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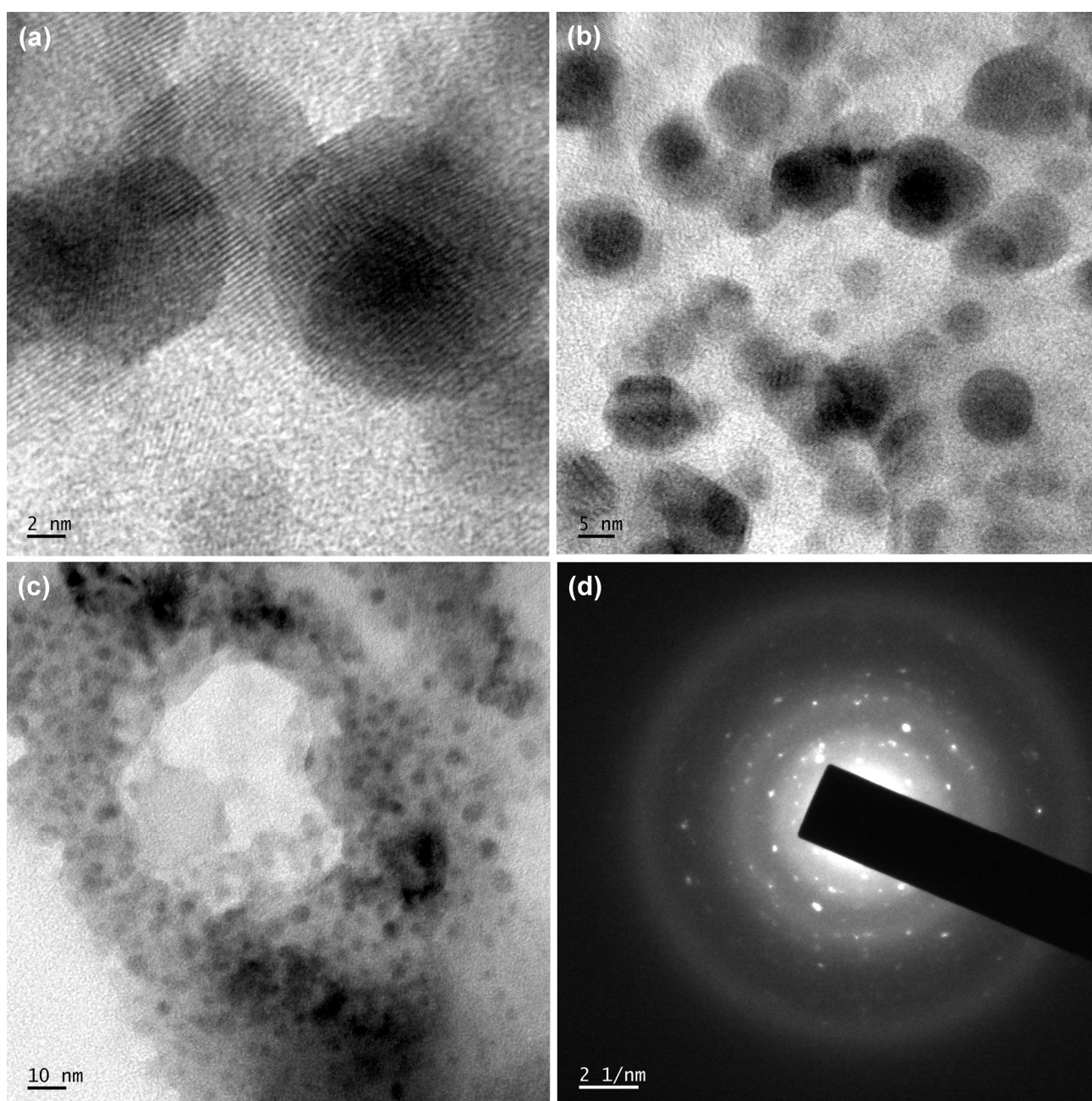
## Supplementary Materials

We examined the drop cast CsPbI<sub>2</sub>Br sample on the TEM copper grid without any annealing process (**Figures S1a-c**). Also **Figure S1d** displays the selected area electron diffraction (SAED) pattern of CsPbI<sub>2</sub>Br displaying the phase impurity, which is different from the XRD pattern in Figures 7a and 8 (see Results and Discussion in *Article*). This indicates that without annealing, the sample has high phase-impurity, e.g., non-perovskite yellow phase (see **Table S1** containing estimated  $2\theta$  data for comparison purpose). Hence, the pseudo-cubic  $\alpha$ -phase CsPbI<sub>2</sub>Br with ‘iodine atoms’ might be kinetically stable (not thermodynamically) due to polymorphism [1,2]. Therefore, ‘iodine’ has double sides, increasing light absorption but causing stability problems under ambient conditions.

**Table S1.** Analysis of selected area electron diffraction (SAED) image for CsPbI<sub>2</sub>Br without annealing at room temperature.

Perovskite	Diameter (1/nm)	Radius (1/nm)	<i>d</i> -Spacing (nm)	$2\theta$ (°)
CsPbI <sub>2</sub> Br	3.91	1.96	0.51	17.3
	5.23	2.63	0.38	23.2
	6.32	3.16	0.32	28.2
	8.05	4.03	0.25	36.1
	10.08	5.04	0.20	45.7
	12.10	6.05	0.17	55.5

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**Figure S1.** HR-TEM image of CsPbI<sub>2</sub>Br with different scale bar: (a) 2 nm, (b) 5 nm, and (c) 10 nm. (d) SAED pattern of CsPbI<sub>2</sub>Br displaying the phase impurity at room temperature.

## References

1. Alaei, A.; Circelli, A.; Yuan, Y.; Yang, Y.; Lee, S.S. Polymorphism in metal halide perovskites. *Mater. Adv.* **2021**, *2*, 47-63.
2. Wang, B.; Novendra, N.; Navrotsky, A. Energetics, Structures, and Phase Transitions of Cubic and Orthorhombic Cesium Lead Iodide (CsPbI<sub>3</sub>) Polymorphs. *J. Am. Chem. Soc.* **2019**, *141*, 14501–14504.