

# Supplementary Information

## For

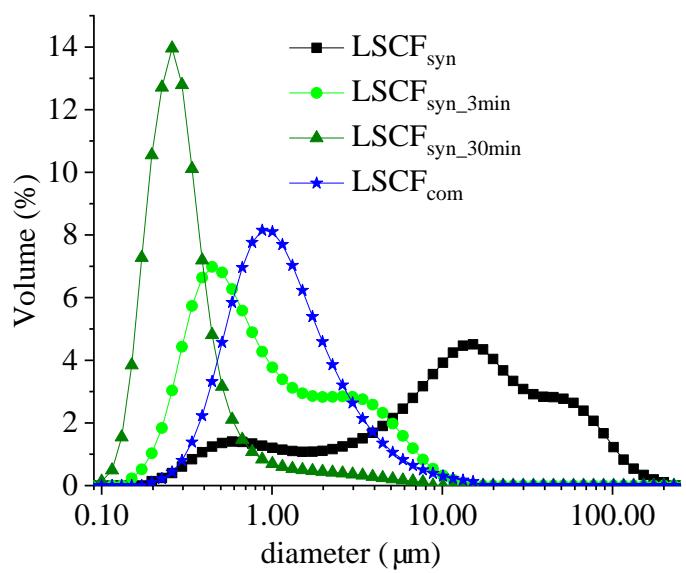
Catalytic and electrochemical properties of Ag infiltrated perovskite  
coatings for propene deep oxidation

T-G. Truong<sup>1,2</sup>, B. Rotonnelli<sup>1</sup>, M. Rieu<sup>3</sup>, Jean-Paul Viricelle<sup>3</sup>, I. Kalaitzidou<sup>2</sup>, D.  
Marinha<sup>1</sup>, L. Burel<sup>2</sup>, A. Caravaca<sup>2</sup>, P. Vernoux<sup>2</sup>, H. Kaper<sup>1</sup>

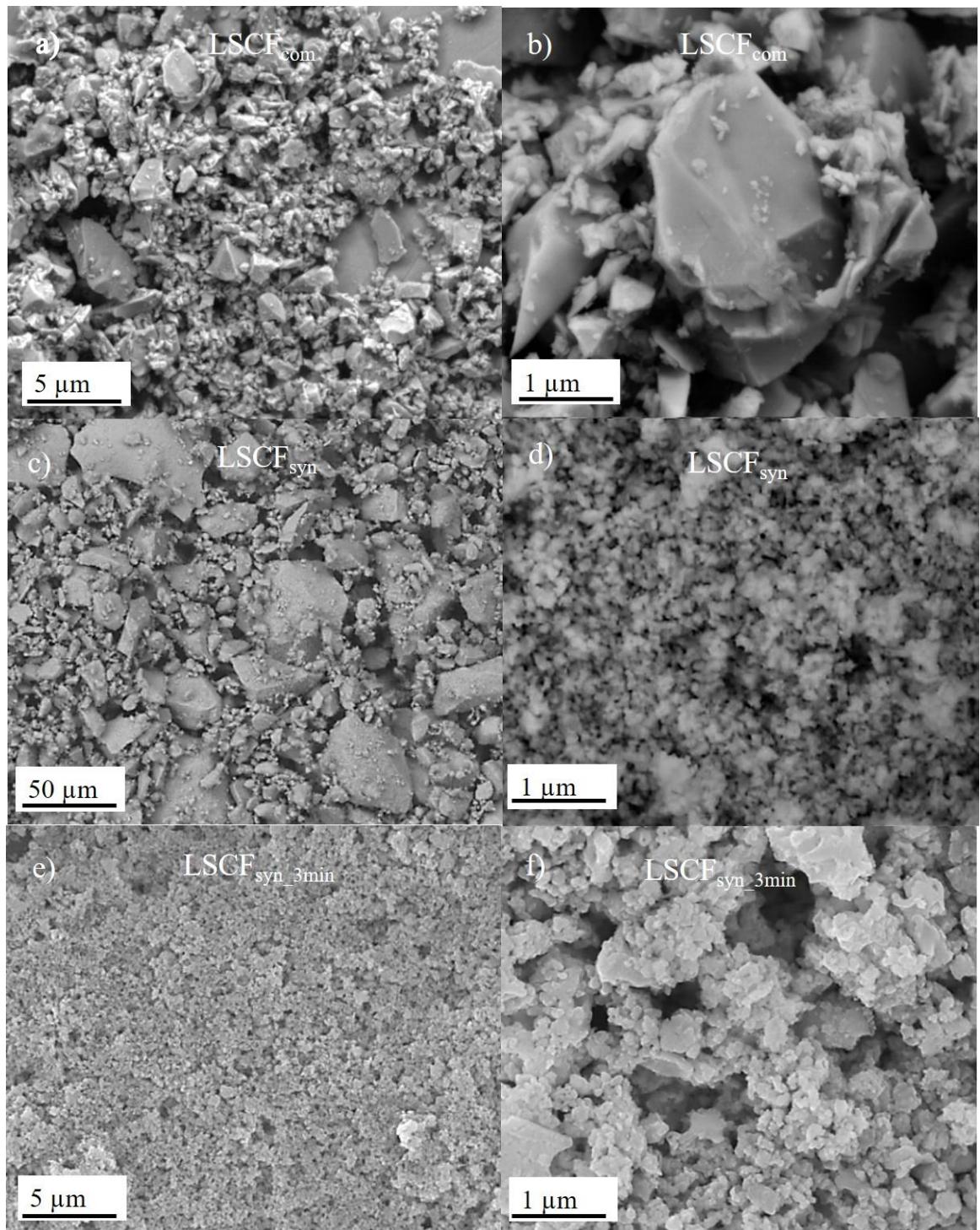
<sup>1</sup>*LSFC Laboratoire de Synthèse et Fonctionnalisation des Céramiques UMR 3080 CNRS / Saint-Gobain CREE, Saint-Gobain Research Provence, 550 avenue Alphonse Jauffret, Cavaillon, France*

<sup>2</sup>*Université de Lyon, Institut de Recherches sur la Catalyse et l'Environnement de Lyon, UMR 5256, CNRS, Université Claude Bernard Lyon 1, 2 avenue A. Einstein, F-69626 Villeurbanne, France*

<sup>3</sup>*Mines Saint-Etienne, Univ Lyon, CNRS, UMR 5307 LGF, Centre SPIN, F - 42023 Saint-Etienne France*



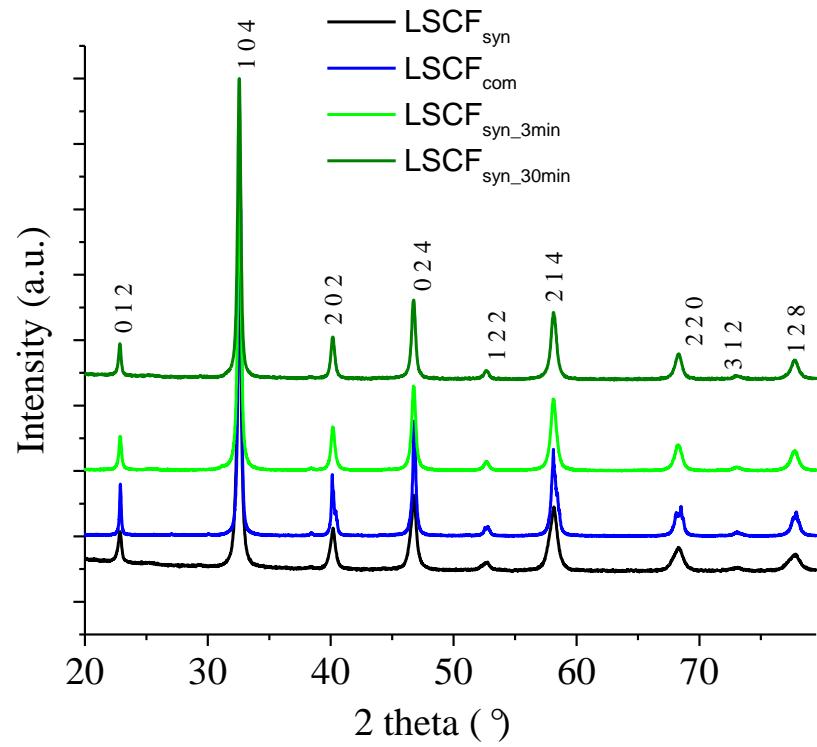
**Figure S1** Grain size distribution of  $\text{LSCF}_{\text{syn}}$ ,  $\text{LSCF}_{\text{syn\_3min}}$ ,  $\text{LSCF}_{\text{syn\_30min}}$  and  $\text{LSCF}_{\text{com}}$ .



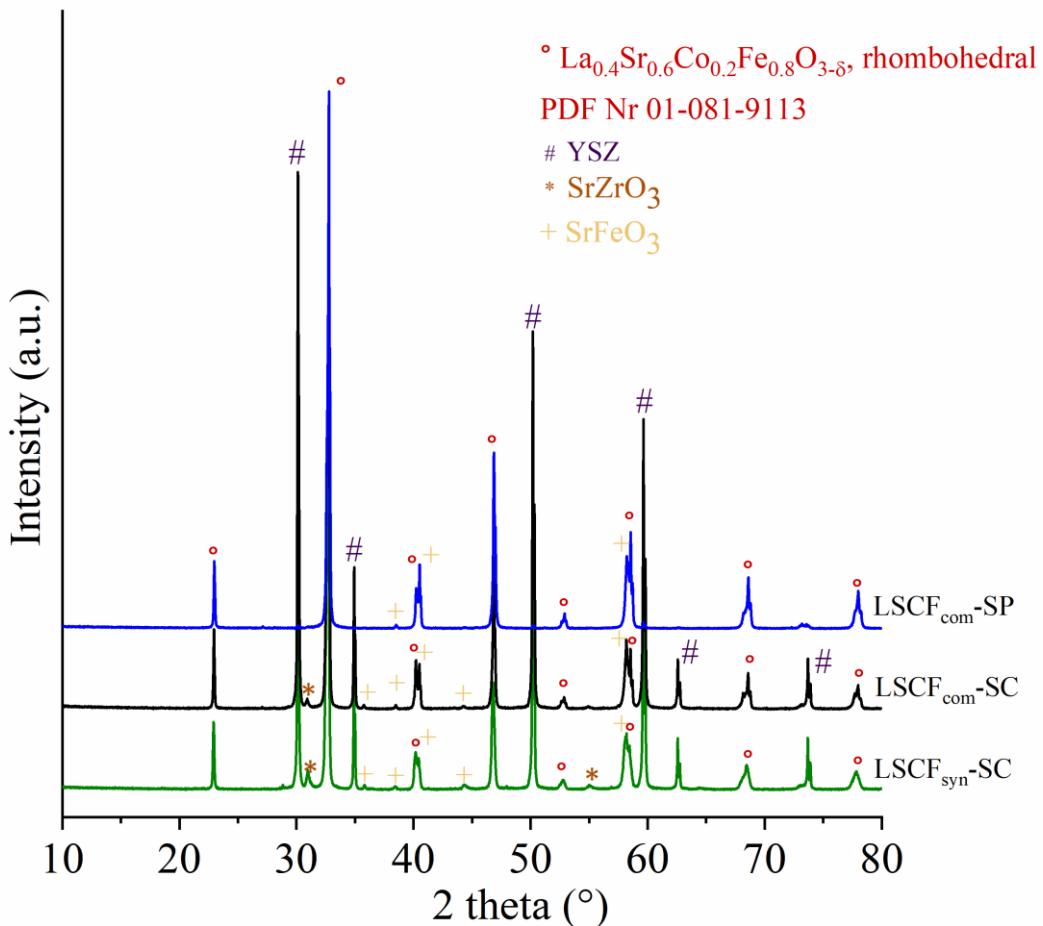
**Figure S2** SEM images of the different LSCF powders. a) and b) LSCF<sub>com</sub>, c) and d) LSCF<sub>syn</sub> and e) and f) LSCF<sub>syn\_3min</sub>.

**Table S1** XRF- and Surface area of LSCF<sub>syn</sub>, LSCF<sub>syn\_3min</sub> and LSCF<sub>com</sub>

	XRF				SSA / m <sup>2</sup> /g
	La	Sr	Co	Fe	
LSCF <sub>syn</sub>	0.6	0.4	0.19	0.8	7
LSCF <sub>syn_3min</sub>	0.6	0.4	0.19	0.8	10
LSCF <sub>com</sub>	0.6	0.4	0.2	0.8	4



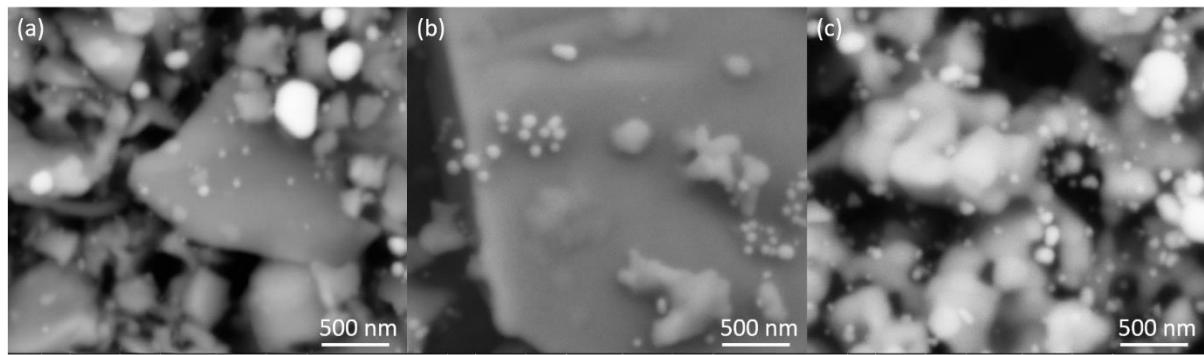
**Figure S3** XRD of  $\text{LSCF}_{\text{syn}}$ ,  $\text{LSCF}_{\text{syn\_3min}}$ ,  $\text{LSCF}_{\text{syn\_30min}}$  and  $\text{LSCF}_{\text{com}}$ .



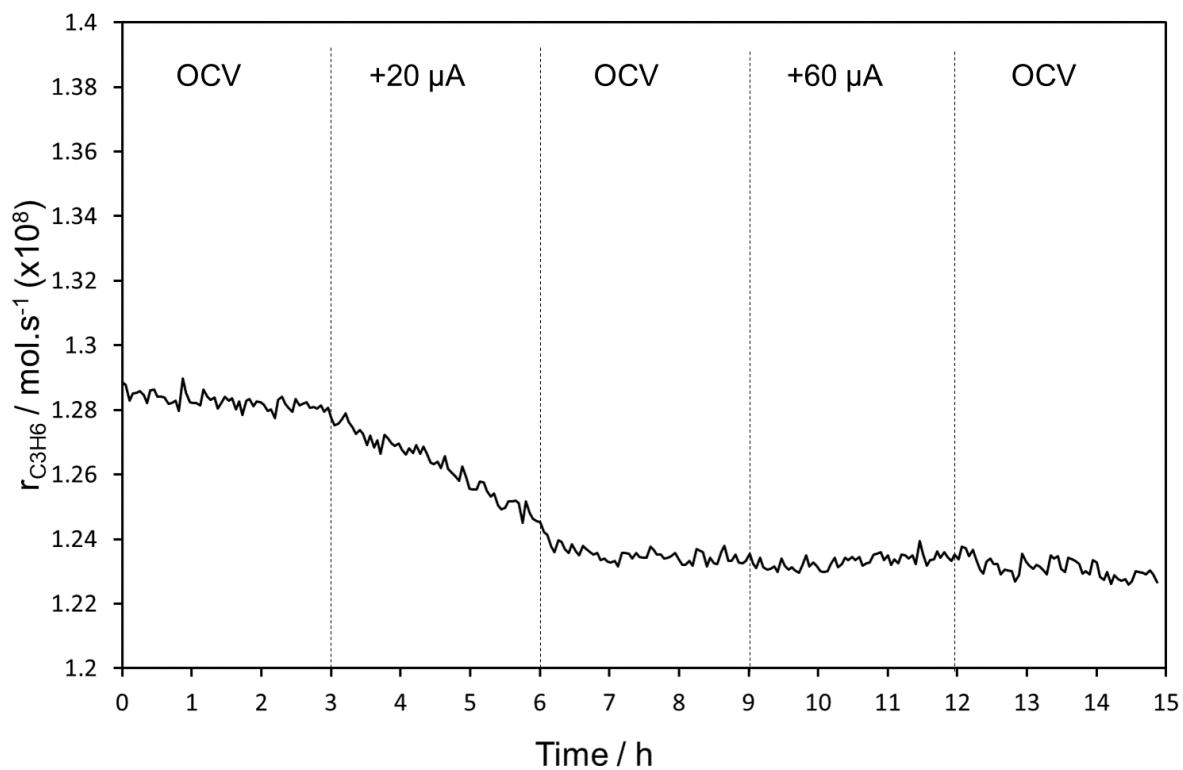
**Figure S4** XRD of thin films of LSCF deposited on YSZ and calcined at 950 °C:  $\text{LSCF}_{\text{syn}}/\text{YSZ}$ ,  $\text{LSCF}_{\text{syn\_3min}}/\text{YSZ}$ , and  $\text{LSCF}_{\text{com}}/\text{YSZ}$ .

**Table S2** Crystallite size of LSCF determined using Scherrer's equation ( $\text{LSCF}_{\text{com}}$ ,  $\text{LSCF}_{\text{syn}}$ ), LSCF deposited on YSZ and calcined at 950 °C ( $\text{LSCF}_{\text{com-SP}}$ ,  $\text{LSCF}_{\text{com-SC}}$  and  $\text{LSCF}_{\text{syn-SC}}$ ) and Ag infiltrated into the LSCF layers (Ag/ $\text{LSCF}_{\text{com-SP}}$ , Ag/ $\text{LSCF}_{\text{com-SC}}$  and Ag/ $\text{LSCF}_{\text{syn-SC}}$ ). For LSCF, the peak at 22.9 ° was used and for Ag the peak at 38.2 °.

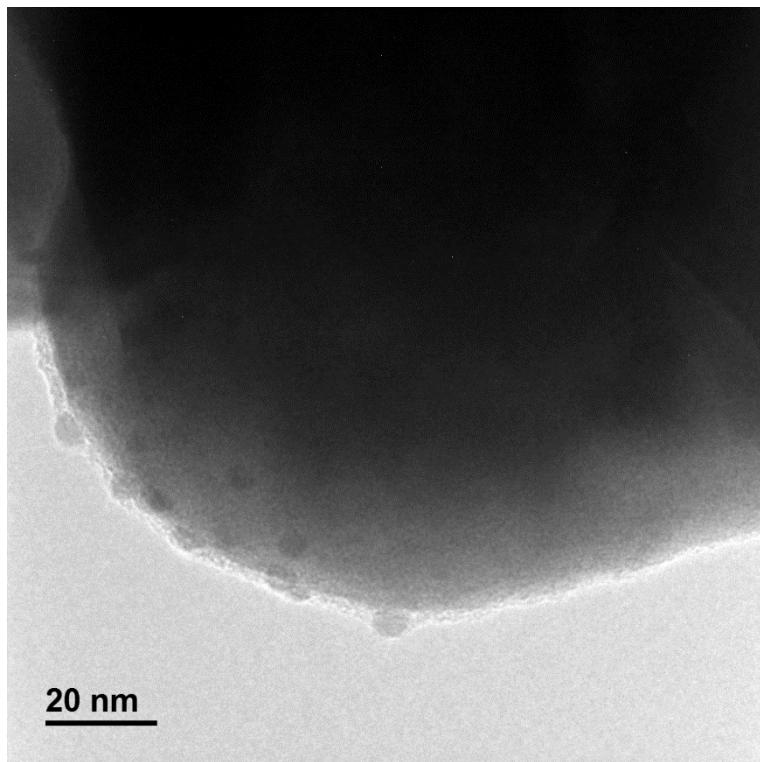
	crystallite size / nm	crystallite size of Ag/nm
$\text{LSCF}_{\text{com}}$	56	
$\text{LSCF}_{\text{syn}}$	26	
$\text{LSCF}_{\text{syn\_3min}}$	34	
$\text{LSCF}_{\text{com-SP}}$	76	
$\text{LSCF}_{\text{com-SC}}$	85	
$\text{LSCF}_{\text{syn-SC}}$	68	
Ag/ $\text{LSCF}_{\text{com-SP}}$	61	50
Ag/ $\text{LSCF}_{\text{com-SC}}$	75	50
Ag/ $\text{LSCF}_{\text{syn-SC}}$	80	46



**Figure S5** SEM images of the surface of Ag-infiltrated LSCF films after reduction at 300 °C for 2 h in 5% H<sub>2</sub>/Ar. Images were taken at the center of each film. a) Ag/LSCF<sub>com</sub>-SP, b) Ag/LSCF<sub>com</sub>-SC and c) Ag/LSCF<sub>syn</sub>-SC. Ag particles can be identified as brighter spots on the images.



**Figure S6** Impact of positive currents on the catalytic rate at 300 °C of Ag/LSCF<sub>com</sub>-SP.



**Figure S7** TEM analysis of Ag/LSCF<sub>com</sub>-SP film after catalytic testing, taken at the centre of the LSCF layers.