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Design, Synthesis and Antitubercular Activity of Certain Nicotinic Acid Hydrazides

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Academic Editor: Derek J. McPhee

Received: 17 April 2015 / Accepted: 12 May 2015 / Published: 15 May 2015

Abstract: Three series of 6-aryl-2-methylnicotinohydrazides **4a–i**, *N'*-arylidene-6-(4-bromophenyl)-2-methylnicotino hydrazides **7a–f**, and *N'*-(un/substituted 2-oxoindolin-3-ylidene)-6-(4-fluorophenyl)-2-methylnicotinohydrazides **8a–c** were synthesized and evaluated for their potential *in vitro* antimycobacterial activity against *M. tuberculosis*. The results showed that isatin hydrazides **8a–c** are remarkably more active than the parent hydrazide **4c**. Hydrazides **8b** and **8c** exhibited the highest activity among all the tested compounds (MIC = 12.5 and 6.25 µg/mL, respectively). Compounds **8b** and **8c** were also devoid of apparent cytotoxicity to HT-29, PC-3, A549, HepG2 and MCF-7 cancer cell lines. Besides, **8b** and **8c** showed good drug-likeness scores of 0.62 and 0.41, respectively. Those two isatin hydrazides could offer an excellent framework for future development to obtain more potent antitubercular agents. The SAR study suggested that lipophilicity of the synthesized derivatives is a crucial element that accounts for their antimycobacterial activity. Finally, a theoretical kinetic study was established to predict the ADME of the active derivatives.

Keywords: synthesis; nicotinic acid; hydrazides; antimycobacterial; ADME

1. Introduction

The Nineteenth World Health Organization (WHO) Tuberculosis Report indicates that TB is one of the world's deadliest communicable diseases [1]. It estimates that in 2013 there were 9.0 million new cases and 1.5 million deaths from TB, including 400,000 deaths associated with co-infection with HIV. The highest rates *per capita* occurred in the African Region (25%), while South-East Asia, the Western Pacific and African Regions account for around 81% of all the total cases [1]. The disease is aggravated by the worldwide continuous emergence of multidrug-resistant strains of *M. tuberculosis* (MDR-TB), extensively drug-resistant tuberculosis (XDR-TB) and totally drug-resistant tuberculosis (TDR-TB) [2,3]. The magnitude and extent of drug-resistant strains have increased concern that TB may once again become an incurable disease [4,5]. Moreover, the increasing incidence of the disease in immunocompromised patients along with the longer durations of therapy emphasize the need for new drugs to extend the range of effective TB treatment options [6–8].

TB treatment is tedious, challenging and time-consuming. It requires the administration of appropriate treatment regimens for at least six months via directly observed therapy (DOT) and follow-up support [2]. Treatment regimens require a minimum six months in two separate phases. The duration of phase one is about two months and involves four drugs (isoniazid, rifampicin, pyrazinamide and ethambutol), followed by four months of phase two (using isoniazid plus rifampicin) [9]. However, the present treatment regimen has some limitations such as drug toxicity and intolerance, drug–drug interactions and poor patient adherence due to the lengthy treatment duration [10]. Therefore, there is an urgent need for the development and more efficient evaluation of new TB drugs and shorter treatment regimens. Isoniazid (INH, Figure 1), a critical frontline drug in TB treatment discovered by Dogmagk, is a prodrug that requires activation *in vivo* by mycobacterial catalase peroxidase (KatG) [11,12]. INH exerts its anti-tubercular activity *via* interference with the synthesis of mycolic acid, one of the essential chemical pathways responsible for the formation of cell walls in *M. tuberculosis* [13].

The enzymatic acetylation of isoniazid by *N*-acetyltransferase (NAT) represents a major metabolic pathway for isoniazid in humans [14], so blocking acetylation *via* chemical modification of the hydrazine unit with a functional group, while preserving potent antimycobacterial action, has the potential to counterbalance the known side effects of INH, improve clinical outcomes and reduce the emergence of acquired isoniazid resistance in patients. Subsequently, numerous studies have pointed out the importance of developing novel INH hydrazides as promising anti-tubercular agents (Figure 1) [15–22].

Recently, Narang *et al.* [23] developed a novel series of nicotinic acid hydrazide derivatives as potential antimycobacterial agents with a general structure represented by **VI** (Figure 1). The results showed that the presence of lipophilic electron-withdrawing halogen groups at the *para* position of the phenyl ring improved the antimycobacterial activity. Concerning the related fused pyridine heterocycles, Adhikari and co-workers reported two studies on the design, synthesis and biological evaluation of two different series of new quinoline-3-carbohydrazone derivatives **VII**, (Figure 1), as

potential antimycobacterial agents [24,25]. Aboul-Fadl *et al.* [26] also explored the anti-tubercular activity of Schiff bases **VIII** of nalidixic acid-3-carbohydrazides and isatins. On the other hand, isatin-based compounds are known to exhibit excellent anti-TB properties [27–30].

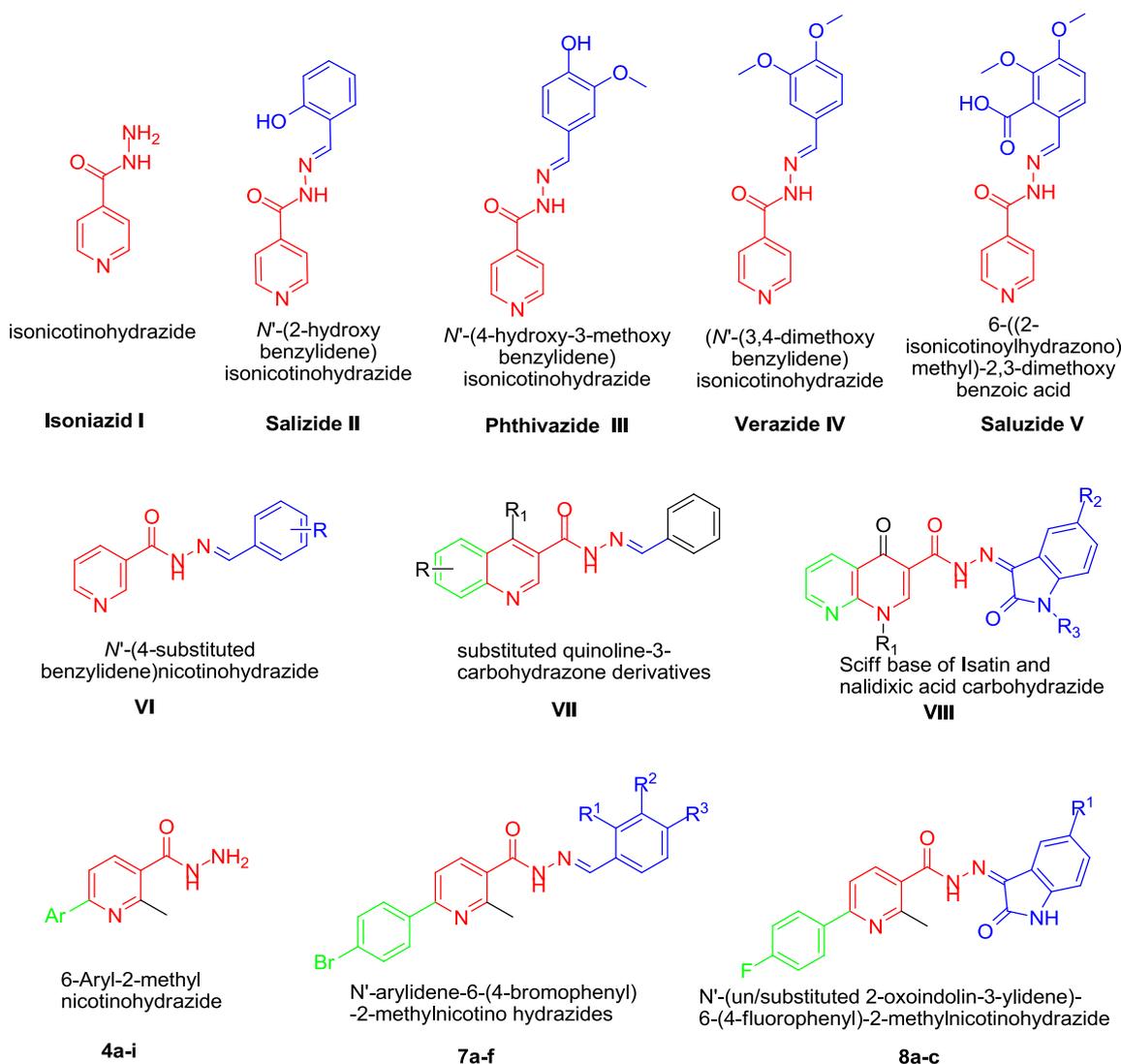


Figure 1. Structures of antitubercular drugs **I–VIII** and the target derivatives **4a–i**, **7a–f** and **8a–c**.

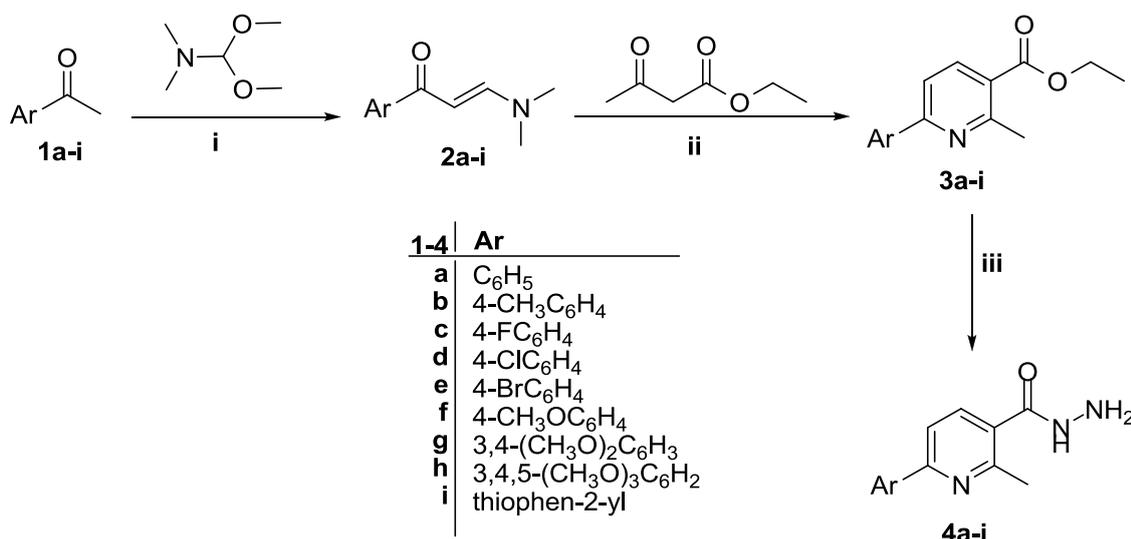
In this work, the aforementioned findings motivated us to synthesize three series of nicotinic acid hydrazide derivatives **4a–i**, **7a–f** and **8a–c** with the aim of obtaining new antimycobacterial agents. The first series comprises different nine 6-aryl-2-methylnicotinohydrazides with free hydrazine units in a similar fashion to INH. Subsequently, two derivatives (compounds **4c** and **4e**) with considerable lipophilicity ($\text{LogP} = 1.53$ and 2.17 , respectively) were chosen for further chemical modification. The non-classical ring opening bioisosterism for structures **VI** and **VIII** was adopted to develop a series of aldehyde hydrazides **7a–f** and a series of isatin hydrazides **8a–c**.

The *in vitro* cytotoxic activity of subset of compounds was evaluated against HT-29, PC-3, A549, HepG2 and MCF-7 cancer cell lines to determine the toxicity of these agents. In addition, we describe an ADME study and SAR description in order to explore the structural requirements controlling these observed antitubercular activities.

2. Results and Discussion

2.1. Chemistry

The synthetic pathways employed to prepare the new targeted derivatives are depicted in Schemes 1 and 2. In a one-pot three-component heterocyclocondensation process, ethyl 2-methyl-6-arylnicotinates **3a–i** was obtained via the reaction of enaminones **2a–i** with ethyl acetoacetate and ammonium acetate in refluxing acetic acid. Preparation of the nicotinic acid hydrazides **4a–i** in 79%–90% yield was achieved via the hydrazinolysis of ester derivatives **3a–i** with refluxing hydrazine hydrate (Scheme 1).



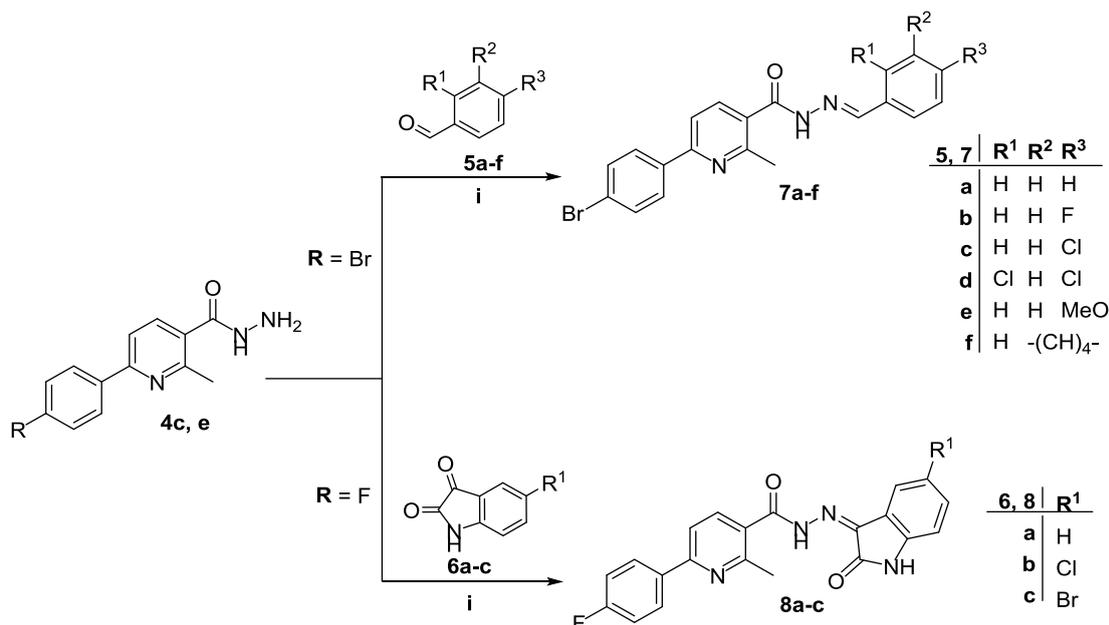
Reagents and conditions: (i) xylene, reflux 8h; (ii) NH₄OAc/AcOH/reflux 5 h; (iii) NH₂NH₂·H₂O/reflux 3 h.

Scheme 1. Synthesis of nicotinic acid hydrazides **4a–i**.

The IR spectra of hydrazides **4a–i** showed absorption bands due to the carbonyl group in the 1635–1664 cm⁻¹ region, in addition to peaks in the region from 3187 to 3414 cm⁻¹, assigned to the NH and NH₂ groups. The ¹H-NMR spectra of **4a–i** showed two singlet D₂O-exchangeable signals attributable to NH and NH₂ protons in the δ 9.58–9.62 and 4.52–4.60 ppm region, respectively, while the methyl (-CH₃) protons appeared as singlets around δ 2.59–2.61 ppm. Furthermore, the ¹³C-NMR spectra of **4f–h** showed signals resonating around δ 167 ppm attributable to the carbons of carbonyl groups, while the carbons of the methyl groups appeared in the δ 23.53–23.59 ppm range.

While the hydrazide **4e** was reacted with different aldehydes **5a–f**, the hydrazide **4c** was reacted with three isatins **6a–c** in ethanol in the presence of a catalytic amount of glacial acetic acid to furnish the target derivatives **7a–f** and **8a–c**, respectively (Scheme 2).

The IR spectra of hydrazides **7a–f** revealed the presence of two peaks in the 3412–3413 and 1635–1654 cm⁻¹ regions, assigned to the NH and carbonyl groups, respectively. The ¹H-NMR spectra of these compounds revealed D₂O-exchangeable signals in the δ 11.80–12.05 ppm region which were assigned to NH protons, in addition to the signal of the methine proton (-CH=N-) in the δ 8.06–8.96 ppm region. Furthermore, the ¹³C-NMR spectra of **7a–f** showed two signals resonating at δ 23.41–23.47 and δ 162.00–170.39 ppm attributable to the methyl (CH₃) and carbonyl (=C-C=O) carbons, respectively.



Reagents and conditions: i, EtOH/AcOH (catalytic)/reflux 4 h.

Scheme 2. Synthesis of nicotinic acid hydrazides **7a–f** and **8a–c**.

2.2. Biological Evaluation

2.2.1. *In Vitro* Antimycobacterial Activity

Biological assays were performed at the Regional Center for Mycology and Biotechnology (RCMB), Al-Azhar University (Cairo, Egypt). Target compounds **4a–i**, **7a–f** and **8a–c** were evaluated *in vitro* for their anti-tubercular activity against *M. tuberculosis* (RCMB 010126) using the microplate Alamar blue assay (MABA) [31]. Isoniazide and pyrazinamide were used as reference drugs. The results as percent inhibition and minimum inhibitory concentration (MIC) are presented in Table 1.

With regard to the issue of activity, valuable information about cell wall structure, host-pathogen interactions and drug targets provide opportunities for rational drug design strategies focused on drug lipophilicity. Increases in lipophilic character, however, result in changes in pathways of diffusion across the cell wall, enhancing the contribution of diffusion through the lipid domain, so increasing the lipophilicity of an antimycobacterial agent enhances its efficacy [32–35]. The calculated lipophilicity (miLogP) values are listed in Table 1.

Drug-likeness model scores were computed for all the compounds using the MolSoft software and are presented in Table 1. Drug-likeness models help to optimize the pharmacokinetic and pharmaceutical properties, for example, solubility, chemical stability, bioavailability and distribution profile of compounds [36]. Counterparts having zero or negative values should not be considered as drug-like candidates. Compounds **7f** and **8a–c** possessed the maximum drug-likeness score ranging from 0.41 to 0.88.

Table 1. Antitubercular activities, LogP measurements and drug-likeness model scores of nicotinic acid hydrazide derivatives.

Compound	Ar	R ¹	R ²	R ³	Mean of Inhibition %	MIC (µg/mL)	LogP ^a	Drug-Likeness Model Score ^b
4a	C ₆ H ₅				42.52 ± 0.63	25	1.36	-0.1
4b	4-CH ₃ C ₆ H ₄				36.33 ± 0.58	25	1.81	-0.36
4c	4-FC ₆ H ₄				NA	NA	1.53	-0.06
4d	4-ClC ₆ H ₄				12.45 ± 0.58	100	2.04	0.05
4e	4-BrC ₆ H ₄				20.63 ± 0.63	50	2.17	-0.25
4f	4-CH ₃ OC ₆ H ₄				42.63 ± 0.16	25	1.42	-0.25
4g	3,4(CH ₃ O) ₂ C ₆ H ₃				22.63 ± 0.16	50	1.01	0.13
4h	3,4,5(CH ₃ O) ₃ C ₆ H ₂				18.32 ± 0.72	50	0.99	0.33
4i	thiophen-2-yl				22.63 ± 0.20	50	1.15	-0.10
7a		H	H	H	NA	NA	4.95	0.12
7b		H	H	F	NA	NA	5.11	0.00
7c		H	H	Cl	NA	NA	5.63	0.09
7d		Cl	H	Cl	13.57 ± 0.72	100	6.23	0.13
7e		H	H	MeO	NA	NA	5.01	0.09
7f		H	-(CH) ₄ -		14.32 ± 0.58	100	6.13	0.46
8a		H			31.44 ± 0.58	25	3.54	0.88
8b		Cl			52.63 ± 0.58	12.5	4.19	0.62
8c		Br			77.42 ± 0.93	6.25	4.32	0.41
Pyrazinamid					93.25 ± 0.63	3.21		
Isoniazide					-	0.75		

^a: Calculated by [37]; ^b: Calculated by [38]; NA: No Activity (>100 µg/mL).

As shown in Table 1, compound **8c** emerged as the most potent analog with good antimycobacterial activity (MIC = 6.25 µg/mL). Compound **8b** also possessed reasonable activity with a MIC value of 12.5 µg/mL. Besides, derivatives **4a**, **4b**, **4f** and **8a** displayed moderate activity (MIC = 25 µg/mL). On the other hand, compounds **4d**, **4e**, **4g–i**, **7d** and **7f** exhibited modest antimycobacterial activity with MIC values ranging from 50 to 100 µg/mL.

2.2.2. Structure Activity Relationships (SAR)

Observing the results, we could deduce valuable data about the structure activity correlations of the tested compounds. Firstly, we explored the impact of substitution of the 4-position of the phenyl group in the first series compounds **4a–i**. Incorporation of an unsubstituted phenyl group led to compound **4a** with good activity against *M. tuberculosis* (MIC = 25 µg/mL). Introduction of a fluorine atom in the

4-position (compound **4c**) led to complete loss of activity, suggesting that the presence of a strongly electron-withdrawing group is not favorable to the activity. Meanwhile, compounds **4d** and **4e** bearing more lipophilic chlorine and bromine substituents at the same position, elicited better activity (MIC = 50 and 100 µg/mL, respectively) than that of analog **4c**. Thence, the order of activities of the halogenated members in the first series, decreased in the order of Br > Cl > F, indicating that the increased lipophilicity is a crucial element for the antitubercular activity. Conversely, substitution at the 4-position with electron-donating groups as methyl and methoxy groups (compounds **4b** and **4f**) retained the activity at MIC = 25 µg/mL. Interestingly, di- and trimethoxy substitutions (compounds **4g** and **4h**) led to decrease in the lipophilicity with subsequent decrease in the antimycobacterial activity (MIC = 50 µg/mL). Furthermore, the bioisosteric replacement of the phenyl group with a 2-thienyl group (compound **4i**) led to a decrease in the activity.

Considering the aldehyde hydrazides of the second series **7a–f**, antimycobacterial activity was only observed in counterparts **7d** and **7f** (MIC = 100 µg/mL). The remaining hydrazides exhibited no activity. Noteworthy, compounds **7d** and **7f** are the most lipophilic analogs in this series with LogP values of 6.23 and 6.13, respectively.

Finally, the effect of substitution of the 5-position of the incorporated isatin moiety in hydrazides **8a–c** was investigated. Compound **8a** with an unsubstituted isatin moiety showed good anti-TB activity (MIC = 25 µg/mL). The introduction of a chlorine substituent at the 5-position (compound **8b**) resulted in increased activity with MIC = 12.50 µg/mL. Moreover, incorporation of a bromine atom (compound **8c**) caused a remarkable increase of activity against *M. tuberculosis* (MIC = 6.25 µg/mL). To summarize, the order of activities of the isatin hydrazides of the third series decreased in the order Br > Cl > H, confirming the importance of lipophilicity for the antimycobacterial activity.

In general, the active members in the second series, compounds **7d** and **7f**, are less active than their parent compound **4e**, indicating that condensation of free hydrazides with aldehydes results in a sharp decrease in activity, whilst, the compounds of the third series are remarkably more potent than their parent compound **4c**, hinting that condensation with isatins greatly enhances the antitubercular activity.

2.2.3. *In Vitro* Cytotoxicity

In vitro cytotoxicity of the most active antitubercular compounds **8a–c** was examined against HT-29 colon cancer and PC-3 prostate cancer cell lines using a sulforhodamine B (SRB) colorimetric assay as described by Skehan *et al.* [39]. Besides, these compounds were evaluated in a previous study for their cytotoxicity against HepG-2 liver cancer, A-549 human lung cancer and MCF-7 breast cancer cell lines [40]. Doxorubicin was included in the experiment as a reference cytotoxic compound. The results were expressed as growth inhibitory concentration (IC₅₀) values which represent the compound concentration required to produce a 50% inhibition of cell growth after 72 h of incubation compared to untreated control (Table 2). Interestingly, none of the tested compounds displayed any significant cytotoxicity, thereby providing a high therapeutic index.

Table 2. Levels of cytotoxicity induced by hydrazides **8a–c** on different cell lines.

Compound	IC ₅₀ (μM)				
	HT-29	PC-3	A549	HepG2	MCF-7
8a	>200	>200	>200	>200	>200
8b	>200	>200	>200	>200	>200
8c	>200	>200	>200	>200	>200
Doxorubicin	7.3 ± 1.11	6.5 ± 1.07	7.6 ± 1.37	6.9 ± 2.05	6.1 ± 1.95

2.3. ADME Study

The ADME of the biologically active counterparts **4a**, **4b**, **4d–i**, **7d**, **7f** and **8a–c** was predicted via a theoretical kinetic study performed by means of the Discovery Studio software (Table 3). Both AlogP98 and PSA_2D descriptors were calculated to evaluate the lipophilicity and polar surface area. Also, solubility, absorption and CYP2D inhibition levels were predicted. Active members of the first series were expected to have good solubility, while compounds **7d** and **7f** showed very low solubility levels and compounds **8a–c** showed low solubility levels. All the examined compounds showed good absorption levels, except compound **7d** that displayed a moderate absorption level. Finally, with exception to compounds **4b–f**, all members were predicted to be CYP2D non-inhibitors.

Table 3. Computer aided ADME study of the active derivatives.

Compound	AlogP98 ^a	PSA_2D ^b	Solubility ^c	Solubility Level ^d	Absorption Level ^e	CYP2D6 ^f	CYP2D6 Probability ^g
4a	1.419	67.912	−2.578	3	0	0	0.306
4b	1.905	67.912	−3.087	3	0	1	0.772
4d	2.083	67.912	−3.496	3	0	1	0.831
4e	2.167	67.912	−3.57	3	0	1	0.712
4f	1.402	76.842	−2.838	3	0	1	0.722
4g	1.386	85.772	−3.101	3	0	0	0.366
4h	1.37	94.702	−3.353	3	0	0	0.316
4i	1.145	67.912	−2.546	3	0	0	0.158
7d	5.839	52.695	−6.834	1	1	0	0.287
7f	5.419	52.695	−6.647	1	0	0	0.306
8a	3.048	82.806	−4.835	2	0	0	0.485
8b	3.713	82.806	−5.608	2	0	0	0.415
8c	3.797	82.806	−5.682	2	0	0	0.495

^a: Lipophilicity descriptor; ^b: Polar surface area; ^c: Solubility parameter. (0~−2 = optimal, −2~−4 = good, −4~−6 = low, −6~−8 = very low); ^d: Solubility level. (0 = extremely low, 1 = very low but possible, 2 = low, 3 = good, 4 = optimal); ^e: Absorption level. (0 = good, 1 = moderate, 2 = low, 3 = very low); ^f: CYP2D inhibition. (0 = non inhibitor, 1 = inhibitor); ^g: CYP2D6 Probability: 0–0.5 = non inhibitor; 0.5–1 = inhibitor.

3. Experimental Section

3.1. General Information

Melting points were measured with a Stuart melting point apparatus and are uncorrected. Infrared (IR) spectra were recorded as KBr disks using a Perkin Elmer FT-IR Spectrum BX apparatus (Akron, OH, USA). NMR Spectra were recorded on a Bruker AV-500 MHz NMR spectrometer (Billerica, MA, USA). ¹H-NMR spectra were run at 500 MHz and ¹³C spectra was run at 125 MHz in deuterated dimethylsulfoxide (DMSO-*d*₆). Chemical shifts are expressed in δ values (ppm) using the solvent peak as internal standard. All coupling constants (*J*) values are given in Hertz. The abbreviations used are as follows: s, singlet; d, doublet; m, multiplet. Microanalyses were carried out using Perkin Elmer PE 2400 CHN Elemental Analyzer and the results were within $\pm 0.4\%$. Reaction courses and product mixtures were routinely monitored by thin layer chromatography (TLC) on silica gel precoated F₂₅₄ plates Merck (Merck KGaA, Darmstadt, Germany). Unless otherwise noted, all solvents and reagents were commercially available and used without further purification.

3.2. Synthesis

3.2.1. Ethyl 2-methyl-6-arylnicotinates **3a–i**

To a solution of the appropriate enaminone **2a–i** (5 mmol) in glacial acetic acid (15 mL), ethyl acetoacetate (5.5 mmol) and ammonium acetate (40 mmol) were added. The reaction mixture was heated under reflux for 5 h. After cooling and pouring into ice-water, the residue obtained was filtered and washed with petroleum ether then with water and finally crystallized from ethanol [41]. The yields of compounds **3a–i** were 80%, 77%, 81%, 84%, 86%, 80%, 78, 76%, 83%, respectively.

3.2.2. 6-Aryl-2-methylnicotinohydrazides **4a–i**

A mixture of the appropriate ester **4a–i** (5 mmol) and 99% hydrazine hydrate (5 mL) was refluxed for 3 h. The solid product obtained upon cooling was filtered off and recrystallized from dioxan to afford the corresponding 6-aryl-2-methylnicotinohydrazides **4a–i**, respectively. The physical properties and spectral data of **4a–d** and **4i** were identical with those reported [40,42]. The yields of **4a–d** and **4i** were 83, 78, 84, 86 and 84%, respectively.

6-(4-Bromophenyl)-2-methylnicotinohydrazide (4e). White crystals (yield 90%), m.p. 213–215 °C; IR (KBr, ν cm⁻¹): 3194, 3294 (NH, NH₂) and 1643 (C=O); ¹H-NMR δ ppm: 2.61 (s, 3H, CH₃), 4.60 (s, 2H, NH₂, D₂O exchangeable), 7.56 (d, *J* = 8.5 Hz, 2H, Ar-H), 7.79 (d, *J* = 8.1 Hz, 1H, H-4 pyridine), 7.87 (d, *J* = 8.1 Hz, 1H, H-5 pyridine), 8.14 (d, *J* = 8.5 Hz, 2H, Ar-H), 9.62 (s, 1H, NH, D₂O exchangeable); Anal. calcd. for C₁₃H₁₂BrN₃O (306.16): C, 51.00; H, 3.95; N, 13.72. Found C, 51.13; H, 4.10; N, 13.78.

6-(4-Methoxyphenyl)-2-methylnicotinohydrazide (4f). White crystals (yield 81%), m.p. 195–197 °C; IR (KBr, ν cm⁻¹): 3292, 3350 (NH, NH₂) and 1635 (C=O); ¹H-NMR δ ppm: 2.59 (s, 3H, CH₃), 3.83 (s, 3H, OCH₃), 4.52 (s, 2H, NH₂, D₂O exchangeable), 7.05 (d, *J* = 8.5 Hz, 2H, Ar-H), 7.70 (d, *J* = 8.1 Hz,

1H, H-4 pyridine), 7.73 (d, $J = 8.1$ Hz, 1H, H-5 pyridine), 8.07 (d, $J = 8.5$ Hz, 2H, Ar-H), 9.58 (s, 1H, NH, D₂O exchangeable); ¹³C-NMR δ ppm: 23.53, 55.72, 114.61, 116.55, 128.54, 129.00, 130.96, 136.78, 155.89, 156.03, 160.85, 167.78; Anal. calcd. for C₁₄H₁₅N₃O₂ (257.12): C, 65.35; H, 5.88; N, 16.33. Found C, 65.54; H, 5.93; N, 16.53.

6-(3,4-Dimethoxyphenyl)-2-methylnicotinohydrazide (4g). White crystals (yield 85%), m.p. 204–205 °C; IR (KBr, ν cm⁻¹): 3187, 3414 (NH, NH₂) and 1648 (C=O); ¹H-NMR δ ppm: 2.60 (s, 3H, CH₃), 3.83 (s, 3H, OCH₃), 3.86 (s, 3H, OCH₃), 4.53 (s, 2H, NH₂, D₂O exchangeable), 7.06 (d, $J = 8.5$ Hz, 1H, Ar-H), 7.68–7.74 (m, 3H, Ar-H), 7.81 (d, $J = 8.5$ Hz, 1H, Ar-H), 9.59 (s, 1H, NH, D₂O exchangeable); ¹³C-NMR δ ppm: 23.59, 56.03, 110.41, 112.19, 116.81, 119.95, 128.76, 131.15, 136.71, 149.37, 150.57, 155.83, 156.09, 167.77; Anal. calcd. for C₁₅H₁₇N₃O₃ (287.13): C, 62.71; H, 5.96; N, 14.63; O, 16.71, Found C, 62.84; H, 6.11; N, 14.89.

2-Methyl-6-(3,4,5-trimethoxyphenyl)nicotinohydrazide (4h). White crystals (yield 85%), m.p. 185–187 °C; IR (KBr, ν cm⁻¹): 3282, 3310 (NH, NH₂) and 1664 (C=O); ¹H-NMR δ ppm: 2.61 (s, 3H, CH₃), 3.73 (s, 3H, OCH₃), 3.87 (s, 6H, OCH₃), 4.55 (s, 2H, NH₂, D₂O exchangeable), 7.41 (s, 2H, Ar-H), 7.75 (d, $J = 8.0$ Hz, 1H, H-4 pyridine), 7.89 (d, $J = 8.0$ Hz, 1H, H-5 pyridine), 9.61 (s, 1H, NH, D₂O exchangeable); ¹³C-NMR δ ppm: 23.58, 56.46, 60.57, 104.53, 117.51, 129.31, 134.02, 136.73, 139.23, 153.61, 155.84, 155.92, 167.66; Anal. calcd. for C₁₆H₂₃N₃O₄ (321.17): C, 59.80; H, 7.21; N, 13.08. Found C, 60.09; H, 7.43; N, 13.22.

3.2.3. General Procedure for Synthesis of *N'*-arylidene-6-(4-bromophenyl)-2-methylnicotinyl Hydrazides **7a–f**

To a stirred solution of the hydrazide **4e** (5 mmol) in hot ethanol (20 mL), aldehydes **5a–f** (5 mmol) and catalytic amount of glacial acetic acid were added. The reaction mixture was heated under reflux for 4 h. The precipitate formed was collected by filtration while hot, washed with hot ethanol, dried and crystallized from ethanol/DMF to afford compounds **7a–f** in 65%–80% yield.

N'-Benzylidene-6-(4-bromophenyl)-2-methylnicotinohydrazide (**7a**). White crystals (yield 75%), m.p. 270–272 °C; IR (KBr, ν cm⁻¹): 3410 (NH), 1635 (C=O); ¹H-NMR δ ppm: 2.67 (s, 3H, CH₃), 7.31–7.49 (m, 4H, Ar-H), 7.72–7.75 (m, 3H, Ar-H), 7.93–8.13 (m, 4H, Ar-H), 8.33 (s, 1H, CH=N), 11.94 (s, D₂O exch., 1H, –CONH–); ¹³C-NMR δ ppm: 23.47, 117.59, 123.67, 127.11, 127.65, 129.27, 130.73, 132.22, 132.26, 134.58, 137.39, 137.50, 148.39, 155.48, 156.47, 164.38; Anal. calcd. for C₂₀H₁₆BrN₃O (393.05): C, 60.93; H, 4.09; N, 10.66. Found C, 61.07; H, 4.30; N, 10.93.

6-(4-Bromophenyl)-N'-(4-fluorobenzylidene)-2-methylnicotinohydrazide (7b). White crystals (yield 65%), m.p. 252–254 °C; IR (KBr, ν cm⁻¹): 3413 (NH), 1647 (C=O); ¹H-NMR δ ppm: 2.66 (s, 3H, CH₃), 7.31 (t, $J = 8.25$ Hz, 2H, Ar-H), 7.72 (d, $J = 8.5$ Hz, 2H, Ar-H), 7.80–7.97 (m, 4H, Ar-H), 8.11–8.13 (m, 2H, Ar-H), 8.33 (s, 1H, CH=N), 11.95 (s, D₂O exch., 1H, –CONH–); ¹³C-NMR δ ppm: 23.41, 116.34 (² $J_{F-C} = 22.5$ Hz), 117.25, 117.58, 123.68, 129.21 (³ $J_{F-C} = 7.5$ Hz), 129.64, 130.48, 131.21, 132.21, 137.38, 143.79, 147.26, 154.75, 162.70 (¹ $J_{F-C} = 211.3$ Hz), 170.39; Anal. calcd. for C₂₀H₁₅BrFN₃O (411.04): C, 58.27; H, 3.67; N, 10.19. Found C, 58.38; H, 3.83; N, 10.25.

6-(4-Bromophenyl)-N'-(4-chlorobenzylidene)-2-methylnicotinohydrazide (**7c**). White crystals (yield 80%), m.p. 265–267 °C; IR (KBr, ν cm^{-1}): 3413 (NH), 1654 (C=O); $^1\text{H-NMR}$ δ ppm: 2.66 (s, 3H, CH_3), 7.20 (d, $J = 8.50$ Hz, 1H, Ar-H), 7.54 (d, $J = 8.50$ Hz, 1H, Ar-H), 7.72 (d, $J = 8.50$ Hz, 2H, Ar-H), 7.77 (d, $J = 8.50$ Hz, 2H, Ar-H), 7.98 (d, $J = 4.5$ Hz, 2H, Ar-H), 8.11 (d, $J = 8.5$ Hz, 2H, Ar-H), 8.32 (s, 1H, $\text{CH}=\text{N}$), 12.01 (s, D_2O exch., 1H, $-\text{CONH}-$); $^{13}\text{C-NMR}$ δ ppm: 23.47, 114.00, 117.59, 129.28, 129.46, 132.26 (2C), 135.16 (2C), 136.50, 137.41, 137.52, 148.00, 152.00, 155.52, 168.00; Anal. calcd. for $\text{C}_{20}\text{H}_{15}\text{BrClN}_3\text{O}$ (427.01): C, 56.03; H, 3.53; N, 9.80. Found C, 56.09; H, 3.76; N, 9.92.

6-(4-Bromophenyl)-N'-(2,4-dichlorobenzylidene)-2-methylnicotinohydrazide (**7d**). White crystals (yield 70%), m.p. 269–271 °C; IR (KBr, ν cm^{-1}): 3413 (NH), 1650 (C=O); $^1\text{H-NMR}$ δ ppm: 2.67 (s, 3H, CH_3), 7.55 (d, $J = 8.50$ Hz, 1H, Ar-H), 7.54 (d, $J = 8.50$ Hz, 1H, Ar-H), 7.72 (d, $J = 8.50$ Hz, 2H, Ar-H), 7.96 (d, $J = 6.50$ Hz, 1H, Ar-H), 8.02 (d, $J = 8.5$ Hz, 1H, Ar-H), 8.05 (d, $J = 8.5$ Hz, 1H, Ar-H), 8.11 (d, $J = 8.5$ Hz, 2H, Ar-H), 7.68 and 8.45 (s, 1H, Ar-H), 8.67 (s, 1H, $\text{CH}=\text{N}$), 12.01 (s, D_2O exch., 1H, $-\text{CONH}-$); Anal. calcd. For $\text{C}_{20}\text{H}_{14}\text{BrCl}_2\text{N}_3\text{O}$ (463.15): C, 51.86; H, 3.05; N, 9.07. Found C, 51.95; H, 3.12; N, 9.17.

6-(4-Bromophenyl)-N'-(4-methoxybenzylidene)-2-methylnicotinohydrazide (**7e**). White crystals (yield 75%), m.p. 251–253 °C; IR (KBr, ν cm^{-1}): 3413 (NH), 1652 (C=O); $^1\text{H-NMR}$ δ ppm: 2.66 (s, 3H, CH_3), 3.74 and 3.83 (s, 3H, $-\text{OCH}_3$), 7.04 (d, $J = 8.50$ Hz, 2H, Ar-H), 7.69 (d, $J = 8.50$ Hz, 2H, Ar-H), 7.72 (d, $J = 8.50$ Hz, 2H, Ar-H), 7.92–7.98 (m, 2H, Ar-H), 8.11 (d, $J = 8.50$ Hz, 2H, Ar-H), 8.26 (s, 1H, $\text{CH}=\text{N}$), 11.80 (s, D_2O exch., 1H, $-\text{CONH}-$); $^{13}\text{C-NMR}$ δ ppm: 23.44, 55.79, 114.85, 117.58, 127.00, 129.00, 129.26 (2C), 132.25 (2C), 135.50, 137.00, 137.34, 138.00, 148.26, 156.41, 162.00; Anal. calcd. for $\text{C}_{21}\text{H}_{18}\text{BrN}_3\text{O}_2$ (423.06): C, 59.45; H, 4.28; N, 9.90; Found C, 59.69; H, 4.41; N, 10.12.

6-(4-Bromophenyl)-2-methyl-N'-(naphthalen-2-ylmethylene)nicotinohydrazide (**7f**). White crystals (yield 73%), m.p. 273–275 °C; IR (KBr, ν cm^{-1}): 3413 (NH), 1648 (C=O); $^1\text{H-NMR}$ δ ppm: 2.71 (s, 3H, CH_3), 7.20–8.95 (m, 14H, Ar-H), 8.96 (s, 1H, $\text{CH}=\text{N}$), 12.03 (s, D_2O exch., 1H, $-\text{CONH}-$); Anal. calcd. for $\text{C}_{24}\text{H}_{18}\text{BrN}_3\text{O}$ (443.06): C, 64.88; H, 4.08; N, 9.46. Found C, 65.07; H, 4.37; N, 9.65.

3.2.4. General Procedure for Preparation of Target Compounds **8a–c**

Indoline-2,3-dione derivative **6a–c** (1 mmol) was added to a suspension of 6-(4-fluorophenyl)-2-methylnicotinohydrazide (**4c**, 1 mmol) in ethanol (10 mL) and a catalytic amount of glacial acetic acid. The reaction mixture was refluxed for 4 h. The precipitate formed was collected by filtration while hot, washed with hot ethanol, dried and crystallized from ethanol/DMF to furnish compounds **8a–c**. The physical properties and spectral data of **8a–c** were identical with those reported [40]. The yields of compounds **4a–c** ranged from 75%–80%.

3.3. Biological Evaluation

3.3.1. Antimycobacterial Activity

The *M. tuberculosis* (RCMB 010126) strain was provided from culture collection of the Regional Center for Mycology and Biotechnology (RCMB), Al-Azhar University (Cairo, Egypt). Isoniazide and pyrazinamide were used as reference drugs. Antimycobacterial activity of the synthesized compounds was evaluated using the microplate Alamar blue assay (MABA) which was performed in black, clear-bottomed, 96 well microplates (in order to minimize background effects). Outer perimeter wells were filled with sterile water to prevent dehydration in experimental wells. Initial compounds dilutions were prepared in dimethyl sulfoxide and subsequent twofold dilutions were performed in the microplates. 0.1 ml of 10^5 CFU/mL *Mycobacterium tuberculosis* inoculum was added to wells, additional control wells consisted of bacteria only (B). Plates were incubated at 37 °C. Starting at day 4 of incubation, 20 µL of alamarBlue solution (Alamar Biosciences/Accumed, Westlake, OH, USA) and 12.5 µL of 20% Tween 80 were added to the entire plate. Plates were then incubated at 37 °C, and results were recorded at 24 h post-reagent addition at 590 nm. Percent inhibition was defined as: $1 - (\text{mean of test well}/\text{mean of B wells}) \times 100$. Visual MICs were defined as the lowest concentration of drug that prevented a color change.

3.3.2. In Vitro Cytotoxic Activity

HT-29 colon cancer and PC-3 prostate cancer cell lines were obtained from the National Cancer Institute (Cairo, Egypt). HT-29 and PC-3, cells were grown in RPMI-1640. Media were supplemented with 10% heat-inactivated FBS, 50 units/mL of penicillin and 50 g/mL of streptomycin and maintained at 37 °C in a humidified atmosphere containing 5% CO₂. The cells were maintained as a “monolayer culture” by serial subculturing. Cytotoxicity was determined using the SRB method as previously described by Skehan *et al.* [39]. Exponentially growing cells were collected using 0.25% trypsin-EDTA and seeded in 96-well plates at 1000–2000 cells/well in supplemented DMEM medium. After 24 h, cells were incubated for 72 h with various concentrations of the tested compounds as well as doxorubicin as the reference compound. Following 72 h of treatment, the cells were fixed with 10% trichloroacetic acid for 1 h at 4 °C. Wells were stained for 10 min at room temperature with 0.4% SRB dissolved in 1% acetic acid. The plates were air dried for 24 h, and the dye was solubilized with Tris-HCl for 5 min on a shaker at 1600 rpm. The optical density (OD) of each well was measured spectrophotometrically at 564 nm with an ELISA microplate reader (ChroMate-4300, Palm City, FL, USA). The IC₅₀ values were calculated according to the equation for Boltzmann sigmoidal concentration-response curve using the nonlinear regression models (Graph Pad, Prism Version 5). The results reported are means of at least three separate experiments. Significant differences were analyzed by one-way ANOVA wherein the differences were considered to be significant at $p < 0.05$.

4. Conclusions

In summary, we have synthesized eighteen derivatives based on the nicotinic acid hydrazide scaffold and evaluated their antimycobacterial activity. From the obtained results, it was obvious that

condensation of free hydrazides with aldehydes sharply decreased the activity, while isatin hydrazides enhanced the activity. Compounds **8b** and **8c** emerged as the most potent counterparts among all the tested compounds (MIC = 12.5 and 6.25 µg/mL, respectively). They also displayed good drug-likeness scores of 0.62 and 0.41, respectively. The cytotoxicity of hydrazides **8a–c** was evaluated against HT-29, PC-3, A549, HepG2 and MCF-7 cancer cell lines. None of the tested hydrazides exhibited cytotoxicity up to 200 µM. The importance of lipophilicity for the antimycobacterial activity was explored via a SAR study. Finally, a theoretical kinetic study was established to predict the ADME of the active derivatives.

Acknowledgments

Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Egyptian Russian University, Cairo, Egypt, is highly appreciated for supporting this research. The authors would like to extend their sincere appreciation to the Deanship of Scientific Research at King Saud University for its funding of this research through the Research Group Project no. RGP-VPP-321.

Author Contributions

Hatem A. Abdel-Aziz and Wagdy M. Eldehna formulated the research idea and participated in the preparation of manuscript; Wagdy M. Eldehna and Mohamed Fares carried out the experimental, interpreted the data and prepared the manuscript; Marwa M. Abdel-Aziz performed the biological screening. All authors have read and approved the final manuscript.

Conflicts of Interest

The authors declare no conflict of interest.

References

1. World Health Organization. *Global Tuberculosis Report*; WHO: Geneva, Switzerland, 2013.
2. Zumla, A.; Nahid, P.; Cole, S.T. Advances in the development of new tuberculosis drugs and treatment regimens, *Nat. Rev. Drug Discov.* **2013**, *12*, 388–404.
3. Goldman, R.C.; Plumley, K.V.; Laughon, B.E. The evolution of extensively drug resistant tuberculosis (XDR-TB): History, status and issues for global control. *Infect. Disord. Drug Targets* **2007**, *7*, 73–91.
4. Benatar, S.R. Extensively drug resistant tuberculosis—Problem will get worse in South Africa unless poverty is alleviated. *Br. Med. J.* **2006**, *333*, doi:10.1136/bmj.333.7570.705-a.
5. Lawn, S.D.; Wilkinson, R. Extensively drug resistant tuberculosis—A serious wake-up call for global health. *Br. Med. J.* **2006**, *333*, 559–560.
6. Dahle, U.R. Extensively drug resistant tuberculosis—Beware patients lost to follow-up. *Br. Med. J.* **2006**, *333*, doi:10.1136/bmj.333.7570.705.
7. Gandhi, N.R.; Moll, A.; Sturm, A.W.; Pawinski, R.; Govender, T.; Lalloo, U.; Zeller, K.; Andrews, J.; Friedland, G. Extensively drug-resistant tuberculosis as a cause of death in patients co-infected with tuberculosis and HIV in a rural area of South Africa. *Lancet* **2006**, *368*, 1575–1580.

8. Manissero, D.; Fernandez, K. Extensive drug-resistant TB: A threat for Europe? *Euro Surveill.* **2006**, *11*, E060928.
9. World Health Organization. *Treatment of Tuberculosis Guidelines*, 4th ed.; WHO: Geneva, Switzerland, 2010.
10. Mitnick, C.D.; McGee, B.; Peloquin, C.A. Tuberculosis pharmacotherapy: Strategies to optimize patient Care. *Expert. Opin. Pharmacother.* **2009**, *10*, 381–401.
11. Domagk, G.; Offe, H.A.; Siefken, W. Ein weiterer Beitrag zur experimentellen Chemotherapie der Tuberkulose (Neoteben). *Dtsch. Med. Wochenschr.* **1952**, *77*, 573–578.
12. Janin, Y.L. Antituberculosis drugs: Ten years of research. *Bioorg. Med. Chem.* **2007**, *15*, 2479–2513.
13. Boechat, N.; Ferreira, V.F.; Ferreira, S.B.; Ferreira, M.G.; de C. da Silva, F.; Bastos, M.M.; dos S. Costa, M.; Lourenc, M.C.S.; Pinto, A.C.; Krettli, A.U.; *et al.* Novel 1,2,3-triazole derivatives for use against *Mycobacterium tuberculosis* H37Rv (ATCC 27294) strain. *J. Med. Chem.* **2011**, *54*, 5988–5999.
14. Sandy, J.; Mushtaq, A.; Kawamura, A.; Sinclair, J.; Sim, E.; Noble, M. The Structure of Arylamine *N*-acetyltransferase from *Mycobacterium smegmatis*—An Enzyme which Inactivates the Antitubercular Drug, Isoniazid. *J. Mol. Biol.* **2002**, *318*, 1071–1083.
15. Lourenço, M.C.S.; Ferreirab, M.L.; Souza, M.V.N.; Peralta, M.A.; Vasconcelos, T.R.A.; das Graças, M.; Henriques, M.O. Synthesis and antimycobacterial activity of (*E*)-*N'* (monosubstituted-benzylidene)isonicotinohydrazide derivatives. *Eur. J. Med. Chem.* **2008**, *43*, 1344–1347.
16. Cardoso, S.H.; de Assis, J.V.; de Almeida, M.V.; Lourenço, M.C.S.; Vicente, F.R.C.; de Souza, M.V.N. Synthesis and antitubercular activity of isoniazid condensed with carbohydrate derivatives. *Quim. Nova* **2009**, *32*, 1557–1560.
17. Bottari, B.; Maccari, R.; Monforte, F.; Ottana, R.; Rotondo, E.; Vigorita, M.G. Isoniazid-related copper(II) and nickel(II) complexes with antimycobacterial *in vitro* activity. part 9. *Bioorg. Med. Chem. Lett.* **2000**, *10*, 657–660.
18. Roye, W.E.; Ewart, G.E. A Preliminary Report on the Safety and Therapeutic Activity of a Salizid INH Derivative. *Dis. Chest* **1958**, *33*, 261–267.
19. Pershin, G.N.; Makeeva, O.O. Chemotherapeutic effect of phthivazide in experimental tuberculosis. *Probl. Tuberk.* **1953**, *2*, 16–20.
20. Rubbo, S.D.; Edgar, J.; Vaughan, G. Chemotherapy of tuberculosis. I. Antituberculous activity of verazide and related hydrazones. *Am. Rev. Tuberc.* **1957**, *76*, 331–345.
21. Soldatov, V.E. Saluzide in the treatment of tuberculous meningitis in adults. *Probl. Tuberk.* **1955**, *6*, 16–21.
22. Judge, V.; Narasimhan, B.; Ahuja, M.; Sriram, D.; Yogeeswari, P.; Clercq, E.D.; Pannecouque, C.; Balzarini, J. Synthesis, antimycobacterial, antiviral, antimicrobial activities, and QSAR studies of isonicotinic acid-1-(substituted phenyl)-ethylidene/cycloheptylidene hydrazides. *Med. Chem. Res.* **2012**, *21*, 1935–1952.
23. Narang, R.; Narasimhan, B.; Sharma, S.; Sriram, D.; Yogeeswari, P.; Clercq, E.D.; Pannecouque, C.; Balzarini, J. Synthesis, antimycobacterial, antiviral, antimicrobial activities, and QSAR studies of nicotinic acid benzylidene hydrazide derivatives. *Med. Chem. Res.* **2012**, *21*, 1557–1576.

24. Eswaran, S.; Adhikari, A.V.; Pal, N.K.; Chowdhury, I.H. Design and synthesis of some new quinoline-3-carbohydrazone derivatives as potential antimycobacterial agents. *Bioorg. Med. Chem. Lett.* **2010**, *20*, 1040–1044.
25. Thomas, K.D.; Adhikari, A.V.; Telkar, S.; Chowdhury, I.H.; Mahmood, R.; Pal, N.K.; Rowd, G.; Sumesh, E. Design, synthesis and docking studies of new quinoline-3-carbohydrazone derivatives as antitubercular agents, *Eur. J. Med. Chem.* **2011**, *46*, 5283–5292.
26. Aboul-Fadl, T.; Bin-Jubair, F.A.S.; Aboul-Wafa, O. Schiff bases of indoline-2,3-dione (isatin) derivatives and nalidixic acid carbohydrazone, synthesis, antitubercular activity and pharmacophoric model building. *Eur. J. Med. Chem.* **2010**, *45*, 4578–4586.
27. Feng, L.; Liu, M.; Zhang, S.; Chai, Y.; Wang, B.; Zhang, Y.; Lv, K.; Guan, Y.; Guo, H.; Xiao, C. Synthesis and *in vitro* antimycobacterial activity of 8-OCH₃ ciprofloxacin methylene and ethylene isatin derivatives. *Eur. J. Med. Chem.* **2011**, *46*, 341–348.
28. Feng, L.; Liu, M.; Wang, B.; Chai, Y.; Hao, X.; Meng, S.; Guo, H. Synthesis and *in vitro* antimycobacterial activity of balofloxacin ethylene isatin derivatives. *Eur. J. Med. Chem.* **2010**, *45*, 3407–3412.
29. Sriram, D.; Yogeewari, P.; Basha, J.S.; Radhab, D.R.; Nagaraja, V. Synthesis and antimycobacterial evaluation of various 7-substituted ciprofloxacin derivatives. *Bioorg. Med. Chem.* **2005**, *13*, 5774–5778.
30. AbouI-Fadl, T.; Mohammed, F.A.; Hassan, E.A. Synthesis, antitubercular activity and pharmacokinetic studies of some schiff bases derived from 1-alkylisatin and isonicotinic acid hydrazide (INH). *Arch. Pharm. Res.* **2003**, *26*, 778–784.
31. Collins, L.A.; Franzblau, S.G. Microplate Alamar Blue Assay versus BACTEC 460 System for High-Throughput Screening of Compounds against *Mycobacterium tuberculosis* and *Mycobacterium avium*. *Antimicrob. Agents Chemother.* **1997**, *41*, 1004–1009.
32. Christensen, H.; Garton, N.; Harobin, R.; Minnikin, D.E.; Barer, M.R. Lipid domains of mycobacteria studied with fluorescent molecular probes. *Mol. Microbiol.* **1999**, *31*, 1561–1572.
33. Maccari, R.; Ottana, R.; Vigorita, M.G. *In vitro* advanced antimycobacterial screening of isoniazid-related hydrazones; hydrazides and cyanoboranes: Part 14. *Bioorg. Med. Chem. Lett.* **2005**, *15*, 2509–2513.
34. Hearn, M.J.; Cynamon, M.H. Design and synthesis of antituberculars: Preparation and evaluation against *Mycobacterium tuberculosis* of an isoniazid Schiff base. *J. Antimicrob. Chemother.* **2004**, *53*, 185–191.
35. Rodrigues, M.O.; Cantos, J.B.; D'Oca, C.R.; Soares, K.L.; Coelho, T.S.; Piovesan, L.A.; Russowsky, D.; da Silva, P.A.; D'Oca, M.G. Synthesis and antimycobacterial activity of isoniazid derivatives from renewable fatty acids. *Bioorg. Med. Chem.* **2005**, *21*, 6910–6914.
36. Vistoli, G.; Pedretti, A.; Testa, B. Assessing drug-likeness—What are we missing? *Drug Discov. Today* **2008**, *13*, 285–294.
37. Molinspiration Cheminformatics. Available online: <http://www.molinspiration.com> (accessed on 15 May 2015).
38. Drug-Likeness and molecular property prediction. Available online: <http://www.molsoft.com/mprop/> (accessed on 15 May 2015).

39. Skehan, P.; Storeng, R.; Scudiero, D.; Monks, A.; McMahon, J.; Vistica, D.; Warren, J.T.; Bokesch, H.; Kenney, S.; Boyd, M.R. New colorimetric cytotoxicity assay for anticancer-drug screening. *J. Natl. Cancer Inst.* **1990**, *82*, 1107–1112.
40. Eldehna, W.M.; Altoukhy, A.; Mahrous, H.; Abdel-Aziz, H.A. Design, synthesis and QSAR study of certain isatin-pyridine hybrids as potential anti-proliferative agents. *Eur. J. Med. Chem.* **2015**, *90*, 684–694.
41. Al-Saleh, B.; Abdel-Khalik, M.M.; Eltoukhy, A.M.; Elnagdi, M.H. Enaminones in Heterocyclic Synthesis: A New Regioselective Synthesis of 2,3,6-Trisubstituted Pyridines, 6-Substituted-3-Aroylpyridines and 1,3,5-Triaroylbenzenes. *J. Heterocycl. Chem.* **2002**, *39*, 1035–1038.
42. Abdel-Aziz, H.A.; Aboul-Fadl, T.; Al-Obaid, A.M.; Ghazzali, M.; Al-Dhfyhan, A.; Contini, A. Design, Synthesis and Pharmacophoric Model Building of Novel Substituted Nicotinic Acid Hydrazones with Potential Antiproliferative Activity. *Arch. Pharm. Res.* **2012**, *35*, 1543–1552.

Sample Availability: Not available.

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