

Supplementary Information

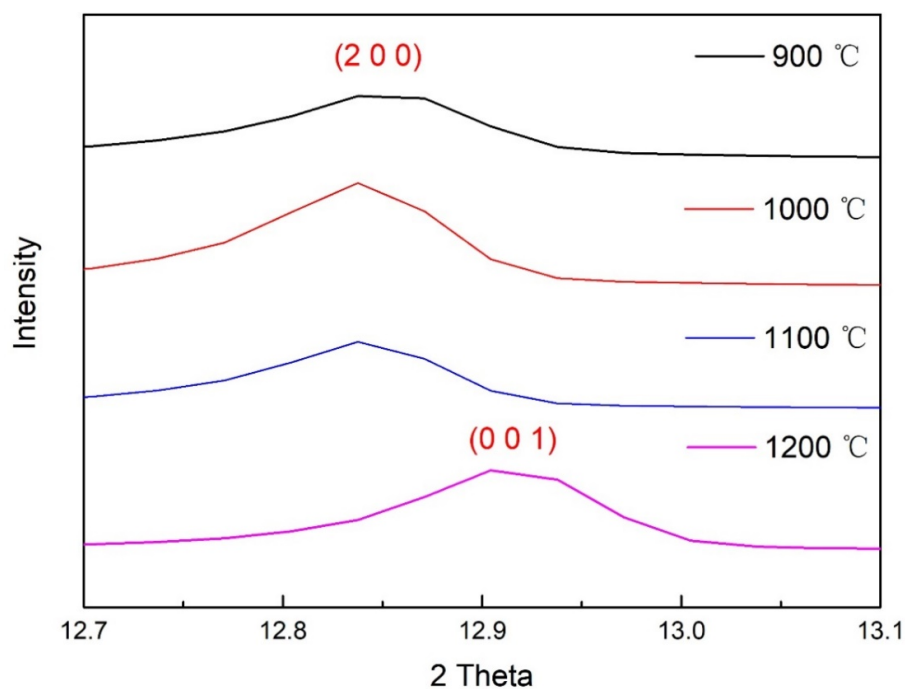


Figure S1. XRD patterns of MoO₃/MoS₂@G-900, MoO₃/MoS₂@G-1000, MoO₃/MoS₂@G-1100 and MoO₃/MoS₂@G-1200 around the (2 0 0) reflection of orthorhombic MoO₃, and the (0 0 1) reflection of monoclinic MoO₃.

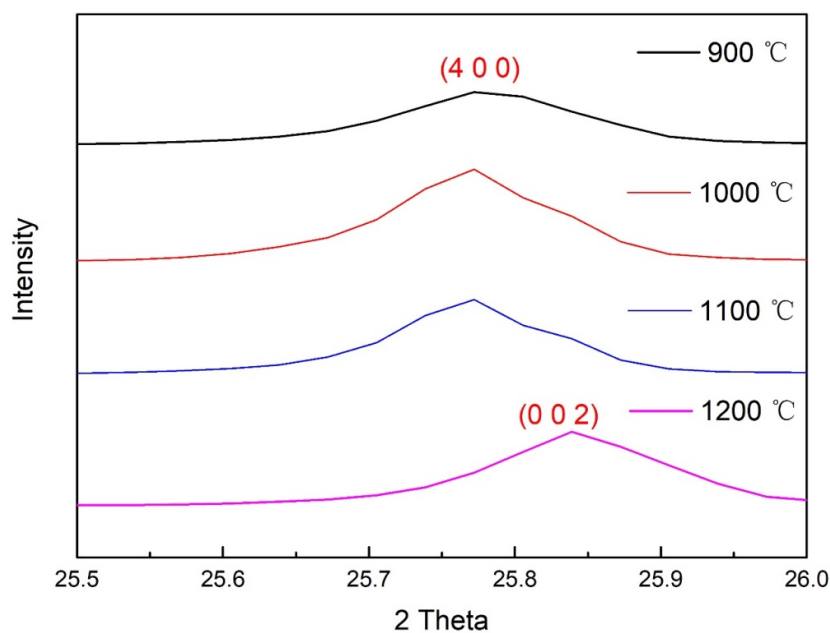


Figure S2. XRD pattern of MoO₃/MoS₂@G-900, MoO₃/MoS₂@G-1000, MoO₃/MoS₂@G-1100 and MoO₃/MoS₂@G-1200, around the (4 0 0) reflection of orthorhombic MoO₃, and the (0 0 2) reflection of monoclinic MoO₃.

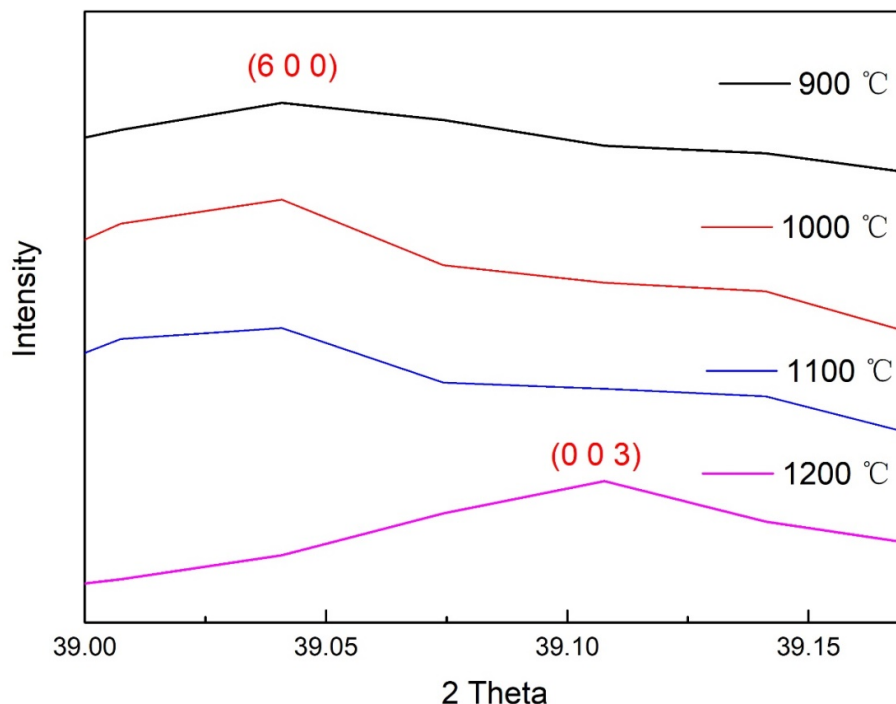


Figure S3. XRD patterns of MoO₃/MoS₂@G-900, MoO₃/MoS₂@G-1000, MoO₃/MoS₂@G-1100 and MoO₃/MoS₂@G-1200, around the (6 0 0) reflection of orthorhombic MoO₃, and the (0 0 3) reflection of monoclinic MoO₃.

Table S1. XRD peak intensity ratio corresponding to various peaks in MoO₃/MoS₂@G-900, MoO₃/MoS₂@G-1000, MoO₃/MoS₂@G-1100 and MoO₃/MoS₂@G-1200.

	Orthorhombic α - MoO ₃ (200)/ MoS ₂ (002)	Orthorhombic α - MoO ₃ (400)/C (002)	Orthorhombic α - MoO ₃ (600)/ MoS ₂ (103)
MoO ₃ /MoS ₂ @G-900	1.34	1.04	6.37
MoO ₃ /MoS ₂ @G-1000	2.56	1.42	7.81
MoO ₃ /MoS ₂ @G-1100	3.30	1.92	8.27
	Monoclinic β - MoO ₃ (0,0,1)/ MoS ₂ (0,0,2)	Monoclinic β - MoO ₃ (0,0,2)/C (0,0,2)	Monoclinic β - MoO ₃ (0,0,3)/ MoS ₂ (1,0,3)
MoO ₃ /MoS ₂ @G-1200	1.02	1.03	4.99

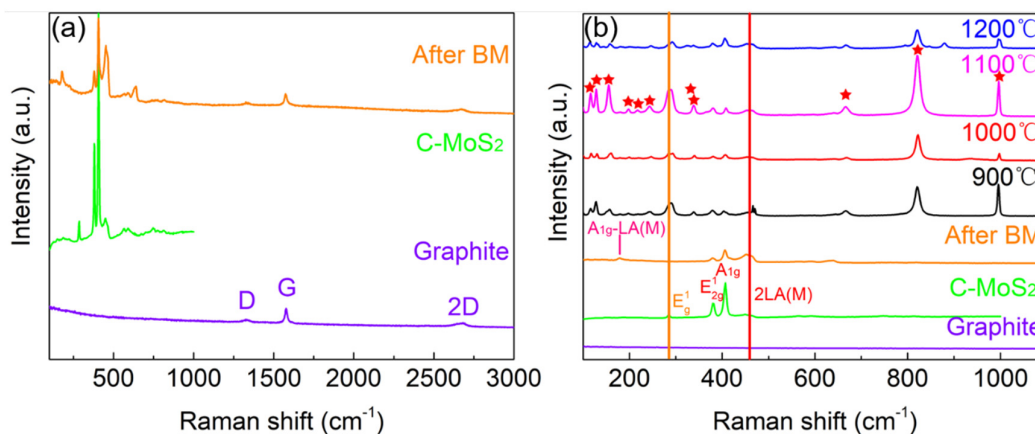


Figure S4. Raman spectra of various precursor materials and products obtained at various temperatures.

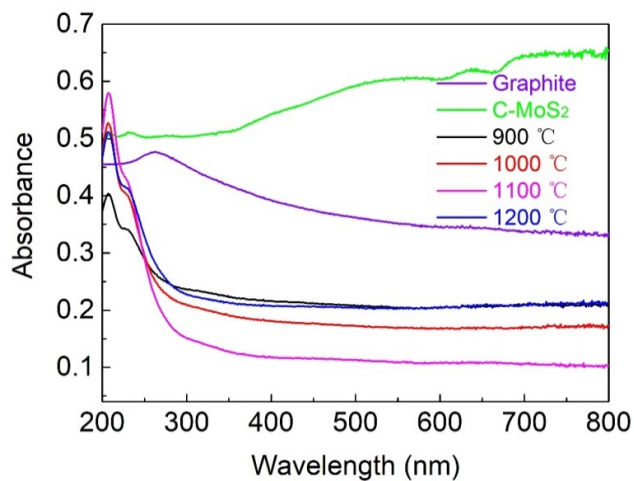


Figure S5. UV-vis spectra of various precursor materials and products obtained at various temperatures.

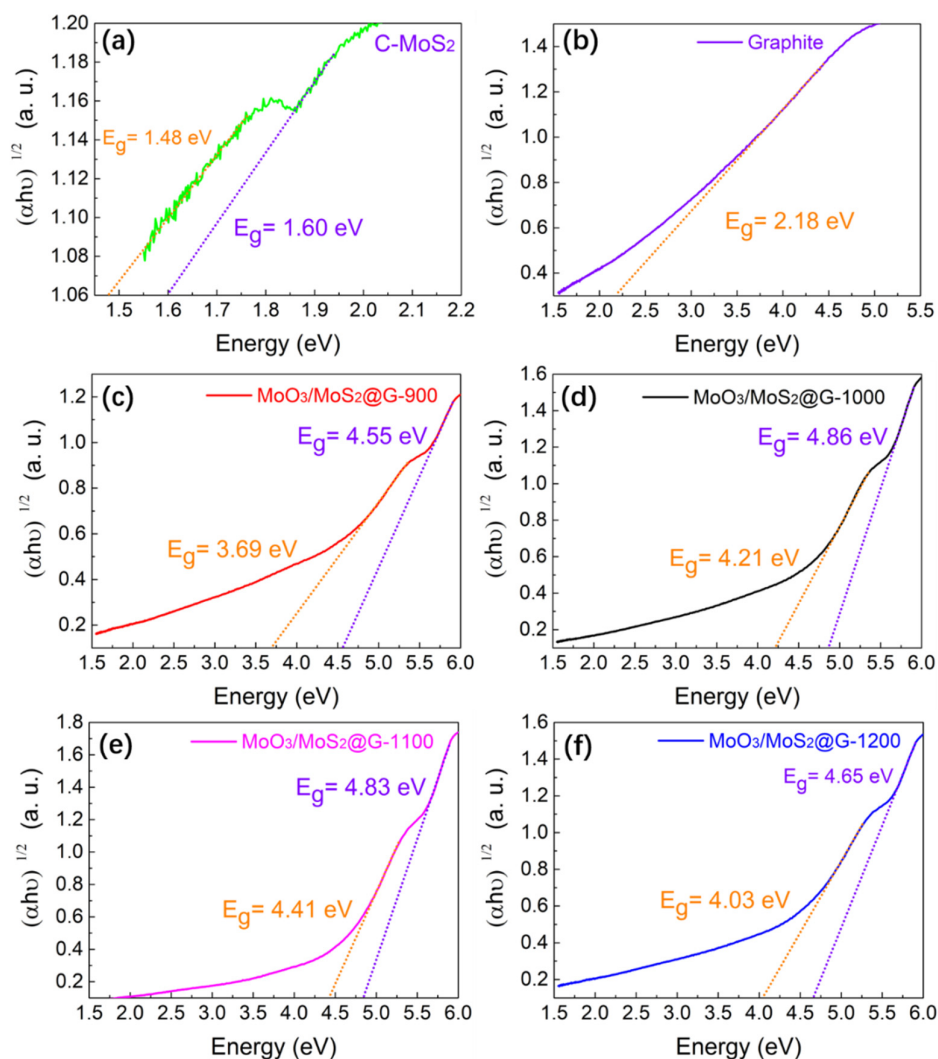


Figure S6. $(\alpha h\nu)^{1/2}$ versus $h\nu$ plots according to UV-vis spectra of various precursor materials and products obtained at various temperatures.

Table S2. Band gap values of various precursor materials and products obtained at various temperatures.

Sample	Eg1 (eV)	Eg2 (eV)
C-MoS ₂	1.48	1.60
Graphite	2.18	
900 °C	3.69	4.55
1000 °C	4.21	4.86
1100 °C	4.41	4.83
1200 °C	4.03	4.65

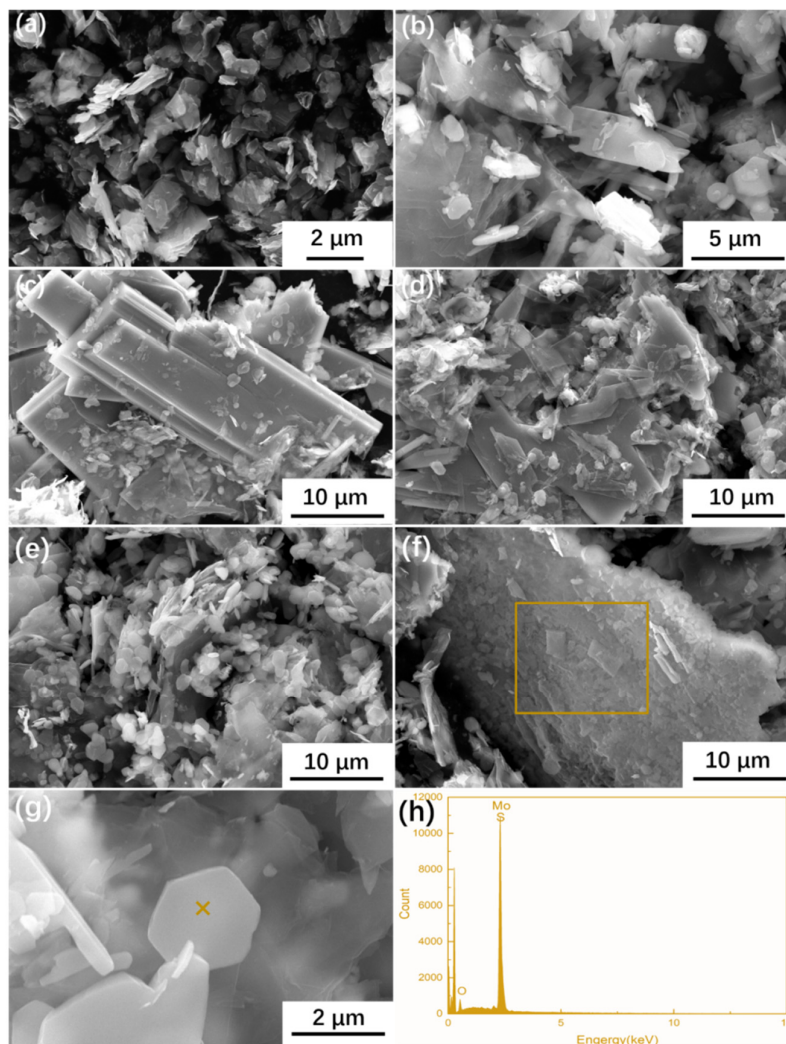


Figure S7. SEM image of (a) commercial MoS₂ and the samples prepared at (b) 900, (c) 1000, (d) 1100, and (e-f) 1200 °C. (g) High magnification image taken on the sample prepared at 1200 °C, and (h) the EDS analysis recorded on hexagonal MoS₂ shown in (g).

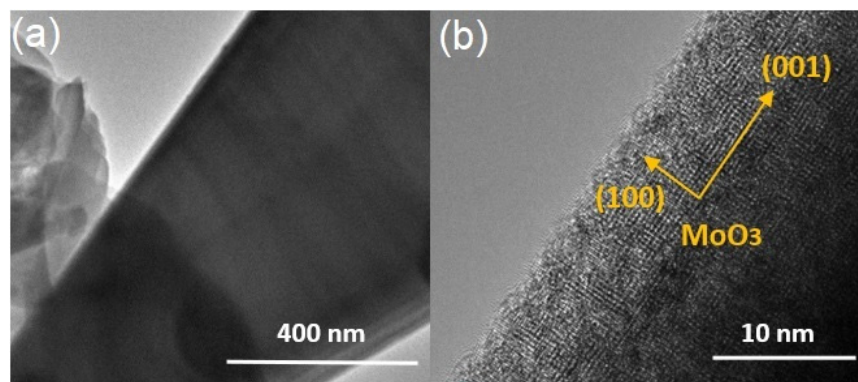


Figure S8. TEM micrographs of rod-like MoO_3 observed in $\text{MoO}_3/\text{MoS}_2@\text{G}-900$.

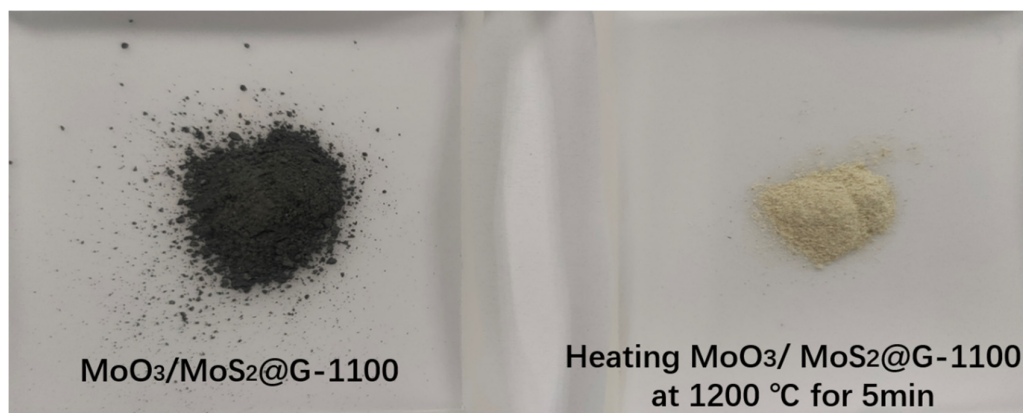


Figure S9. Appearance of $\text{MoO}_3/\text{MoS}_2@\text{G}-1100$, the same sample after heating at $1200\text{ }^\circ\text{C}$ for 5 min.

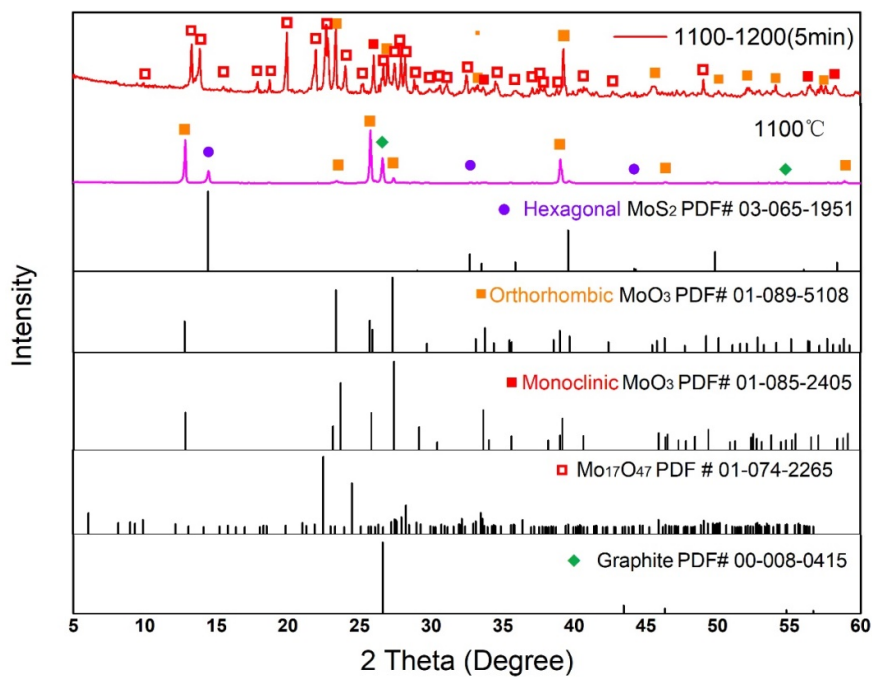


Figure S10. XRD patterns of $\text{MoO}_3/\text{MoS}_2@\text{G}-1100$ and the sample obtained by heating of $\text{MoO}_3/\text{MoS}_2@\text{G}-1100$ at $1200\text{ }^\circ\text{C}$ for 5 min.

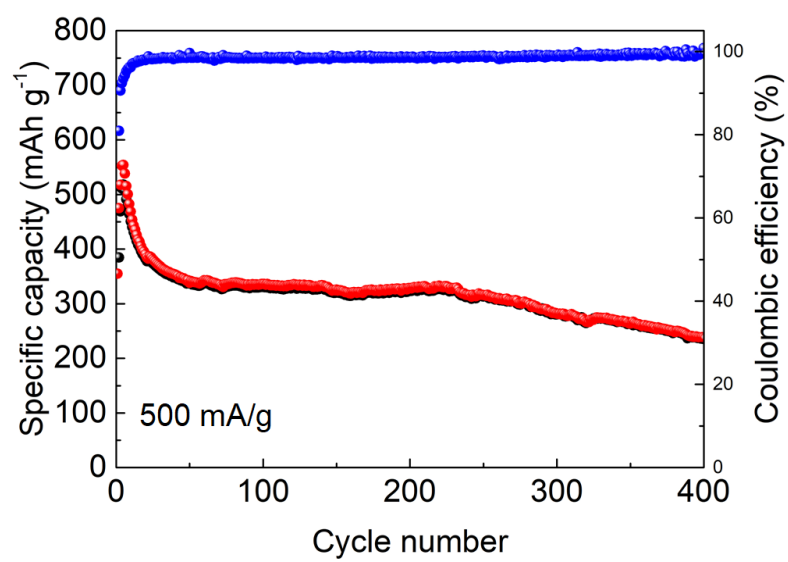


Figure S11. Cycling performance of MoO₃/MoS₂@G-1100 at 500 mA g⁻¹.

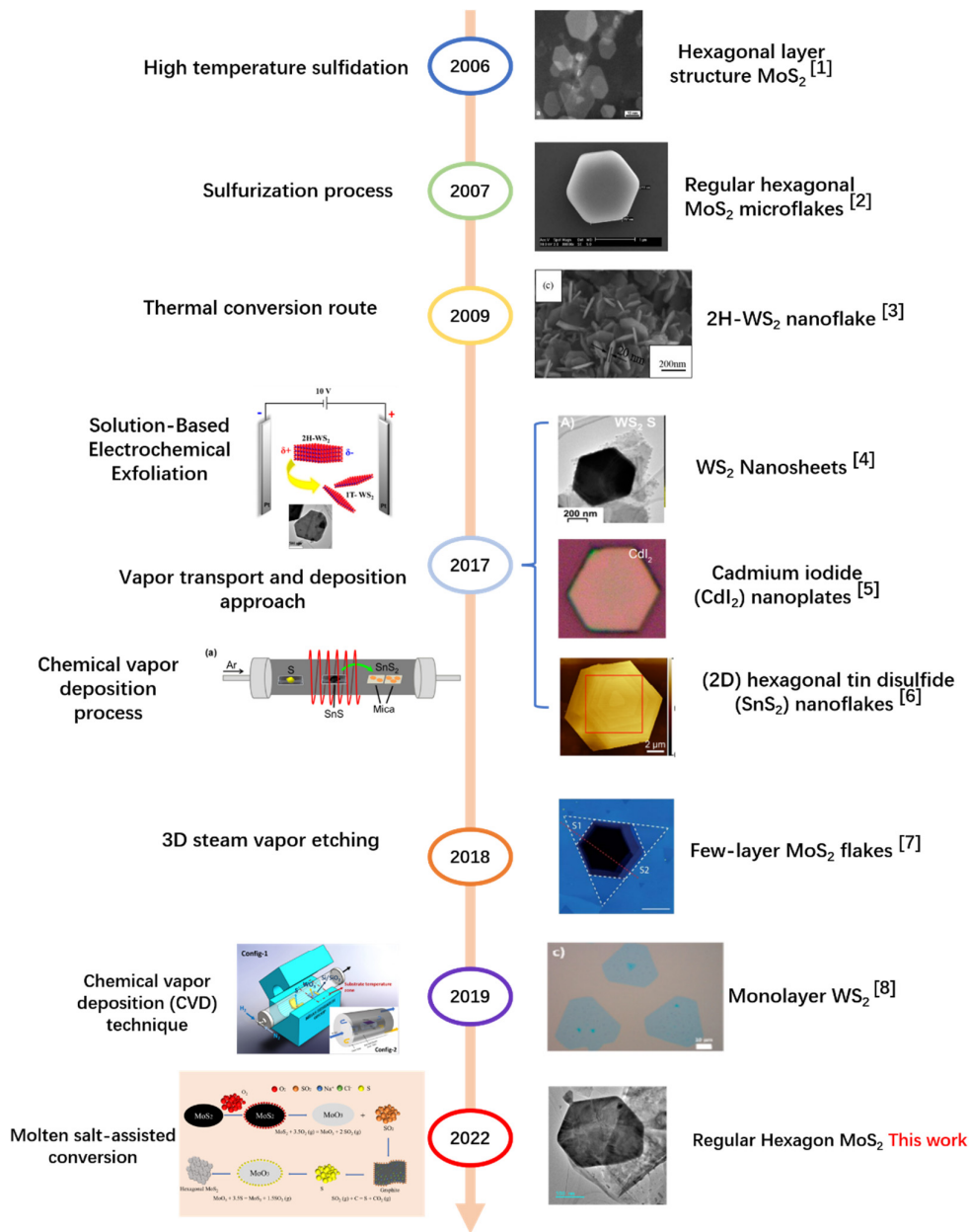


Figure S12. Summary of methods proposed for the preparation of hexagonally-shaped nanocrystals extracted from the literature.

Table S3. A comparison between precursor materials and the methods used for the preparation of hexagonal crystals extracted from the literature, with those produced in the current study.

Hexagonal crystal	Precursor materials	Preparation process
Hexagonal layer structure MoS ₂ [31]	(NH ₄) ₂ MoS ₄ , Ni(acetate) ₂	Sulfidation at 800 °C (6 h)
Hexagonal MoS ₂ microflakes [32]	Pure Mo, S	Sulfurization at 750 °C (8 h), 350-110°C (4–10 h)
2H-WS ₂ nanoflake [33]	Tungsten oxide rod, S	Thermal treatment at 750 °C (4 h)
WS ₂ nanosheets [34]	WS ₂ , Na ₂ SO ₄	Electrochemical exfoliation (48 h)
CdI ₂ nanoplates [35]	WS ₂ , SiO ₂ /Si substrate, Ar	Vapor transport/deposition at 320 °C (40 min)
Hexagonal SnS ₂ nanoflakes [36]	SnS, Ar	CVD at 200–550°C

(36 min)		
Few-layer MoS ₂ flakes [77]	Si/SiO ₂ substrate, MoS ₂ , Ar	Mechanical exfoliation and steam etching at 700 °C (30 min)
Monolayer WS ₂ [78]	WO ₃ , Si/SiO ₂ substrate, Ar	CVD at 100–180 °C
Hexagonal MoS ₂ nanocrystals [This work]	MoS ₂ , natural graphite, NaCl	Molten salt-assisted conversion at 1100°C (20 min)