

Proceedings

Detection of Adrenaline Based on Bioelectrocatalytical System to Support Tumor Diagnostic Technology [†]

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Abstract: An amperometric biosensor based on the bioelectrocatalytic measurement principle for the detection of adrenaline has been developed. The adrenaline sensor has been prepared by modification of a platinum thin-film electrode with a pyrroloquinoline quinone-dependent glucose dehydrogenase. The enzyme was immobilized via cross-linking method. Lower detection limit of 1 nM of adrenaline has been achieved by measuring at physiological level at pH 7.4.

Keywords: chip-based biosensor; adrenaline; glucose dehydrogenase; bioelectrocatalysis

1. Introduction

Adrenal vein sampling (AVS) is the method of choice for characterizing aldosterone-producing adenoma of the adrenal gland and to distinguish unilateral from bilateral diseases [1,2]. While patients with bilateral disease are treated with life-long medication, patients with unilateral disease are offered adrenalectomy which leads to significant improvement and has the potential of cure [3,4]. However, AVS is an invasive and laborious technique because both adrenal veins must be sampled for meaningful comparison. Furthermore, placing the catheter tip within the right adrenal vein is quite challenging. Therefore, measuring a specific biomarker during catheterization could reduce X-ray exposure time and enhance result quality of the medical examination. In this case, adrenal veins' adrenaline (epinephrine) concentrations can be measured to confirm successful catheterization of the adrenal veins: there is a gradient of the adrenaline concentration between adrenal blood (>100 nM) and peripheral blood (~1 nM) that can be employed as an indicator for correct catheter tip position.

Several sensors have been proposed for the detection of adrenaline or other catecholamines by using recycling processes for achieving signal amplification and to improve lower detection limit [5–9]. However, the optimum working pH of the described sensors does not comply with the pH value of blood that limits their application for the adrenaline detection in real blood samples.

In this contribution, a high-sensitive adrenaline thin-film biosensor based on the bioelectrocatalytic measurement principle has been realized by using a pyrroloquinoline quinone (PQQ)-dependent glucose dehydrogenase (GDH), which is stable in a broad pH range. Figure 1 shows the

bioelectrocatalytical amplification principle. The recycling system implies an amplification of the sensor signal. The results of investigation of adrenaline detection in buffer at physiological level of pH 7.4 are presented.

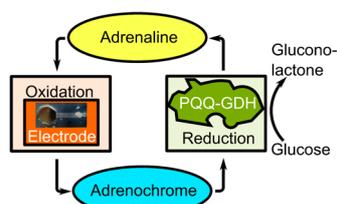


Figure 1. Scheme of bioelectrocatalytical measurement principle for amplification of the adrenaline sensor signal. In a first reaction step, adrenaline is oxidized to adrenochrome. Then, reduction of adrenochrome was catalyzed by the enzyme PQQ-GDH.

2. Materials and Method

2.1. Chemicals

The enzyme GDH was bought from Sorachim SA (Switzerland, 757 U/mg) [10] and PQQ was purchased from Wako (Japan). Glutaraldehyde, bovine serum albumin (BSA), glycerol, CaCl_2 and the phosphate buffer (PBS) components (monosodium phosphate and disodium phosphate) were obtained from Sigma-Aldrich. Adrenaline solution (1 mg/mL) was purchased from Infectopharm (Germany). Adrenaline stock solution of 1 μM and 100 μM were stored at 4 °C in the dark.

2.2. Preparation of the Adrenaline Biosensor

The adrenaline biosensor was realized by modification of a platinum thin-film electrode with an enzyme membrane (see Figure 2). The sensor chip was fabricated by means of conventional silicon and thin-film technologies. For the preparation of the enzyme membrane, 4 μL of GDH solution solved in PBS, together with 20 μM PQQ, 1 mM CaCl_2 [11], 8 μL of BSA (10 vol %) and 8 μL of a glutaraldehyde (2 vol %)/glycerol (10 vol %) solution were mixed. The membrane cocktail was then dropped onto the platinum electrode. After drying, the chip was stored at 4 °C until required.

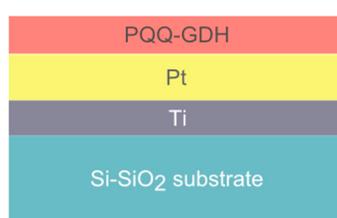


Figure 2. Adrenaline thin-film biosensor arrangement with enzyme membrane.

2.3. Electrochemical Sensor Characterization

For the electrochemical characterization, the adrenaline biosensor was connected to a potentiostat (PalmSens, Palm Instruments BV, The Netherlands). A three-electrode arrangement was used, where a conventional liquid-junction Ag/AgCl electrode (Metrohm, Herisau, Switzerland) was utilized as a reference electrode and a platinum wire as a counter electrode. For the oxidation reaction of adrenaline, a constant potential of +450 mV vs. Ag/AgCl was applied to the platinum working electrode and the resulting current was recorded over time.

3. Results and Discussion

Figure 3 depicts the calibration curve of the developed adrenaline biosensor. Different adrenaline concentrations in the range of 1 nM to 150 nM were measured in buffer solution of pH 7.4 in the presence of 20 mM glucose. A lower detection limit of ~1 nM adrenaline could be reached.

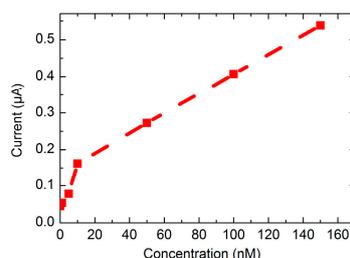


Figure 3. Calibration curve of adrenaline biosensor in the concentration range of 1 nM–150 nM.

4. Conclusions and Outlook

In this work, an amperometric thin-film biosensor for the detection of adrenaline has been presented. Using the bioelectrocatalysis measurement principle for the signal amplification, a lower detection limit of 1 nM in buffer solution at physiological level of pH 7.4 has been achieved. Ongoing experiments focus on the investigation of this biosensor in Ringers' solution as well as real blood samples. We see potential for application in adrenal venous sampling for adrenal gland disease, for catheter-based detection of paragangliomas and also in stress-research.

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Conflicts of Interest: The authors declare no conflict of interest. The founding sponsors had no role in the design of the study; in the collection, analyses, or interpretation of data; in the writing of the manuscript, and in the decision to publish the results.

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