

Article

## Two New Chiratane-Type Triterpenoids from *Swertia kouitchensis*

Luo-Sheng Wan <sup>1</sup>, Ting-Ting Liu <sup>2</sup>, Xiao-Jun Lin <sup>3</sup>, Qiu-Xia Min <sup>1</sup> and Jia-Chun Chen <sup>1,4,\*</sup>

<sup>1</sup> Hubei Key Laboratory of Natural Medicinal Chemistry and Resource Evaluation, School of Pharmacy, Tongji Medical College, Huazhong University of Science and Technology, Wuhan 430030, China

<sup>2</sup> The First Affiliated Hospital, College of Medicine, Zhejiang University, Hangzhou 310003, China

<sup>3</sup> Shantou Institute for Drug Control, Shantou 515041, China

<sup>4</sup> State Clinical Research Center of TCM, Hubei 430061, China

\* Author to whom correspondence should be addressed; E-Mail: homespringchen@mail.hust.edu.cn; Tel./Fax: +86-27-8369-2793.

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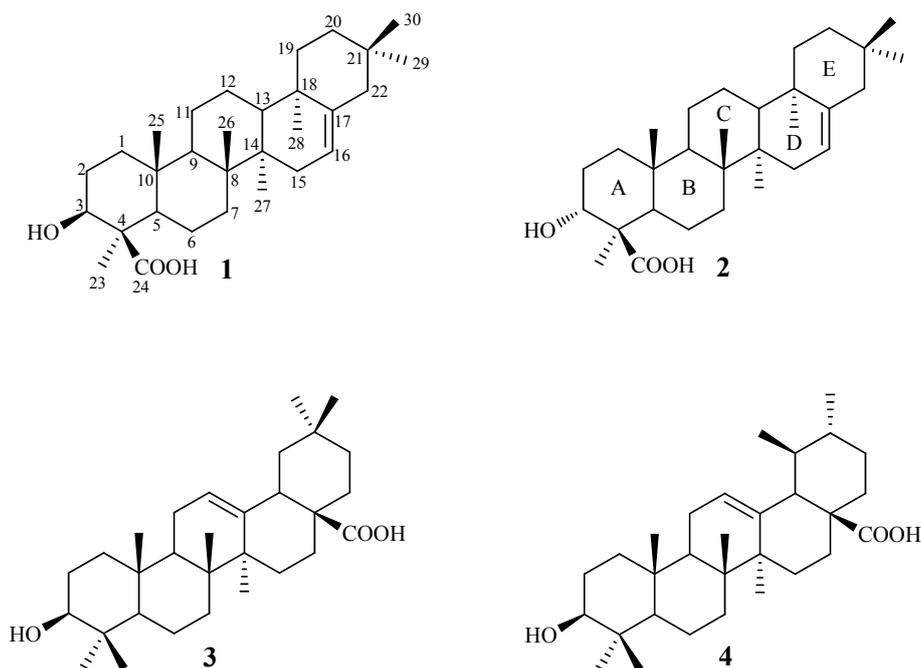
**Abstract:** Two rare new chiratane-type triterpenoids, kouitchenoids A and B (**1**, **2**), together with oleanolic acid (**3**) and ursolic acid (**4**), were isolated from ethanol extract of *Swertia kouitchensis*. The new structures were determined by the analysis of MS and NMR data. In addition, compounds **1–4** showed moderate inhibitory activity against the  $\alpha$ -glucosidase (with IC<sub>50</sub> values ranging from 1,812 to 2,027  $\mu$ M).

**Keywords:** *Swertia kouitchensis*; diabetes; chiratane; triterpenoid;  $\alpha$ -glucosidase

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### 1. Introduction

*Swertia kouitchensis* Franch. (Gentianaceae), widely distributed in China, has been used for the treatment of hepatitis and diabetes [1,2]. Previous study revealed that the ethanol extract of *S. kouitchensis* showed  $\alpha$ -glucosidase inhibitory effect [3]. Thus, we initiated a study on the subject. As a result, two rare chiratane-type triterpenoids, kouitchenoids A and B (**1** and **2**), along with oleanolic acid (**3**) [4] and ursolic acid (**4**) (Figure 1) [5] were isolated and identified. All of these compounds were evaluated for their inhibitory activities against  $\alpha$ -glucosidase. Described herein are the isolation, structure elucidation, and biological activities of these compounds.

**Figure 1.** Structures of compounds 1–4.

## 2. Results and Discussion

The 95% ethanol extract of *S. kouitchensis* whole plants was suspended in water and successively partitioned with petroleum ether,  $\text{CH}_2\text{Cl}_2$ , EtOAc and *n*-butanol. The  $\text{CH}_2\text{Cl}_2$  fraction was subjected to column chromatography and partitioned as described in the Experimental section to afford two new triterpenoids **1** and **2**, along with two known compounds **3** and **4**.

Compound **1** was obtained as white amorphous powder, gave a molecular formula of  $\text{C}_{30}\text{H}_{48}\text{O}_3$  by HRESIMS ( $m/z$  455.3525  $[\text{M}-\text{H}]^-$ , calcd. for  $\text{C}_{30}\text{H}_{47}\text{O}_3$ , 455.3531). Its IR spectrum exhibited absorptions at 3,453 and  $1,696\text{ cm}^{-1}$ , assignable to hydroxyl and carboxyl groups, respectively. The  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra of **1** (Table 1) closely resembled those of chiratenol [6,7], except for the appearance of a carboxyl carbon signal at  $\delta_{\text{C}}$  181.0 instead of a methyl (C-24) signal of chiratenol, and the downfield shift (+10.8 ppm) for C-4. The position of this carboxyl group was confirmed by HMBC spectrum (Figure 2), in which cross-peaks were observed between H-3 ( $\delta_{\text{H}}$  3.41) and C-2, C-4, C-23, and C-24, so that the carboxyl group is assigned to C-24. And also, H-3 was assigned as  $\alpha$ -orientation by the NOESY correlation (Figure 2) between H-3 and H-5. Therefore, **1** was deduced to be 3 $\beta$ -hydroxy-chirat-16-en-24-oic acid, and named kouitchenoid A.

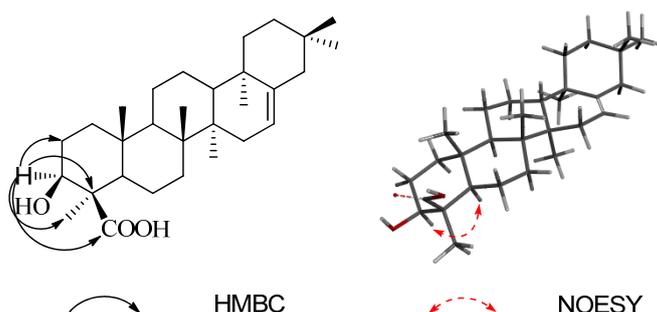
Compound **2** was obtained as white amorphous powder. Its molecular formula was assigned to be  $\text{C}_{30}\text{H}_{48}\text{O}_3$  based on the HRESIMS spectrum at  $m/z$  455.3525  $[\text{M}-\text{H}]^-$  (calcd. for  $\text{C}_{30}\text{H}_{47}\text{O}_3$ , 455.3531). IR spectrum of **2** exhibited absorptions at 3,448 and  $1,701\text{ cm}^{-1}$ , assignable to hydroxyl and carboxyl functions, respectively. Its  $^1\text{H}$ -NMR and  $^{13}\text{C}$ -NMR shifts for the B, C, D, and E rings were almost the same as those of compound **1**, while those of the A ring differed. The main differences in the A ring between these two compounds were that the oxygenated proton H-3 at  $\delta_{\text{H}}$  3.41 (1H, *dd*,  $J = 12.0, 4.4\text{ Hz}$ ) in **1** was changed into  $\delta_{\text{H}}$  4.75 (1H, *brs*,  $W_{1/2} = 5.7\text{ Hz}$ ) in **2** and the oxygenated C-3 carbon at  $\delta_{\text{C}}$  78.7 in **1** was shifted upfield to  $\delta_{\text{C}}$  71.1 in **2**, suggesting that the H-3 in a  $\beta$ -orientation in **2**. This assignment could be further supported by the missing correlation between H-3 and H-5 in the NOESY

spectrum of **2**. Together with its  $^1\text{H}$ - $^1\text{H}$  COSY and HMBC spectra, which also were similar to those of **1**, compound **2** was determined to be  $3\alpha$ -hydroxy-chirat-16-en-24-oic acid and named kouitchenoid B.

**Table 1.**  $^1\text{H}$ -NMR (400 MHz) and  $^{13}\text{C}$ -NMR (100 MHz) Spectral Data of Compounds **1** and **2** in  $\text{C}_5\text{D}_5\text{N}$  ( $\delta$  in ppm,  $J$  in Hz).

Position	<b>1</b>		<b>2</b>	
	$\delta_{\text{C}}$	$\delta_{\text{H}}$	$\delta_{\text{C}}$	$\delta_{\text{H}}$
1	40.2	1.87 ( <i>dt</i> , $J = 3.2, 12.8$ ), 1.09 ( <i>m</i> )	35.3	1.91 ( <i>dt</i> , $J = 3.2, 12.1$ ), 1.73 ( <i>m</i> )
2	29.7	2.51 ( <i>m</i> ), 2.02 ( <i>m</i> )	28.1	2.80 ( <i>m</i> ), 2.05 ( <i>m</i> )
3	78.7	3.41 ( <i>dd</i> , $J = 12.0, 4.4$ )	71.1	4.75 ( <i>brs</i> , $W_{1/2} = 5.7$ )
4	49.6		48.8	
5	57.1	1.05 ( <i>m</i> )	49.7	2.08 ( <i>m</i> )
6	21.0	2.22 ( <i>m</i> ), 2.06 ( <i>m</i> )	20.9	2.38 ( <i>m</i> ), 2.07 ( <i>m</i> )
7	34.6	1.51 ( <i>m</i> ), 1.37 ( <i>m</i> )	34.7	1.61 ( <i>m</i> ), 1.40 ( <i>m</i> )
8	41.5		41.8	
9	50.8	1.29 ( <i>m</i> )	50.8	1.53 ( <i>m</i> )
10	37.5		38.7	
11	22.4	1.59 ( <i>m</i> ), 1.29 ( <i>m</i> )	22.3	1.69 ( <i>m</i> ), 1.32 ( <i>m</i> )
12	24.1	1.59 ( <i>m</i> ), 1.45 ( <i>m</i> )	24.1	1.58 ( <i>m</i> ), 1.43 ( <i>m</i> )
13	46.0	1.58 ( <i>m</i> )	46.0	1.59 ( <i>m</i> )
14	41.0		41.1	
15	32.8	2.18 ( <i>m</i> ), 1.53 ( <i>m</i> )	32.7	2.18 ( <i>m</i> ), 1.51 ( <i>m</i> )
16	120.9	5.34 ( <i>d</i> , $J = 5.1$ )	121.0	5.33 ( <i>d</i> , $J = 5.0$ )
17	139.7		139.7	
18	37.5		37.4	
19	38.9	1.61 ( <i>m</i> ), 1.18 ( <i>m</i> )	38.9	1.60 ( <i>m</i> ), 1.18 ( <i>m</i> )
20	35.8	1.53 ( <i>m</i> ), 1.18 ( <i>m</i> )	35.8	1.51 ( <i>m</i> ), 1.16 ( <i>m</i> )
21	33.0		33.0	
22	46.8	2.28 ( <i>m</i> ), 1.65 ( <i>m</i> )	46.8	2.27 ( <i>m</i> ), 1.63 ( <i>m</i> )
23	25.0	1.74 ( <i>m</i> )	25.7	1.80 ( <i>m</i> )
24	180.1		181.0	
25	15.1	1.11 ( <i>s</i> )	14.9	1.21 ( <i>s</i> )
26	17.3	1.03 ( <i>s</i> )	17.5	1.09 ( <i>s</i> )
27	16.7	1.06 ( <i>s</i> )	16.7	0.97 ( <i>s</i> )
28	18.0	1.00 ( <i>s</i> )	18.0	0.97 ( <i>s</i> )
29	25.0	0.84 ( <i>s</i> )	25.1	0.83 ( <i>s</i> )
30	32.8	0.96 ( <i>s</i> )	32.8	0.95 ( <i>s</i> )

**Figure 2.** Key HMBC, and NOESY correlations of compound **1**.



Compounds **1–4** were evaluated for their  $\alpha$ -glucosidase inhibitory activity using *p*-nitrophenyl- $\alpha$ -D-glucopyranoside (PNPG) as the substrate [8]. Although, not stronger than the activity of the reference drug acarbose, these compounds still exerted mild inhibitory activity against  $\alpha$ -glucosidase (Table 2).

**Table 2.** Inhibitory effects of compounds **1–4** and acarbose against  $\alpha$ -glucosidase <sup>a,b</sup>.

Compound	IC <sub>50</sub> ( $\mu$ M)	Compound	IC <sub>50</sub> ( $\mu$ M)
<b>1</b>	1932 $\pm$ 97	<b>4</b>	2017 $\pm$ 101
<b>2</b>	1812 $\pm$ 85	acarbose	627 $\pm$ 28
<b>3</b>	1858 $\pm$ 76		

<sup>a</sup> IC<sub>50</sub>, the concentration that inhibits cell growth by 50%; <sup>b</sup> Each value represents the mean  $\pm$  S.D. (n = 3).

### 3. Experimental

#### 3.1. General Procedures

Optical rotations were measured on an AA10R digital polarimeter. IR Spectra were detected on Avater-360 spectrophotometer with KBr pellets, and are reported in  $\text{cm}^{-1}$ . 1D and 2D NMR spectra (all in  $\text{C}_5\text{D}_5\text{N}$ ) were recorded on a Bruker AV-400 spectrometer, and chemical shifts are expressed in  $\delta$  (ppm) and referenced to the solvent peaks at  $\delta_{\text{H}}$  (8.74, 7.59, 7.22) and  $\delta_{\text{C}}$  (150.3, 135.9, 123.9) for  $\text{C}_5\text{D}_5\text{N}$ , respectively, and coupling constants are in Hz. HR-ESI-MS were determined on a Agilent 6520 Q-TOF LC-MS mass spectrometer. Semi-Preparative HPLC was performed on a Hitachi Spectra Series HPLC system equipped with an L-2130 pump and a UV L-2400 detector in a YMC-ODS column (10 mm  $\times$  250 mm, 5  $\mu$ m; flow rate at 2.0 mL/min; wavelength detection at 208 nm; retention time 34.2 min for **1**, 38.0 min for **2**, 18.7 min for **3**, and 20.1 min for **4**). Column chromatography (CC) was performed on  $\text{SiO}_2$  (200–300 mesh, Qingdao Marine Chemical Factory, Qingdao, China) and Toyopearl HW-40C (Tosoh Bioscience Shanghai Co., Ltd., Shanghai, China). Analytical TLCs were run on silica gel plates (GF<sub>254</sub>, Yantai Institute of Chemical Technology, Yantai, China). Fractions were monitored by TLC, and spots were visualized by heating TLC sprayed with 10%  $\text{H}_2\text{SO}_4$ .

#### 3.2. Plant Material

The whole plant of *S. kouitchensis* was collected in Enshi, Hubei province, China, in October 2010, and identified by Prof. Jiachun Chen (Tongji Pharmaceutical School of HUST, Wuhan, China). A voucher specimen (*S.k*-2010-1010) has been deposited in the University herbarium for future reference.

#### 3.3. Extraction and Isolation

The chopped, dried whole plants of *S. kouitchensis* (15 kg) were refluxed twice with 120 L of 95% (v/v) EtOH– $\text{H}_2\text{O}$ , two hours each time. After filtration, the filtrate was concentrated under reduced pressure to yield a brownish residue (3.0 kg). Part of the residue (2.5 kg) were suspended in water and partitioned successively with petroleum ether,  $\text{CH}_2\text{Cl}_2$ , EtOAc, and *n*-butanol to afford five fractions. The  $\text{CH}_2\text{Cl}_2$ -soluble part (about 400 g) was subjected to CC ( $\text{SiO}_2$ , 200–300 mesh, 3.0 kg, 12  $\times$  100 cm, petroleum ether/acetone 100:0  $\rightarrow$  0:100) to yield five fractions A–E. Fraction B (87.4 g) was subjected to HW 40C ( $\text{CHCl}_3/\text{MeOH}$  1:1) to give four subfractions B<sub>1–4</sub>. B<sub>2</sub> was subjected to CC

(SiO<sub>2</sub>, CHCl<sub>3</sub>/ EtoAc 20:1→10:1) to give two subfractions B<sub>2a</sub> and B<sub>2b</sub>. B<sub>2a</sub> was purified by semi-preparative HPLC (MeOH/H<sub>2</sub>O 90:10) to yield compound **3** (78.5 mg) and compound **4** (13.0 mg). B<sub>2b</sub> was purified by semi-preparative HPLC (MeOH/H<sub>2</sub>O 95:5) to yield compound **1** (4.1 mg) and **2** (5.7 mg).

*3β-Hydroxy-chirat-16-en-24-oic acid (1)*. White amorphous powder.  $[\alpha]_D^{25} +67.2^\circ$  ( $c = 0.3$ , pyridine); IR (KBr)  $\nu_{\max}$  3453, 2940, 1696, 1461, 1379, 1035 cm<sup>-1</sup>; <sup>1</sup>H-NMR and <sup>13</sup>C-NMR see Table 1; HRESIMS  $m/z$  455.3525 [M-H]<sup>-</sup> (calcd. for C<sub>30</sub>H<sub>47</sub>O<sub>3</sub>, 455.3531).

*3α-Hydroxy-chirat-16-en-24-oic acid (2)*. White amorphous powder;  $[\alpha]_D^{25} +35.3^\circ$  ( $c = 0.2$ , pyridine); IR (KBr)  $\nu_{\max}$  3448, 2934, 1701, 1456, 1378, 1036 cm<sup>-1</sup>; <sup>1</sup>H-NMR and <sup>13</sup>C-NMR see Table 1; HRESIMS  $m/z$  455.3525 [M-H]<sup>-</sup> (calcd. for C<sub>30</sub>H<sub>47</sub>O<sub>3</sub>, 455.3531).

### 3.4. In Vitro Inhibitory Activity against α-Glucosidase

α-Glucosidase (from *Saccharomyces cerevisiae*, Sigma-Aldrich, St. Louis, MO, USA) inhibitory activities were determined by using *p*-nitrophenyl-α-D-glucopyranoside (PNPG) as the substrate, according to the reported method [8]. Briefly, 20 μL of enzyme solution [0.6 U/mL α-glucosidase in 0.1 M potassium phosphate buffer (pH 6.8)] and 120 μL of the test compound in water containing 0.5% DMSO were mixed, and was preincubated for 15 min at 37 °C prior to initiation of the reaction by adding the substrate. After preincubation, PNPG solution 20 μL [5.0 mM PNPG in 0.1 M potassium phosphate buffer (pH 6.8)] was added and then incubated together at 37 °C for incubation. After the incubation, 80 μL 0.2 M Na<sub>2</sub>CO<sub>3</sub> in 0.1 M potassium phosphate buffer was added to the test tube to stop the reaction. The amount of PNP released was quantified using a UVmax Kinetic Microplate Reader (Bio Tek, Synergy 2, Winooski, VT, USA) at 405 nm.

## 4. Conclusions

Phytochemical investigation of CH<sub>2</sub>Cl<sub>2</sub>-soluble part of *S. kouitchensis* afforded two new chiratan-type triterpenoids, kouitchenoids A (**1**) and B (**2**), together with two known triterpenoids, oleanolic acid (**3**) and ursolic acid (**4**). Their structures were elucidated on the basis of spectral analysis and literature comparisons. All isolated compounds **1–4** exhibited moderate inhibitory activities against α-glucosidase *in vitro*, comparable with that of acarbose.

## Acknowledgments

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## Conflict of Interest

The authors declare no conflict of interest.

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*Sample Availability:* Samples of the compounds **1**, **3** and **4** are available from the authors.

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