

Supplementary Information

Catalyst design: Counter anion effect on Ni nanocatalysts anchored on hollow carbon spheres

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Results

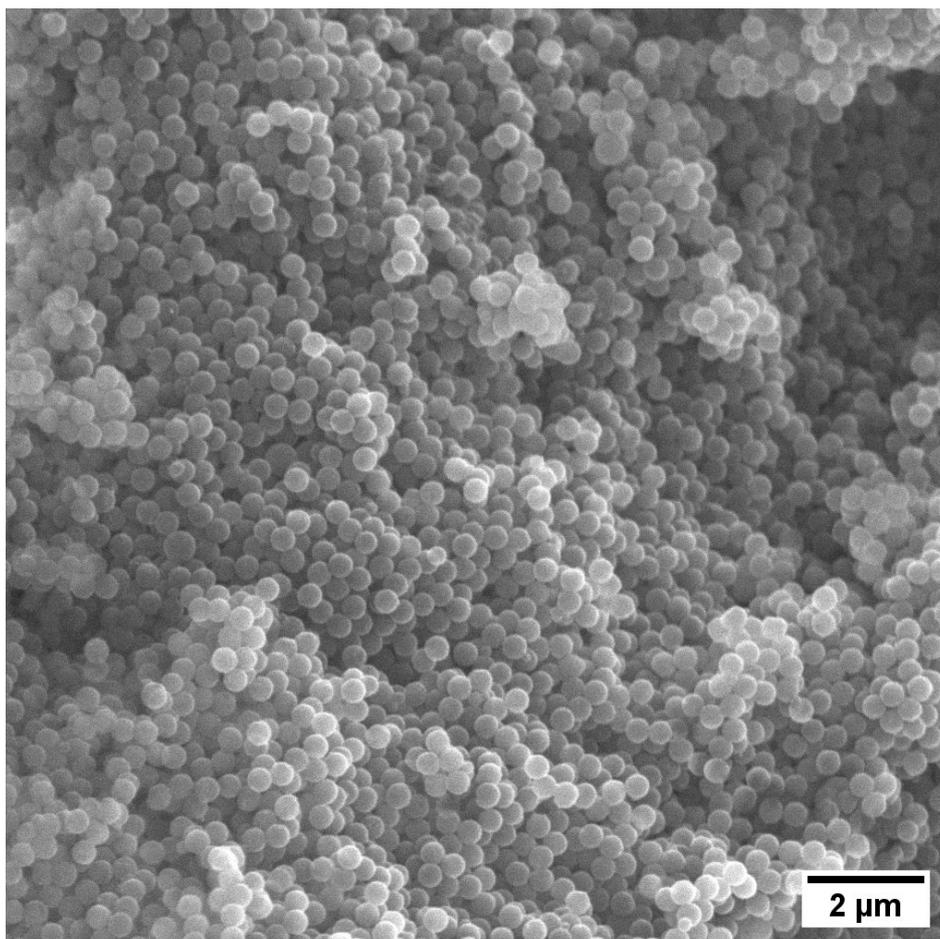


Figure S1: SEM micrographs of the PSS template.

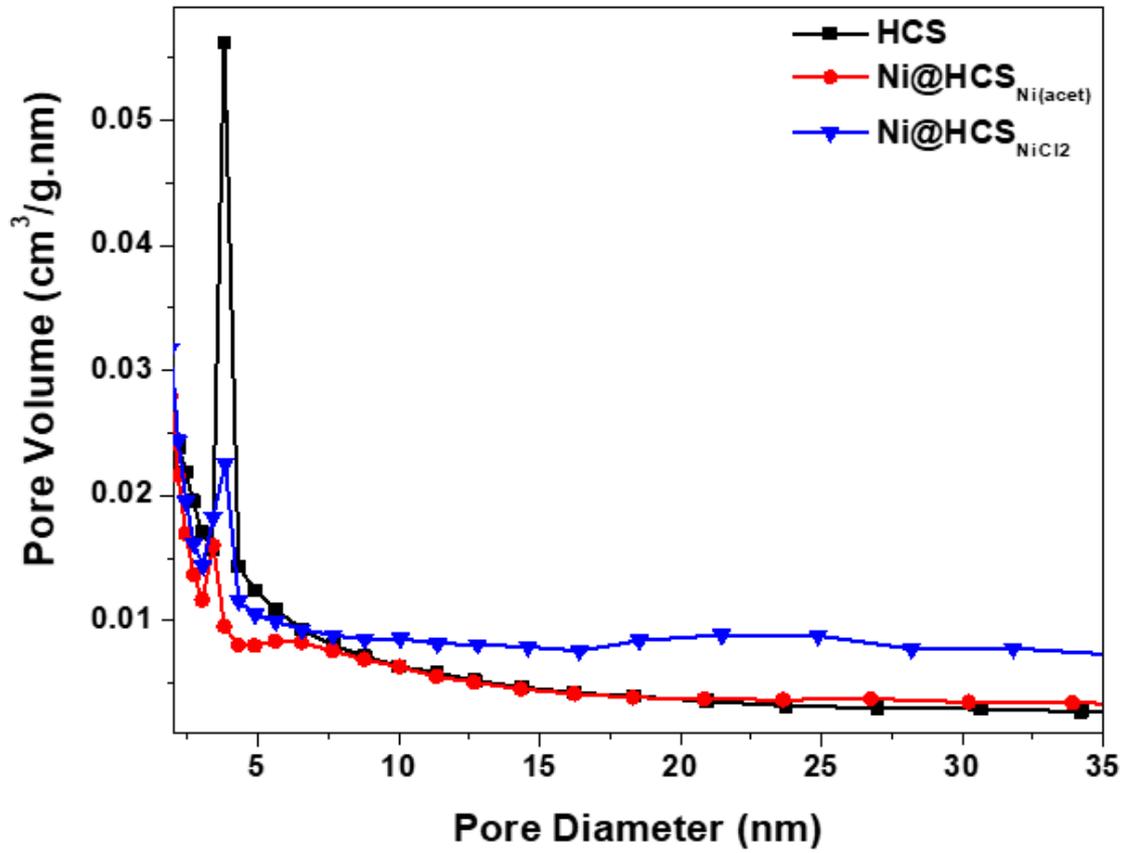


Figure S2: Pore size distributions of the pristine HCSs and Ni@HCSs nanocatalysts.

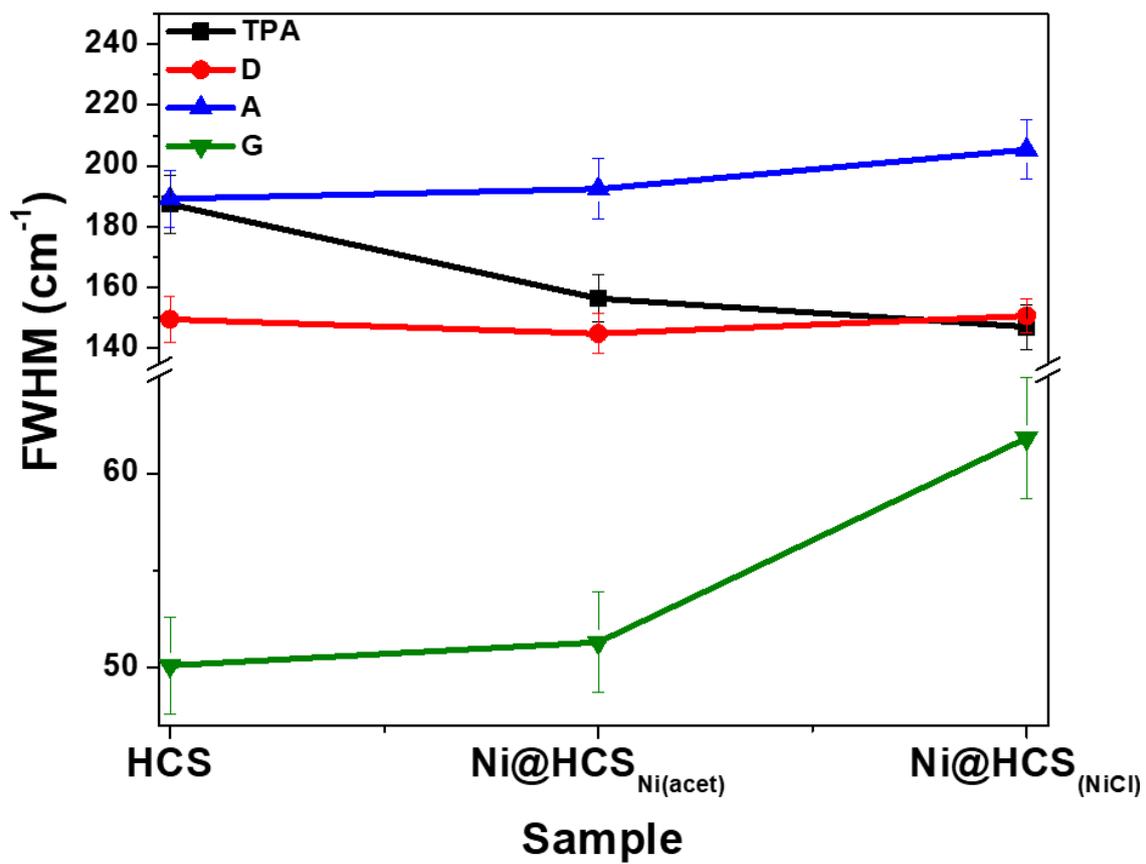


Figure S3: Peak broadening for all Raman-active bands for the pristine HCSs and nanocatalysts.

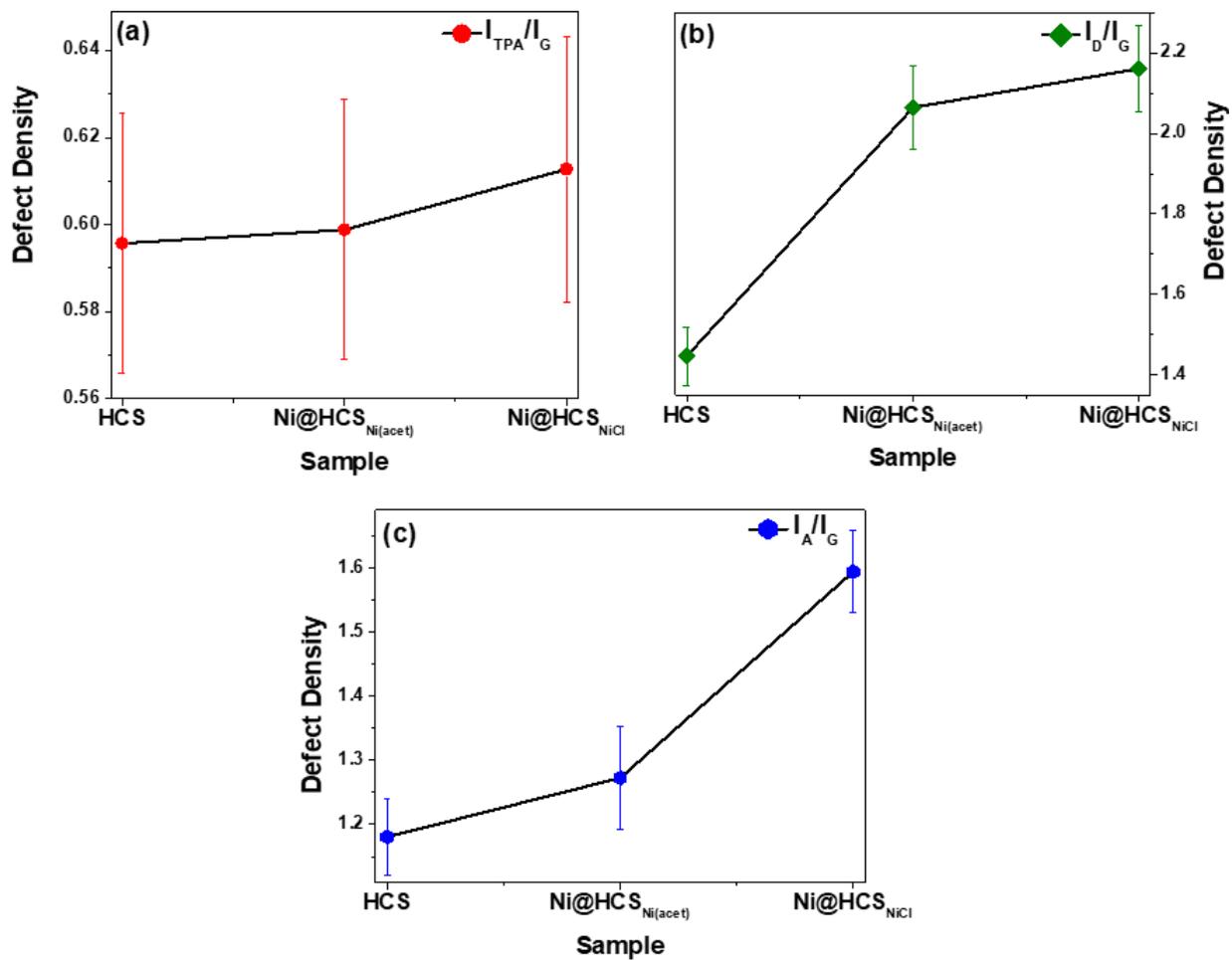


Figure S4: Defect density ratios (a) I_{TPA}/I_G , (b) I_D/I_G and (c) I_A/I_G of the pristine and Ni@HCSs samples.

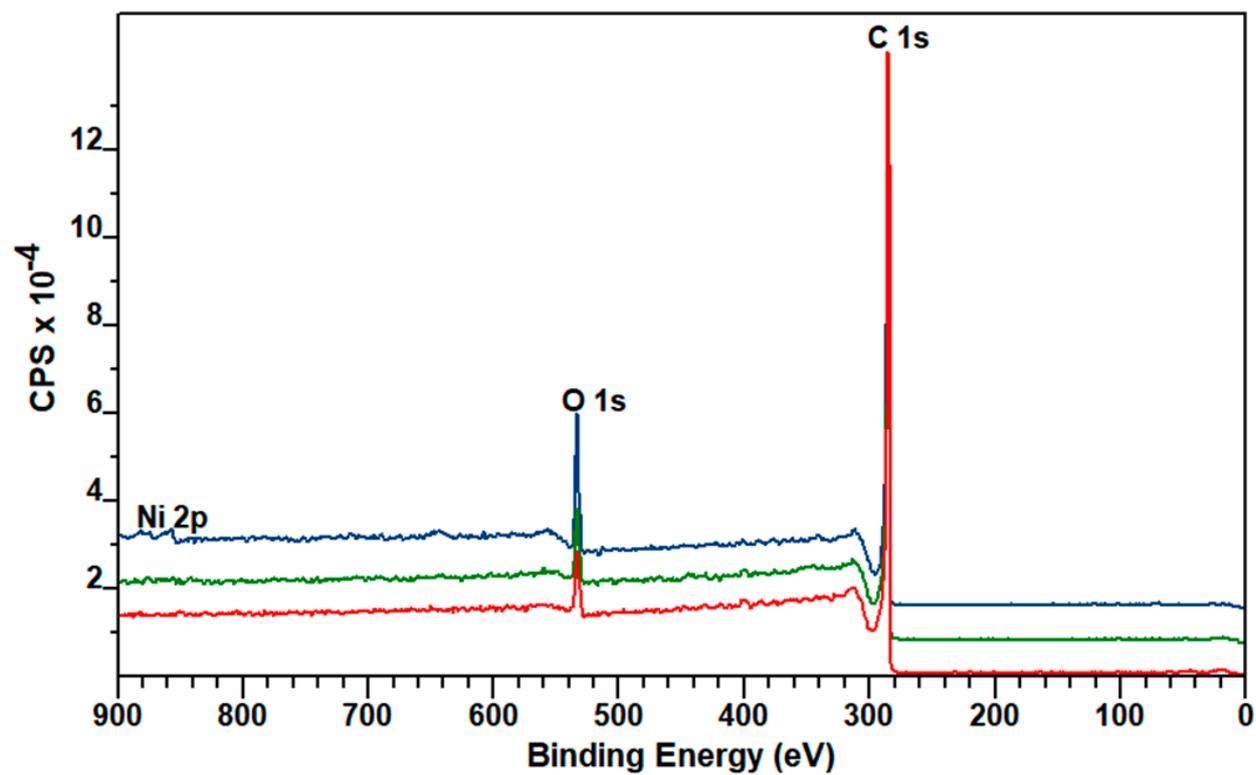


Figure S5: Survey XPS spectra of pristine HCSs (red), Ni@HCSsNi(acet)₂ (blue), and Ni@HCSsNiCl₂ (green) nanocatalysts

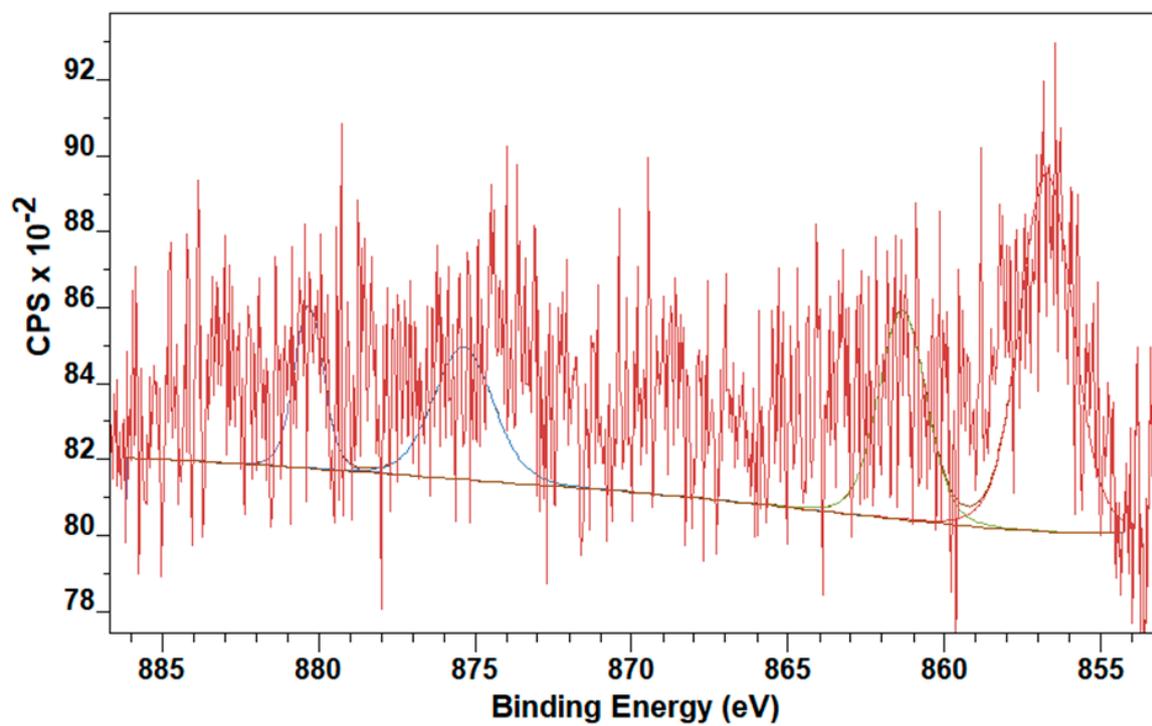


Figure S6: Deconvoluted Ni 2p spectrum of Ni@HCSNiCl₂ nanocatalysts.

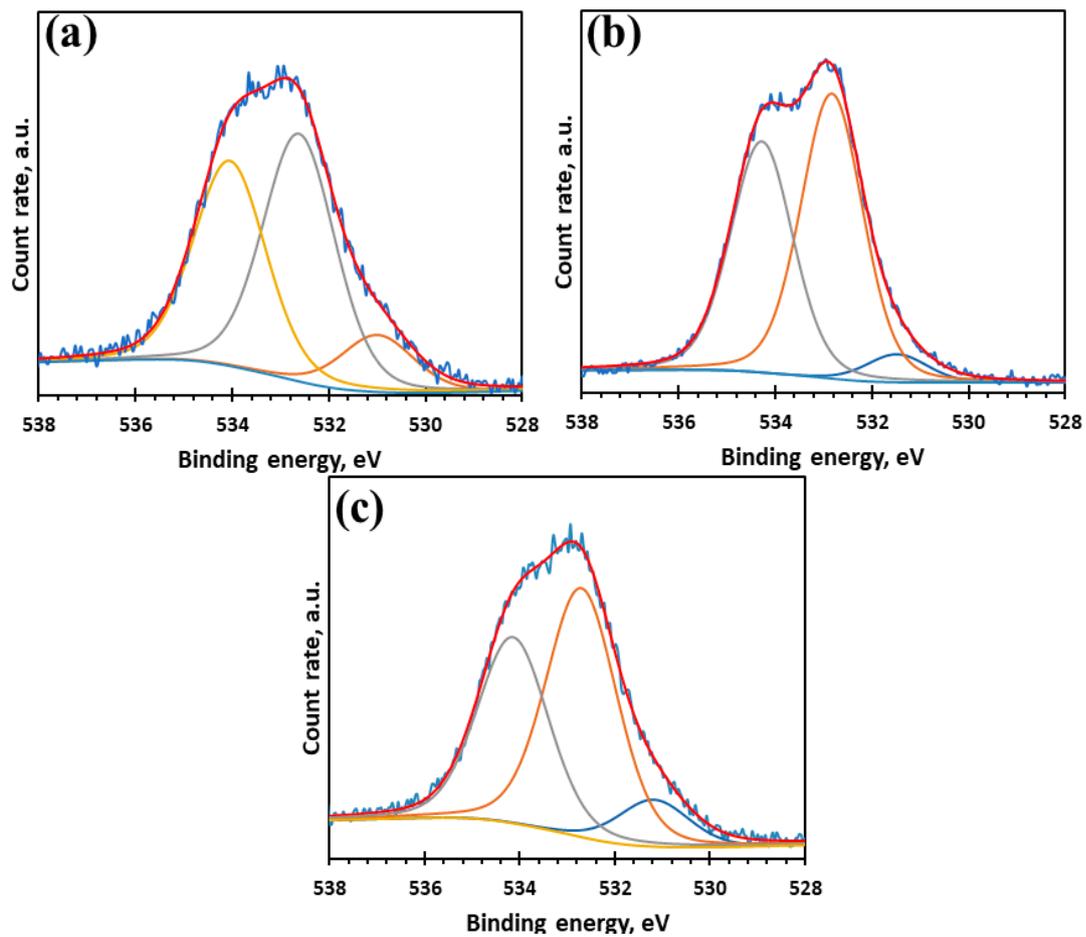


Figure S7: Deconvoluted O1s spectra of (a) pristine HCSs, (b) Ni@HCS_SNi_(acet)₂, and (c) Ni@HCS_SNiCl₂ nanocatalysts.

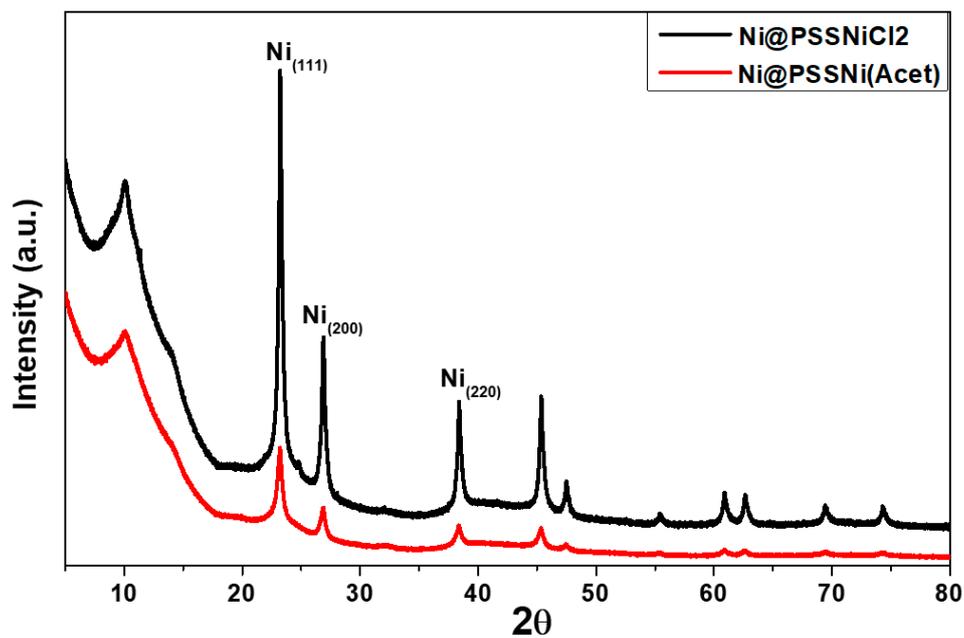


Figure S8: XRD patterns of the Ni nanoparticles on the PSS templates prior to RF coating.

Table S1: Morphological parameters based on TEM micrographs

Sample	Size (nm)				
	Diameter	Cavity	Shell	Ni nanoparticle	
				TEM	XRD
HCSs	342 ± 17	327 ± 16	10 ± 2.1	-	-
Ni@HCSs _{NiCl2}	332 ± 25	281 ± 21	23 ± 4.7	12 ± 1.9	10.9
Ni@HCSs _{Ni(acet)}	326 ± 25	245 ± 19	42 ± 6.9	6.4 ± 1.2	8.6

Table S2: Thermal stability properties of pristine and Ni@HCSs samples

<i>Sample</i>	<i>Onset Temp (°C)</i>	<i>Decomp. Temp. (°C)</i>			<i>% Residue</i>
		(i)	(ii)	(iii)	
<i>HCSs</i>	568.1		642.7		0.76
<i>Ni@HCSs_{Ni(acet)}</i>	452.2	489.3	582.6	630.5	6.84
<i>Ni@HCSs_{NiCl2}</i>	426.1	484.5	552.5	592.9	10.2