

Supplementary Information

Efficient cross-coupling of acetone with linear aliphatic alcohols over supported copper on a fluorite-type $\text{Pr}_2\text{Zr}_2\text{O}_7$

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Table S1. Conversion of butanol measured under different reaction conditions, where the reactant was 20 g with the butanol/acetone mole ratio of 2 and the catalyst used was 1 g.

Catalyst	Nominal Cu loading [wt%]	Temp. [°C]	Time [h]	Water content [wt%]	Number of cycle	Butanol conversion [mol%]	Associated figure
Cu/ <i>m</i> -ZrO ₂	5					56.2	Fig. 4
Cu/PrO _{1.83}	5					91.0	Fig. 4
	5		20			82.6	Fig. 4
	1					60.2	Fig. 6
	3					72.2	Fig. 6
		240	3	0	-	40.7	Fig. 8
			6			54.4	Fig. 8
			12			74.0	Fig. 8
			24			82.4	Fig. 8
				3		76.9	Fig. 9
Cu/Pr ₂ Zr ₂ O ₇	5			5		73.1	Fig. 9
				7		66.2	Fig. 9
				10		58.4	Fig. 9
		200	20		1 st	50.9	Fig. 10a
					2 nd	48.6	Fig. 10a
					3 rd	50.0	Fig. 10a
		240		0	2 nd	79.3	Fig. 10b
					3 rd	56.4	Fig. 10b
					4 th	73.4	Fig. 10b

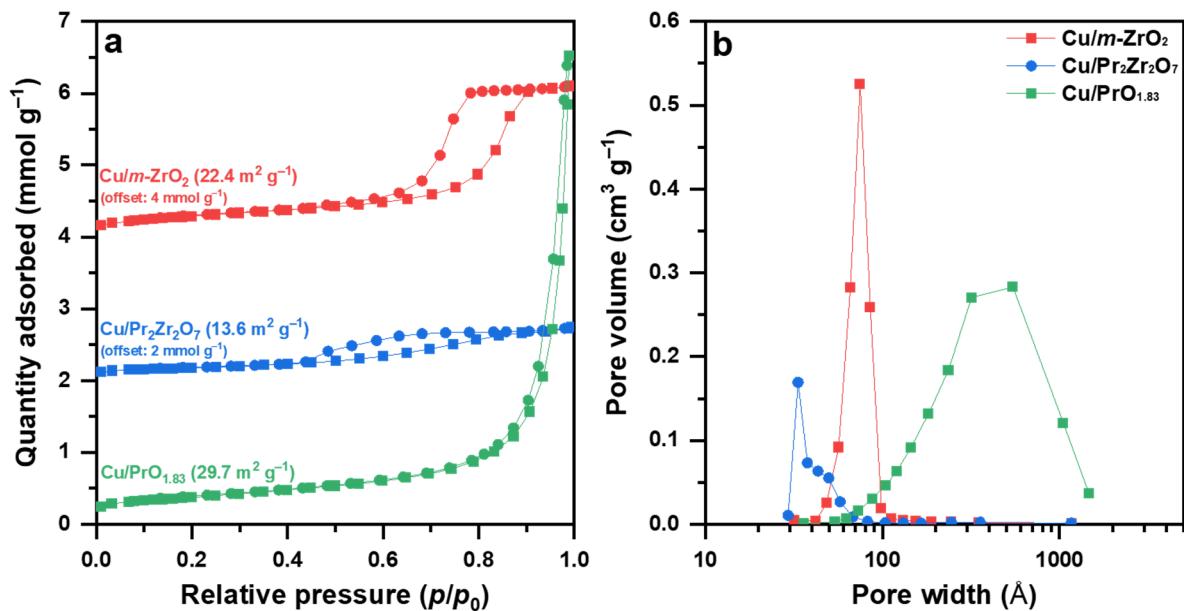


Figure S1. N_2 physisorption results of $\text{Cu}/m\text{-ZrO}_2$, $\text{Cu}/\text{PrO}_{1.83}$, and $\text{Cu}/\text{Pr}_2\text{Zr}_2\text{O}_7$ listed in Table 1: (a) BET isotherms and (b) pore size distribution curves.

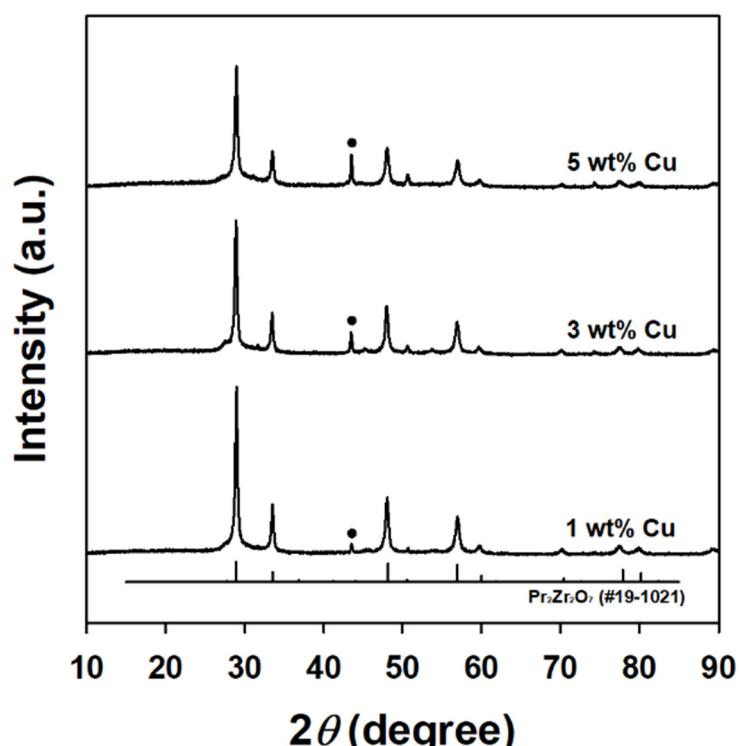
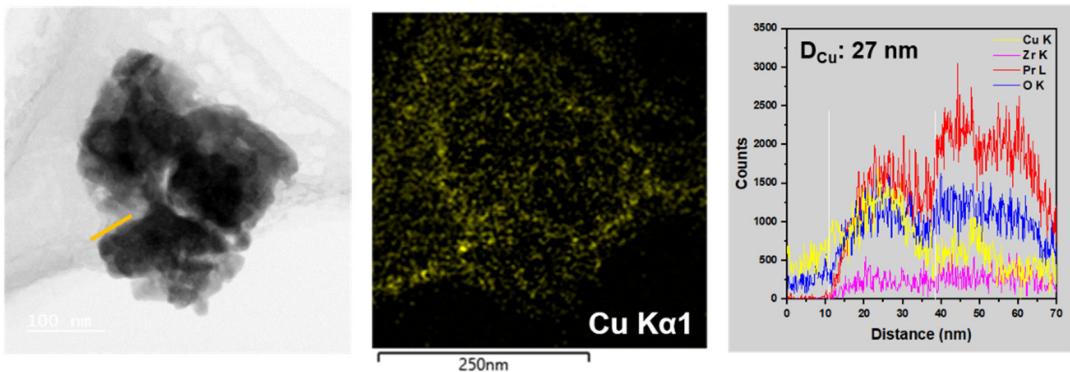
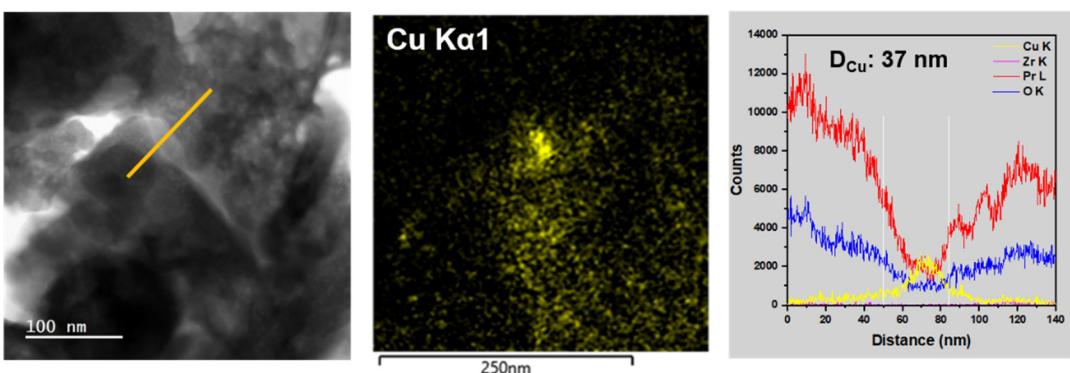


Figure S2. XRD patterns of the reduced $\text{Cu}/\text{Pr}_2\text{Zr}_2\text{O}_7$ of 1, 3, and 5 wt% Cu. The diffraction peak of Cu (111) is represented by the square symbol.

a $1\text{Cu}/\text{Pr}_2\text{Zr}_2\text{O}_7$



b $3\text{Cu}/\text{Pr}_2\text{Zr}_2\text{O}_7$



c $5\text{Cu}/\text{Pr}_2\text{Zr}_2\text{O}_7$

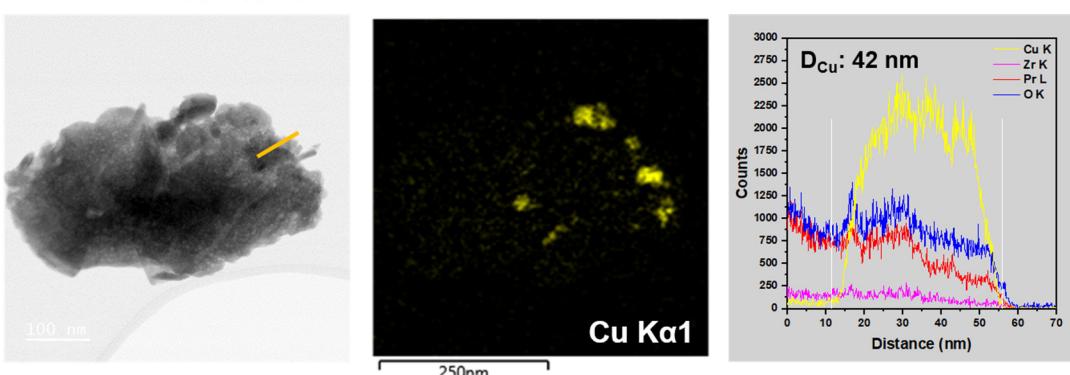


Figure S3. TEM images, Cu EDS mapping images, and line scanning results (obtained by tracking the orange line in the first image): (a) $1\text{Cu}/\text{Pr}_2\text{Zr}_2\text{O}_7$, (b) $3\text{Cu}/\text{Pr}_2\text{Zr}_2\text{O}_7$, and (c) $5\text{Cu}/\text{Pr}_2\text{Zr}_2\text{O}_7$.