Photoreactive Carbon Dots Modified g-C₃N₄ for Effective Photooxidation of Bisphenol-A Under Visible Light Irradiation

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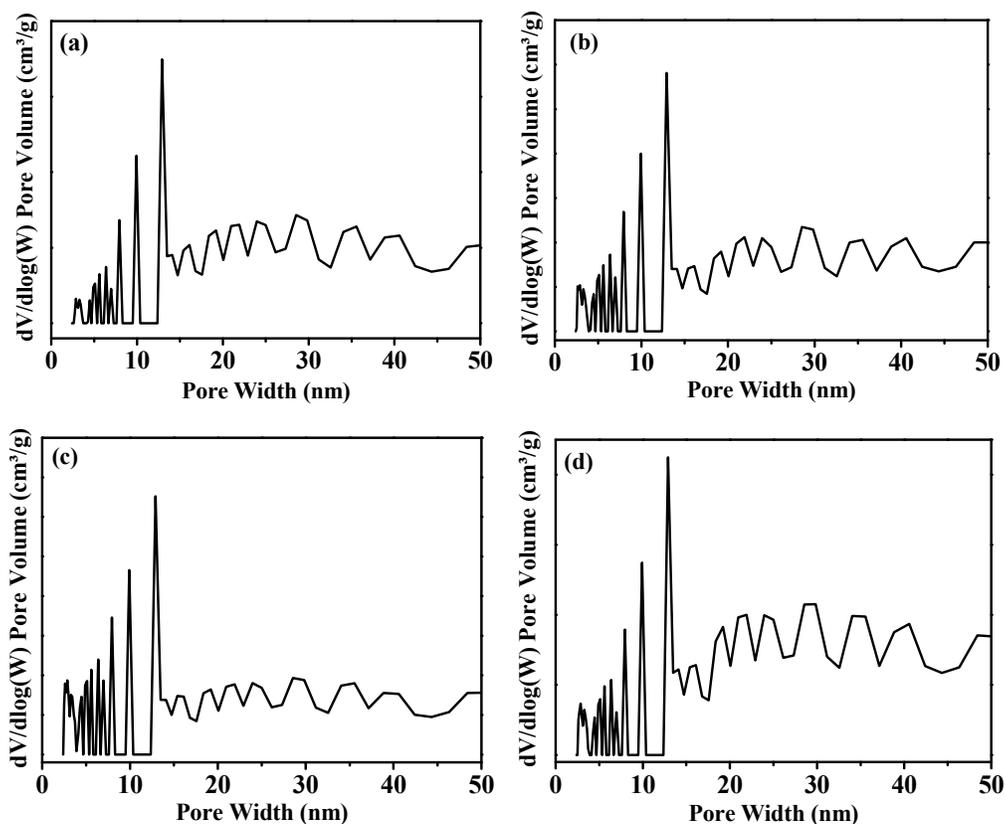
Figure S1. The DFT pore size distribution of (**a**) pure g-C₃N₄, (**b**) 0.5CDs/g-C₃N₄, (**c**) 1.0CDs/g-C₃N₄ and (**d**) 1.5CDs/g-C₃N₄.
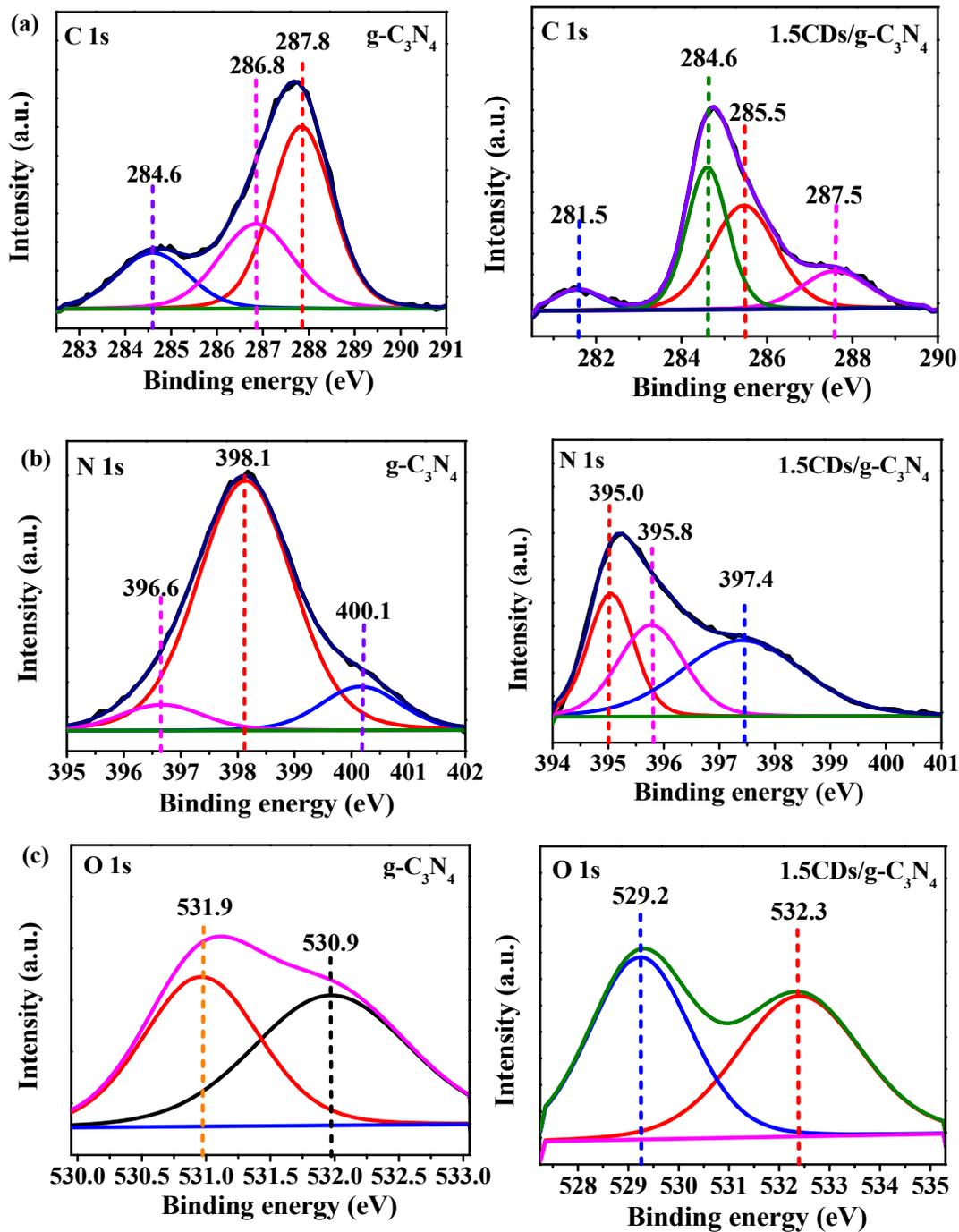
Figure S2. The deconvoluted (**a**) C 1s (**b**) N 1 s and (**c**) O 1s XPS spectra of 1.5CDs/g-C_3N_4.
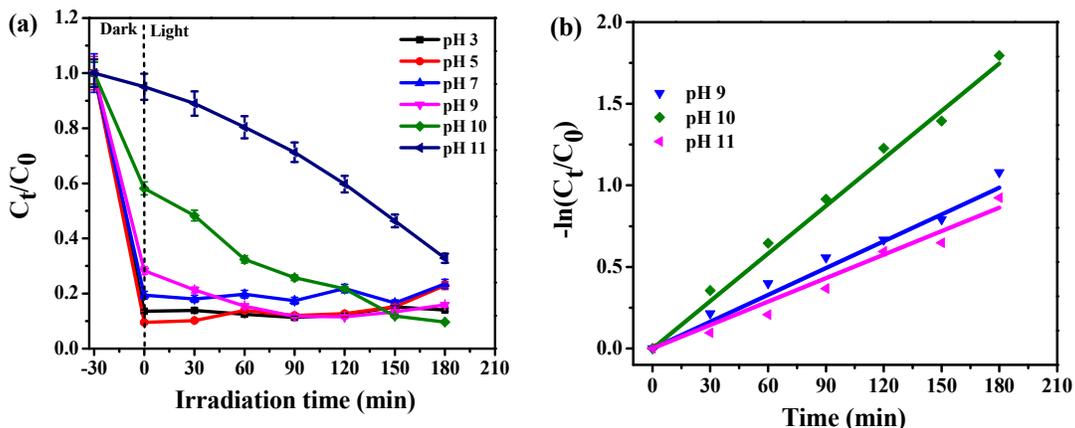
Figure S3. The (**a**) effect of different pH and (**b**) pseudo first-order kinetic fitting curves under visible light irradiation of BPA on the degradation of BPA. Reaction conditions: 1.5CDs/g-C$_3$N$_4$ dosage = 30 mg L^{-1}, BPA concentration =10 mg L^{-1} and solution volume =100 mL.
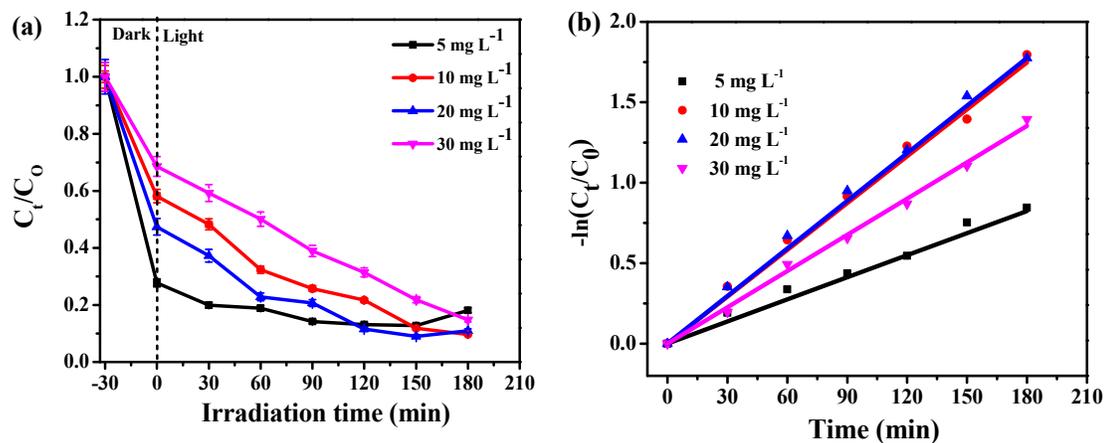

Figure S4. The (**a**) effect of initial concentration of BPA and (**b**) pseudo first-order kinetic fitting curves under visible light irradiation of BPA on the degradation of BPA. Reaction conditions: 1.5CDs/g-C$_3$N$_4$ dosage = 30 mg L^{-1}, solution volume = 100 mL and pH= 10.
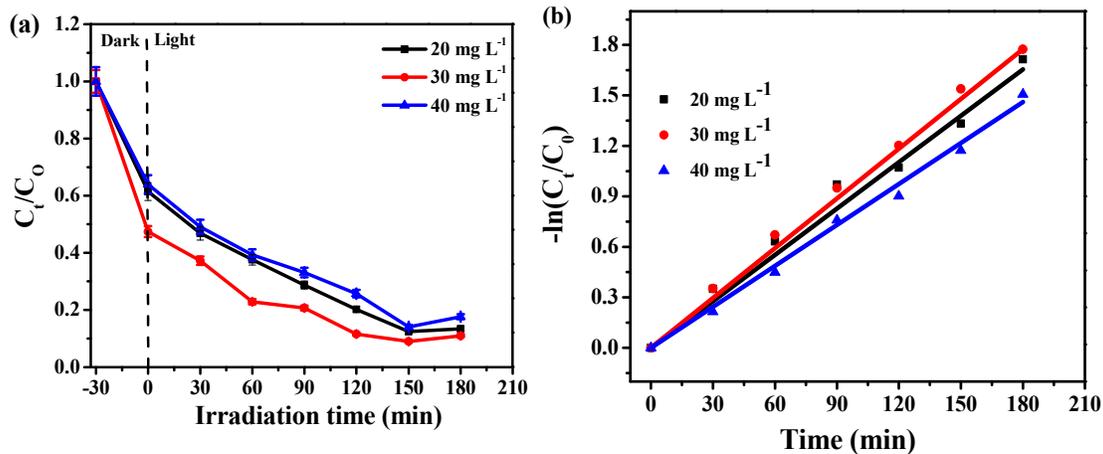
Figure S5. The (**a**) effect of catalyst dosage and (**b**) pseudo first-order kinetic fitting curves under visible light irradiation of BPA on the degradation of BPA. Reaction conditions: BPA concentration = 20 mg L^{-1}, solution volume =100 mL and pH= 10.
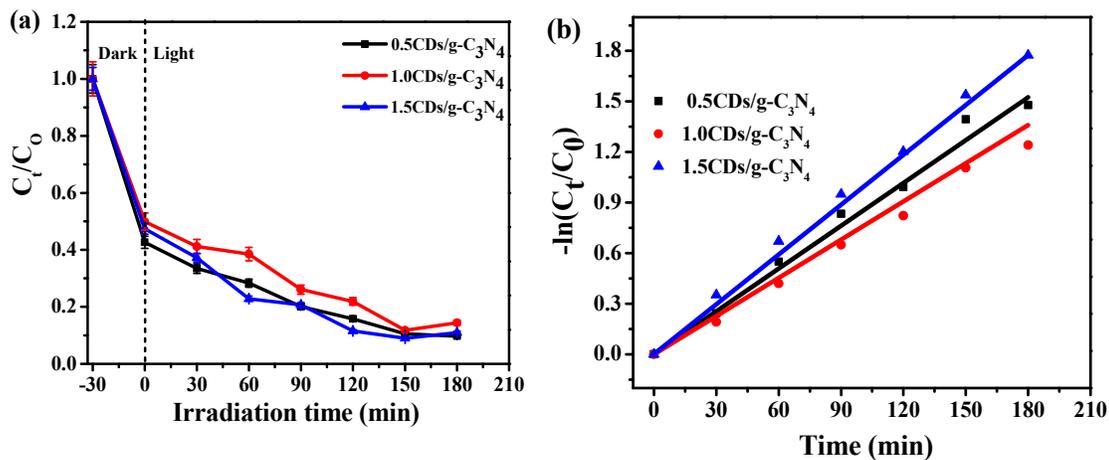
Figure S6. The (**a**) effect of CDs loading and (**b**) pseudo first-order kinetic fitting curves under visible light irradiation of BPA on the degradation of BPA. Reaction conditions: catalyst dosage = 30 mg L^{-1}, BPA concentration = 20 mg L^{-1}, solution volume =100 mL and pH= 10.
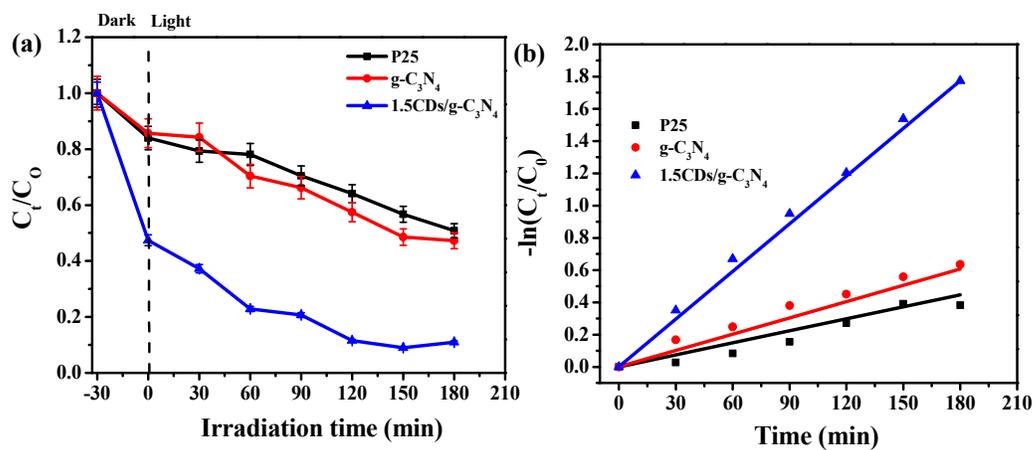
Figure S7. The (**a**) effect of different photocatalysts and (**b**) pseudo first-order kinetic fitting curves under visible light irradiation of BPA on the degradation of BPA. Reaction conditions: catalyst amounts = 30 mg L^{-1}, BPA concentration = 20 mg L^{-1}, solution volume =100 mL and pH= 10.