

Supplementary Materials: *P*-Stereogenic Phosphines for the Stabilisation of Metal Nanoparticles. A Surface State Study

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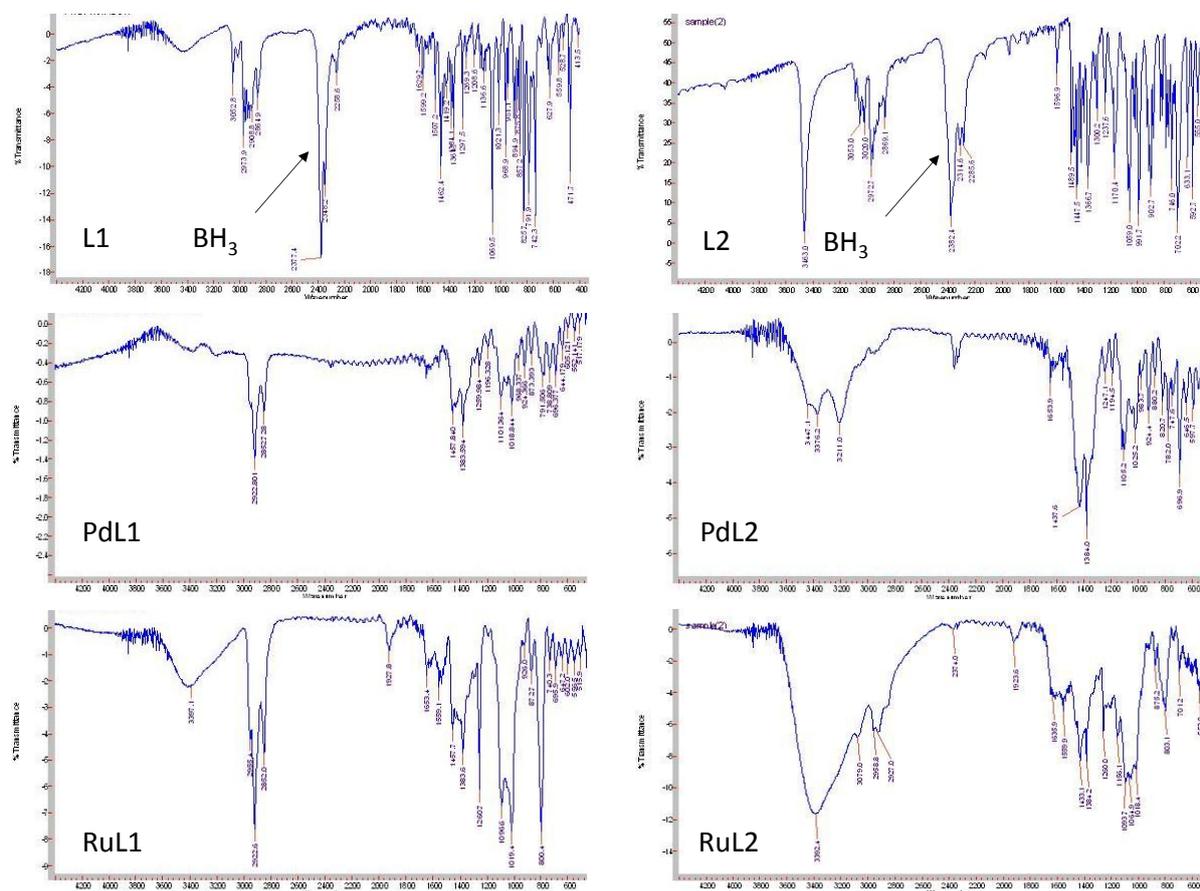


Figure S1. IR spectra (KBr pellets) of ligands L1 and L2 and the corresponding MNPs: PdL1, PdL2, RuL1 and RuL2.

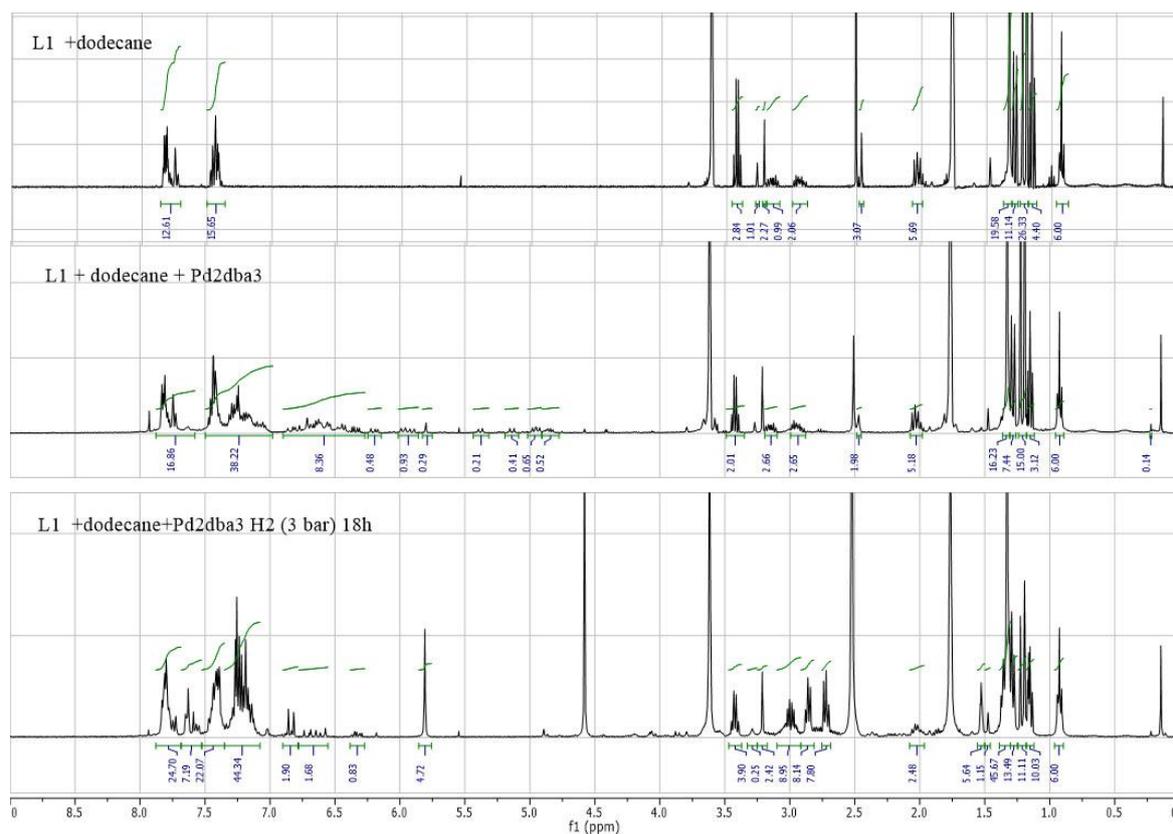


Figure S2. ¹H NMR (400 MHz, solvent, 298 K) spectra corresponding to the monitoring of the formation of PdL1.

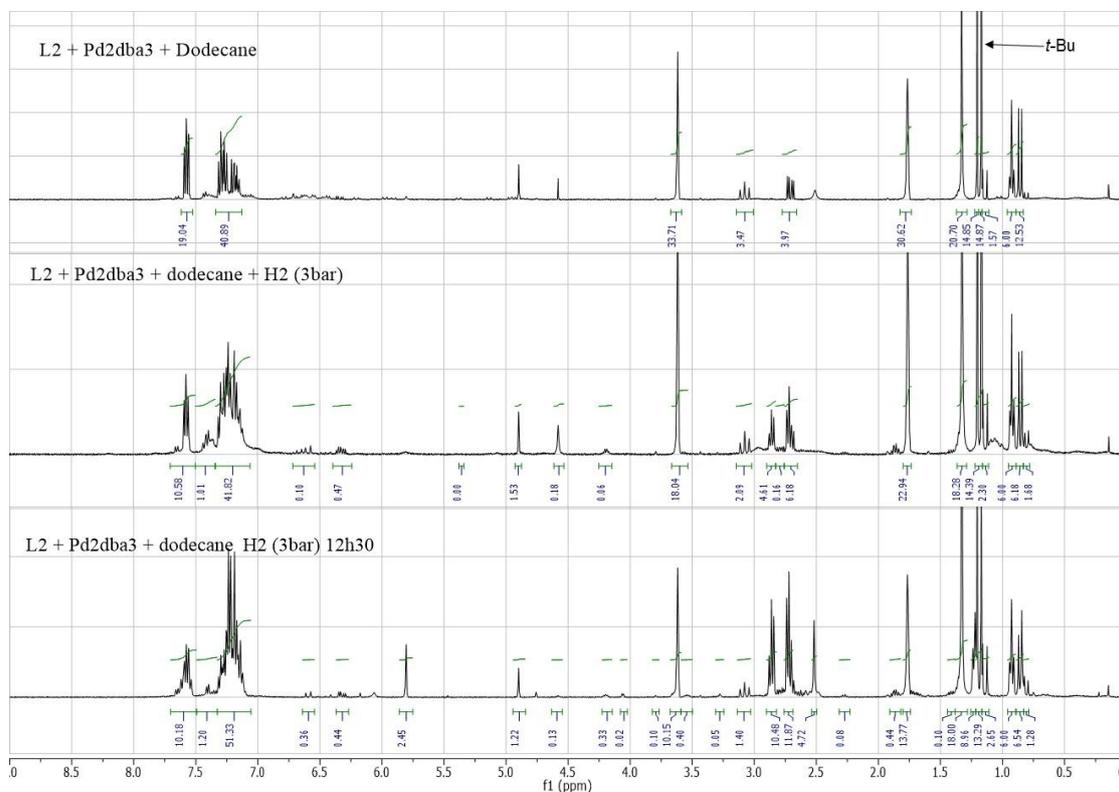


Figure S3. ¹H NMR (400 MHz, solvent, 298 K) spectra corresponding to the monitoring of the formation of PdL2.

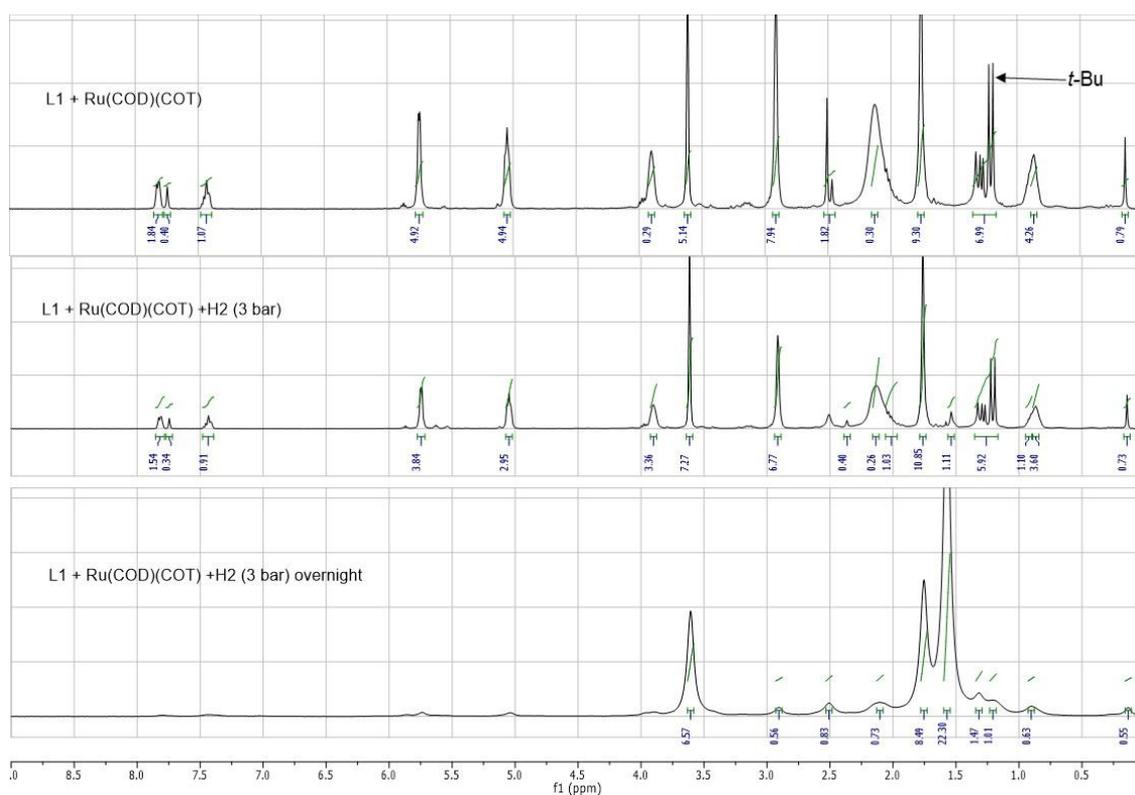


Figure S4. ^1H NMR (400 MHz, solvent, 298 K) spectra corresponding to the monitoring of the formation of **RuL1**.

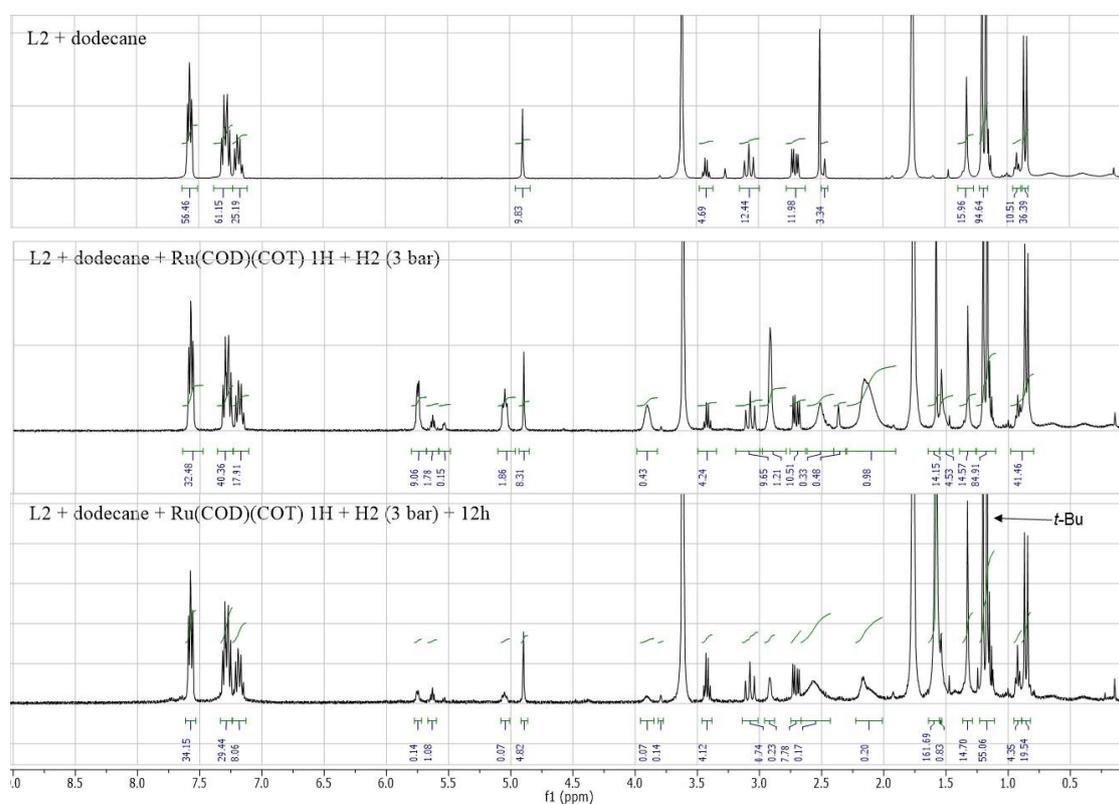


Figure S5. ^1H NMR (400 MHz, solvent, 298 K) spectra corresponding to the monitoring of the formation of **RuL2**.

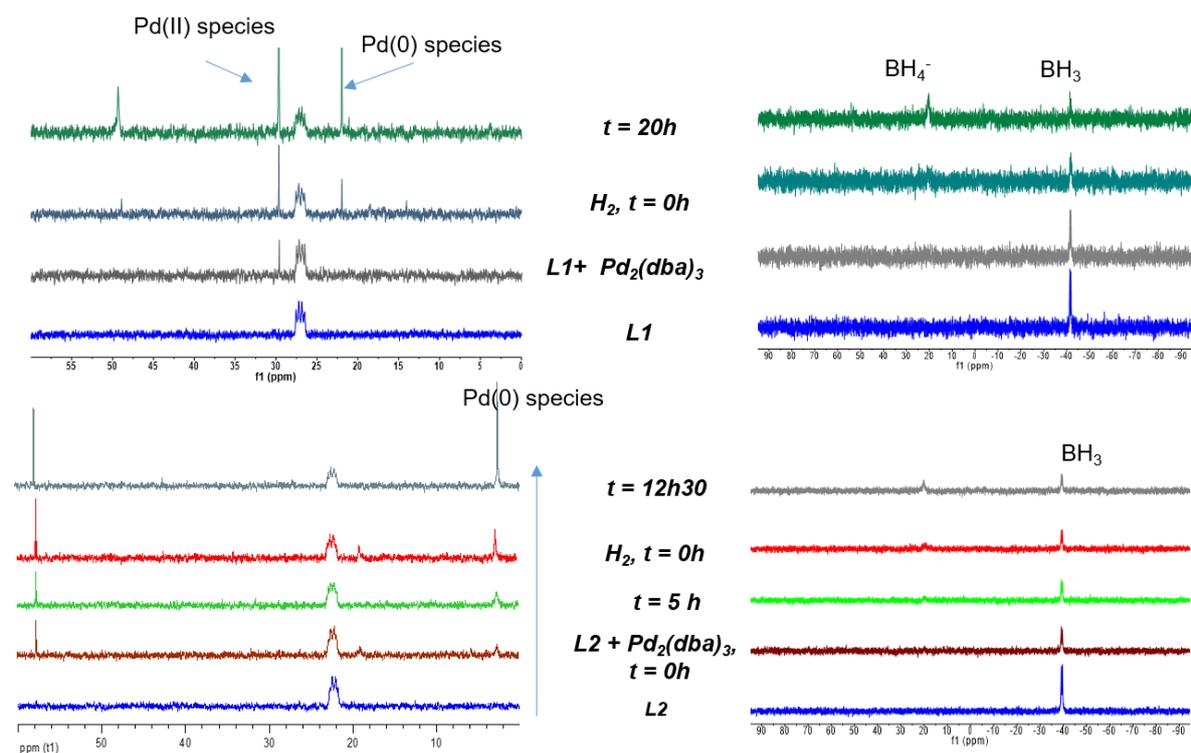


Figure S6. ^{31}P (162 MHz, solvent, 298 K) and ^{11}B (128 MHz, solvent, 298 K) NMR monitoring of the formation of PdL1 (top) and PdL2 (bottom).

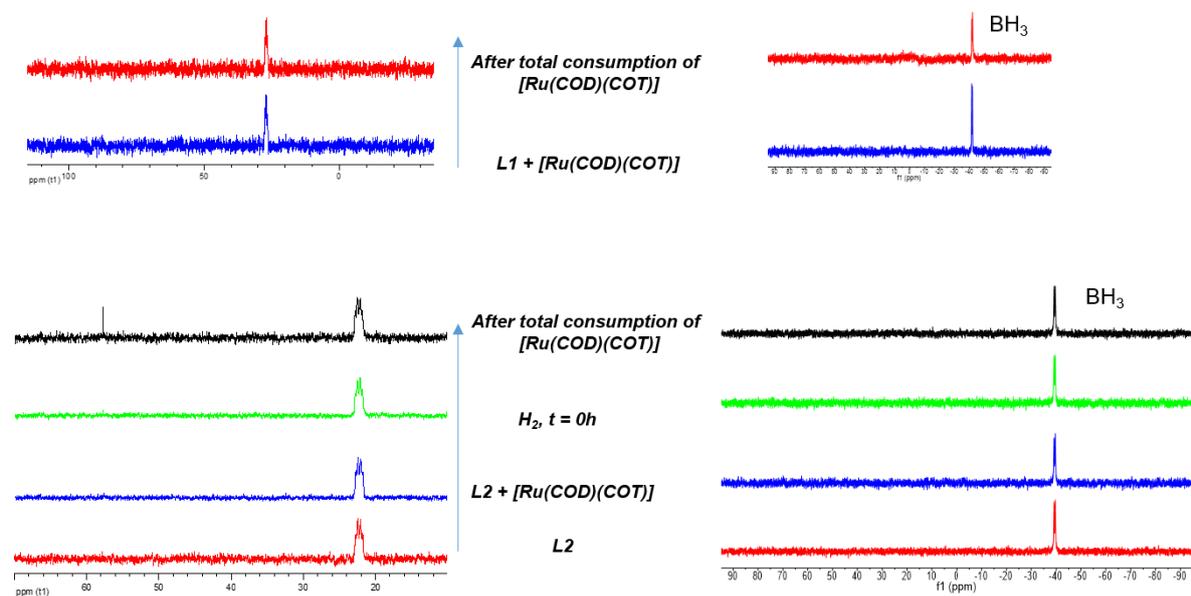


Figure S7. ^{31}P (162 MHz, solvent, 298 K) and ^{11}B (128 MHz, solvent, 298 K) NMR monitoring of the formation of RuL1 and RuL2.

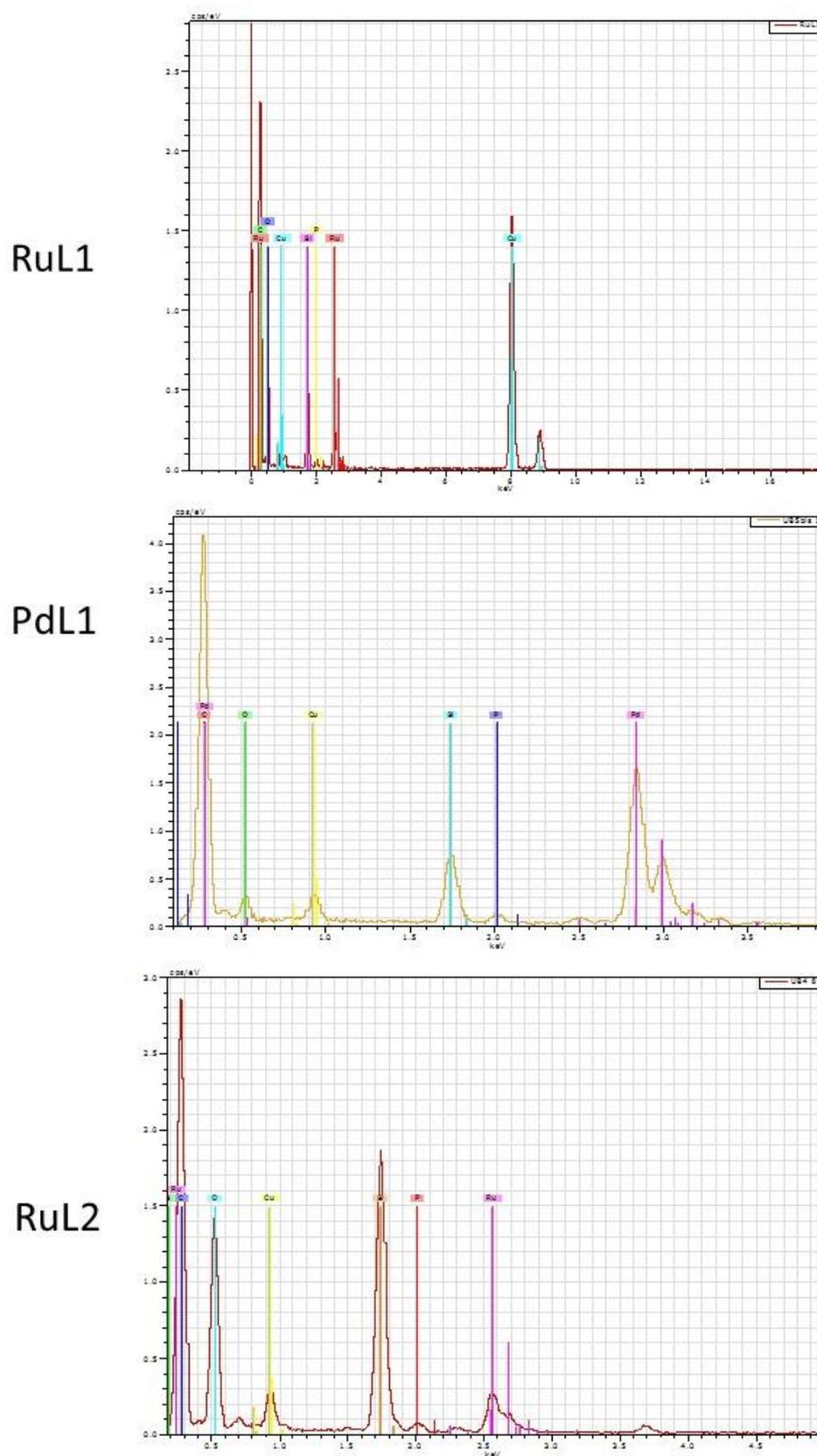


Figure S8. EDX (Energy Dispersive X-ray) analyses of PdL1, RuL1 and RuL2.

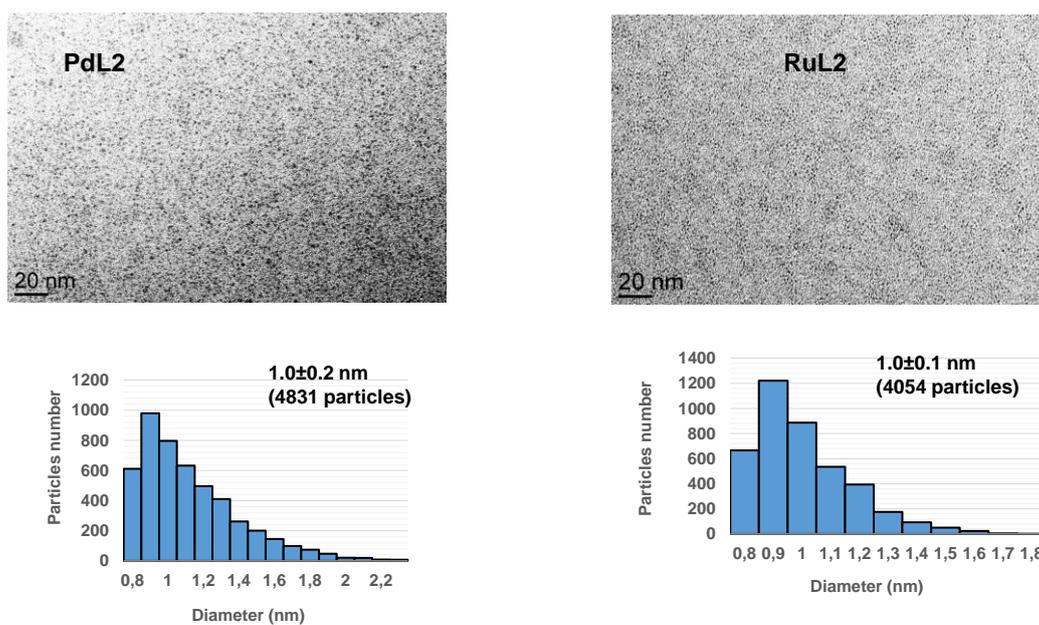


Figure S9. TEM (Transmission Electron Microscopy) images of PdL2 and RuL2 NPs stabilised with borane-free phosphine ligand, with the corresponding size distribution diagrams.

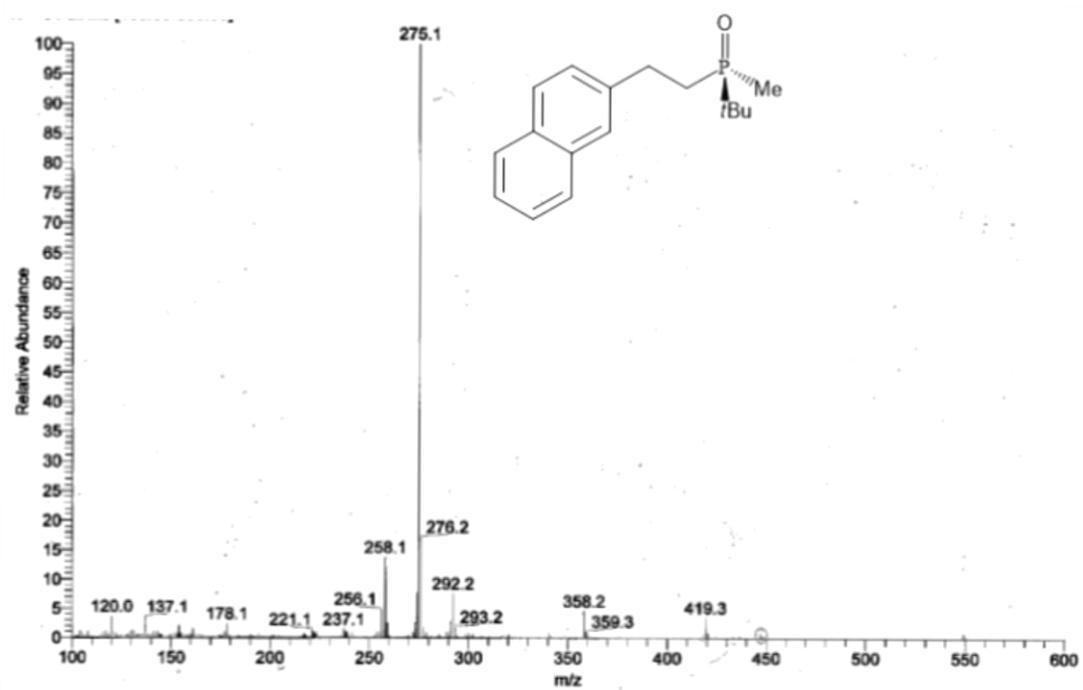


Figure S10. Cont.

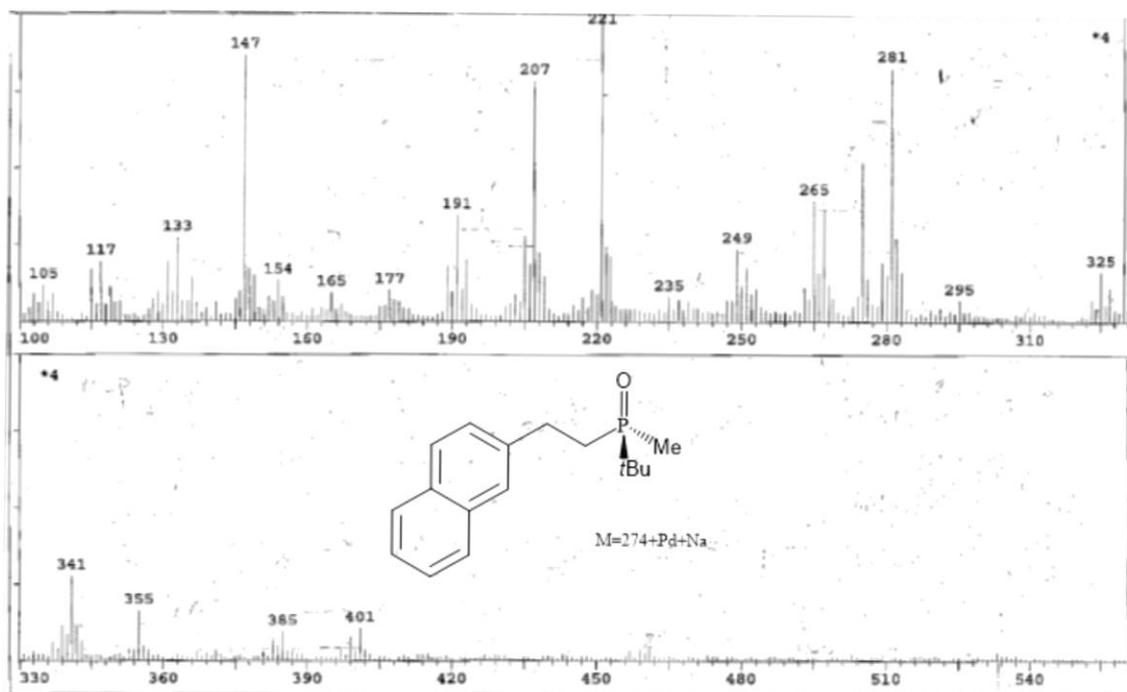


Figure S10. Mass spectrometry analyses of an organic phase of **PdL1** treated with H₂O₂(aq) and extracted with CH₂Cl₂ (FAB (**bottom**) and DCI/NH₃ analyses (**top**)).

For **PdL1**, we observed a peak at 275.1 corresponding to the borane-free phosphine ($M = 258$ g/mol) + NH₃ by DCI/NH₃ mass spectrometry and a peak at 401 corresponding to Pd(L1) + Na + O by FAB, which corresponds to the signal at +56 ppm in the ³¹P NMR spectrum.

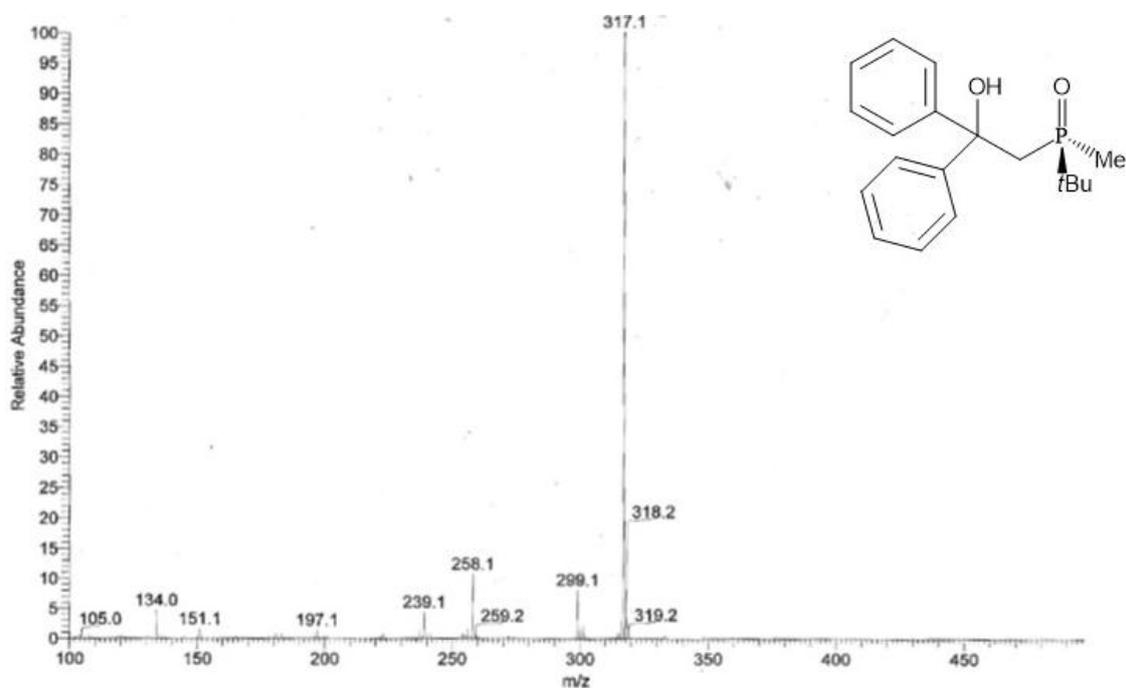


Figure S11. Cont.

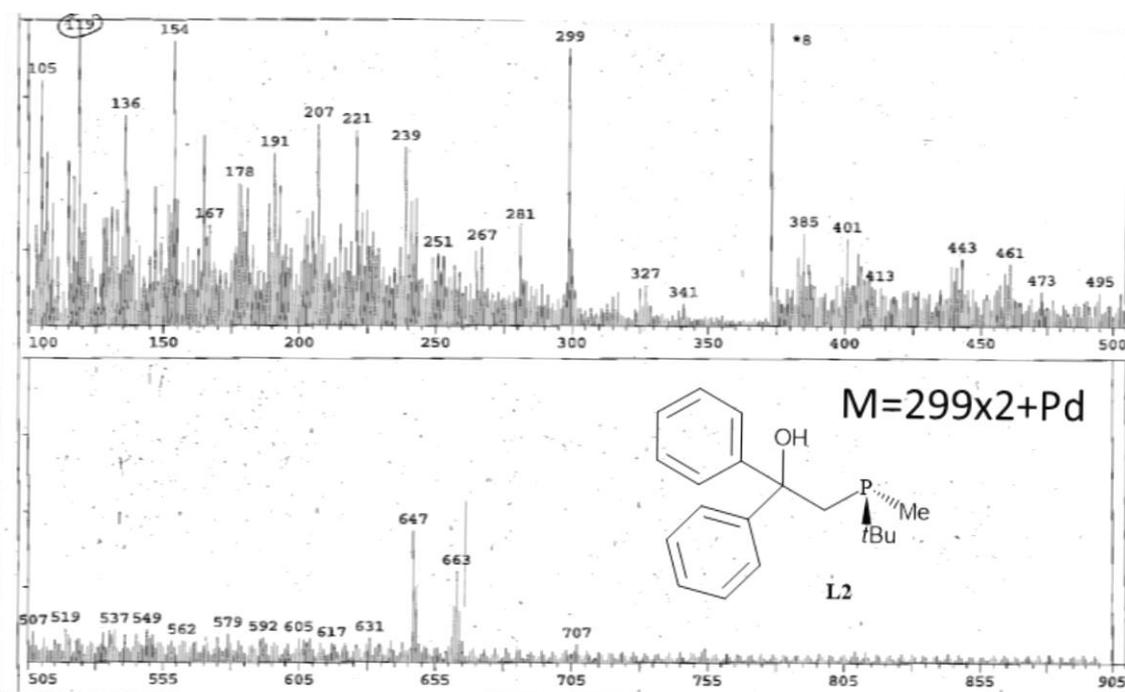


Figure S11. Mass spectrometry analyses of an organic phase of **PdL2** treated with H₂O₂(aq) and extracted with CH₂Cl₂ (FAB (**bottom**) and DCI/NH₃ analyses(**top**)).

For **PdL2**, we observed a peak at 317.1 corresponding to the borane-free phosphine **L2** ($M = 299$ g/mol) + NH₃ by DCI/NH₃ mass spectrometry and a peak at 707 corresponding to Pd(L₂)₂ by FAB, which corresponds to the signal at +60 ppm in the ³¹P NMR spectrum.

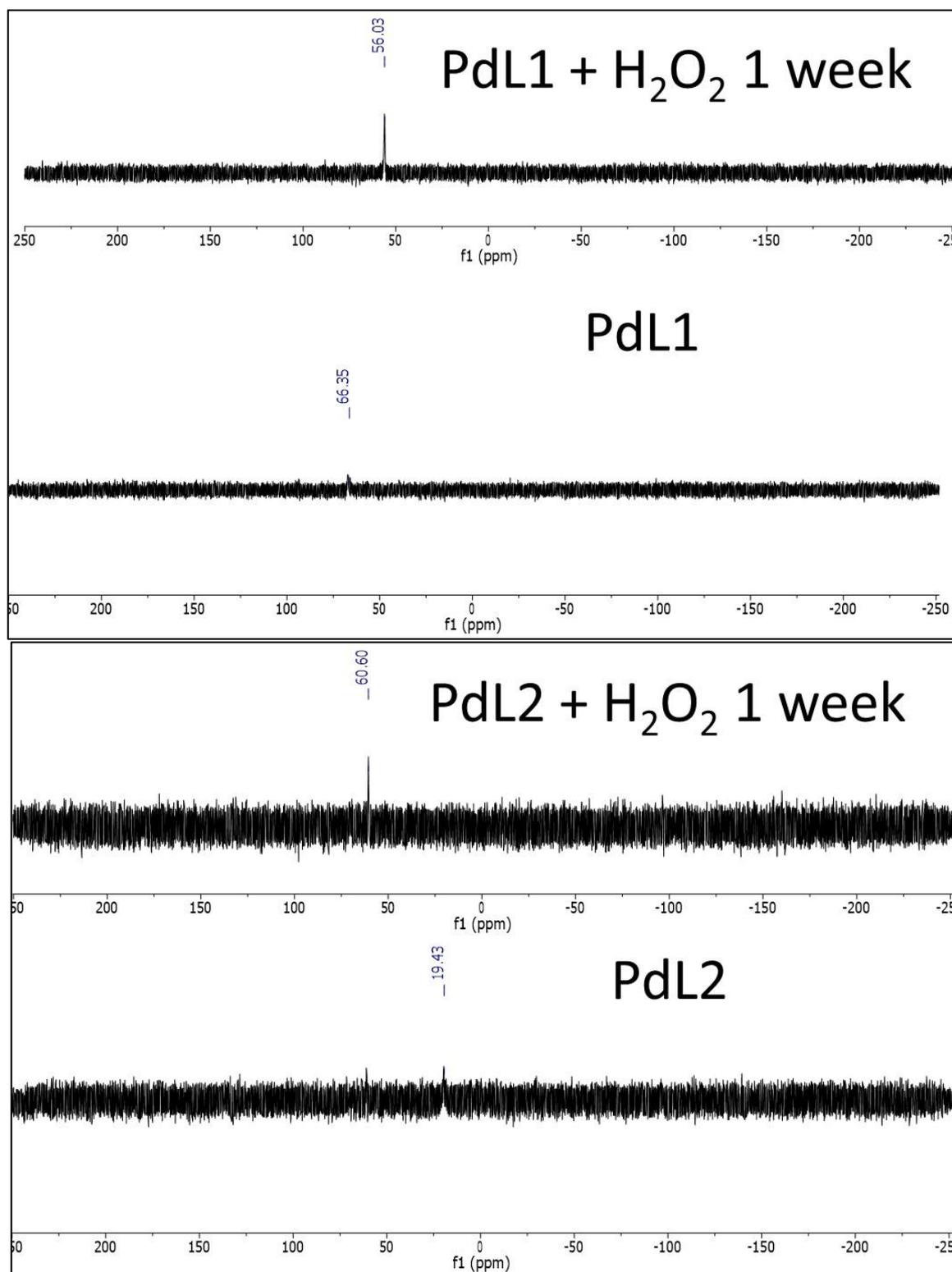
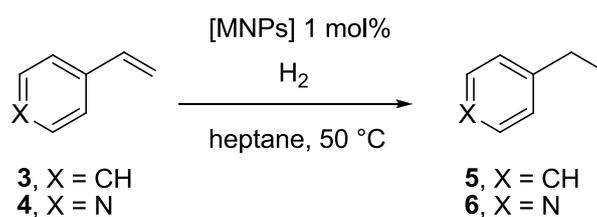


Figure S12. ^{31}P NMR spectra in CDCl_3 of PdNPs before and after addition of $\text{H}_2\text{O}_2(\text{aq})$.

Table S1. Hydrogenation of styrene and 4-vinylpyridine catalysed by PdL and RuL (L = L1, L2).¹

MNPs = PdL1, PdL2, RuL1, RuL2

Entry	Catalyst	Substrate	H ₂ Pressure (bar)	Time (h)	Conversion (%) ²
1	PdL1	3 or 4	3	1	100
2	PdL2	3 or 4	3	1	100
3	RuL1	3 or 4	3	16	100
4	RuL2	3 or 4	3	16	100
5	RuL1	3 or 4	20	16	100
6	RuL2	3 or 4	20	16	100
7	RuL2	3 or 4	40	16	96 ³
8	[Ru(COD)(COT)]	3	20	16	100 (64/36) ⁴

¹ Reaction conditions: to form the metal nanoparticles: 5 mg of [Pd₂(dba)₃] or 3 mg of [Ru(COD)(COT)] were mixed with 2 mg of ligand (L1 or L2) in heptane at room temperature overnight under 3 bar of H₂. Then 1 mmol of substrate (4-Vinylpyridine (105 mg) or styrene (104 mg)) with 142 mg of decane as internal standard were added. Then the reaction was run at 50 °C for 1 to 16h under 3–40 bar H₂;

² Determined by GC using decane as internal standard; ³ <5% of ethylcyclohexane or ethylpiperidine;

⁴ Mixture of ethylbenzene (64%) and ethylcyclohexane (36%).