

Supporting Information

Comparing the Performance of Supported Ru Nanocatalysts Prepared by Chemical Reduction of RuCl₃ and Thermal Decomposition of Ru₃(CO)₁₂ in the Sunlight-Powered Sabatier Reaction

Daria Burova ^{1,2,3}, Jelle Rohlfs ⁴, Francesc Sastre ⁴, Pau Martínez Molina ⁴, Nicole Meulendijks ⁴, Marcel A. Verheijen ^{5,6}, An-Sofie Kelchtermans ^{1,2,3}, Ken Elen ^{1,2,3}, An Hardy ^{1,2,3}, Marlies K. Van Bael ^{1,2,3,*} and Pascal Buskens ^{1,4,*}

¹ Institute for Materials Research, Design and Synthesis of Inorganic Materials (DESINe), Hasselt University, Agoralaan Building D, B-3590 Diepenbeek, Belgium; daria.burova@uhasselt.be (D.B.); ansofie.kelchtermans@uhasselt.be (A.-S.K.); ken.elen@uhasselt.be (K.E.); an.hardy@uhasselt.be (A.H.)

² IMEC VZW, IMOMEC Associated Laboratory, Wetenschapspark 1, B-3590 Diepenbeek, Belgium

³ EnergyVille, Thor Park 8320, B-3600 Genk, Belgium

⁴ The Netherlands Organisation for Applied Scientific Research (TNO), High Tech Campus 25, 5656AE Eindhoven, The Netherlands; jelle.rohlfs@tno.nl (J.R.); francesc.sastrecalabuig@tno.nl (F.S.); martinezmolina.pau@gmail.com (P.M.M.); nicole.meulendijks@tno.nl (N.M.)

⁵ Department of Applied Physics, Eindhoven University of Technology, 5600MB Eindhoven, The Netherlands ; marcelverheijen@eurofinseag.com

⁶ Eurofins Materials Science, High Tech Campus 11, 5656AE Eindhoven, The Netherlands

* Correspondence: marlies.vanbael@uhasselt.be (M.K.V.B.); pascal.buskens@tno.nl (P.B.)

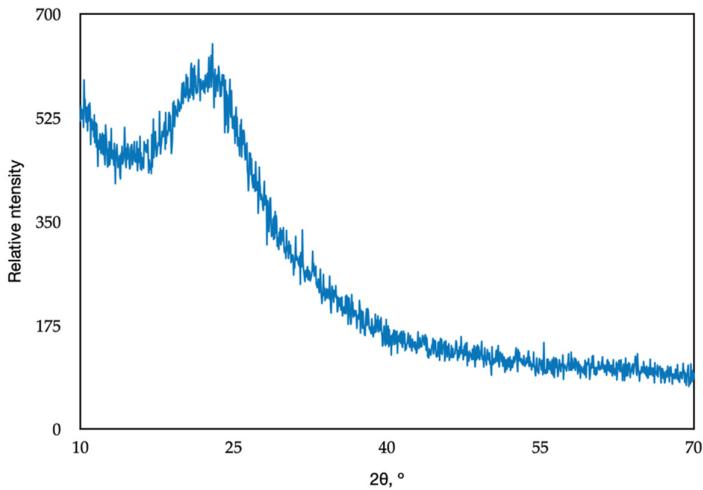


Figure S1. X-ray diffractogram of SiO_2 supported Ru catalyst prepared by reduction of RuCl_3 .

The powder X-ray diffraction pattern was recorded using a Bruker AXS D8 Discover diffractometer ($\text{Cu K}\alpha$ radiation, LynxEye detector). The broad peak centered at $2\theta = 23.8^\circ$ represents the short range order in amorphous SiO_2 . The X-ray diffraction analysis neither shows diffraction peaks of crystalline Ru nanoparticles on the SiO_2 support (expected at $2\theta \approx 37^\circ$ and $2\theta \approx 45^\circ$ [1]), nor reflections corresponding to crystalline RuO_2 [1] (expected at $2\theta \approx 28^\circ$, $2\theta \approx 35^\circ$ and $2\theta \approx 54^\circ$). Ergo, due to the low Ru loading and small particle size, Ru reflections could not be detected.

[1] JCPDS reference patterns: Ru 00-006-0663; RuO_2 00-040-1290.

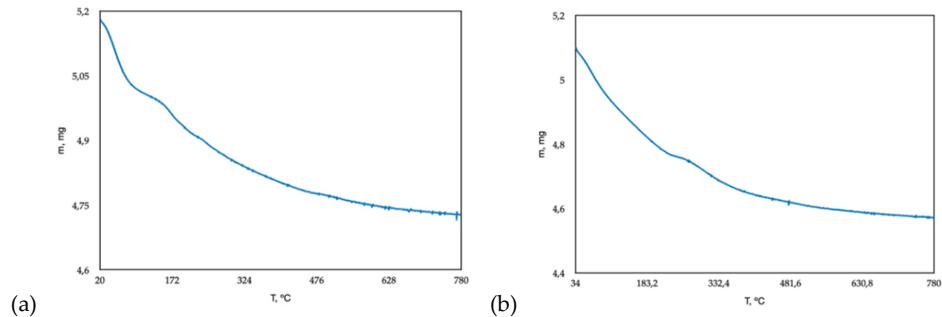


Figure S2. Thermogravimetric analyses of (a) SiO_2 -supported and (b) Al_2O_3 -supported Ru catalyst prepared by reduction of RuCl_3 , under air.

The thermal decomposition profiles of the samples were studied by thermogravimetric analysis (TGA, TA instruments Q500). The samples (~5 mg) were heated up to 800 °C at a heating rate of 10 °C/min under the air atmosphere. In both cases, the TGA analysis shows a mass loss of about 10%, which is likely due to loss of adsorbed water and progressive condensation of Si-OH and Al-OH groups.

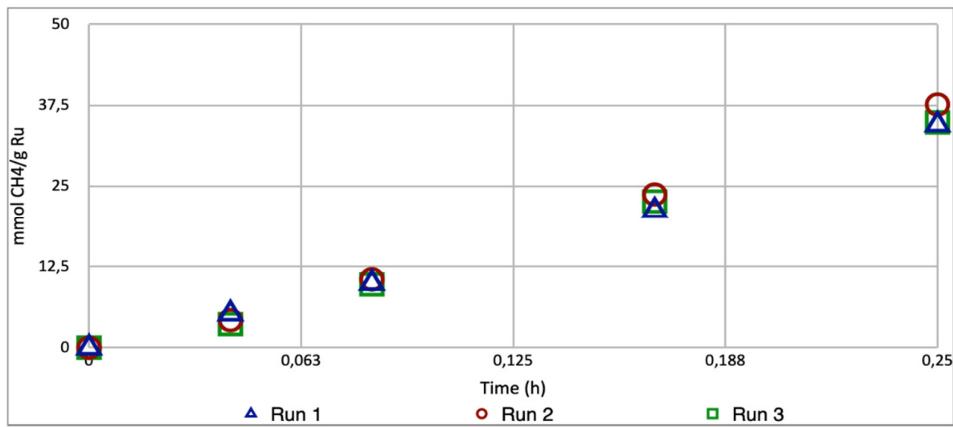


Figure S3. Conversion-time profile for three sequential runs of the sunlight-powered Sabatier reaction with Ru/SiO₂-TD. Ru loading is 3.49% w/w. Reaction conditions for all experiments: reaction mixture of H₂/CO₂/N₂ (4.5:1:1) at 3.5 ± 0.2 bar pressure, 200 mg of Ru/SiO₂-TD catalyst, light intensity of 6.66 suns, catalyst bed temperature approximately 220°C.

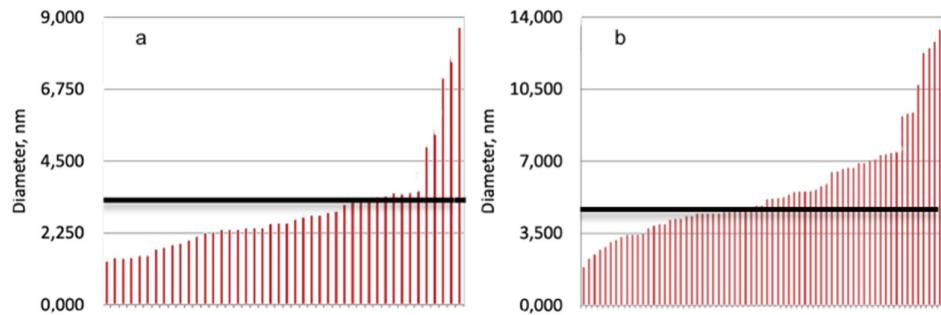


Figure S4. Comparison of mean diameter of Ru nanoparticles and numbers of agglomerates before (a) and after (b) reaction for Ru/SiO₂-CR catalysts with a Ru loading on silica of 3.34% w/w. The black line in both diagrams indicates the border between single particles and agglomerates.

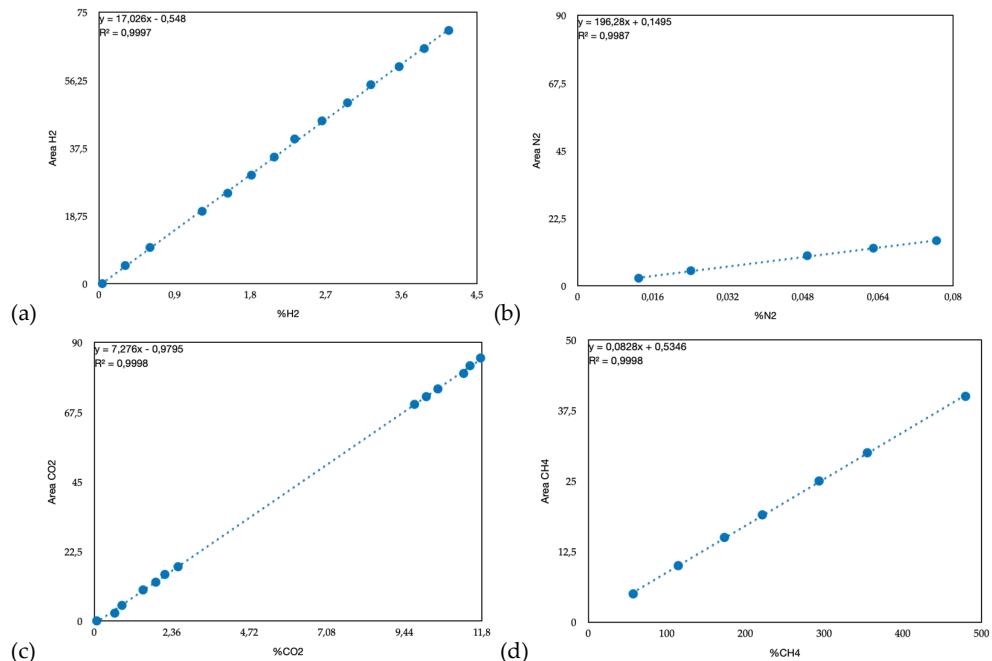


Figure S5. Calibration curves for GC detection of (a) H₂, (b) N₂, (c) CO₂ and (d) CH₄.

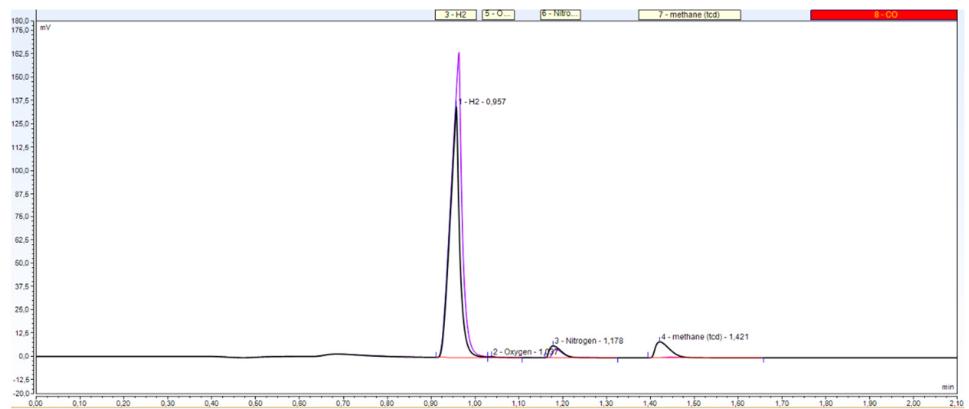


Figure S6. Prototypical gas chromatogram for catalytic conversion of CO₂ and H₂ to CH₄ in a mixture diluted with N₂.