

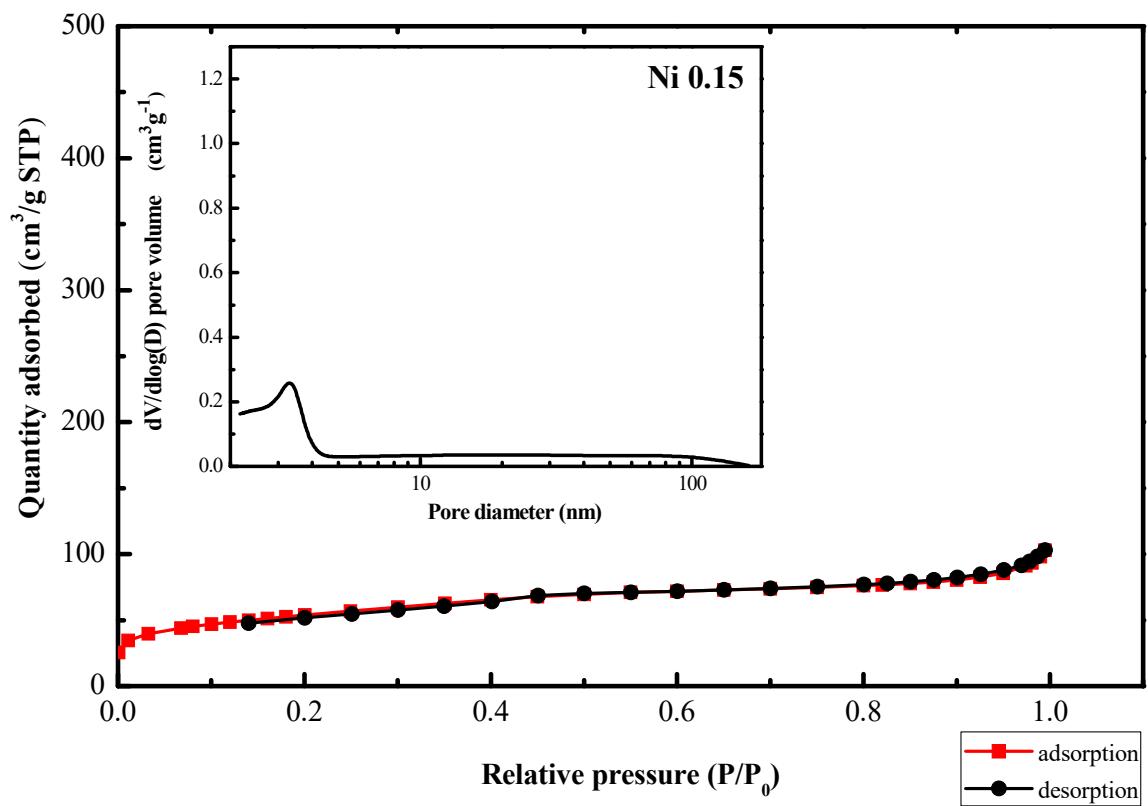
## List of Supplementary data

**Supplemental figure S1.** Isotherm and pore size distribution (inset) of Ni-Ce-ZrO<sub>2</sub> catalysts

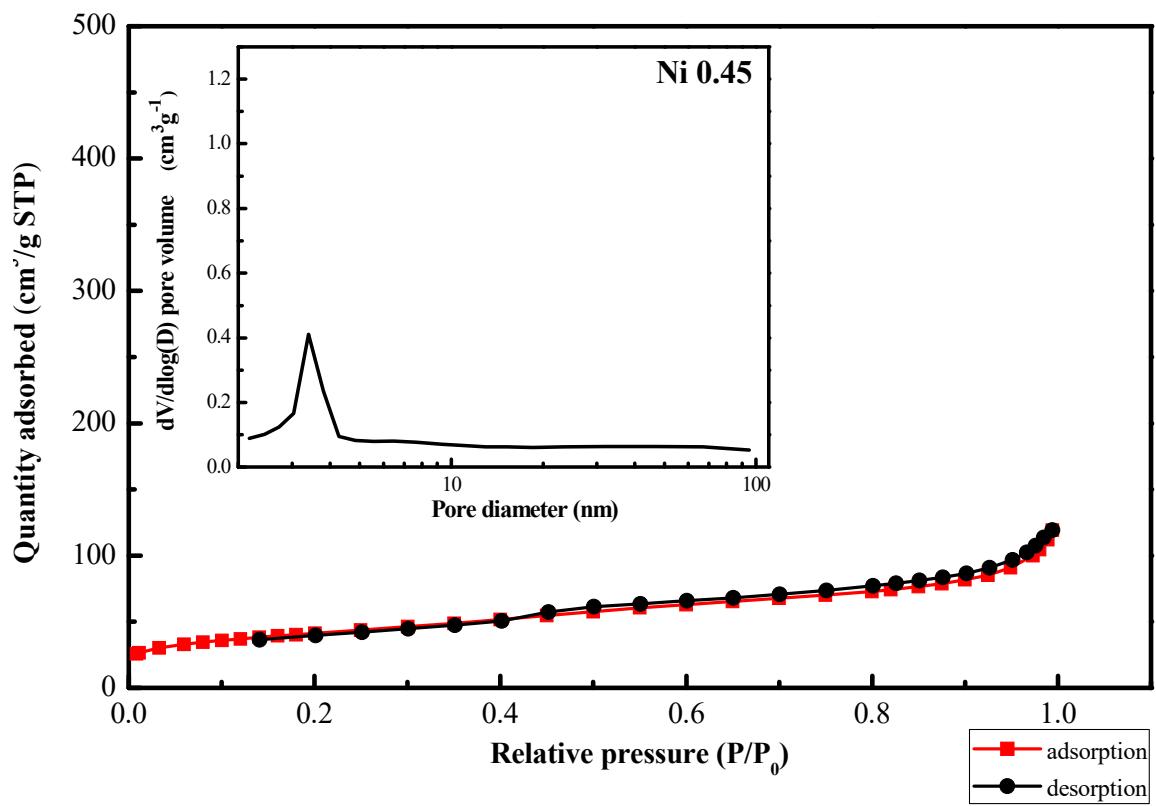
**Supplemental table S2.** XPS core level electron binding energy of Ni-Ce-ZrO<sub>8</sub> catalysts for different Ni content

**Supplemental figure S3.** The stability of Ni6.0 catalyst at 300 °C for 48 h

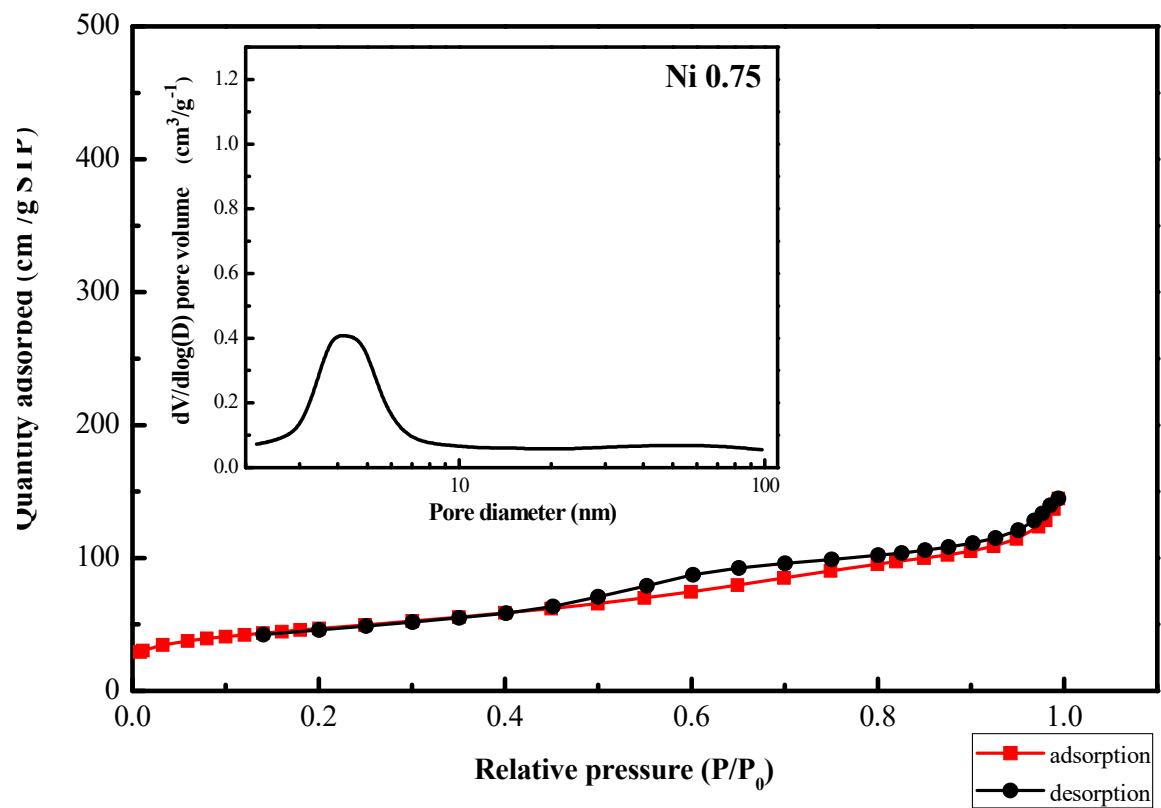
**Supplemental figure S4.** TEM images of (a) fresh and (b) spent Ni6.0 catalysts. Reaction conditions: GHSV = 10,000 h<sup>-1</sup>, H<sub>2</sub>/CO<sub>2</sub> = 4, reaction temperature = 300 °C and reaction time 30 min.



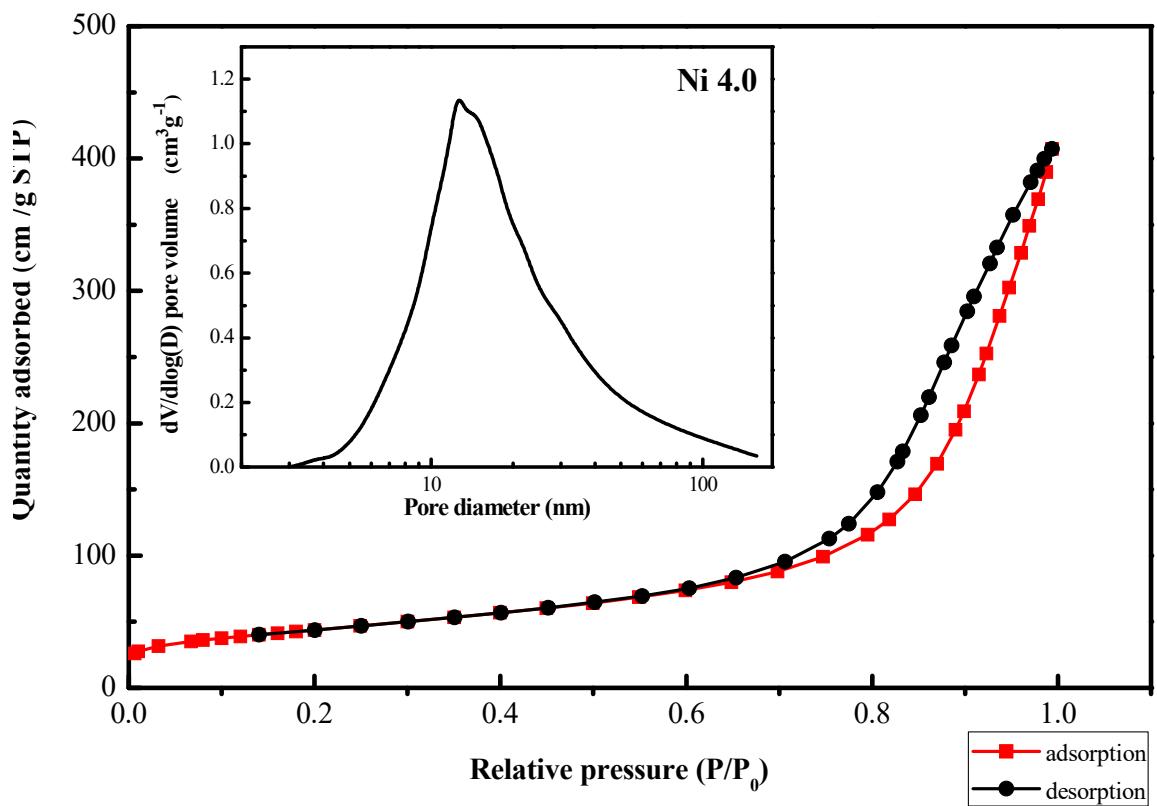
**Supplemental figure S1.** Isotherm and pore size distribution (inset) of Ni-Ce-ZrO<sub>2</sub> catalysts



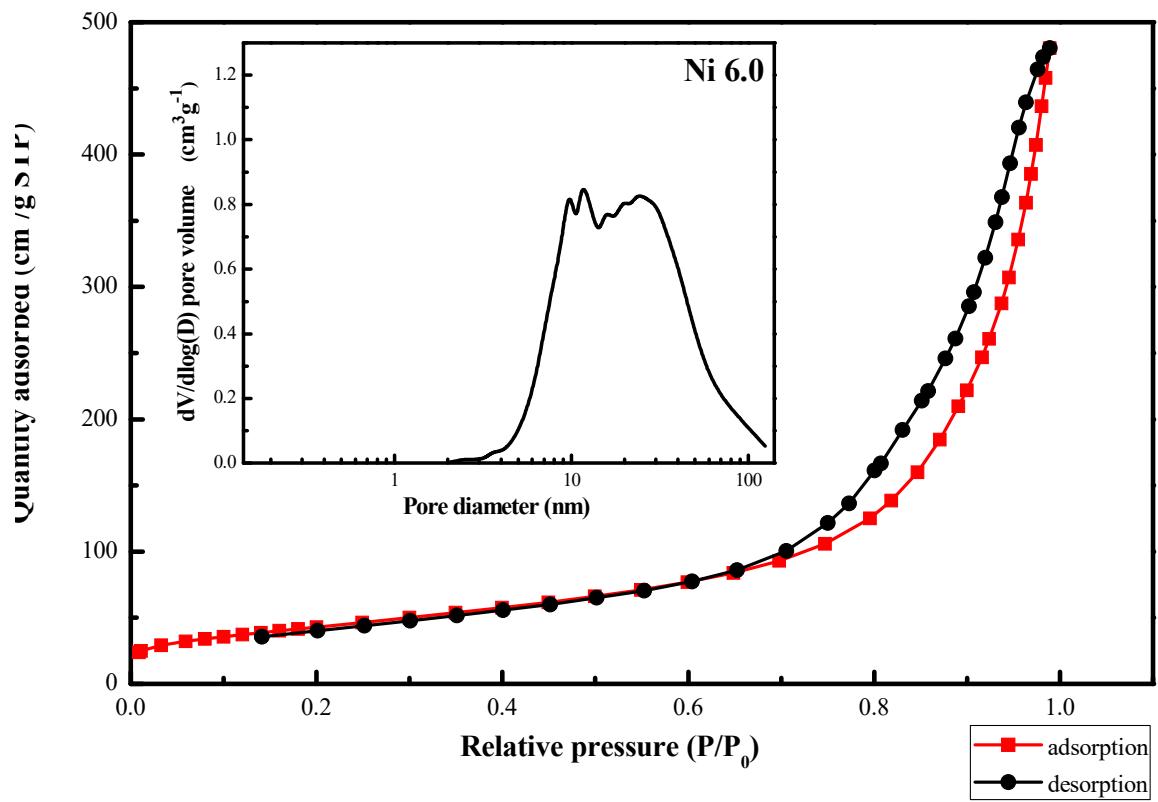
**Supplemental figure S1 (cont.).** Isotherm and pore size distribution (inset) of Ni-Ce-ZrO<sub>2</sub> catalysts



**Supplemental figure S1 (cont.).** Isotherm and pore size distribution (inset) of Ni-Ce-ZrO<sub>2</sub> catalysts



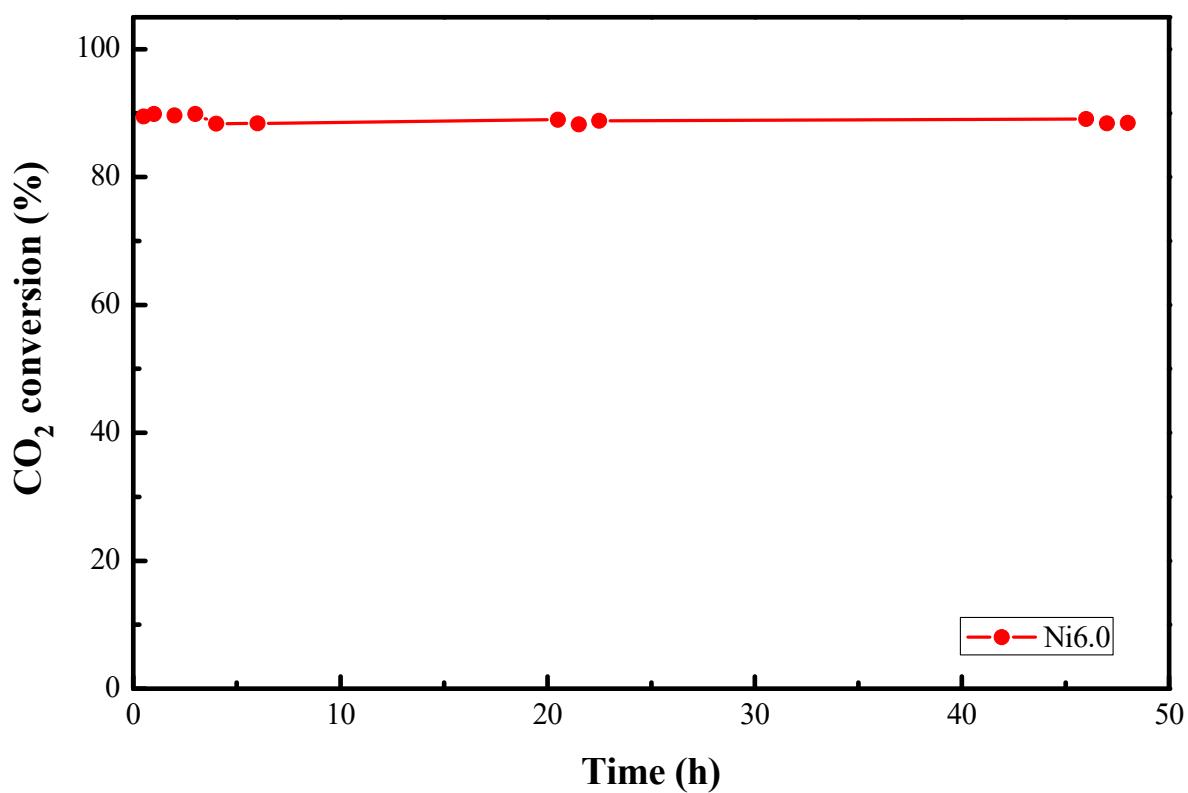
**Supplemental figure S1 (cont.).** Isotherm and pore size distribution (inset) of Ni-Ce-ZrO<sub>2</sub> catalysts



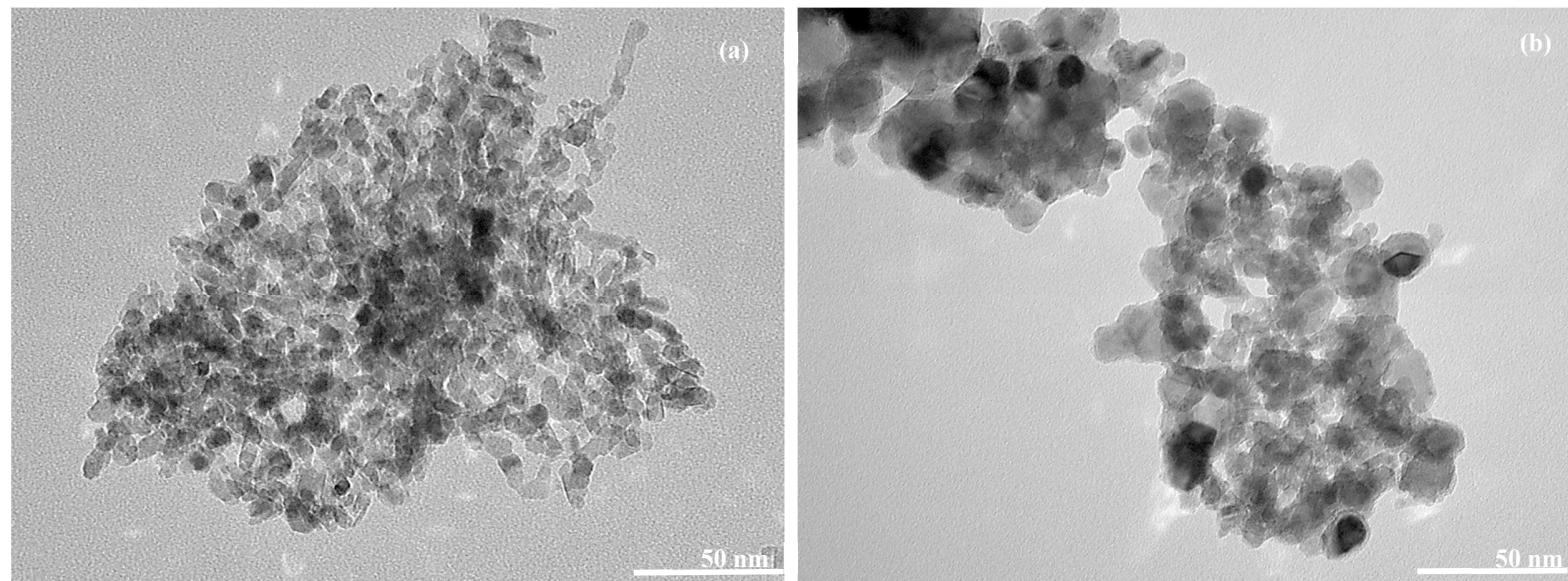
**Supplemental figure S1 (cont.).** Isotherm and pore size distribution (inset) of Ni-Ce-ZrO<sub>2</sub> catalysts

**Supplemental table S2.** XPS core level electron binding energy of Ni-Ce-ZrO<sub>δ</sub> catalysts for different Ni content

Catalysts	Binding energy (eV)						NiO/Ni(OH) <sub>2</sub> ratio	Ce <sup>4+</sup> /Ce <sup>3+</sup> ratio
	Zr3d <sub>5/2</sub>	Zr3d <sub>3/2</sub>	O1s	Ni2p <sub>1/2</sub>	Ce3d <sub>5/2</sub>	Ce3d <sub>3/2</sub>		
Ni0.15	181.74	184.06	528, 530.61	855	882, 888, 898	901, 906, 917	1.03	4.90
Ni0.75	182.09	184.38	529.13, 530.83	855, 856.5	882, 888, 898	901, 906, 917	1.15	2.50
Ni6.0	181.46	183.75	528.85, 531.10	853, 855, 860.5	880, 898	901, 917	1.41	2.40



**Supplemental figure S3.** The stability of Ni6.0 catalyst at 300 °C for 48 h



**Supplemental figure S4.** TEM images of (a) fresh and (b) spent Ni6.0 catalysts. Reaction conditions: GHSV = 10,000 h<sup>-1</sup>, H<sub>2</sub>/CO<sub>2</sub> = 4, reaction temperature = 300 °C and reaction time 30 min.