Supporting Information

S1 NMR Spectra



Figure S1.¹ H-NMR spectrum (500 MHz, benzene-*d*₆) of 1.



Figure S2. ¹³C {¹H} NMR spectrum (126 MHz, benzene- d_6) of 1.



Figure S3. ¹¹B $\{^{1}H\}$ NMR spectrum (160 MHz, benzene-*d*₆) of **1**.

Figure S4. ¹³C-¹H HSQC NMR spectrum (126 MHz–500 MHz, benzene-*d*₆) of 1.

Figure S5. ¹H-NMR spectrum (500 MHz, benzene-*d*₆) of **2**.

Figure S6. ${}^{13}C{}^{1}H$ NMR spectrum (126 MHz, chloroform-*d*) of **2**.

Figure S7. ¹¹B $\{^{1}H\}$ NMR spectrum (160 MHz, benzene-*d*₆) of **2**.

Figure S8. ¹H NMR spectrum (500 MHz, benzene-*d*₆) of **3**.

Figure S9. ¹³C {¹H} NMR spectrum (126 MHz, benzene- d_6) of 3.

Figure S10. ¹¹B $\{^{1}H\}$ NMR spectrum (160 MHz, benzene-*d*₆) of **3**.

Figure S11. ¹³C-¹H HSQC NMR (126 MHz–500 MHz, benzene-*d*₆) spectra of **3**.

Room Temperature

Figure S13. ¹H-NMR (500 MHz, benzene- d_6) spectra of **1** after heating with paraformaldehyde for 24 h.

Figure S14. FTIR spectra of 1.

Figure S15. FTIR spectra of 3.

S3 Calculations

General Information

All the calculations were performed on the full structures of the reported compounds. Calculations were performed with the GAUSSIAN 09 suite of programs [1]. The ω B97XD functional [2] was qualified as promising by Grimme [3] and was used to accurately describe the mechanism of FLP mediated hydrogenation of alkynes [4] and was thus used in combination with the 6-31G** basis set for all atoms [5,6]. The transition states were located and confirmed by frequency calculations (single imaginary frequency). The stationary points were characterized as minima by full vibration frequencies calculations (no imaginary frequency). All geometry optimizations were carried out without any symmetry constraints. The energies were then refined by single point calculations to include solvent effects using the SMD solvation model [7] with the experimental solvent (benzene) at the ω B97XD/6-31++G** level of theory [8]. All structures with their associated free enthalpy and Gibbs free energies as well as their cartesian coordinates are fully detailed in the following section.

С Н C H С Н C B С Н С Η Η С Η Η С Ĥ Н С Н С Η Η С Η Н С Η Η

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H B C C C C C C H C C

H H C H

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С

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N C C H

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0.73193000	10.23040500	0.85919100
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4.29184800	10.15846900	-2.34800500
4.94526700	10.15172000	-0.72165400
1.38791100	6.73130100	-2.48477700
2.09407600	5.95370200	-2.81077600
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1.77315400	7.70062900	-2.82813000

 $\begin{array}{c} N\\ C\\ C\\ H\\ C\\ C\\ C\\ C\\ C\\ C\\ C\\ H\\ H\end{array}$

Н Н Н Н Н Н Η Н Н Н Н Н Н Н Н Н С Н Н Н

1-HCOOH (3)

Sum of electronic and thermal Enthalpies= -1166.690769 Sum of electronic and thermal Free Energies= -1166.768467

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Н	-2.67028100	4.90040000 1.0	0309000	С	-1.21731500	3.32767400	0.76205200
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С	0.16234600	3.94951400 -2.0	6775000	Н	-1.42517900	2.74100600	1.66614100
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С	1.27750400	5.16214200	2.42834200
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C C C C C C H H H H H H

H H H H H H

H H H H

Н

1-CH₂O

Sum of electronic and thermal Enthalpies= -1091.430292 Sum of electronic and thermal Free Energies= -1091.503354

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С	0.11228400	-0.97266400	3.76484500	Н	2.35363900	-1.40554500	-0.06036300
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Ν	1.13644000	1.05280900	0.11529200	Н	-4.48498800	-0.85675800	0.72447500
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Н	1.16562600	3.01764100	2.44081600	Н	-4.01003500	-0.50206800	-1.80200800
Н	2.64586200	3.15932600	1.47670700	Н	-4.20105300	-3.10085500	0.06242100
С	3.72613700	0.88499600	0.62394500	Н	-3.98433800	-3.11455100	1.78726400
Н	3.64425700	0.14697000	1.42005900	Н	-3.70823800	-2.87160500	-1.80638100
Н	4.66059200	0.68015700	0.09585300	Н	-2.93544000	-2.29525000	-3.25584300
Н	3.82827000	1.87654700	1.05919100	0	-0.08174200	-0.70907900	-1.12843900
С	-0.67436100	2.79276900	0.65042400	С	0.13169100	0.59093900	-1.08283400
Н	-1.32607000	2.79640100	-0.22397400	Н	0.62409300	0.94837800	-1.98441700
Н	-0.76817700	1.17024300	-0.88976300				

S4 Crystallographic data

General Information

Single crystals with suitable size of all compounds were mounted on CryoLoops with Paratone-N and optically aligned on a Bruker SMART APEX–II X–ray diffractometer with 1K CCD detector using a digital camera. Initial intensity measurements were performed using a fine–focused sealed tube, graphite-monochromated, X-ray source (Mo K α , $\lambda = 0.71073$ Å) at 50 kV and 30 mA. Standard APEX-II [9] software package was used for determining the unit cells, generating the data collection strategy, and controlling data collection. SAINT [10] was used for data integration including Lorentz and polarization corrections. Semi-empirical absorption corrections were applied using SCALE (SADABS) [11]. The structures of all compounds were solved by direct methods and refined by full-matrix least-squares methods with SHELX-97 [12] in the SHELXTL6.14 package. As the solvent molecules in some compounds are highly disordered, the SQUEEZE subroutine of the PLATON [13] software suit was used to remove the scattering contributions from the highly disordered guest molecules. The resulting new HKL files were used to further refine the structures. All of the H atoms (on C atoms) were generated geometrically and refined in riding mode. Crystallographic information for all obtained phases is summarized in Table S1. Atomic coordinates and additional structural information are provided in the cif files of the Supporting Information.

	1	2
Empirical formula	C23 H36 B N	C23 H38 B N O
Formula weight	337.34	355.35
Temperature	150(2) K	296(2) K
Wavelength	0.71073 Å	0.71073 Å
Crystal system	Monoclinic	Orthorhombic
Space group	P21/c	P212121
	a = 7.6005(6) Å	a = 11.546(4) Å
	b = 18.9799(16) Å	b = 13.277(5) Å
Luit cell dimensions	c = 13.7261(12) Å	c = 13.602(5) Å
Unit cell dimensions	$\alpha = 90^{\circ}$	$\alpha = 90^{\circ}$
	$\beta = 92.6030(10)^{\circ}$	$\beta = 90^{\circ}$
	$\gamma = 90^{\circ}$	$\gamma = 90^{\circ}$
Volume	1978.0(3) Å ³	2085.1(14) Å ³
Ζ	4	4
Density (calculated)	1.133 Mg/m ³	1.132 Mg/m ³
Absorption coefficient	0.063 mm^{-1}	0.067 mm^{-1}
F(000)	744	784
Crystal size	$0.580 \times 0.460 \times 0.280 \ mm^3$	$0.480 \times 0.300 \times 0.220 \text{ mm}^3$
Theta range for data collection	1.832 to 28.215°	2.144 to 30.530°
Index ranges	$-10 \le h \le 10, -25 \le k \le 25, -18 \le l \le 18$	$-14 \le h \le 16, -16 \le k \le 18, -19 \le l \le 19$
Reflections collected	20848	16231
Independent reflections	4871 [R(int) = 0.0300]	6365 [R(int) = 0.0200]
Completeness (theta = 25.242°)	100.0 %	100.0 %
Max. and min. transmission	0.982 and 0.965	0.985 and 0.976
Data/restraints/parameters	4871/0/230	6365/0/244
Goodness-of-fit on F ²	1.043	1.051
Final R indices [I > 2sigma(I)]	R1 = 0.0434, $wR2 = 0.1160$	R1 = 0.0354, wR2 = 0.0891
R indices (all data)	R1 = 0.0470, wR2 = 0.1193	R1 = 0.0394, w $R2 = 0.0919$
Largest diff. peak and hole	0.420 and -0.195 e·Å ⁻³	0.307 and -0.163 e·Å ⁻³

Table S1. Crystal data and structure refinement for 1 and 2.

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