

Supplementary Materials

Colorimetric Detection of Chromium(VI) Ions in Water Using Unfolded-Fullerene Carbon Nanoparticles

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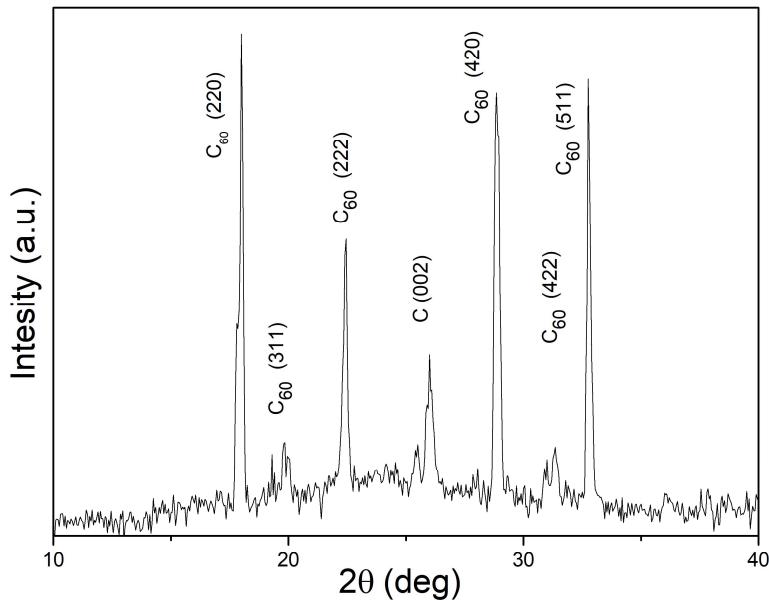


Figure S1. X-ray powder diffraction pattern of synthesized N-CDs.

Table S1. Main IR peaks of the pristine fullerene-C₆₀, mixture water-THF (W-THF) N-CDs-W, N-CDs-W-THF and corresponding assignments; s=strong, m=medium, w=weak, br=broad, sp=sharp, sh=shoulder.

Fullerene	W-THF	N-CDs-W	N-CDs-W-THF	Assignments
	3430 m br	3390 m br	-O-H stretching	
	3240 m br	3205m br	-N-H asymmetric stretching	
3360 br			-O-H stretching	
	3192 m sh		-N-H symmetric stretching	
	3047 m br	3053 m br	-CH aromatic stretching	
		2953 m br	-CH _x stretching	
2974 w			-CH ₂ asymmetric stretching	
2877 w			-CH ₂ symmetric stretching	
	2866 sh	2882 w br	-CH _x stretching	
	1753 vw		-C=O carboxylic stretching	
		1707 s sp	-C=O carboxylic stretching	
	1672 vw br		-C=O primary amide stretching	
1635m			-O-H bending	
	1630 m sp		-C=C, -C=O stretching, -NH ₂ scissoring	
	1593 vw br		-C=C, -C=O stretching, -NH ₂ scissoring	
		1548w br	-C=C, -C=O stretching, -NH ₂ scissoring	
1465 w			-CH ₂ deformation bending	
1427 m sp			-C=C- tangential displacement	
	1409 s br	1406 s br	-C-N amide III stretching, -C-O hydroxyl stretching	
		1329 s br	-C-OH carboxylic acid stretching	
1367 w			-CH ₂ deformation bending	
	1307 s br		-C-OH carboxylic acid stretching	
		1245 m br	-CH _x wagging	
		1209 m sh	-CH _x wagging	
		1178 m sh	-C-OH carboxyl stretching	
1190 w			O-H···O-C stretching	
1180 m sp			-C=C- tangential displacement	
	1094 w sh		-C-O-C asymmetric ether stretching	
1055 br		1055 s br	-C-O-C- asymmetric stretching	
	1042 m sp	1032 m sh	-C-O-C asymmetric ether stretching	
		1022 m sh	-C-O-C asymmetric ether stretching	
914 w			-CH ₂ -CH ₂ - stretching	
		935 w sp	-NH wagging, -C-O-C symmetric ether stretching	
		905 w sp	-C-O-C symmetric ether stretching,	
		874 vw br	-NH wagging, -C-O-C symmetric ether stretching	
	827 m sp	826 s sp	-C-O-C epoxy bending	
807 w				
		714 m sp	-NH ₂ wagging	
572 s sp			-C-H radial displacement	
522 s sp			-C-H radial displacement	

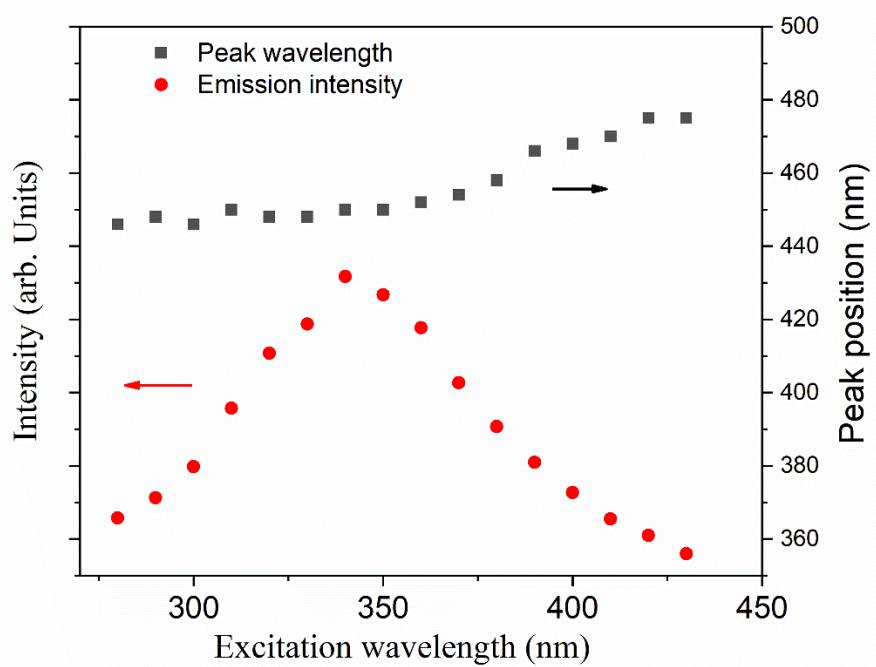


Figure S2. Dependence on the excitation wavelength of fluorescent emission intensity and peak wavelength in N-CDs-W-THF.

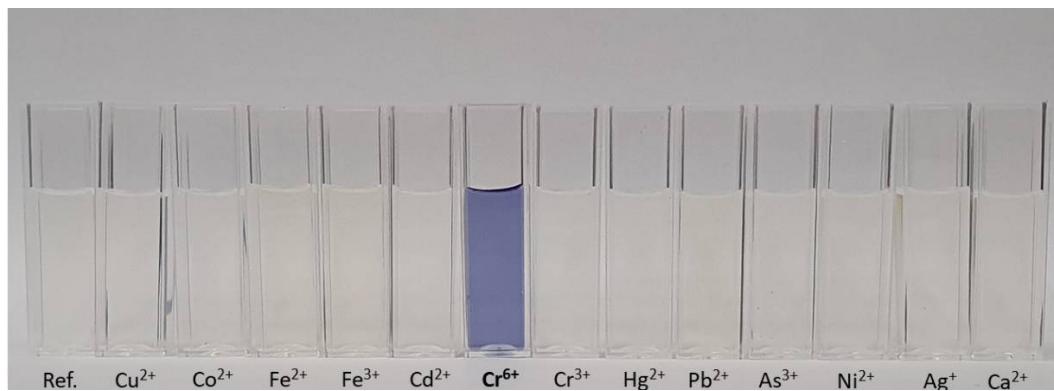


Figure S3. White-light image of the pristine N-CDs-W-THF sensing solution (Ref) and after the addition of different HM ions at a concentration of 100 μ M and calcium ions at 1 mM.

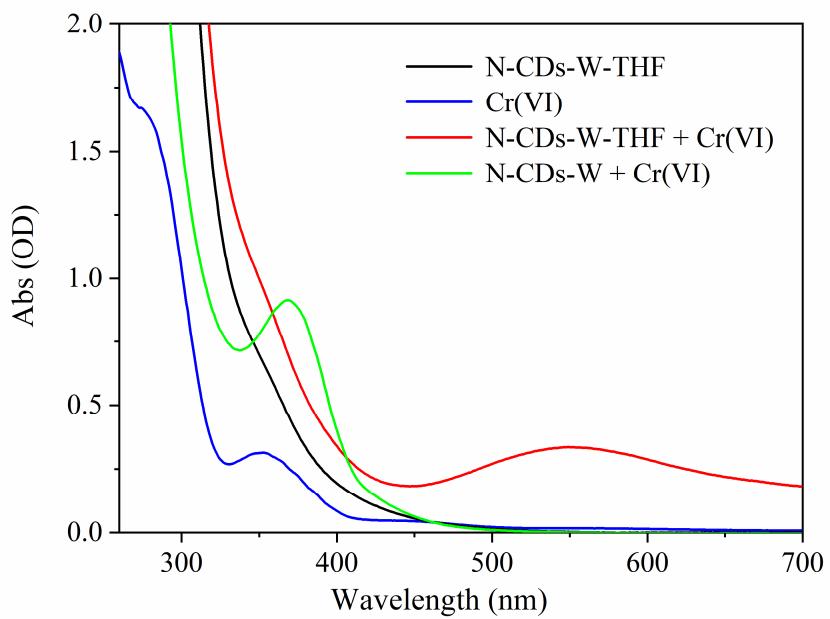


Figure S4. UV-vis absorption spectra of N-CDs-W-THF reference solution (black curve), Cr(VI) in DI water at 100 μM (blue), N-CDs-W-THF solution (red) and N-CDs-W solution (green) in the presence of Cr(VI) ions at a concentration of 100 μM .

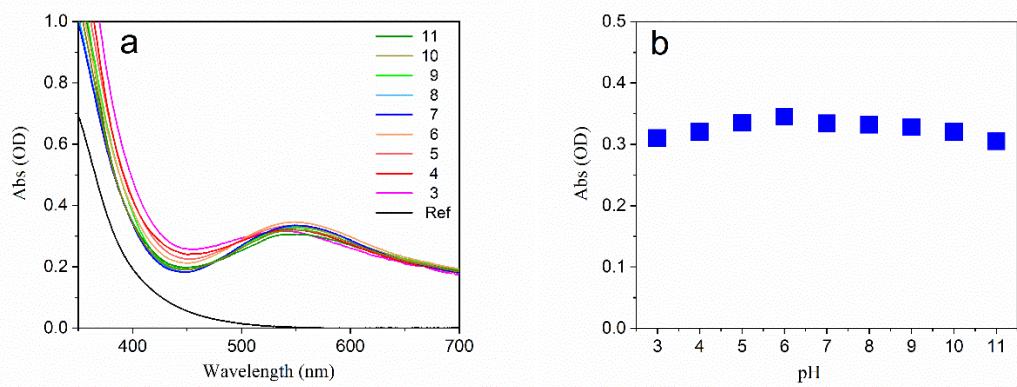


Figure S5. (a) UV-vis absorption spectra of N-CDs-W-THF and (b) absorbance at 550 nm of the sensing solution upon the addition of DI water with 100 μM of Cr(VI) ions at different values of pH in the optimized volume ratio 2:1.

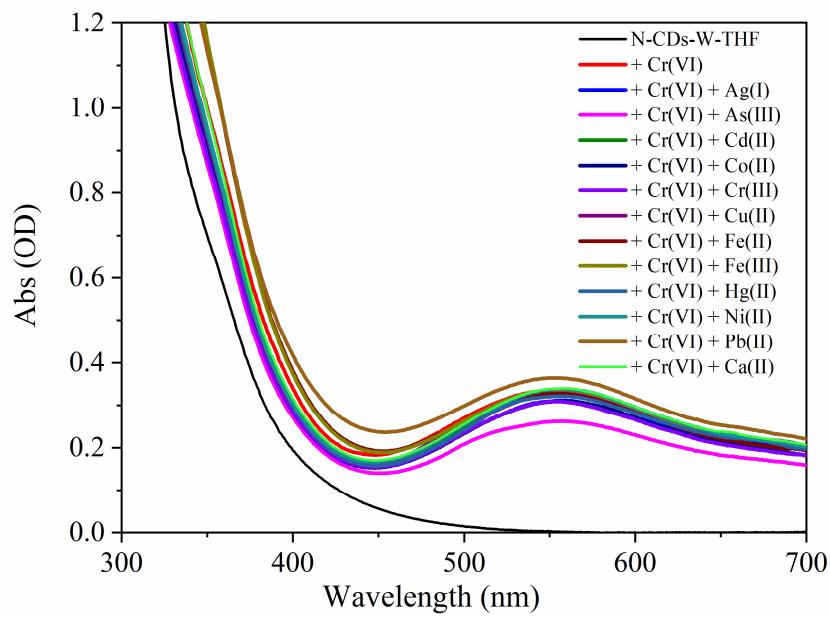


Figure S6. UV-vis absorption spectra of N-CDs-W-THF reference solution upon the addition of Cr(VI) and other interfering HM ions at a concentration of 100 μM and calcium ions at 1 mM.

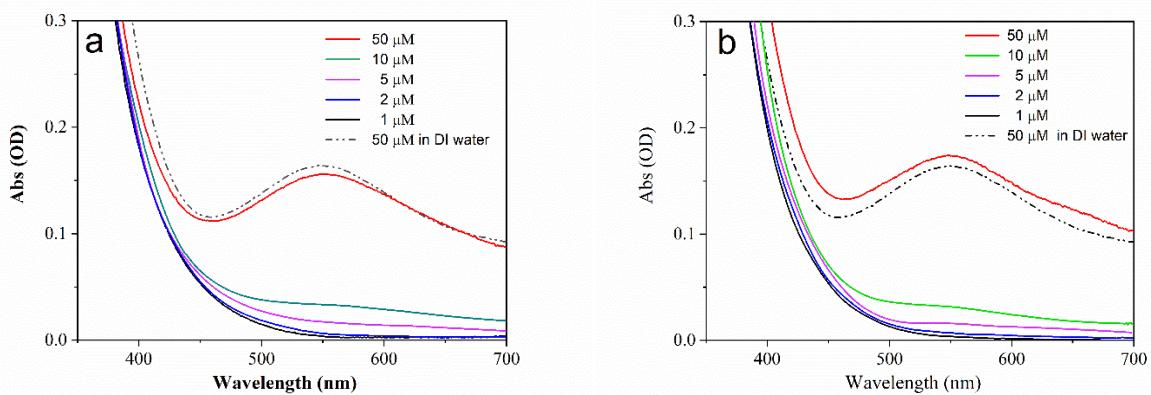


Figure S7. UV-vis absorption spectra of N-CDs-W-THF reference solution upon the addition of (a) tap water and (b) lake water spiked with different concentrations of Cr(VI).

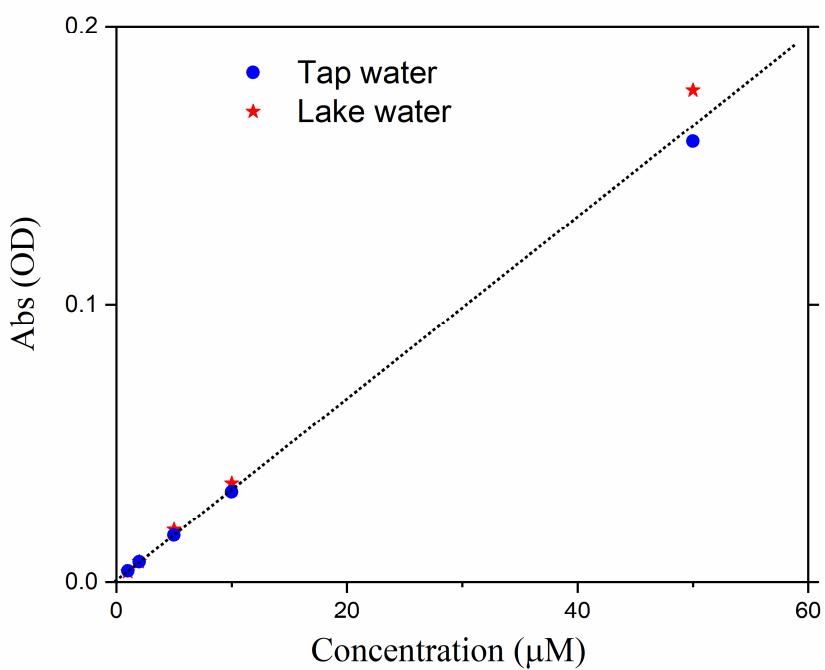


Figure S8. Absorbance at 550 nm as a function of Cr(VI) concentration in tap water samples (circles) and lake water samples (stars). The dotted line is the calibration curve obtained with DI water.