

Article

Silica-Supported Polyphosphoric Acid in the Synthesis of 4-Substituted Tetrahydroisoquinoline Derivatives

Stanimir Manolov, Stoyanka Nikolova and Iliyan Ivanov *

Department of Organic Chemistry, University of Plovdiv, 24 Tzar Assen str., 4000 Plovdiv, Bulgaria

* Author to whom correspondence should be addressed; E-Mail: iiiliyan@abv.bg or ivanov@uni-plovdiv.bg; Tel.: +359-32-261-349; Fax: +359-32-261-403.

Received: 19 December 2012; in revised form: 21 January 2013 / Accepted: 22 January 2013 /

Published: 1 February 2013

Abstract: We report herein an application of an α -amidoalkylation reaction, as an alternative efficient synthesis of 4-aryl- and 4-methyl-1,2,3,4-tetrahydroisoquinoline derivatives. The amides required for this purpose would result from reaction of aminoacetaldehyde dimethylacetal with different substituted benzenes in polyphosphoric acid, followed by acylation of the obtained amines with different acid chlorides or sulfochlorides. We compared the cyclisation step using conventional (milieu of acetic-trifluoroacetic acid = 4:1) and solid supported reagents (SiO_2 /PPA), as recovered, regenerated and reused without loss of its activity catalyst. We found that in comparison to conventional methods, the yields of the reaction are greater and the reaction time is shorter.

Keywords: 4-substituted tetrahydroisoquinoline; cherylline analogues; α -amidoalkylation reaction; silica-supported polyphosphoric acid (SiO_2 -PPA)

1. Introduction

Very rare natural 4-aryl-1,2,3,4-tetrahydroisoquinoline alkaloids have been isolated from *Crinum powellii* var. *alba* and other *Crinum* species [1]. Due to the uniqueness of the structure and potential medicinal properties of the 4-aryliquinoline derivatives [2–5], many synthetic routes for these compounds [6–15] and especially cherylline have been reported. The Amaryllidaceae [6–16] have been one of the most studied families of plants because of their alkaloids composition. In 1970, Brossi *et al.* [1] isolated from *Crinum powelli* an alkaloid with a 4-aryl-1,2,3,4-tetrahydroisoquinoline structure named cherylline. Some of them are valuable medicines with centrally stimulating,

thymoleptic, and antiarrhythmic actions [2,17]. Others are calcium antagonistic [18], and exhibit broad spectrum of activity such as antibacterial [19], antiplasmodial [3,4,16,20,21], estrogen agonist/antagonist activity [22], and serotonin (5-HT) re-uptake [23].

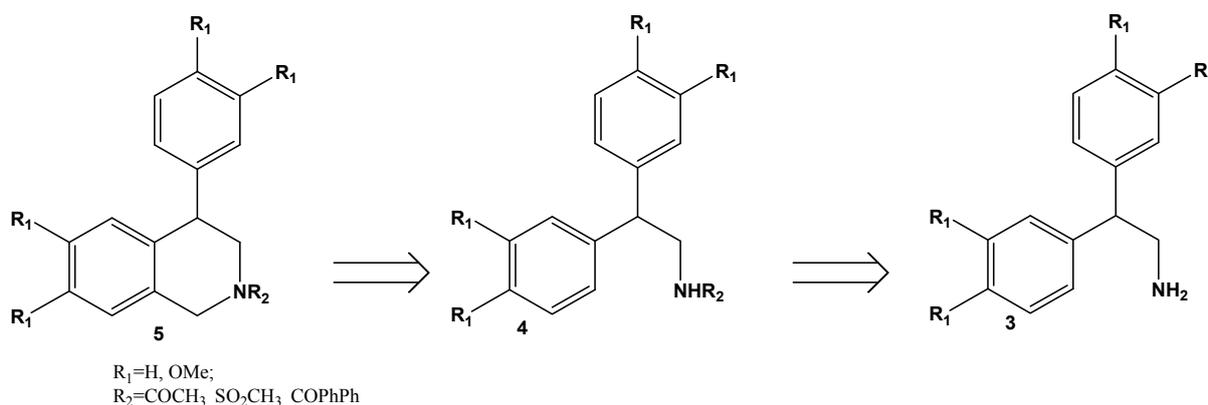
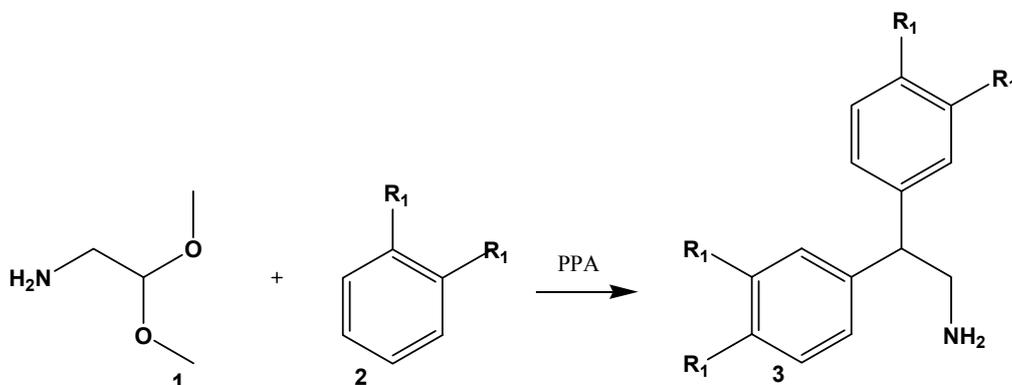
The importance of these heterocyclic compounds has generated a considerable number of synthetic approaches. Hence, the formation of the nitrogen-containing ring of 4-substituted 1,2,3,4-tetrahydroisoquinolines has been achieved with several reactions such as Bischler-Napieralski [24–26], intramolecular Horner [27], Friedel-Crafts [21,27–32] Pictet-Spengler [33], and oxazoline-driven chemistry [34], among others [25,35–42]. Most of the reported methods for the synthesis of these compounds are either multistep or low yielding.

In our previous investigations we report application of polyphosphoric acid (PPA) as a cyclisation agent for the construction of tetrahydroisoquinoline ring systems. Polyphosphoric acid is a strong mineral acid which has powerful dehydrating properties and is widely used for intramolecular and intermolecular acylations, heterocyclic synthesis, and acid-catalyzed reactions. For instance, in recent reports the cyclization reactions of *N,N'*-bis(oxotrifluoroalkenyl)-1,3-phenylenediamines in PPA medium [43], the acylation of benzene and its derivatives with 2-, 3-, 4-aminobenzoic and 4-aminophenylacetic acid in PPA to aminobenzophenones [44], and the reaction of tryptamine with carboxylic acid in PPA to afford 1-substitued-3,4-dihydro-9*H*- β -carboline derivatives [45], *etc.*, were described. However, the use of PPA has several drawbacks: since 10- to 50-fold excess is generally employed, it is difficult to pour and stir at room temperature, and it is necessary to carefully neutralize the reaction mixtures before the product extraction.

Recently, PPA/SiO₂ has been used as an efficient heterogeneous catalyst for many organic transformations. PPA/SiO₂ has some advantages including its low cost, ease of preparation, and ease of handling. In addition, the catalyst can be easily separated from the reaction mixtures by simple filtration and is reusable. Previously [46], the conversion of carbonyl compounds into oxathioacetals or dithioacetals using PPA/SiO₂ and a convenient method for the synthesis of isoxazole derivatives using PPA/SiO₂ as a reusable catalyst have been reported [47]. In the last several years the development of non-toxic, low cost, eco-friendly, recycable catalyst systems which give high productivity under mild reaction conditions has received much attention in organic synthesis [48]. Solid supported catalysts [49,50] have gained much prominence due to their inherent economic and environmental benefits, ease of handling, easy catalyst separation and regeneration, thermal stability and long catalytic life [51]. Since the activity and selectivity of a reagent dispersed on the surface of the support is improved as the effective surface area of reagent can be increased manifold, they are expected to perform better than the individual reagents [52]. Low toxicity, moisture, air tolerance and low price are other common features that make the use of solid supported reagents attractive alternative to the conventional catalysts.

2. Results and Discussion

Our retrosynthetic analysis of 4-substituted 1,2,3,4-tetrahydroisoquinolines is depicted in Scheme 1. We anticipated that **5** could be constructed from amides **4** via an α -amidoalkylation reaction. The required amides **4** can be prepared from the amines **3** via acylation with different acid chlorides. The amines **3** would arise from reaction [53] of aminoacetaldehyde dimethylacetal (**1**) with benzene or substituted benzenes **2** (Scheme 2).

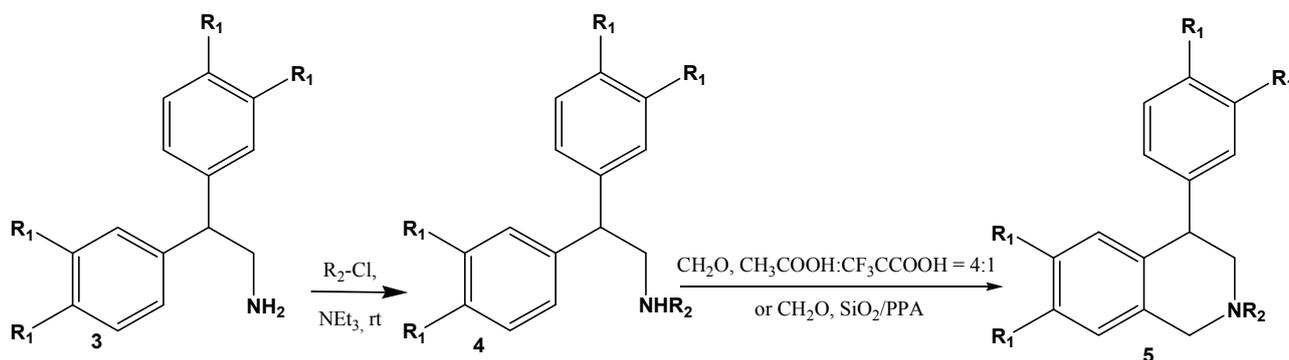
Scheme 1. Retrosynthetic scheme for the synthesis of 4-substituted tetrahydroisoquinolines.**Scheme 2.** Synthesis of amines **3**.

For the synthesis of the required starting materials Klumpp's [53] protocol using amino acetals and benzene in TfOH to give the corresponding products with good yields can be used. We carried out the same reaction in polyphosphoric acid instead of TfOH. We found that aminoacetaldehyde dimethylacetal reacts with excess of 1,2-dimethoxybenzene (veratrol) in PPA at rt for 5 to 10 min to give a good (82%) yield of the corresponding product **3**. The same product with unsubstituted benzene was obtained after 30 min at 80 °C with 42% yield (Scheme 2, Table 1).

Table 1. Synthesis of amines **3**.

3	R₁		Yield, %	M.p., °C
a	H	H	42	47–49
b	OCH ₃	OCH ₃	82	oil

The amides **4** required for the next step of our synthesis, were prepared by acylation of the amines **3** with different acid chlorides or sulfochlorides. The next step was cyclization of the newly synthesized amides (Scheme 3, Table 2).

Scheme 3. Synthesis of 4-aryl tetrahydroisoquinolines.**Table 2.** Synthesis of amides **4**.

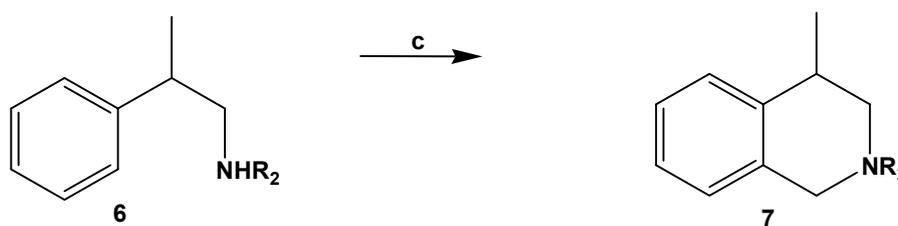
4	R	R₁	R₂	Yield, %	M.p., °C
a	H	H	COCH ₃	80	85–86
b	H	H	SO ₂ CH ₃	98	138–139
c	H	H	COPhPh	94	192–193
d	OCH ₃	OCH ₃	COCH ₃	91	126–128
e	OCH ₃	OCH ₃	SO ₂ CH ₃	90	49–52
f	OCH ₃	OCH ₃	COPhPh	97	200–202

We compared the cyclisation step using either mixture of acetic-trifluoroacetic acid = 4:1 or silica-supported PPA (SiO₂-PPA). We found that with SiO₂-PPA the reaction yields are greater and the reaction times are shorter (Table 3). The shortest time and best yield were achieved at 80 °C, using 0.06 g of catalyst and 1 h reflux. As described earlier [54] the catalyst was recovered quantitatively and used in the reaction three times. The recovered catalyst did not show any significant loss of activity.

Table 3. Comparative yields of the compounds **5** synthesized through two conventional methods.

5	R	R₁	R₂	Yield, %	
				(CH₃COOH:CF₃COOH = 4:1)	(SiO₂/PPA)
a	H	H	COCH ₃	90	94
b	H	H	SO ₂ CH ₃	80	91
c	H	H	COPhPh	84	96
d	OCH ₃	OCH ₃	COCH ₃	85	96
e	OCH ₃	OCH ₃	SO ₂ CH ₃	87	97
f	OCH ₃	OCH ₃	COPhPh	87	98

The same protocol was applied for the synthesis of 4-methyl-1,2,3,4-tetrahydroisoquinoline derivatives which are interesting as potential anaesthetics and antispasmodics [55] (Scheme 4, Table 4).

Scheme 4. Synthesis of 4-methyl tetrahydroisoquinolines.

Reagents and Conditions: (c) CH_2O , $\text{CH}_3\text{COOH}:\text{CF}_3\text{CCOOH} = 4:1$ rt or CH_2O , SiO_2/PPA .

Table 4. Synthesis of amides **6** and 4-methyl-1,2,3,4-tetrahydroisoquinolines **7**.

6	R₂	Yield, %	M.p., °C
a	COCH ₃	98	Oil
b	SO ₂ CH ₃	98	Oil
c	COPhPh	98	156–157
7	R₂	Yield, %	m.p., °C
a	COCH ₃	89	Oil
b	SO ₂ CH ₃	93	66–67
c	COPhPh	96	Oil

3. Experimental

3.1. General

Reagents and chemicals were purchased from commercial sources (Sigma-Aldrich S.A. and Riedel-de Haën) and used as received. Melting points were determined on a Boetius hot stage apparatus and are uncorrected. Spectra were recorded on a Bruker Avance DRX250 spectrometer (BAS-IOCCP, Sofia). ¹H-NMR and ¹³C-NMR spectra were taken in CDCl₃ (unless otherwise specified) at 250 MHz and 62.5 MHz respectively. Chemical shifts were given in part per million (ppm) relative and were referenced to TMS ($\delta = 0.00$ ppm) as an internal standard and coupling constants are indicated in Hz. All the NMR spectra were taken at rt (ac. 295 K). Elemental analyses were performed with a vario EL III. TLC was carried out on precoated 0.2 mm Fluka silica gel 60 plates, using diethyl ether/*n*-hexane = 1:1 as chromatographic system. Merck silica gel 60 (0.063–0.2 mm) was used for column chromatographic separation. Polyphosphoric acid was obtained from 85% phosphoric acid and P₂O₅ (1:1 w/w).

3.2. Preparation of PPA/SiO₂ Catalyst General Procedure

PPA (4.0 g) was charged in the round-bottom flask, and CHCl₃ (100 mL) was added. After the mixture was stirred at 50 °C for 1 h, followed by SiO₂ (16.0 g, 70–230 mesh) was added to the solution, and the mixture was stirred for another 1 h. CHCl₃ was removed by evaporation, and the resulting solid was dried in vacuo at room temperature for 3 h. Used PPA/SiO₂ was regenerated as follows: PPA/SiO₂ was recovered by filtration from the reaction mixture, and then it was put in the 50 mL round-bottom flask and dried in vacuum at 100 °C for 2 h.

3.3. Typical Procedure for the Synthesis of 2-Amino-1,1-diphenylethanes 3

Aminoacetaldehyde dimethylacetal (**1**, 0,210 g, 2 mmol) was dissolved in the corresponding benzene (22.47 mmol) in an open flask and polyphosphoric acid (3 g) was added. The reaction mixture was stirred for 20 min at rt. After completion of the reaction, the mixture was poured over ice. The organic layer was removed and after that the water layer the solution was neutralized with sodium hydroxide, then extracted with CH₂Cl₂ (3 × 20 mL). Combined extracts were dried (Na₂SO₄) and concentrated. The products, after evaporation of the solvent, were purified by column chromatography on silica gel using Et₂O as eluent.

3.4. 2,2-Diphenylethanamine (3a): Known Compound [53]

2,2-Bis(3,4-dimethoxyphenyl)ethanamine (**3b**). ¹H-NMR: 1.97 (broad s, 2H, NH₂), 3.29 (d, 2H, *J* = 7.6 Hz), 3.85 (s, 6H), 3.87 (s, 6H), 3.93 (ddd, 1H, *J* = 4.0, 6.0, 10.0 Hz), 6.76–6.86 (m, 6H); ¹³C-NMR: 149.0, 147.7, 135.3, 119.7, 111.6, 111.3, 55.9, 53.8, 47.1; Anal. calc. for C₁₈H₂₃NO₄: C, 68.12; H, 7.30; N, 4.41; found C, 68.18; H, 7.21; N, 4.38.

3.5. Acylation of Amines 3: Typical Procedure for the Synthesis of Amides 4 and 6

To solution of amine **3** (1 mmol) in dichloromethane (15 mL) equal amount of acetyl chloride, methanesulfonyl chloride or biphenyl-4-carbonyl chloride was added. After 10 min a little excess of triethylamine was added. After 30 min the solution was washed with diluted hydrochloric acid, saturated solution of Na₂CO₃ and water. The organic layer was dried (Na₂SO₄), concentrated and filtered on short column with neutral Al₂O₃.

N-(2,2-Diphenylethyl)acetamide (**4a**). ¹H-NMR: 1.87 (s, 3H), 3.88 (dd, 2H, *J* = 5.9, 7.9 Hz), 4.17 (t, 1H, *J* = 8.0 Hz), 5.46 (s, 1H, NH), 7.22–7.25 (m, 6H), 7.28–7.31 (m, 4H); ¹³C-NMR: 170.1, 141.9, 128.8, 128.1, 126.9, 50.6, 43.9, 23.3; Anal. calc. for C₁₆H₁₇NO: C, 80.30; H, 7.16; N, 5.85; found C, 80.26; H, 7.23; N, 5.78.

N-(2,2-Diphenylethyl)methanesulfonamide (**4b**). ¹H-NMR: 2.81 (s, 3H), 3.76 (d, 2H, *J* = 7.9 Hz), 4.20 (t, 1H, *J* = 8.0 Hz), 7.20–7.28 (m, 6H), 7.29–7.36 (m, 4H); ¹³C-NMR: 140.9, 129.0, 128.0, 127.3, 51.4, 47.6, 40.5; Anal. calc. for C₁₅H₁₇NO₂S: C, 65.43; H, 6.22; N, 5.09; S, 11.64; found C, 65.37; H, 6.29; N, 5.01; S, 11.71.

N-(2,2-Diphenylethyl)-[1,1'-biphenyl]-4-carboxamide (**4c**). ¹H-NMR (CF₃COOD): 4.36 (d, 2H, *J* = 7.8 Hz), 4.48 (dd, 1H, *J* = 7.0, 9.4 Hz), 7.26–7.36 (m, 3H), 7.35 (s, 5H), 7.38–7.49 (m, 5H), 7.55–7.72 (m, 6H); ¹³C-NMR: 148.4, 139.9, 138.4, 128.7, 128.6, 128.5, 127.7, 127.5, 127.4, 127.2, 126.6, 121.0, 112.0, 107.5, 49.8, 46.5; Anal. calc. for C₂₇H₂₃NO: C 85.91; H, 6.14; N, 3.71; found C 85.97; H, 6.08; N, 3.78.

N-(2,2-Bis(3,4-dimethoxyphenyl)ethyl)acetamide (**4d**). ¹H-NMR: 1.91 (s, 3H), 3.82–3.90 (m, 2H), overlapped with 3.85 (s, 6H), 3.87 (s, 6H), 4.08 (t, 1H, *J* = 7.9 Hz), 5.46 (t, 1H, *J* = 5.7 Hz, NH), 6.76 (dd, 2H, *J* = 1.9, 5.8 Hz), 6.81 (d, 2H, *J* = 1.9 Hz), 6.83 (s, 1H), 6.86 (s, 1H); ¹³C-NMR: 170.0, 149.1,

147.9, 134.5, 119.7, 111.4, 111.3, 55.9, 55.8, 49.6, 44.0, 23.3; Anal. calc. for C₂₀H₂₅NO₅: C, 66.83; H, 7.01; N, 3.90; found C, 66.75; H, 7.09; N, 3.86.

N-(2,2-Bis(3,4-dimethoxyphenyl)ethyl)methanesulfonamide (**4e**). ¹H-NMR: 2.84 (s, 3H), 3.67 (d, 2H, *J* = 7.7 Hz), 3.83 (s, 3H), 3.85 (s, 3H), 4.09 (t, 1H, *J* = 7.8 Hz), 4.32 (broad s, 1H, NH), 6.74 (dd, 2H, *J* = 1.8, 9.8 Hz), 6.80 (d, 2H, *J* = 1.9 Hz), 6.83 (d, 2H, *J* = 8.2 Hz); ¹³C-NMR: 149.2, 148.2, 133.6, 119.6, 111.5, 111.4, 56.0, 55.9, 50.4, 47.7, 40.4; Anal. calc. for C₁₉H₂₅NO₆S: C, 57.70; H, 6.37; N, 3.54; S, 8.11; found C, 57.78; H, 6.32; N, 3.61; S, 8.01.

N-(2,2-Bis(3,4-dimethoxyphenyl)ethyl)-[1,1'-biphenyl]-4-carboxamide (**4f**). ¹H-NMR: 3.82 (s, 6H), 3.86 (s, 6H), 4.05 (dd, 2H, *J* = 5.7, 7.7 Hz), 4.24 (t, 1H, *J* = 7.9 Hz), 6.20 (t, 1H, *J* = 5.5 Hz, NH), 6.82 (dd, 6H, *J* = 1.1, 10.9 Hz), 7.36–7.47 (m, 3H), 7.54–7.70 (m, 6H); ¹³C-NMR: 167.2, 149.2, 148.1, 144.3, 139.9, 134.6, 133.2, 128.9, 128.0, 127.3, 120.0, 111.6, 111.4, 55.9, 50.0, 44.6; Anal. calc. for C₃₁H₃₁NO₅: C, 74.83; H, 6.28; N, 2.81; found C, 74.75; H, 6.34; N, 2.79.

N-(2-Phenylpropyl)acetamide (**6a**). ¹H-NMR: 1.20 (d, 3H, *J* = 7.0 Hz), 1.87 (s, 3H), 2.86 (ddd, 1H, *J* = 2.2, 7.0, 13.2 Hz), 3.14 (ddd, 1H, *J* = 4.8, 8.8, 13.4 Hz), 3.56 (ddd, 1H, *J* = 6.1, 7.0, 13.1 Hz), 5.30 (broad s, 1H, NH), 7.14 (ddd, 3H, *J* = 1.8, 4.3, 5.5 Hz), 7.25 (dt, 2H, *J* = 2.0, 3.5 Hz); ¹³C-NMR: 170.0, 144.1, 128.8, 127.2, 126.8, 46.2, 39.7, 23.3, 19.5; Anal. calc. for C₁₁H₁₅NO: C, 74.54; H, 8.53; N, 7.90; found C, 74.48; H, 8.60; N, 7.85.

N-(2-Phenylpropyl)methanesulfonamide (**6b**). ¹H-NMR: 1.24 (d, 3H, *J* = 7.0 Hz), 2.69 (s, 3H), 2.89 (dd, 1H, *J* = 6.7, 14.7 Hz), 3.11–3.19 (m, 1H), 3.21–3.33 (m, 1H), 4.22 (t, 1H, *J* = 6.8 Hz, NH), 7.13–7.21 (m, 3H), 7.23–7.30 (m, 2H); ¹³C-NMR: 143.1, 128.9, 127.3, 127.2, 50.0, 40.4, 40.3, 19.0; Anal. calc. for C₁₀H₁₅NO₂S: C, 56.31; H, 7.09; N, 6.57; S, 15.03; found C, 56.26; H, 7.15; N, 6.49; S, 15.09.

N-(2,2-Bis(3,4-dimethoxyphenyl)ethyl)-[1,1'-biphenyl]-4-carboxamide (**6c**). ¹H-NMR: 1.28 (d, 3H, *J* = 7.0 Hz), 2.95–3.09 (m, 1H), 3.35 (ddd, 1H, *J* = 4.9, 8.7, 13.5 Hz), 3.78 (ddd, 1H, *J* = 6.1, 6.9, 13.2 Hz), 6.00 (t, 1H, *J* = 5.8 Hz, NH), 7.15–7.20 (m, 3H), 7.25–7.32 (m, 3H), 7.33–7.39 (m, 2H), 7.59–7.63 (m, 2H); ¹³C-NMR: 167.2, 144.2, 144.1, 140.0, 133.4, 128.93, 128.85, 128.0, 127.34, 127.29, 127.2, 127.0, 46.7, 39.9, 19.3; Anal. calc. for C₃₁H₃₁NO₅: C, 74.83; H, 6.28; N, 2.81; found C, 74.79; H, 6.33; N, 2.78.

3.6. Cyclization of Amides **4** and **6** in Acetic/Trifluoroacetic Acid Milieu: Typical Procedure

2-Phenylethylamides (3 mmol) and paraformaldehyde (5 mmol) were dissolved in a mixture of CH₃COOH/CF₃COOH = 4:1 (5 mL) at 80–100 °C for 1 h. The solution was poured on the crushed ice and extracted with CH₂Cl₂ (3 × 20 mL). The extract was washed with 20% aq. Na₂CO₃ (2 × 30 mL) and dried (Na₂SO₄). The solvent was distilled and the products were purified by recrystallization (Et₂O).

3.7. Cyclization of Amides **4** and **6** in SiO₂-Supported Milieu: Typical Procedure

2-Phenylethylamides (3 mmol) and paraformaldehyde (5 mmol) were dissolved in a C₂H₄Cl₂ (10 mL) at 80 °C, 0.06 g of catalyst (SiO₂/PPA) and 1h reflux. After the completion of the reaction the reaction mixture was cooled and the catalyst was separated by simple filtration.

1-(4-Phenyl-3,4-dihydroisoquinolin-2(1H)-yl)ethanone (5a). $^1\text{H-NMR}$: 1.67 (s, 3H), 3.82 (d, 2H, $J = 4.4$ Hz), 4.23 (t, 1H, $J = 4.2$ Hz), 4.52 (d, 1H, $J = 17.1$ Hz), 5.23 (d, 1H, $J = 17.4$ Hz), 6.96–7.06 (m, 3H), 7.22–7.29 (m, 6H); $^{13}\text{C-NMR}$: 169.9, 142.7, 136.0, 133.6, 129.5, 128.7, 128.5, 127.1, 126.6, 51.2, 45.4, 44.3, 20.8; Anal. calc. for $\text{C}_{17}\text{H}_{17}\text{NO}$: C, 81.24; H, 6.82; N, 5.57; O, 6.37; found C, 81.28; H, 6.77; N, 5.60.

2-(Methylsulfonyl)-4-phenyl-1,2,3,4-tetrahydroisoquinoline (5b). $^1\text{H-NMR}$: 3.47 (dd, 1H, $J = 7.4, 12.5$ Hz), 3.86 (ddd, 1H, $J = 1.0, 5.0, 12.4$ Hz), 4.33 (t, 1H, $J = 6.3$ Hz), 4.57 (q, 2H, $J = 15.4, 15.38$ Hz), 6.94–6.98 (m, 1H), 7.11–7.18 (m, 4H), 7.21–7.31 (m, 4H); $^{13}\text{C-NMR}$: 142.5, 136.2, 132.2, 129.9, 128.9, 128.7, 127.2, 127.0, 126.3, 50.9, 47.6, 45.1, 36.4; Anal. calc. for $\text{C}_{16}\text{H}_{17}\text{NO}_2\text{S}$: C, 66.87; H, 5.96; N, 4.87; S, 11.16; found C, 66.91; H, 5.91; N, 4.90; S, 11.11.

Biphenyl-4-yl(4-phenyl-3,4-dihydroisoquinolin-2(1H)-yl)methanone (5c). $^1\text{H-NMR}$ (CF_3COOD): 4.36 (d, 2H, $J = 7.8$ Hz), 4.48 (dd, 1H, $J = 7.0, 9.4$ Hz), 7.26–7.36 (m, 3H), 7.35 (s, 5H), 7.38–7.49 (m, 5H), 7.55–7.72 (m, 6H); $^{13}\text{C-NMR}$: 148.4, 139.9, 138.4, 128.7, 128.6, 128.5, 127.7, 127.5, 127.4, 127.2, 126.6, 121.0, 112.0, 107.5, 49.8, 46.5; Anal. calc. for $\text{C}_{28}\text{H}_{23}\text{NO}$: C, 86.34; H, 5.95; N, 3.60; found C, 86.39; H, 5.89; N, 3.58.

1-(4-(3,4-Dimethoxyphenyl)-6,7-dimethoxy-3,4-dihydroisoquinolin-2(1H)-yl)ethanone (5d). $^1\text{H-NMR}$ (CDCl_3): 1.66 (s, 3H), 3.74 (s, 3H), 3.81 (s, 3H), 3.87 (s, 3H), 3.90 (s, 3H), 4.90 (t, 1H, $J = 0.96$ Hz), 4.93 (d, 2H, $J = 1.09$ Hz), 5.34–5.35 (m, 2H), 6.47 (s, 1H), 6.53 (s, 1H), 6.68–6.69 (m, 2H), 6.82 (s, 1H); $^{13}\text{C-NMR}$ (CDCl_3): 169.9, 149.0, 148.3, 148.1, 148.0, 147.9, 135.5, 120.7, 111.8, 111.4, 111.2, 109.2, 108.6, 56.2, 56.0, 55.93, 55.89, 51.5, 44.6, 43.9, 20.9. Anal. calc. for $\text{C}_{21}\text{H}_{25}\text{NO}_5$: C, 67.91; H, 6.78; N, 3.77; found C, 67.96; H, 6.74; N, 3.82.

4-(3,4-Dimethoxyphenyl)-6,7-dimethoxy-2-(methylsulfonyl)-1,2,3,4-tetrahydroisoquinoline (5e). $^1\text{H-NMR}$ (CDCl_3): 2.72 (s, 3H), 3.41 (dd, 1H, $J = 7.42, 12.37$ Hz), 3.72 (s, 3H), 3.81–3.86 (m, 1H), overlapped with 3.84 (s, 3H), 3.89 (s, 3H), 3.90 (s, 3H), 4.21 (dd, 1H, $J = 5.58, 6.24$ Hz), 4.51 (q, 2H, $J = 6.25, 15.0$ Hz), 6.46 (s, 1H), 6.64 (s, 1H), 6.67–6.71 (m, 2H), 6.81–6.85 (m, 1H); $^{13}\text{C-NMR}$ (CDCl_3): 149.0, 148.2, 148.1, 135.0, 128.2, 124.1, 121.0, 112.1, 111.9, 111.2, 108.5, 56.0, 55.9, 51.1, 47.3, 44.3, 36.3. Anal. calc. for $\text{C}_{20}\text{H}_{25}\text{NO}_6\text{S}$: C, 58.95; H, 6.18; N, 3.44; S, 7.87; found C, 58.99; H, 6.14; N, 3.45; S, 7.81.

Biphenyl-4-yl[4-(3,4-dimethoxyphenyl)-6,7-dimethoxy-3,4-dihydroisoquinolin-2(1H)-yl]methanone (5f). $^1\text{H-NMR}$: 3.73 (s, 3H), 3.76 (s, 3H), 3.91 (s, 3H), 3.94 (s, 3H), 4.09 (s, 1H), 4.65 (d, 1H, $J = 17.0$ Hz), 5.34 (d, 1H, $J = 16.6$ Hz), 6.51 (s, 2H), 6.81 (d, 2H, $J = 7.7$ Hz), 6.92 (d, 2H, $J = 6.8$ Hz), 7.36–7.50 (m, 6H), 7.57–7.65 (m, 4H); $^{13}\text{C-NMR}$: 171.3, 149.2, 148.5, 148.14, 148.07, 142.2, 140.2, 135.2, 134.5, 128.9, 127.5, 127.4, 127.1, 126.8, 125.3, 112.0, 111.7, 111.2, 108.8, 56.03, 55.99, 55.94, 55.87, 52.2, 51.8, 44.6. Anal. calc. for $\text{C}_{32}\text{H}_{31}\text{NO}_5$: C, 75.42; H, 6.13; N, 2.75; found C, 75.50; H, 6.10; N, 2.81.

4. Conclusions

In conclusion, we have developed a highly efficient SiO_2 -PPA catalyzed method for the construction of 4-aryl- or 4-methyl-1,2,3,4-tetrahydroisoquinoline ring systems, as analogues of biologically-active compounds. The catalyst is completely recoverable and the efficiency of the

catalyst remains unaltered even after three to four cycles. It is also noticed that the cyclisation using PPA–SiO₂ proceeds rapidly and is superior to the reported procedures with respect to yield and amount of the catalyst employed.

Acknowledgments

We acknowledge financial support from the National Fund for scientific research DO-02-131 and DO-02-195.

References

1. Brossi, A.; Grethe, G.; Tietel, S.; Wildman, W.C.; Bailey, D.T. Cherylline, a 4-phenyl-1,2,3,4-tetrahydroisoquinoline alkaloid. *J. Org. Chem.* **1970**, *35*, 1100–1104.
2. Jacob, J.N.; Nichols, D.E.; Kohli, J.D.; Glock, D. Dopamine agonist properties of *N*-alkyl-4-(3,4-dihydroxyphenyl)-1,2,3,4-tetrahydroisoquinolines. *J. Med. Chem.* **1981**, *24*, 1013–1015.
3. Zara-Kaczian, E.; Gyorgy, L.; Deak, G.; Seregi, A.; Doda, M. Synthesis and pharmacological evaluation of some new tetrahydroisoquinoline derivatives inhibiting dopamine uptake and/or possessing a dopaminomimetic property. *J. Med. Chem.* **1986**, *29*, 1189–1195.
4. Maryanoff, B.E.; Vaught, J.L.; Shank, R.P.; McComsey, D.F.; Costanzo, M.J.; Nortey, S.O. Pyrroloisoquinoline antidepressants. 3. A focus on serotonin. *J. Med. Chem.* **1990**, *33*, 2793–2797.
5. Kihara, M.; Ikeuchi, M.; Yamauchi, A.; Nukatsuka, M.; Matsumoto, H.; Toko, T. Synthesis and biological evaluation of 7-hydroxy-3,4-diphenyl-1,2-dihydroisoquinolines as new 4-hydroxytamoxifen analogues. *Chem. Pharm. Bull.* **1997**, *45*, 939–943.
6. Venkov, A.P.; Vodenicharov, D.; A New Synthesis of 1,2,3,4-Tetrahydro-2-methyl-4-phenylisoquinolines. *Synthesis* **1990**, 253–255.
7. Kihara, M.; Kashimoto, M.; Kobayashi, Y.; Kobayashi, S. A new intramolecular barbiere reaction of *N*-(2-iodobenzyl)phenacylamines: A convenient synthesis of 1,2,3,4-tetrahydroisoquinolin-4-ols. *Tetrahedron Lett.* **1990**, *31*, 5347–5348.
8. Venkov, A.P.; Vodenicharov, D.M.; Ivanov, I.I. 4-Aryl-1,4-dihydro-3(2*H*)-isoquinolinones are obtained by oxidative cyclization of *N*-benzyl-*N*-alkylarylacetyl amides with lead tetra-acetate in acetic acid/trifluoroacetic acid. *Synthesis* **1991**, 476–478.
9. Kihara, M.; Kashimoto, M.; Kobayashi, Y.; Nagao, Y. A new synthesis of 7,12-Dihydro-12-phenyl-5*H*-6,12-methanodibenz[*c,f*]azocines via *N*-Benzyl-1,2,3,4-tetrahydro-4-phenylisoquinolin-4-ols. *Heterocycles* **1992**, *34*, 747–756.
10. Coskun, N.; Sümengen, D. A new synthesis of *N*-Substituted-1,2,3,4-tetrahydro-4-phenylisoquinolin-3-ones. *Synth. Commun.* **1993**, *23*, 1393–1402.
11. Miller, R.B.; Svoboda, J.J. An efficient synthesis of 4-aryl-1,2,3,4-tetrahydroisoquinolines. *Synth. Commun.* **1994**, *24*, 1187–1194.
12. Meda, N.; Selvakumar, N.; Kraus, G.A. An efficient synthesis of 4-aryl kainic acid analogs. *Tetrahedron* **1999**, *55*, 943–954.
13. Kuster, G.J.; Kalmoua, F.; Scheeren, H.W.; de Gelder, R. A simple entry towards novel bi- and tricyclic *N*-oxy- β -lactams by high pressure promoted tandem [4+2]/[3+2] cycloadditions of enol ethers and β -nitrostyrene. *Chem. Commun.* **1999**, 855–856.

14. Nolellino, L.; D'Ischia, M.; Prota, G. Expedient synthesis of 5,6-dihydroxyindole and derivatives via an improved Zn(II)-assisted 2, β -dinitrostyrene approach. *Synthesis* **1999**, 793–796.
15. Kihara, M.; Andoh, J.I.; Yoshida, C. New reduction reaction of benzylic alcohols with acid and proof of the intermolecular hydride shift mechanism. *Heterocycles* **2000**, 53, 359–372.
16. Labrana, J.; Machocho, A.K.; Kricsfalusy, V.; Brun, R.; Codina, C.; Viladomat, F.; Bastida, J. Alkaloids from *Narcissus angustifolius* subsp. *transcarpathicus* (Amaryllidaceae). *Phytochemistry* **2002**, 60, 847–856.
17. Elgorashi, E.E.; Stafford, G.J.; Jäger, A.K.; van Staden, J. Inhibition of [3H] citalopram binding to the rat brain serotonin transporter by amaryllidaceae alkaloids. *Planta Med.* **2006**, 72, 470–473.
18. Dong, H.; Sheng, J.Z.; Lee, C.M.; Wong, T.M. Calcium antagonistic and antiarrhythmic actions of CPU-23, a substituted tetrahydroisoquinoline. *Br. J. Pharmacol.* **1993**, 109, 113–119.
19. Bernan, V.S.; Montenegro, D.A.; Korshalla, J.D.; Maiese, W.M.; Steinberg, D.A.; Greenstein, M. Bioxalomycins, new antibiotics produced by the marine *Streptomyces* sp. LL-31F508: Taxonomy and fermentation. *J. Antibiot.* **1994**, 47, 1417–1424.
20. Ratsimamanga-Urverg, S.; Rasoanaivo, P.; Rafatro, H.; Robijaona, B.; Rakoto-Ratsimamanga, A. *In vitro* antiplasmodial activity and chloroquine-potentiating action of three new isoquinoline alkaloid dimers isolated from *Hernandia voyronii* Jumelle. *Ann. Trop. Med. Parasitol.* **1994**, 88, 271–277.
21. Riggs, R.M.; Nichols, D.E.; Foreman, M.M.; Truex, L.L. Effect of β -alkyl substitution on D-1 dopamine agonist activity: Absolute configuration of β -methyldopamine. *J. Med. Chem.* **1987**, 30, 1887–1897.
22. Chesworth, R.; Zawistoski, M.P.; Lefker, B.A.; Cameron, K.O.; Day, R.F.; Mangano, F.M.; Rosati, R.L.; Colella, S.; Petersen, D.N.; Brault, A.; *et al.* Tetrahydroisoquinolines as subtype selective estrogen agonists/antagonists. *Bioorg. Med. Chem. Lett.* **2004**, 14, 2729–2733.
23. Keith, J.M.; Barbier, A.J.; Wilson, S.J.; Miller, K.; Boggs, J.D.; Fraser, I.C.; Mazur, C.; Lovenberg, T.W.; Carruthers, N.I. Dual serotonin transporter inhibitor/histamine H3 antagonists: Development of rigidified H3 pharmacophores. *Bioorg. Med. Chem. Lett.* **2007**, 17, 5325–5329.
24. Brossi, A.; Teitel, S. Total synthesis of racemic cherylline. *Tetrahedron Lett.* **1970**, 11, 417–419.
25. Ruchirawat, S.; Tontoolarug, S.; Sahakitpichan, P. Synthesis of 4-aryltetrahydroiso-quinolines: Application to the synthesis of cherylline. *Heterocycles* **2001**, 55, 635–640.
26. Takano, S.; Akiyama, M.; Ogasawara, K. Total synthesis of (\pm)- and (+)-latifine. *J. Chem. Soc. Perkin Trans. 1* **1985**, 2447–2453.
27. Couture, A.; Deniau, E.; Lebrun, S.; Grandclaudeon, P. Total syntheses of (+/-)-cherylline and (+/-)-latifine. *J. Chem. Soc. Perkin Trans. 1* **1999**, 789–794.
28. Cuevas, J.-C.; Snieckus, V. α' -Silylated tertiary benzamides as dual ortho- and α' -carbanion synthons. Carbodesilylative routes to isoquinoline and dibenzoquinolizidine derivatives. *Tetrahedron Lett.* **1989**, 30, 5837–5840.
29. Philippe, N.; Denivet, F.; Vasse, J.-L.; Sopkova-de Olivera Santos, J.; Levacher, V.; Dupas, G. Highly stereoselective Friedel–Crafts type cyclization. Facile access to enantiopure 1,4-dihydro-4-phenyl isoquinolinones. *Tetrahedron* **2003**, 59, 8049–8056.
30. Hara, H.; Shirai, R.; Hoshino, O.; Umezawa, B. A Facile Synthesis of 4-aryl-1,2,3,4-tetrahydroisoquinolines: A total synthesis of (\pm)-cherylline. *Heterocycles* **1983**, 20, 1945–1950.

31. Toda, J.; Sonobe, A.; Ichikawa, T.; Saitoh, T.; Horiguchi, Y.; Sano, T. Synthesis of 4-aryl-2-methyl-1,2,3,4-tetrahydroisoquinolines via Pummerer-type cyclization of *N*-(arylmethyl)-*N*-methyl-2-aryl-2-(phenylsulfinyl) acetamides. *Arcivoc* **2000**, *1*, 165–180.
32. Hara, H.; Shirai, R.; Hoshino, O.; Umezawa, B. Studies on tetrahydroisoquinolines. XXV. A synthesis of 4-aryl-1,2,3,4-tetrahydroisoquinolines; Total synthesis of (±)-cherylline. *Chem. Pharm. Bull* **1985**, *33*, 3107–3112.
33. Couture, A.; Deniau, E.; Grandclaudon, P.; Lebrun, S. Asymmetric synthesis of (+)- and (–)-latifine. *Asymmetry* **2003**, *14*, 1309–1316.
34. Seijas, J.A.; Vázquez-Tato, M.P.; Martínez, M.M.; Pizzolatti, M.G. Oxazoline as a useful tool in organic synthesis: Preparation of 4-aryl-1,2,3,4-tetrahydroisoquinoline alkaloid skeleton. *Tetrahedron Lett.* **2005**, *46*, 5827–5830.
35. Irie, H.; Shiina, A.; Fushimi, T.; Katakawa, J.; Fujii, N.; Yajima, H. New synthesis of isoquinoline alkaloids, thalifoline, corypalline, and cherylline. *Chem. Lett.* **1980**, 875–878.
36. Katakawa, J.; Yoshimatsu, H.; Yoshida, M.; Zhang, Y.; Irie, H.; Yajima, H. Synthesis of Amaryllidaceae Alkaloids, (±)-Cherylline and (±)-Latifine. *Chem. Pharm. Bull.* **1988**, *36*, 3928–3932.
37. Kessar, V.; Singh, P.; Chawla, R.; Kumar, P. Cyclization of ortho-halogenated *N*-acylbenzylamines: A formal synthesis of (±)-cherylline. *Chem. Soc. Chem. Commun.* **1981**, 1074–1075.
38. Schwartz, M.A.; Scott, S.W. Biogenetically patterned synthesis of (+–)-cherylline. *J. Org. Chem.* **1971**, *36*, 1827–1829.
39. Kametani, T.; Takahashi, K.; Van Loc, C. Studies on the syntheses of heterocyclic compounds—DXCI: Total synthesis of (±)-cherylline and corgoine through quinonoid intermediates. *Tetrahedron* **1975**, *31*, 235–238.
40. Honda, T.; Namiki, H.; Saton, F. Palladium-catalyzed intramolecular δ -lactam formation of aryl halides and amide-enolates: Syntheses of cherylline and latifine. *Org. Lett.* **2001**, *3*, 631–633.
41. Freter, K.; Dubois, E.; Thomas, A. A new tetrahydroisoquinoline synthesis. *J. Heterocycl. Chem.* **1970**, *7*, 159–169.
42. Lebrun, S.; Couture, A.; Deniau, E.; Grandclaudon, P. A new synthesis of (+)- and (–)-cherylline. *Org. Biomol. Chem.* **2003**, *1*, 1701–1706.
43. Bonacorso, H.G.; Andrighetto, R.; Zanatta, N.; Martins, M.A.P. The unexpected cyclization routes of *N,N'*-bis(oxotrifluoroalkenyl)-1,3-phenylenediamines in polyphosphoric acid medium. *Tetrahedron Lett.* **2010**, *51*, 3752–3755.
44. Ivanov, I.; Nikolova, S.; Statkova-Abeghe, S. Efficient one-pot Friedel-Crafts acylation of benzene and its derivatives with unprotected aminocarboxylic acids in polyphosphoric acid. *Synth. Commun.* **2006**, *36*, 1405–1411.
45. Ivanov, I.; Nikolova, S.; Statkova-Abeghe, S. A simple method for the synthesis of 1-substituted beta-carboline derivatives from tryptamine and carboxylic acids in polyphosphoric acid. *Heterocycles* **2005**, *65*, 2483–2492.
46. Aoyama, T.; Takido, T.; Kodomari, M. Silica gel-supported polyphosphoric acid (PPA/SiO₂) as an efficient and reusable catalyst for conversion of carbonyl compounds into oxathioacetals and dithioacetals. *Synlett* **2004**, *13*, 2307–2310.

47. Itoh, K.-I.; Aoyama, T.; Satoh, H.; Fujii, Y.; Sakamaki, H.; Takido, T.; Kodomari, M. Application of silica gel-supported polyphosphoric acid (PPA/SiO₂) as a reusable solid acid catalyst to the synthesis of 3-benzoylisoxazoles and isoxazolines. *Tetrahedron Lett.* **2011**, *52*, 6892–6895.
48. Zhang, Q.; Zhang, S.; Deng, Y. Recent advances in ionic liquid catalysis. *Green. Chem.* **2011**, *13*, 2619–2637.
49. Ansari, M.I.; Hussain, M.K.; Yadav, N.; Gupta, P.K.; Hajela, K. Silica supported perchloric acid catalysed rapid N-formylation under solvent-free conditions, *Tetrahedron Lett.* **2012**, *53*, 2063–2065.
50. Verma, R.S. Solvent-free organic syntheses using supported reagents and microwave irradiation. *Green. Chem.* **1999**, *1*, 43–55.
51. Corma, A.; Garcia, H. Silica-bound homogenous catalysts as recoverable and reusable catalysts in organic synthesis. *Adv. Synth. Catal.* **2006**, *348*, 1391–1412.
52. Kantevari, S.; Bantu, R.; Nagarapu, L. HClO₄-SiO₂ and PPA-SiO₂ catalyzed efficient one-pot Knoevenagel condensation, Michael addition and cyclo-dehydration of dimedone and aldehydes in acetonitrile, aqueous and solvent free conditions: Scope and limitations. *J. Mol. Catal. A-Chem.* **2007**, *269*, 53–57.
53. Klumpp, D.A.; Sanchez, G.V., Jr.; Aguirre, S.L.; Zhang, Y.; de Leon, S. Chemistry of dicationic electrophiles: Superacid-catalyzed reactions of amino acetals. *J. Org. Chem.* **2002**, *67*, 5028–5031.
54. Khojastehnezhad, A.; Moeinpour, F.; Davoodnia, A. PPA-SiO₂ catalyzed efficient synthesis of polyhydroquinoline derivatives through Hantzsch multicomponent condensation under solvent-free conditions. *Chin. Chem. Lett.* **2011**, *22*, 807–810.
55. Vittorio, F.; Santagati, N.A.; Duro, R.; Duro, F.; Caruso, A.; Amico Roxas, M.; Trombadore, S. Alkyl derivatives of isoquinoline. III. Synthesis and pharmacologic activity of dialkylaminoalkyl amides of 1-chloro- and 1-methoxy-3-carboxy-4-methylisoquinoline and of 3-carboxy-2,4-dimethyl-1,2-dihydro-1-oxoisoquinoline. *Farmaco Sci.* **1984**, *39*, 229–245.

Sample Availability: Samples of the compounds are available from the authors.

© 2013 by the authors; licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution license (<http://creativecommons.org/licenses/by/3.0/>).