σ ||Fo| - |Fc|| / Σ |Fo| (I > 2σ (I)). ^b wR₂ = [Σ(w(Fo² - Fc²)² / Σw(Fo²)²]^{1/2} (I > 2σ (I)). ^c GoF = [Σ(w(Fo² - Fc²)² / Σ(Nr - Np)²].

Table S6. Selected bond lengths, angles, and torsion angles of **10**.

Bond lengths (Å)	
Si2–C6	1.859(3)
Si2–C7	1.860(3)
Si2–C8	1.869(3)
Si2–C19	1.869(3)
Angles (°)	
C6–Si2–C7	109.10(12)
C3–Si2–C8	114.72(11)
C6–Si2–C19	111.57(11)
C7–Si2–C8	112.77(12)
C7–Si2–C19	116.28(11)

7. Copies of ^1H -NMR and $^{13}\text{C}[^1\text{H}]$ NMR of 1–10

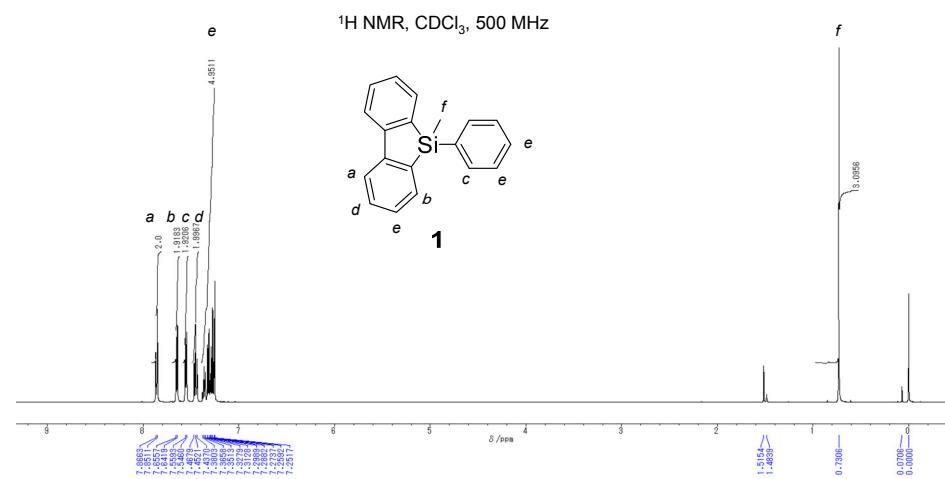


Figure S12. ^1H -NMR of **1**.

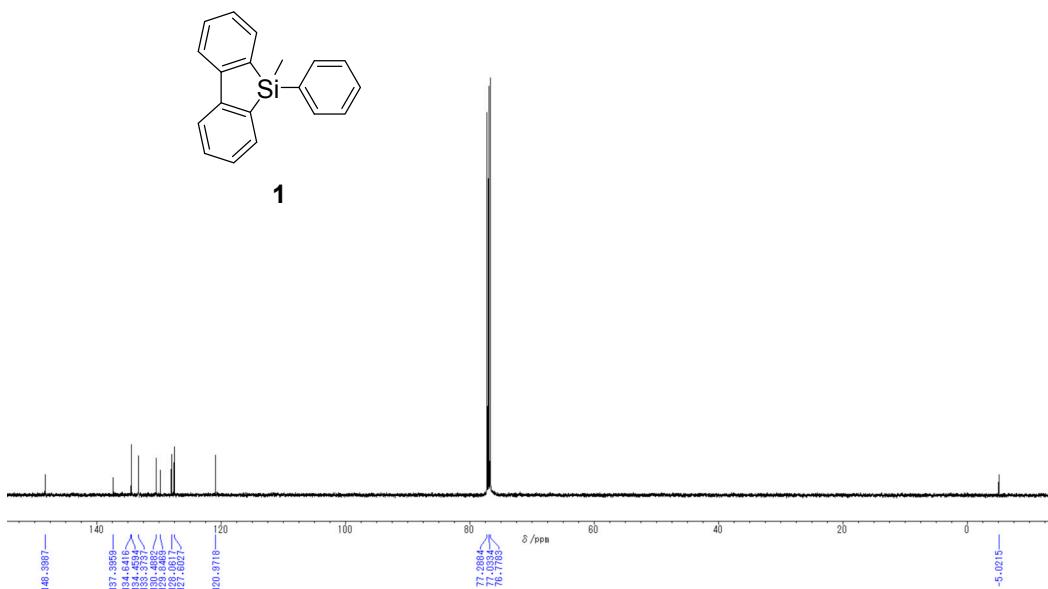


Figure S13. $^{13}\text{C}[^1\text{H}]$ -NMR of **1**.

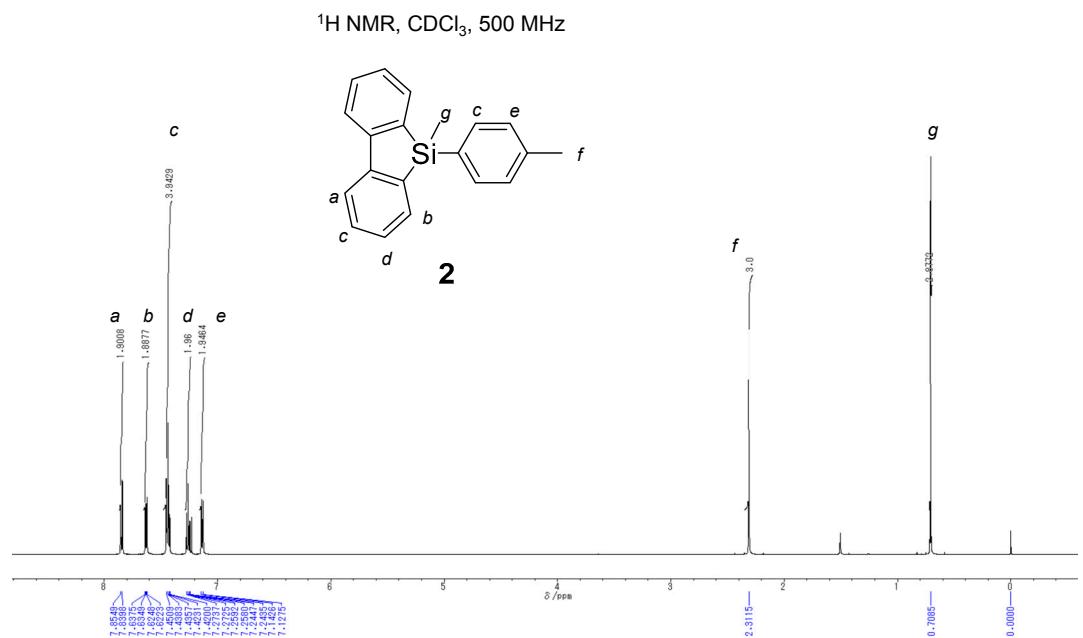


Figure S14. ^1H -NMR of 2.

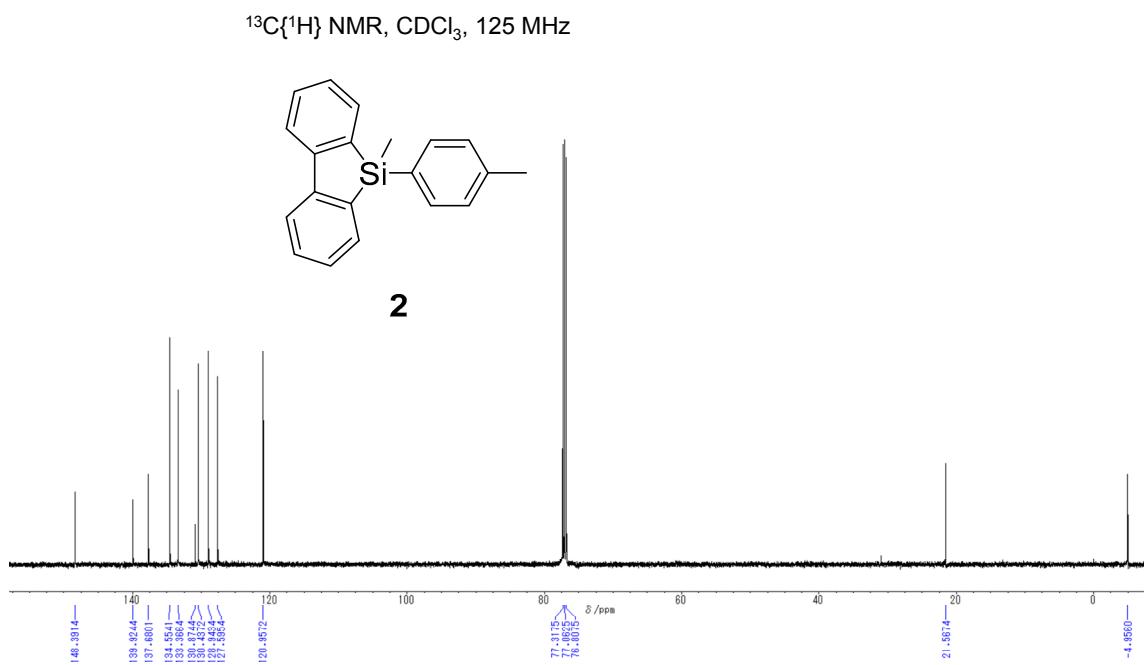
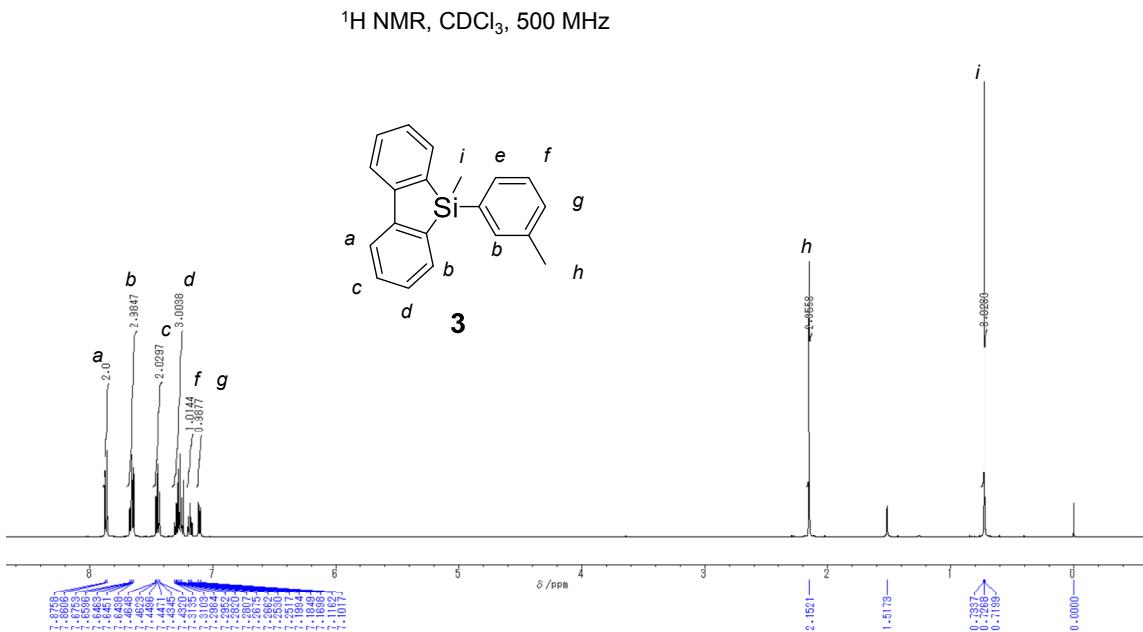
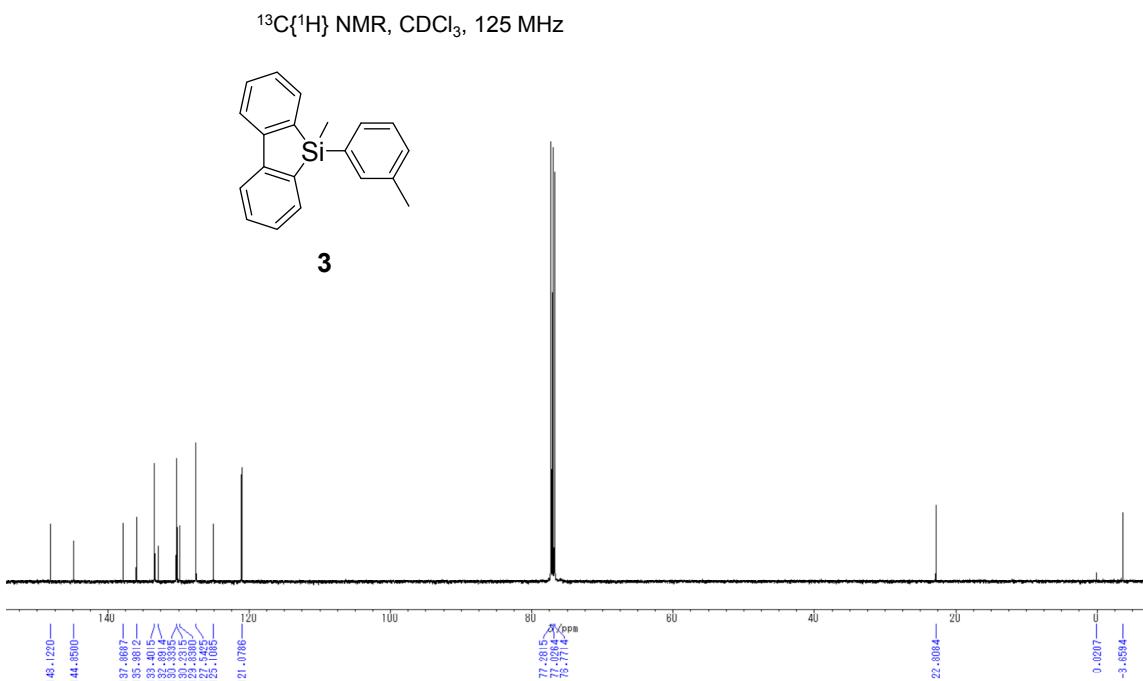


Figure S15. $^{13}\text{C}\{\text{H}\}$ -NMR of **2**.

Figure S16. ¹H-NMR of 3.Figure S17. ¹³C{¹H}-NMR of 3.

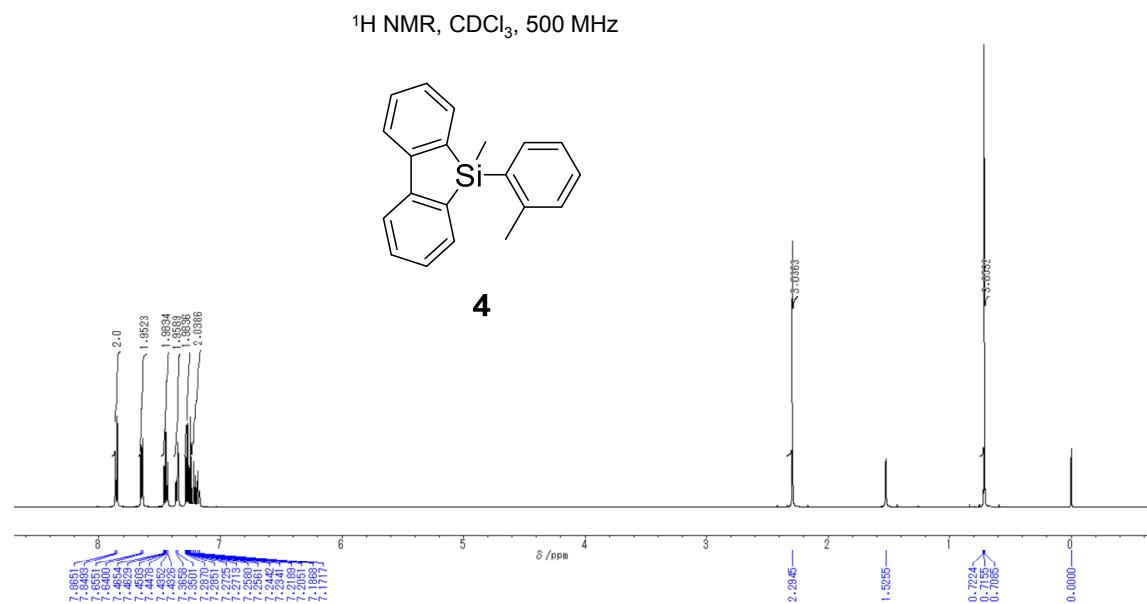


Figure S18. ^1H -NMR of 4.

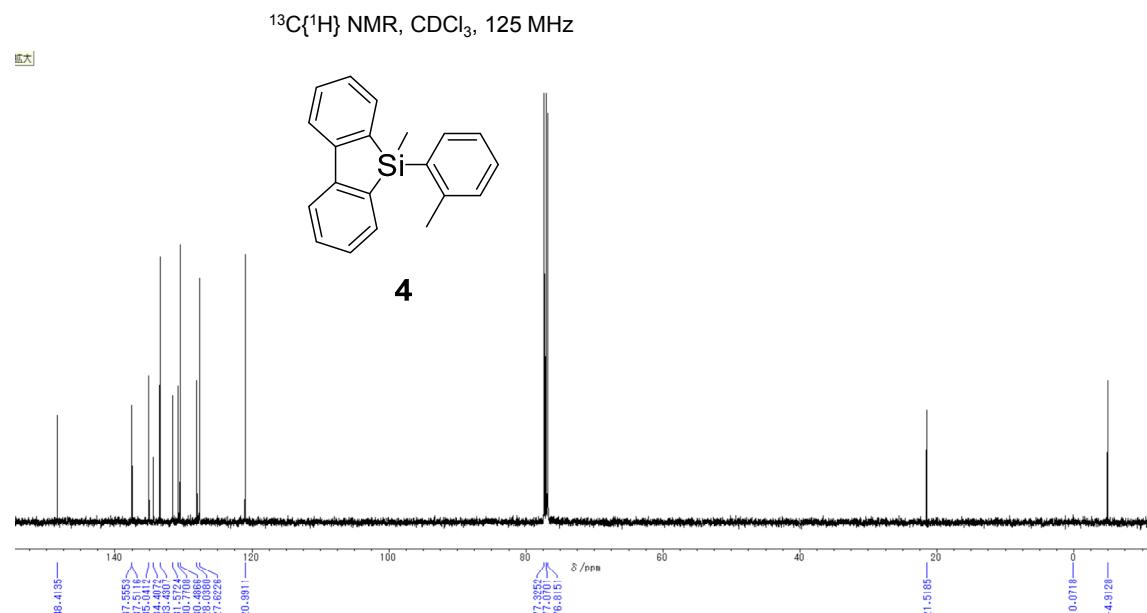
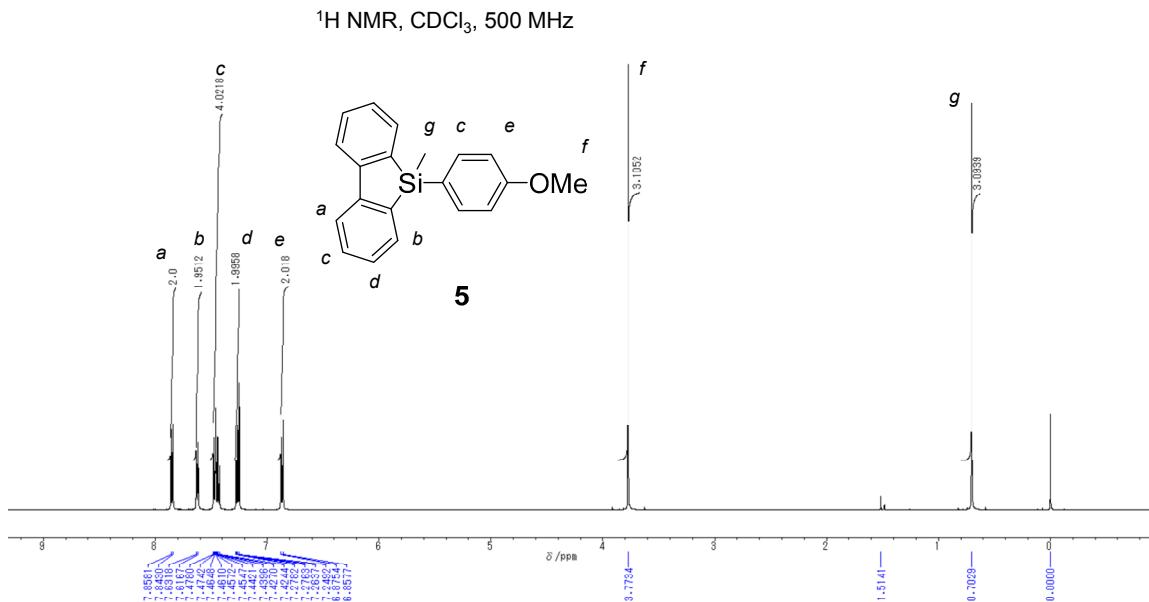
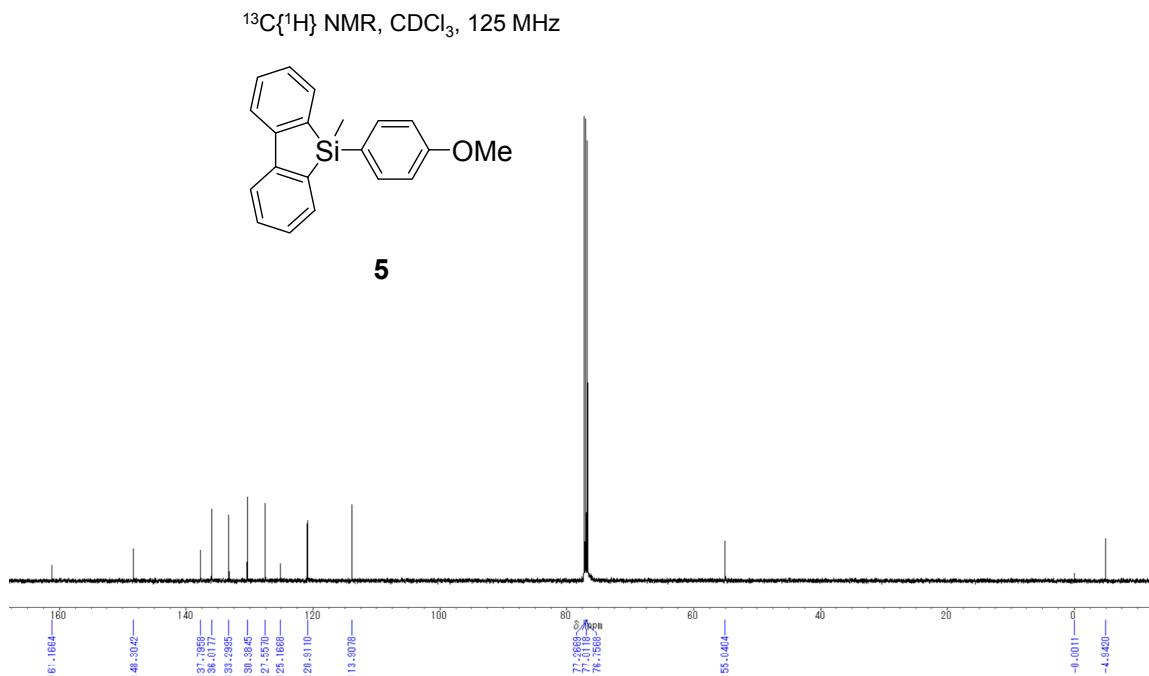


Figure S19. $^{13}\text{C}\{\text{H}\}$ -NMR of 4.

Figure S20. ¹H-NMR of 5.Figure S21. ¹³C{¹H}-NMR of 5.

¹H NMR, CDCl₃, 500 MHz

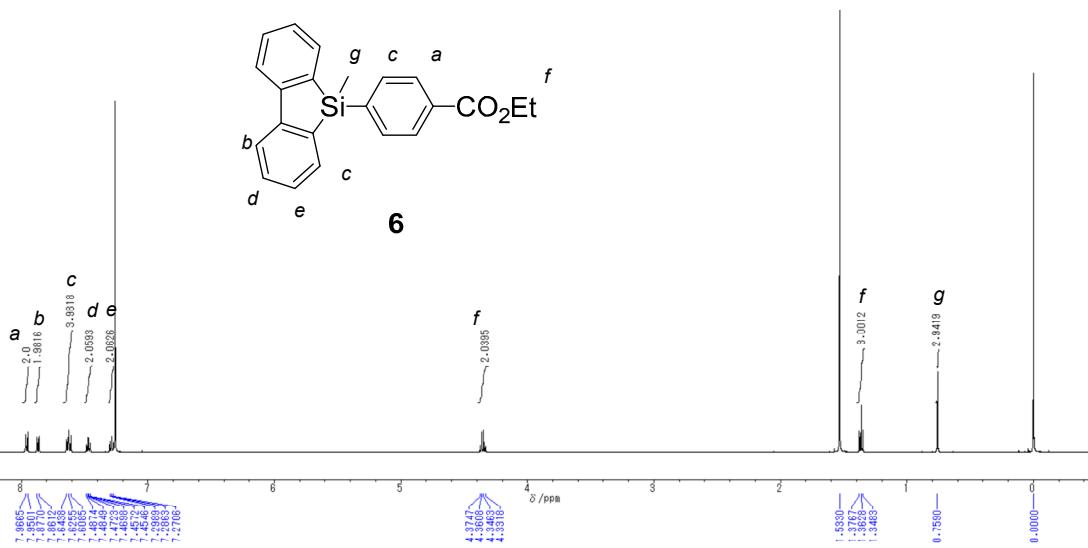


Figure S22. ^1H -NMR of 6.

$^{13}\text{C}\{^1\text{H}\}$ NMR, CDCl_3 , 125 MHz

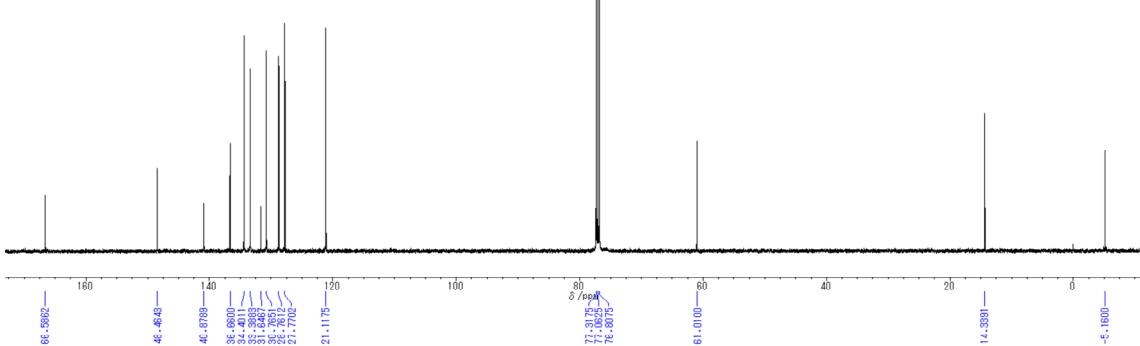
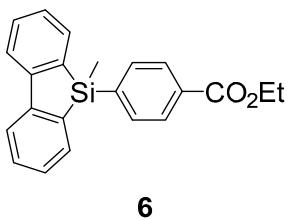


Figure S23. $^{13}\text{C}\{\text{H}\}$ -NMR of 6.

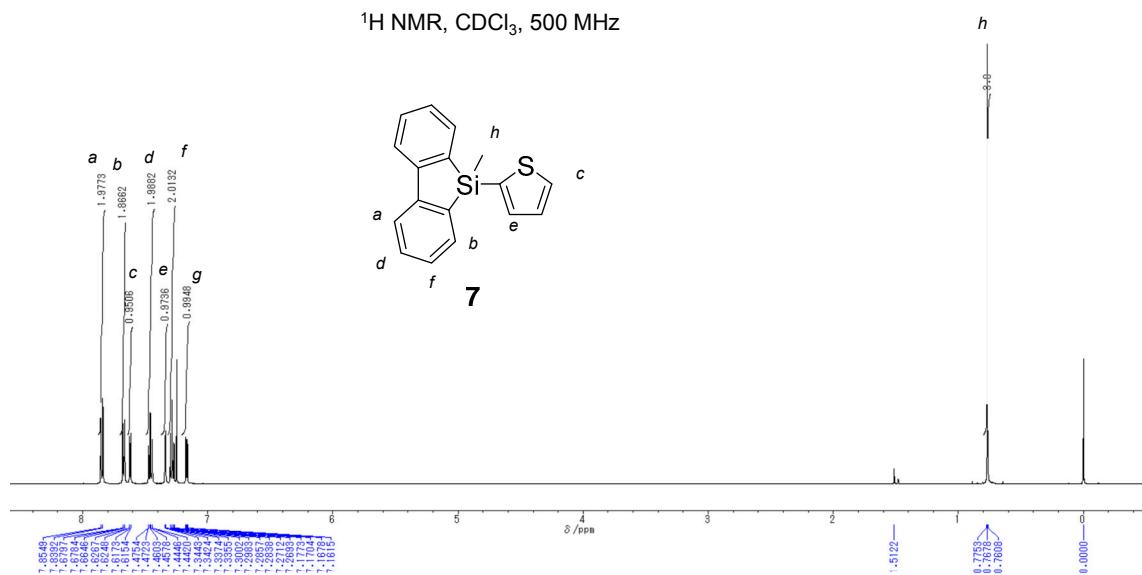


Figure S24. ^1H -NMR of 7.

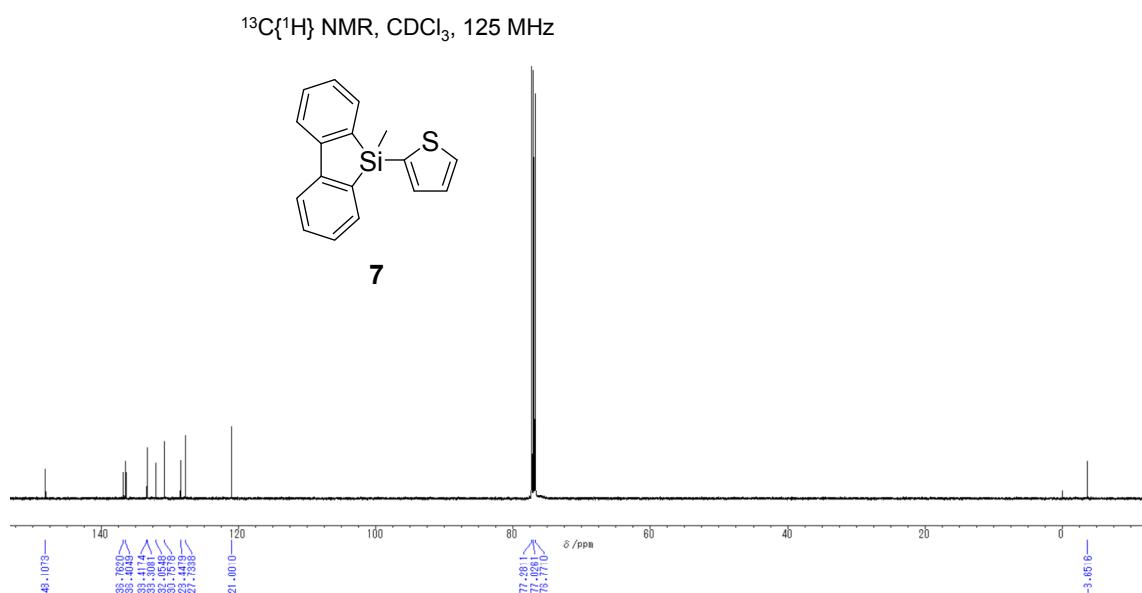


Figure S25. $^{13}\text{C}\{\text{H}\}$ -NMR of 7.

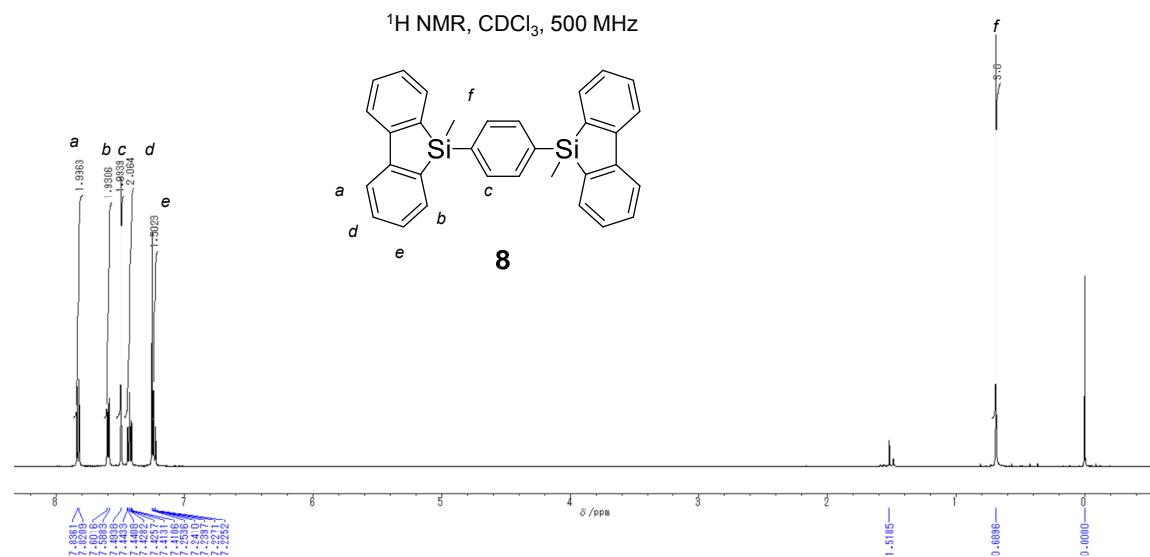


Figure S26. ^1H -NMR of 8.

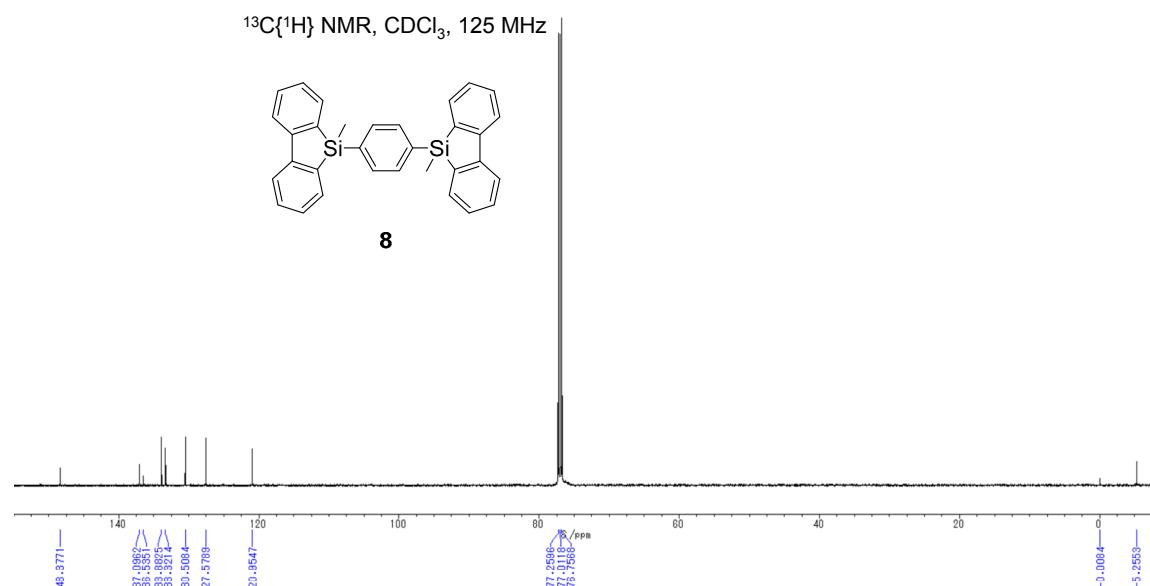


Figure S27. $^{13}\text{C}\{\text{H}\}$ -NMR of 8.

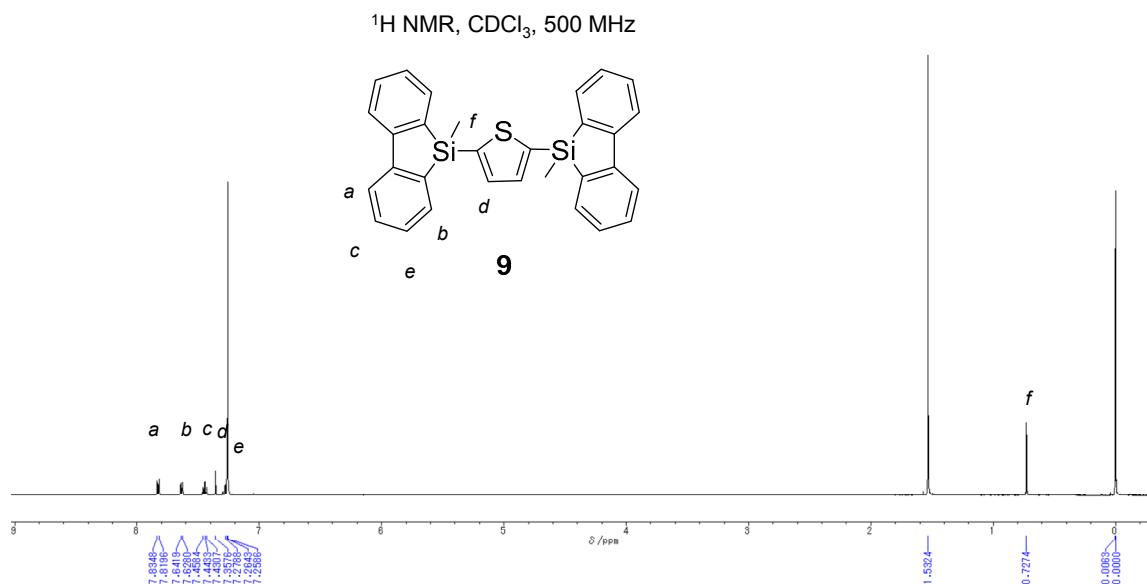


Figure S28. ^1H -NMR of 9.

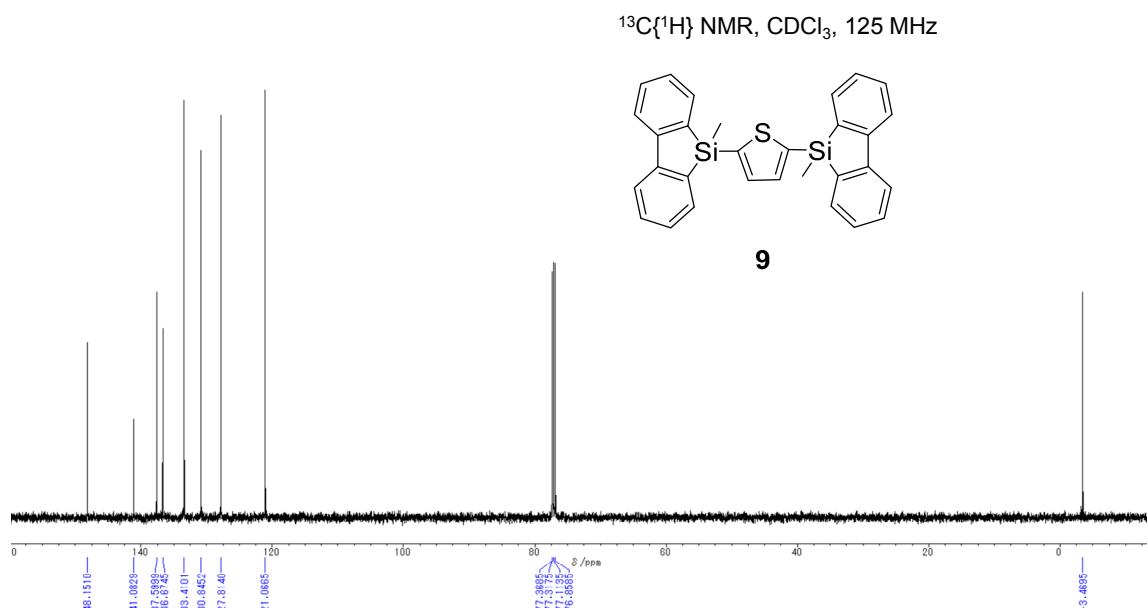
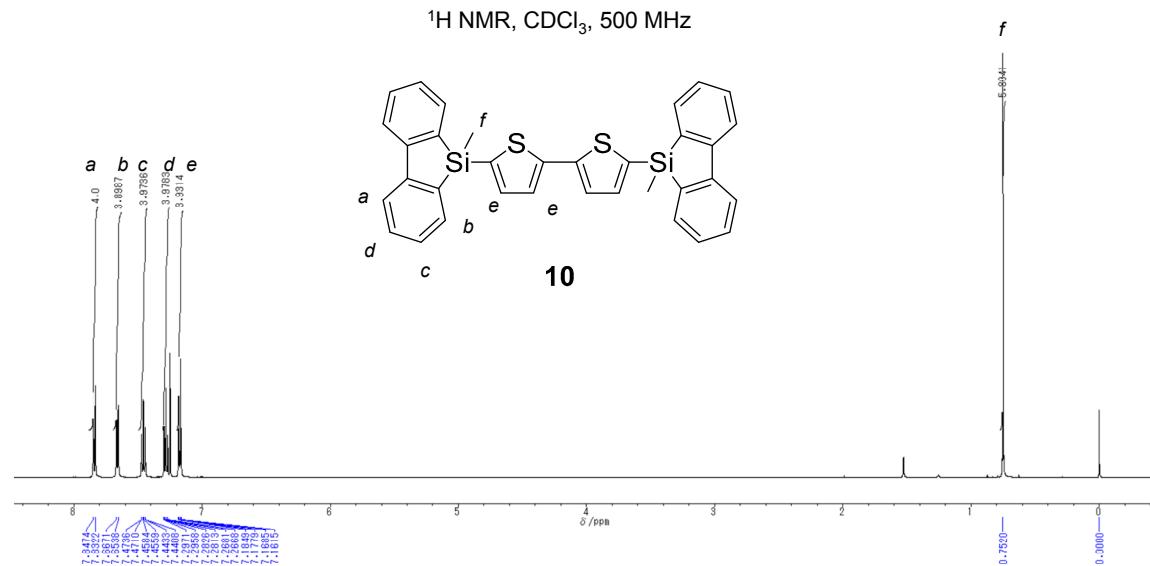
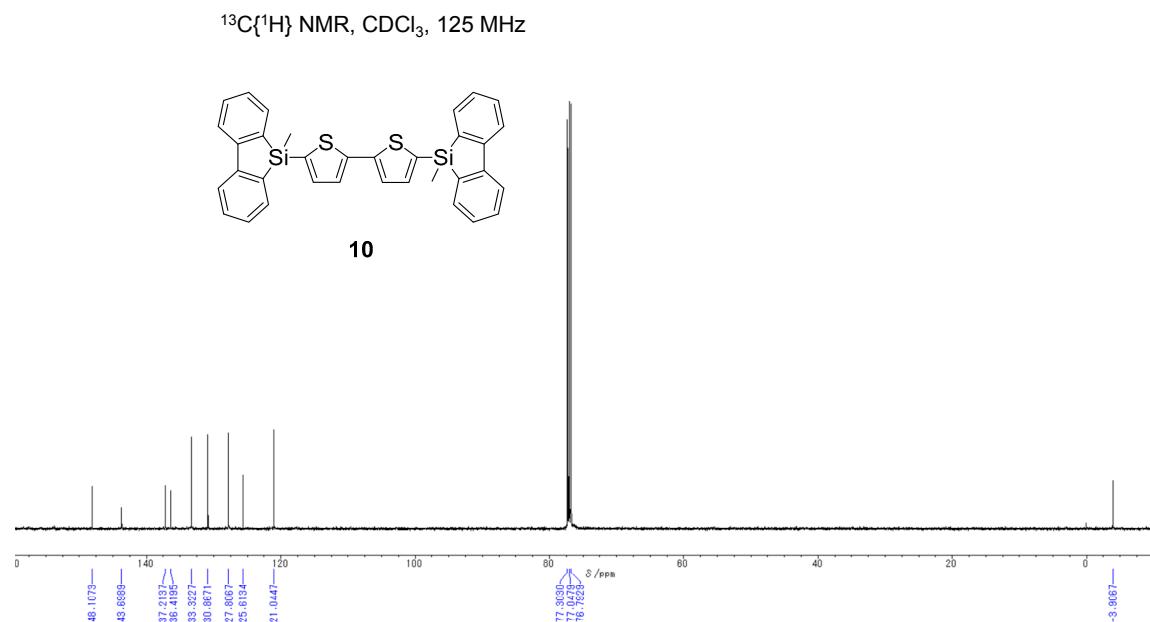


Figure S29. $^{13}\text{C}\{^1\text{H}\}$ -NMR of 9.

**Figure S30.** ¹H-NMR of **10**.**Figure S31.** ¹³C{¹H}-NMR of **10**.