

Supplementary Information

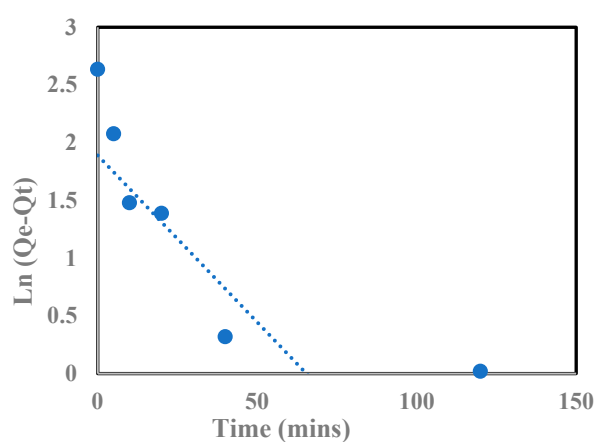
Investigating the potential of Greener-Porous Graphene for the treatment of organic pollutants in wastewater

Bhavya Joshi^{a*}, Ahmed M.E. Khalil^a, Shaowei Zhang^{a#}, Fayyaz A. Memon^a

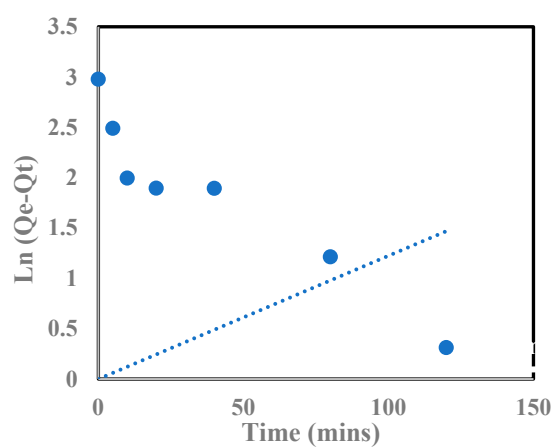
^a Department of Engineering, University of Exeter, UK EX4 4QF

Corresponding authors' e-mail addresses: bj300@exeter.ac.uk, f.a.memon@exeter.ac.uk, s.zhang@exeter.ac.uk,

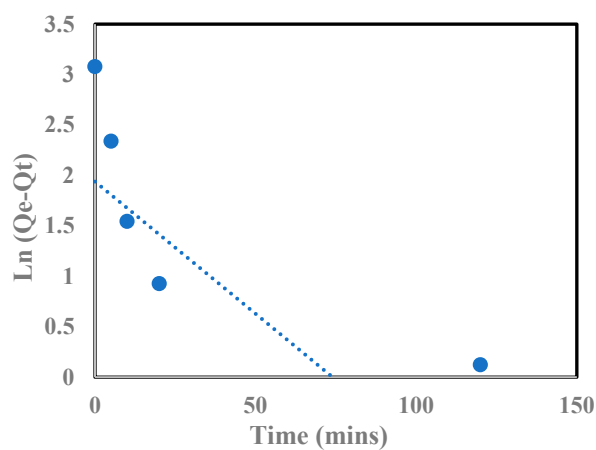
S1. Kinetic Pseudo 1st order Modelling Graphs



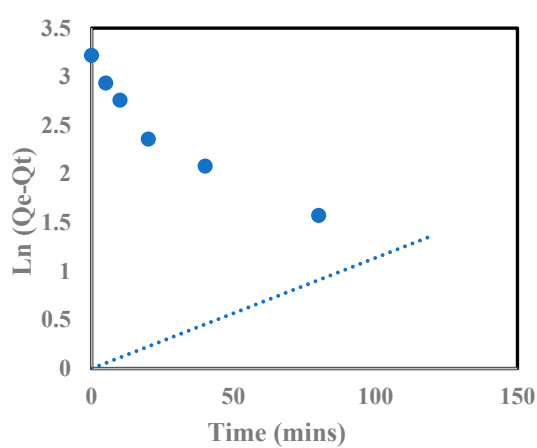
(a)



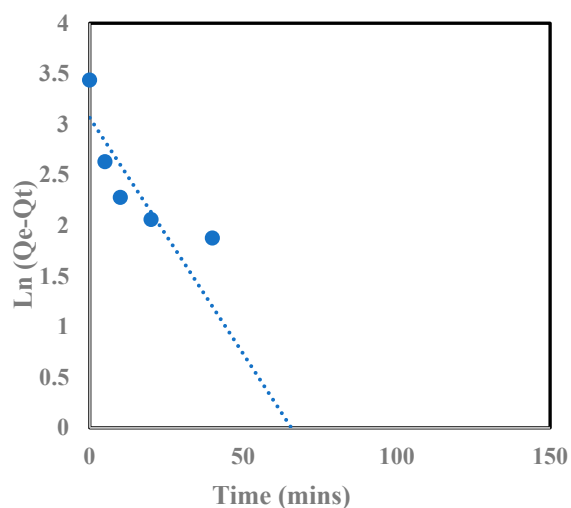
(b)



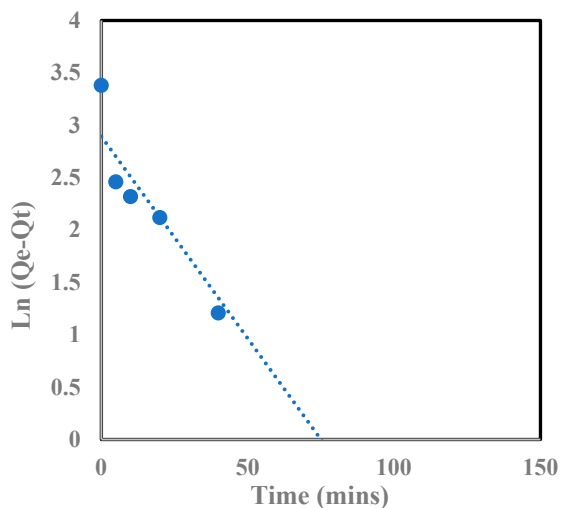
(c)



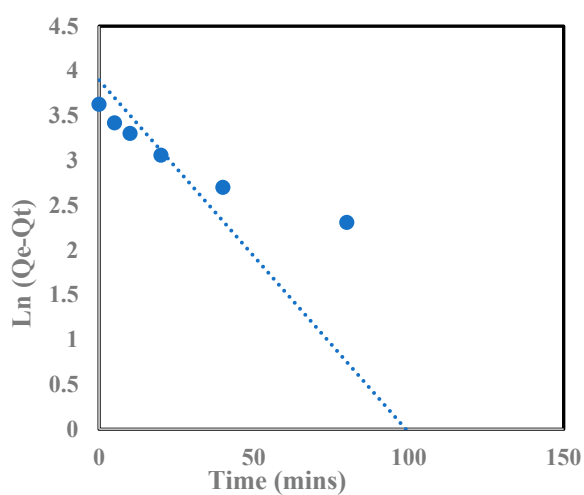
(d)



(e)



(f)



(g)

Figure S1: Pseudo 1st order model for: a) Contaminant AT; b) Contaminant CB; c) Contaminant CPF; d) IBU; e) Contaminant MeO; f) Contaminant MeR; g) Contaminant RD.

Dosages (mL)	MeR R.E%	MeO R.E%	RD R.E%	AT R.E%	CPF R.E%	CB R.E%	IBU R.E%
0.5	38.51	72.96	51.41	15.15	58.40	15.32	11.06
1.0	73.36	77.71	59.5	34.87	54.2	49.17	62.53
1.5	75.73	93.44	75.42	37.14	56.74	62.04	83.72
2.0	84.75	94.86	83.3	49.82	71.49	64.35	94.14
2.5	89.03	92.91	91.72	49.88	73.88	68.04	92.43
3.0	92.76	93.83	87.64	53.31	77.52	70.72	94.28

Table S1: Summary of Equilibrium study of greener PG on seven emerging contaminants

SA ^a (m ² g ⁻¹)	PV ^b (cm ³ g ⁻¹)	PD ^c (nm)
289.146m ² g ⁻¹	0.255 cc/g	3.169nm

Table S2: Parameters of the greener PG sample obtained from the N₂ adsorption-desorption isotherms.

a: BET surface area

b: Pore volume-DFT desorption cumulative volume of the pores

c: Pore diameter- average pore diameter determined by the BJH method.

Pore width: 3.169nm

Mesopore vol: 0.255 cc/g

Total pore volume: 2.738e-01 cc/g for

CONTAMINANT	Freundlich Isotherm Model	Langmuir Isotherm Model	Temkin Isotherm Model
ATENOLOL (AT)	$K_f (\{ \text{mg/g} \} \{ \text{mg/l} \}^{1/m}) = 1889117.276$ $m = 1/n = -2.02$ $R^2 = 0.93$	$Q_{\text{max}} (\text{mg/g}) = 3.79$ $K_L (\text{l/mg}) = -0.24$ $R^2 = 0.77$	$K_t (\text{L/mg}) = 0.90$ $B_t (\text{J/mol}) = -25.32$ $R^2 = 0.92$
CARBAMAZEPINE (CB)	$K_f (\{ \text{mg/g} \} \{ \text{mg/l} \}^{1/m}) = 154.54$ $m = 1/n = -1.43$ $R^2 = 0.92$	$Q_{\text{max}} (\text{mg/g}) = 5.15$ $K_L (\text{l/mg}) = -9.77$ $R^2 = 0.73$	$K_t (\text{L/mg}) = 0.083$ $B_t (\text{J/mol}) = -20.99$ $R^2 = 0.98$
IBUPROFEN (IBU)	$K_f (\{ \text{mg/g} \} \{ \text{mg/l} \}^{1/m}) = 34.05894$ $m = 1/n = -0.69$ $R^2 = 0.764$	$Q_{\text{max}} (\text{mg/g}) = 7.61$ $K_L (\text{l/mg}) = -0.0104$ $R^2 = 0.390$	$K_t (\text{L/mg}) = 0.04$ $B_t (\text{J/mol}) = -11.55$ $R^2 = 0.89$
CIPROFLOXACIN (CPF)	$K_f (\{ \text{mg/g} \} \{ \text{mg/l} \}^{1/m}) = 47.75$ $m = -0.507$ $R^2 = 0.98$	$Q_{\text{max}} (\text{mg/g}) = 17.15$ $K_L (\text{l/mg}) = -0.003$ $R^2 = 0.95$	$K_t (\text{L/mg}) = 0.041$ $B_t (\text{J/mol}) = -13.26$ $R^2 = 0.99$
METHYL RED (MeR)	$K_f (\{ \text{mg/g} \} \{ \text{mg/l} \}^{1/m}) = 4227.29$ $m = 1/n = -0.39$ $R^2 = 0.82$	$Q_{\text{max}} (\text{mg/g}) = 19.799$ $K_L (\text{l/mg}) = -0.0011$ $R^2 = 0.46$	$K_t (\text{L/mg}) = 0.024$ $B_t (\text{J/mol}) = -9.911$ $R^2 = 0.89$

METHYL ORANGE (MeO)	$K_f (\{ \text{mg/g} \} \{ \text{mg/l} \}^{1/m}) = 34.74$ $m=1/n= -0.15$ $R^2 = 0.98$	$Q_{\text{max}} (\text{mg/g}) = 28.52$ $K_L (\text{l/mg}) = -0.00018$ $R^2 = 0.92$	$K_t (\text{L/mg}) = 0.001$ $B_t (\text{J/mol}) = -5.31$ $R^2 = 0.988$
RHODAMINE-B (Rd)	$K_f (\{ \text{mg/g} \} \{ \text{mg/l} \}^{1/m}) = 156.3$ $m=1/n= -0.046$ $R^2 = 0.98$	$Q_{\text{max}} (\text{mg/g}) = 155.54$ $K_L (\text{l/mg}) = 24197.99$ $R^2 = 0.94$	$K_t (\text{L/mg}) = 2.9e^{-10}$ $B_t (\text{J/mol}) = -7.122$ $R^2 = 0.98$

Table S3: Linear adsorption isotherm parameters for all seven contaminants, removal by greener PG.

