

## **Supporting Information**

### **Versatile Bifunctional and Supported IrNi Oxide Catalyst for Photoelectrochemical Water Splitting**

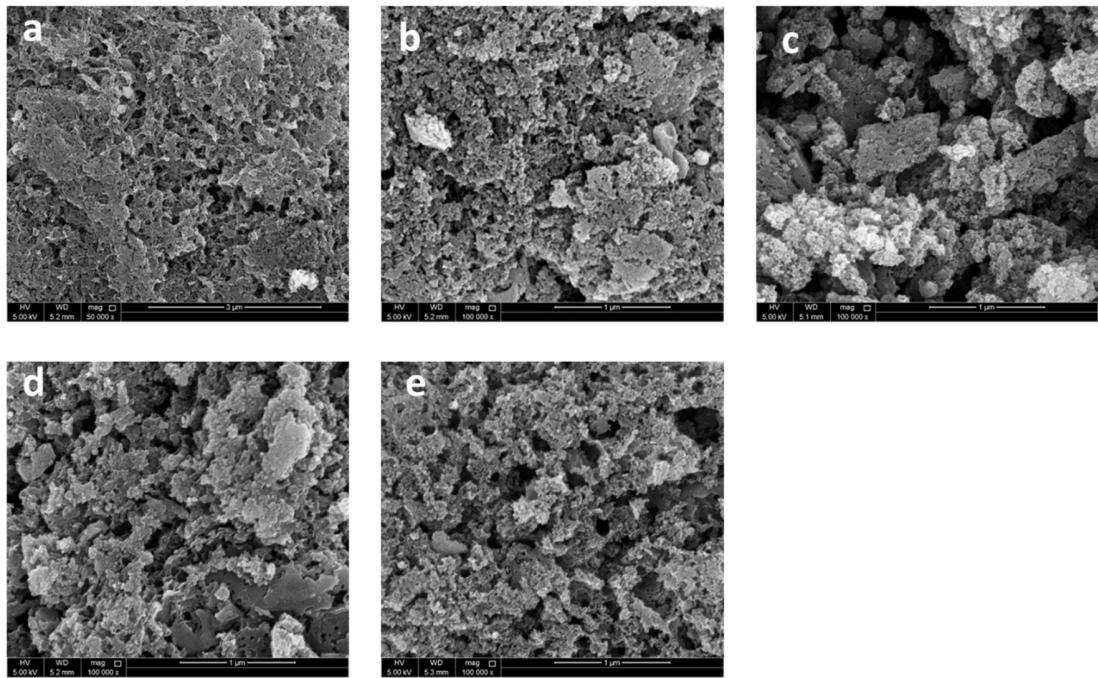
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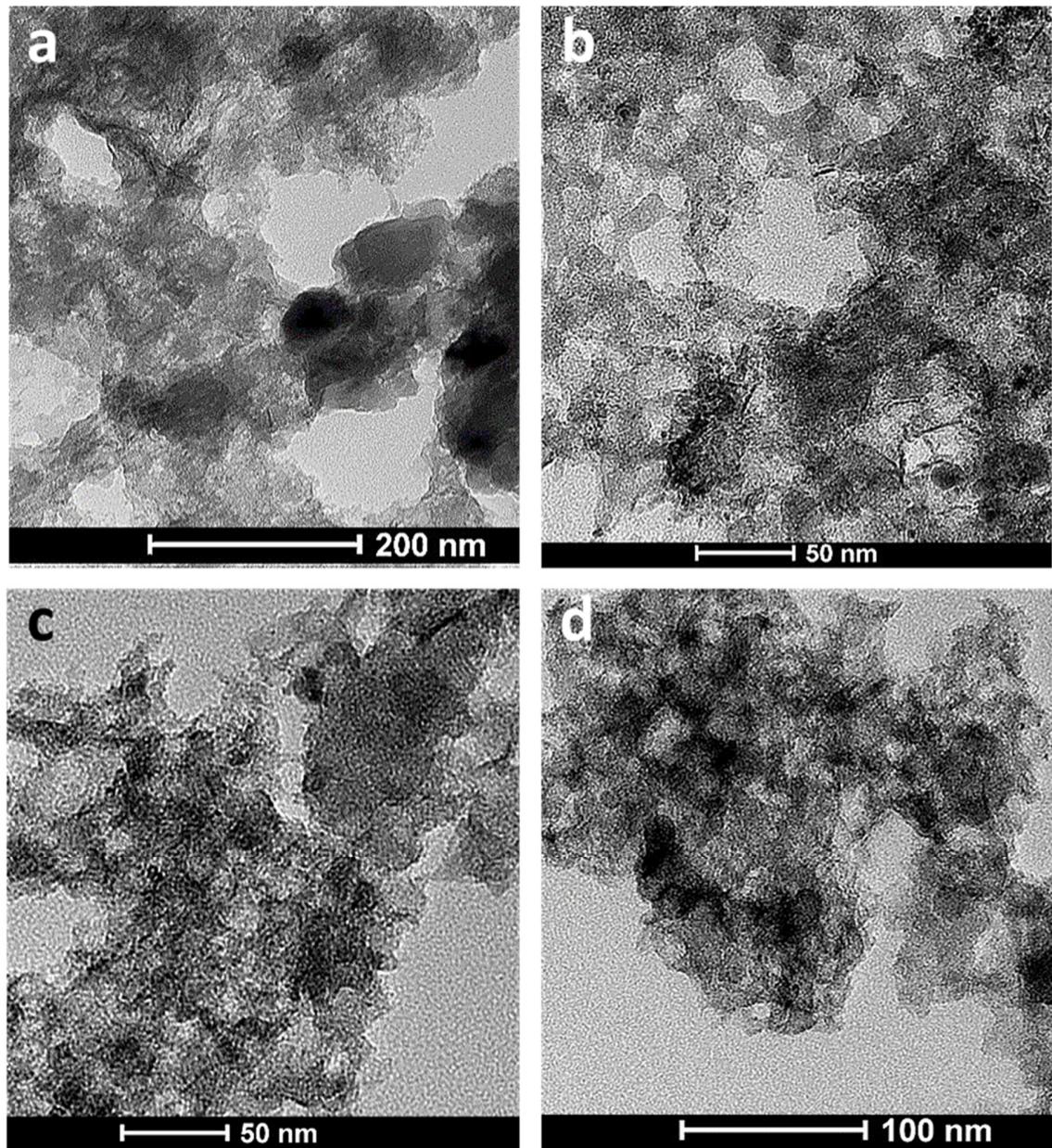
<sup>2</sup> Department of Chemical Engineering, Qatar University, Doha P.O. Box 2713, Qatar

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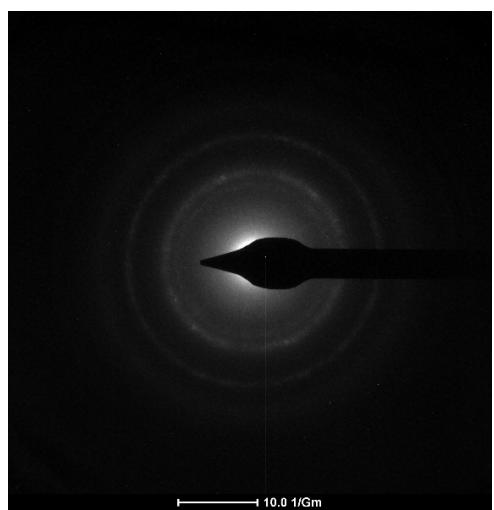
\* Correspondence: dhan@qu.edu.qa



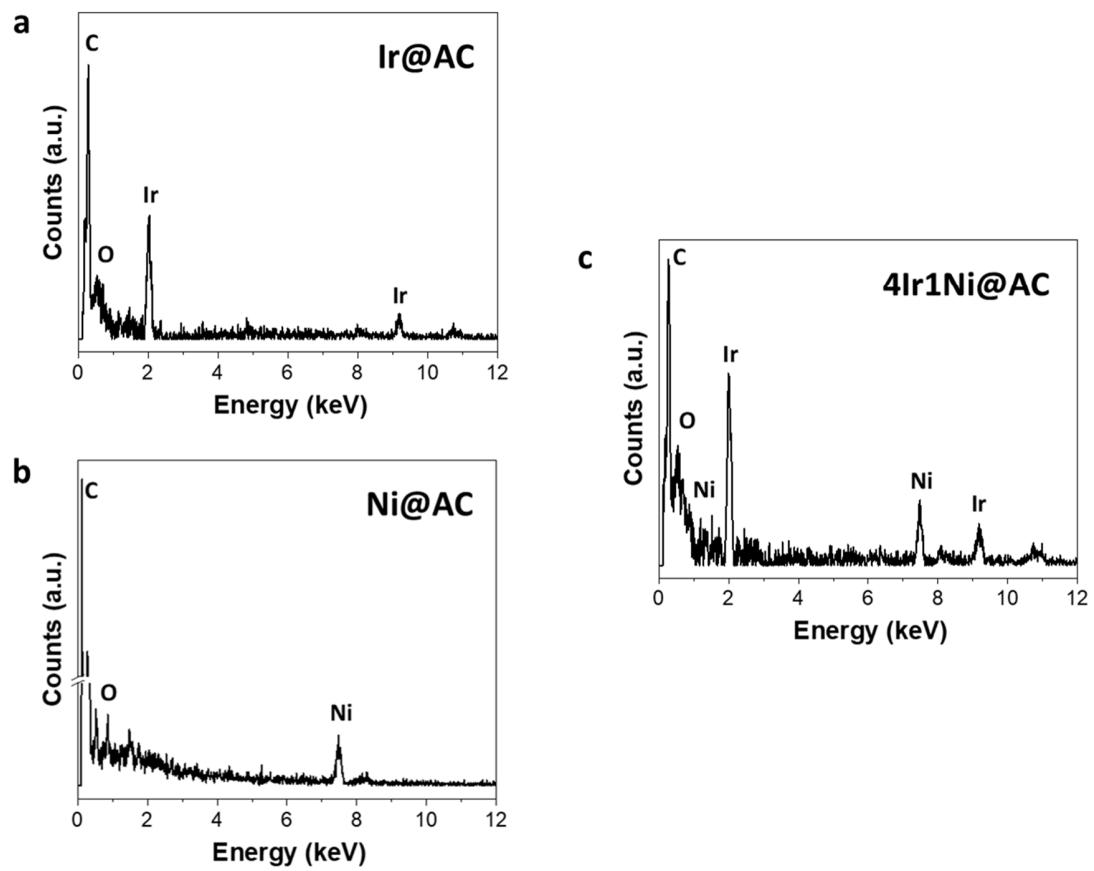
**Figure S1.** SEM images of (a) Ni@AC, (b) 1Ir4Ni@AC, (c) 1Ir1Ni@AC, (d) 4Ir1Ni@AC and (e) Ir@AC, scalar bar in the images refer to 1  $\mu\text{m}$ .



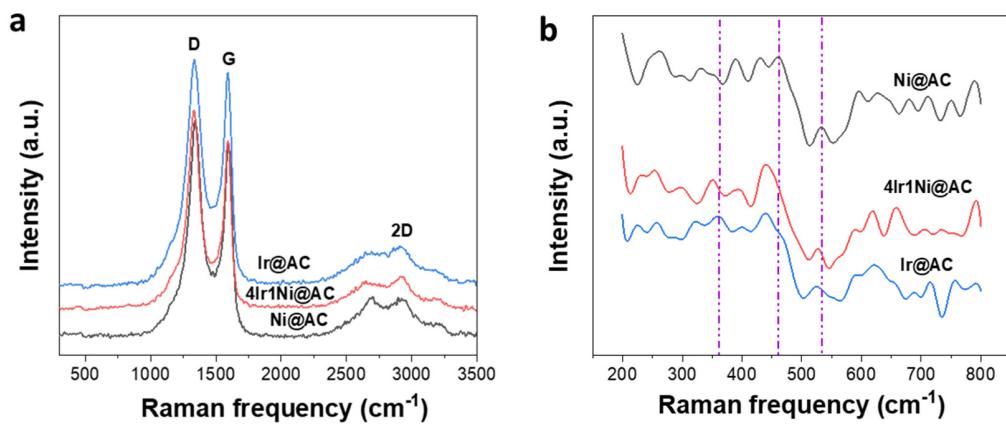
**Figure S2.** TEM images of (a) Ni@AC, (b) 1Ir4Ni@AC, (c) 1Ir1Ni@AC, and (d) Ir@AC, scalar bar is given in each image.



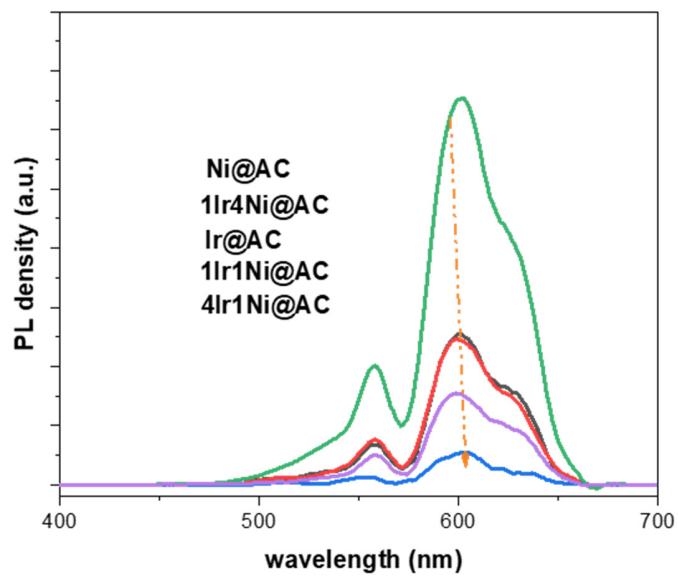
**Figure S3.** SAED ring pattern of Ni@AC



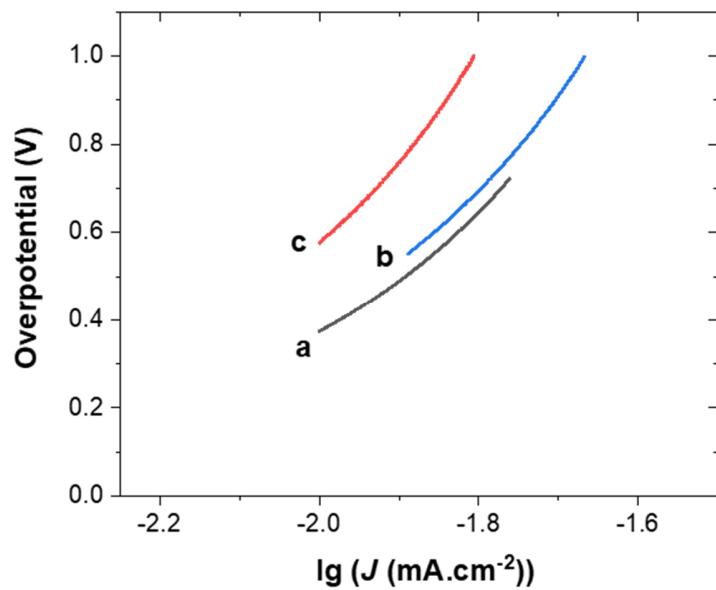
**Figure S4.** SEM-EDX spectra of (a) Ir@AC (b) Ni@AC and (c) 4Ir1Ni@AC showing the elements present in each catalyst material.



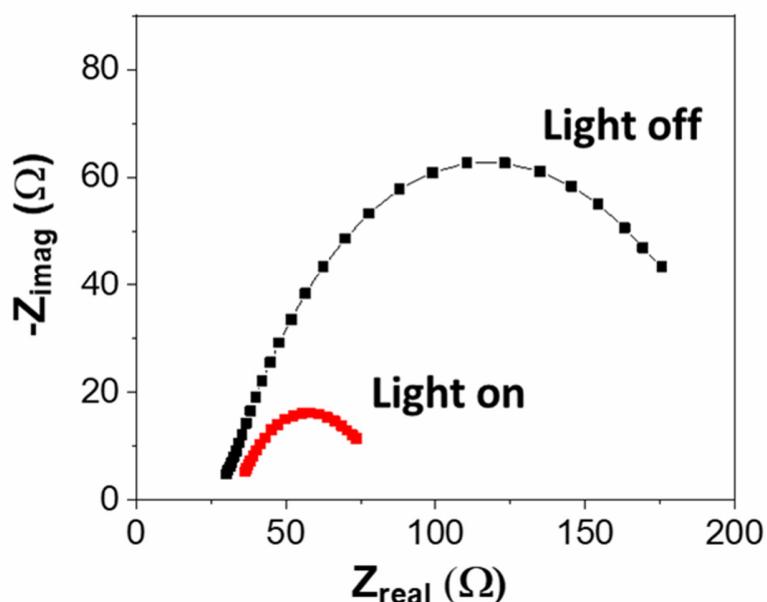
**Figure S5.** Raman spectra of Ir@AC, Ni@AC and 4Ir1Ni@AC (a) 300-3500  $\text{cm}^{-1}$  and (b) 200-800  $\text{cm}^{-1}$  frequency range



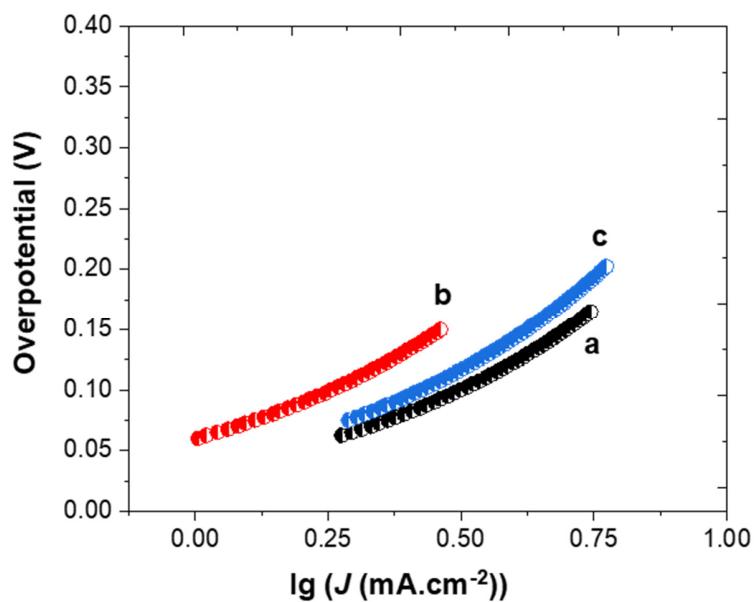
**Figure S6.** PL emission spectra of metal and alloy nanoparticles of Ir and Ni, after excitation at 350 nm.



**Figure S7.** Tafel slope for HER reaction under 1.5 G AM light illumination in a. NaOH, b. Na<sub>2</sub>SO<sub>4</sub> and c. H<sub>2</sub>SO<sub>4</sub> electrolyte solutions.



**Figure S8.** EIS spectra of 4Ir1Ni@AC electrode in 0.1 M Na<sub>2</sub>SO<sub>4</sub> electrolyte measured from 10 mHz to 100 kHz frequency range at 5 mV AC amplitude.



**Figure S9.** Tafel slope for OER reaction under 1.5 G AM light illumination in a. NaOH, b. Na<sub>2</sub>SO<sub>4</sub> and c. H<sub>2</sub>SO<sub>4</sub> electrolyte solutions.

**Table S1.** Relative mass ration between Ir and Ni

alloy code	Ir: Ni weight ratio
1ir1Ni@AC	<b>1.2±0.1:1</b>
1Ir4Ni@AC	<b>1:3.5±0.2</b>
4Ir1Ni@AC	<b>4.2±0.1:1</b>

**Table S2.** Summary of recently reported Ir based electrocatalysts for water splitting performance

Reaction	Catalyst	Electrolyte	Overpotential at 10 mA cm <sup>-2</sup> / mV	Ref
HER	Er <sub>2</sub> Si <sub>2</sub> O <sub>7</sub> :IrO <sub>2</sub>	1.0 M KOH	170	<i>ACS Catal.</i> <b>2018</b> , 8, 8830.
	Ir–Ni thin films	1.0 M KOH	60	<i>Energy Technol.</i> <b>2019</b> , 2, 442.
	Ir NPs	1.0 M KOH	109	<i>J. Am. Chem. Soc.</i> <b>2015</b> , 137, 4347.
	<a href="#"><u>4Ir1Ni@AC</u></a>	0.1 M NaOH	430	This work
OER	Ir	1.0 M KOH	430	<i>J. Am. Chem. Soc.</i> <b>2015</b> , 137, 4347.
	IrO <sub>x</sub>	1.0 M NaOH	320	<i>J. Am. Chem. Soc.</i> <b>2013</b> , 135, 16977.
	Mesoporous IrO <sub>x</sub>	0.1 M NaOH	320	<i>J. Phys. Chem. C</i> <b>2009</b> , 113, 12958.
	IrNi oxide	0.1 M HClO <sub>4</sub>	310	<i>J. Am. Chem. Soc.</i> <b>2015</b> , 137, 13031.
	<a href="#"><u>4Ir1Ni@AC</u></a>	0.1 M NaOH	250	This work