

## Supplementary Materials

# Metal Oxide/TiO<sub>2</sub> Hybrid Nanotubes Fabricated through the Organogel Route

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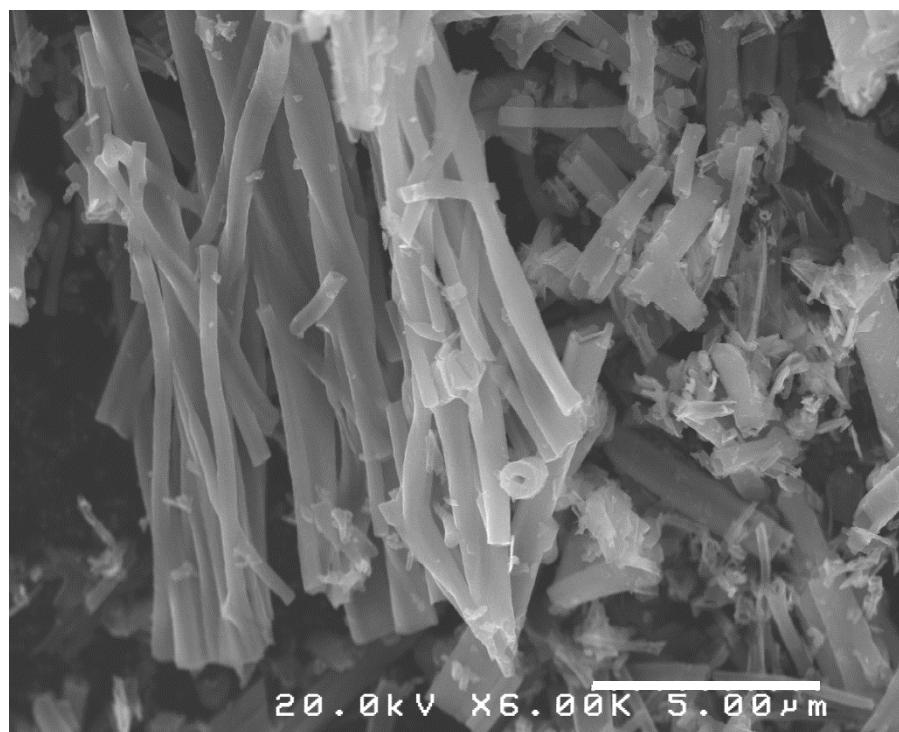
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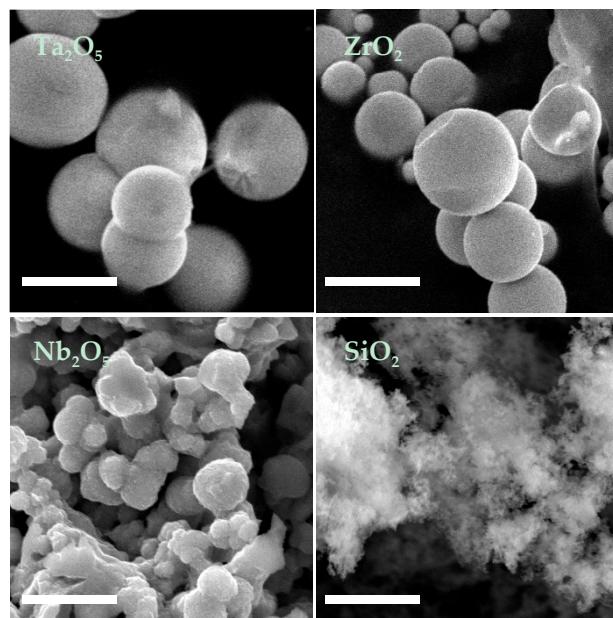
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### Synthesis of gelator 1

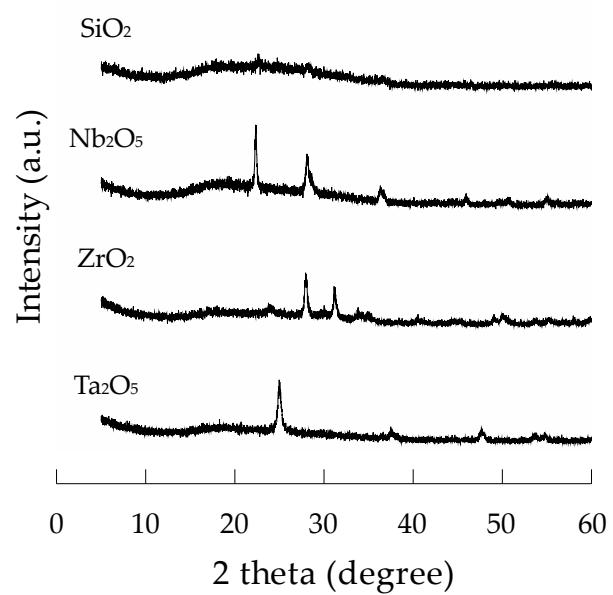
N<sup>ε</sup>-Bis(hexylaminocarbonyl)-L-lysine methyl ester (10 mmol) was dissolved in MeOH-H<sub>2</sub>O (400 mL–20 mL) and 4M NaOH (5 mL) was added. The reaction mixture was heated at 70°C for 24 h. The resulting solution was evaporated to ca. 100 mL, and water (600 mL) was added. The aqueous solution was adjusted to ca. pH = 1 by addition of 6M HCl with vigorous stirring. The white precipitate was filtered, washed with water, and then dried. The product was obtained by recrystallization from MeOH-ethyl acetate (92%). m.p. = 167–169 °C. IR (KBr): 3334 cm<sup>-1</sup> ( $\nu$ N–H), 1782 cm<sup>-1</sup>, 1701 cm<sup>-1</sup> ( $\nu$ C=O, CO<sub>2</sub>H), 1621 cm<sup>-1</sup> ( $\nu$ C=O, urea), 1582 cm<sup>-1</sup> ( $\delta$ N–H); <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>, TMS):  $\delta$  = 0.87 (t, J = 6.6Hz, 6H), 1.27 (br, 14H), 1.38–1.51 (m, 6H), 1.71–1.81 (m, 2H), 3.11 (br, 6H), 4.26 (br, 1H), 5.41 (br, 1H), 5.64 (br, 1H), 5.89 (br, 1H), 6.19 (br, 1H). Elemental analysis calcd (%) for C<sub>20</sub>H<sub>40</sub>N<sub>4</sub>O<sub>4</sub> (400.56): C, 59.97; H, 10.07; N, 13.99. Found: C, 60.66; H, 10.61; N, 14.13.



**Figure S1.** SEM image of TiO<sub>2</sub> nanotube and small fragments of TiO<sub>2</sub>.

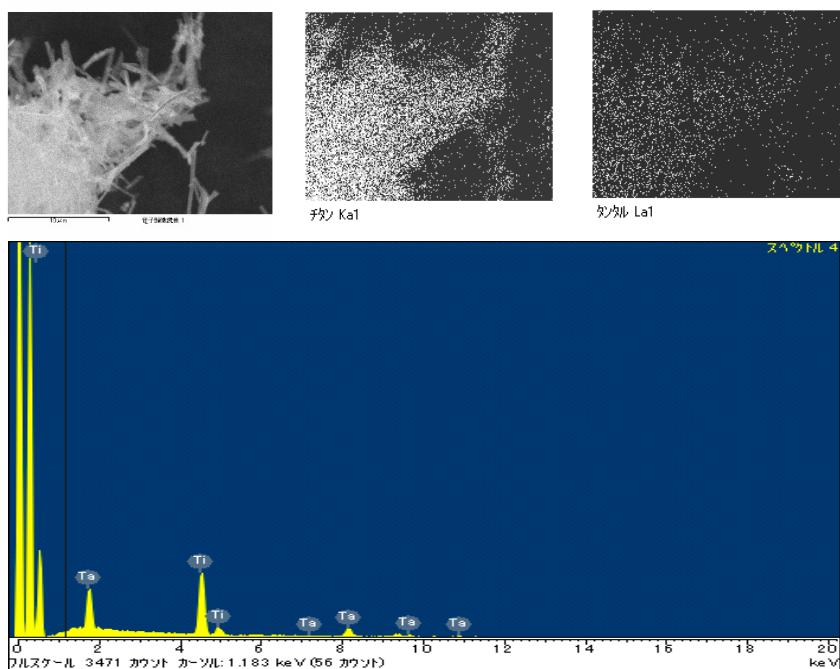


**Figure S2.** SEM images of Ta<sub>2</sub>O<sub>5</sub>, ZrO<sub>2</sub>, Nb<sub>2</sub>O<sub>5</sub>, and SiO<sub>2</sub> fabricated without gelator. Scale bars are 5 μm.

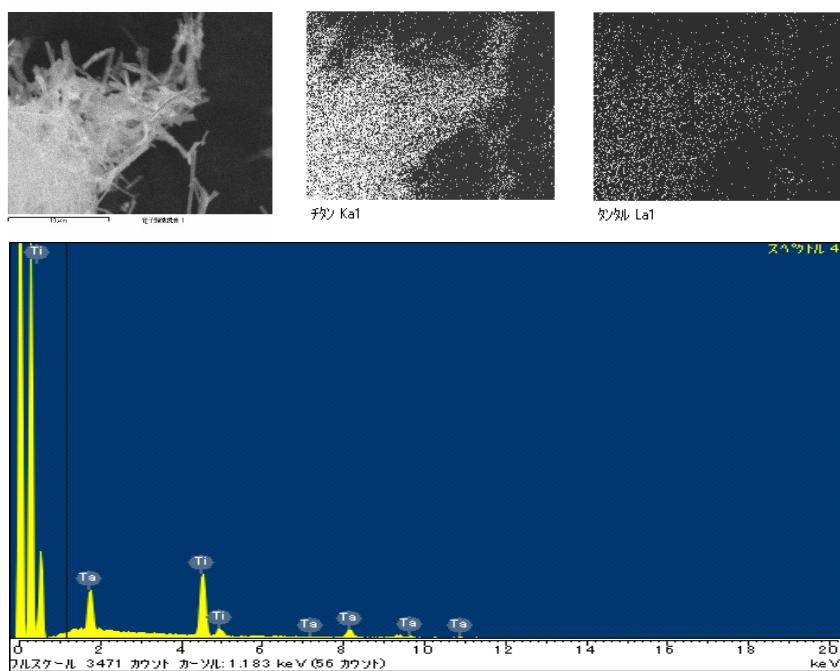


**Figure S3.** XRD patterns of  $\text{Ta}_2\text{O}_5$ ,  $\text{ZrO}_2$ ,  $\text{Nb}_2\text{O}_5$ , and  $\text{SiO}_2$  nanotubes fabricated in ethanol gel.

Ti:Ta = 9:1

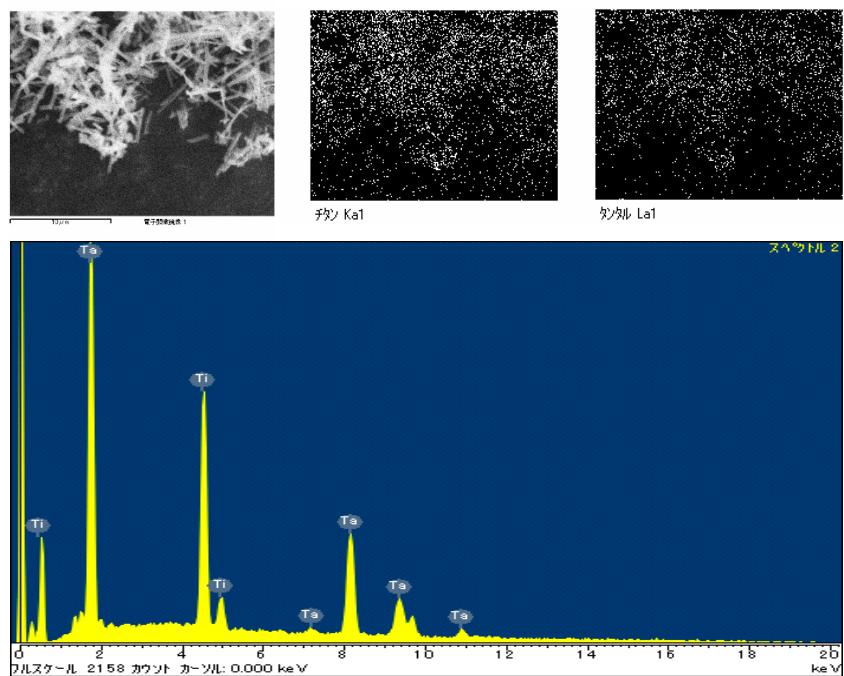


Ti:Ta = 8:2

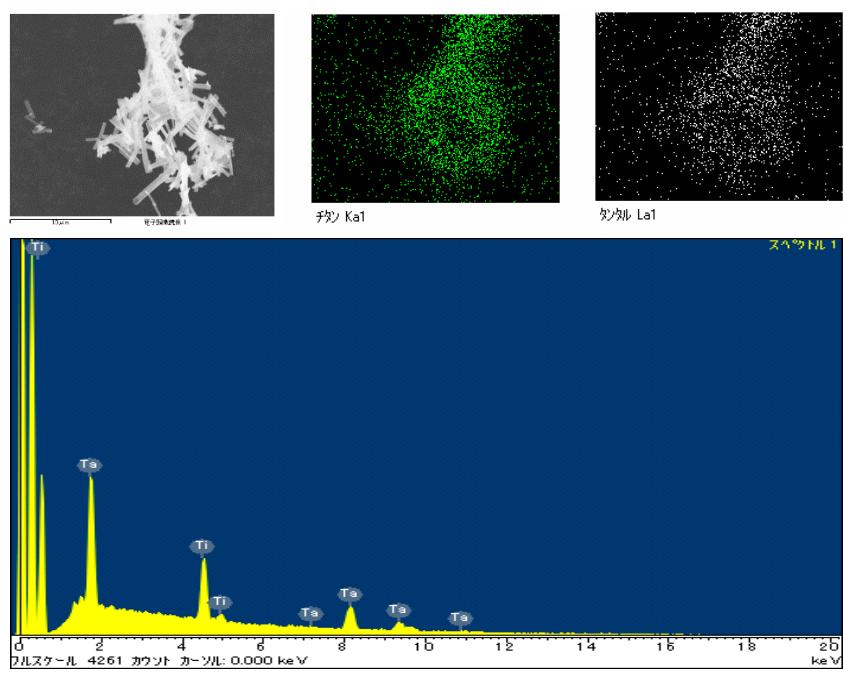


**Figure S4.** EDX spectra and element mappings of  $\text{Ta}_2\text{O}_5/\text{TiO}_2$  hybrid nanotubes. Upper: Ti:Ta = 9:1; lower: Ti:Ta = 8:2. Scale bars are 10 μm.

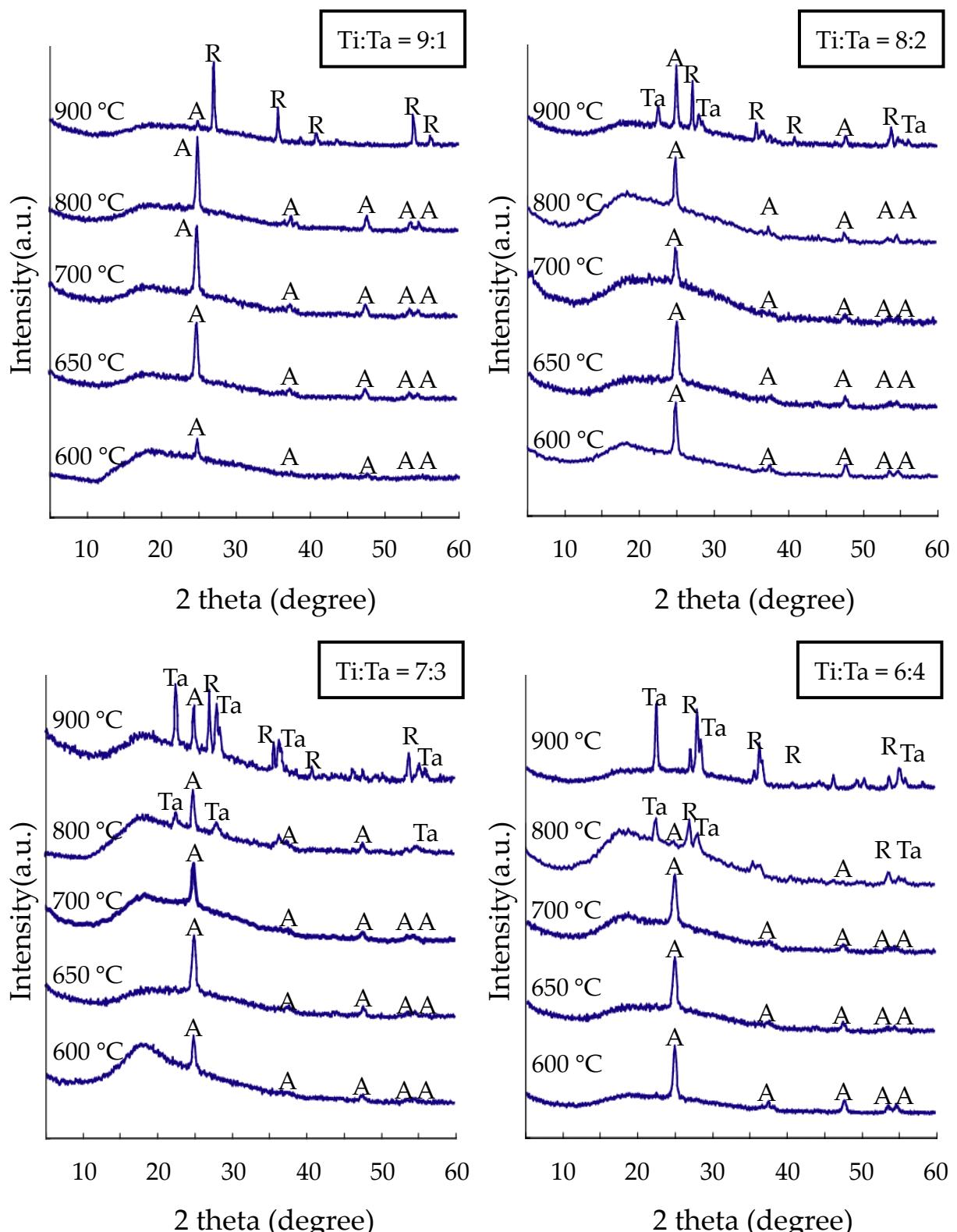
Ti:Ta = 7:3



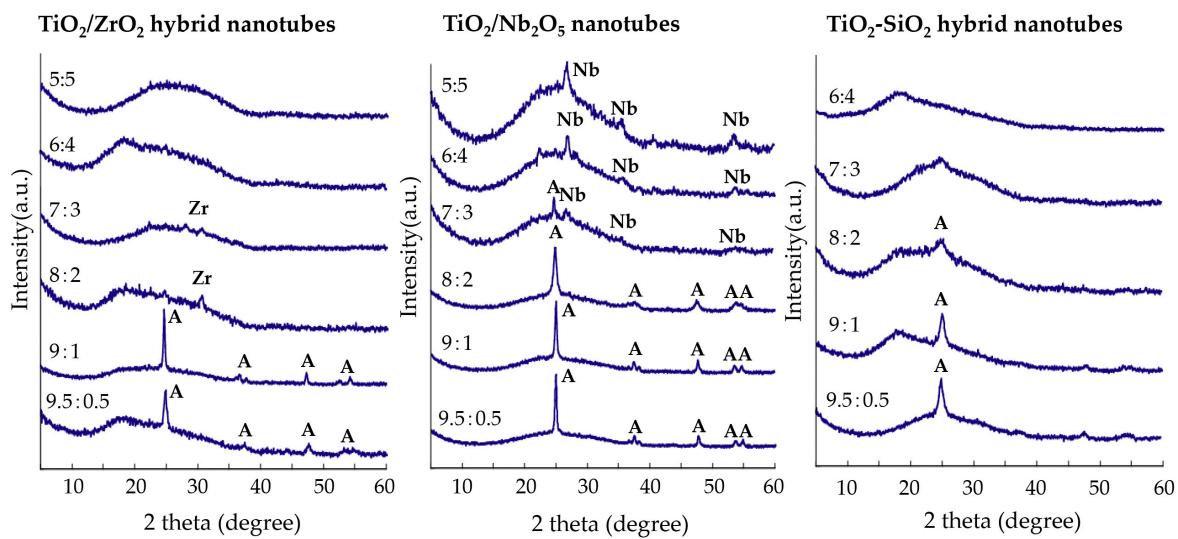
Ti:Ta = 6:4



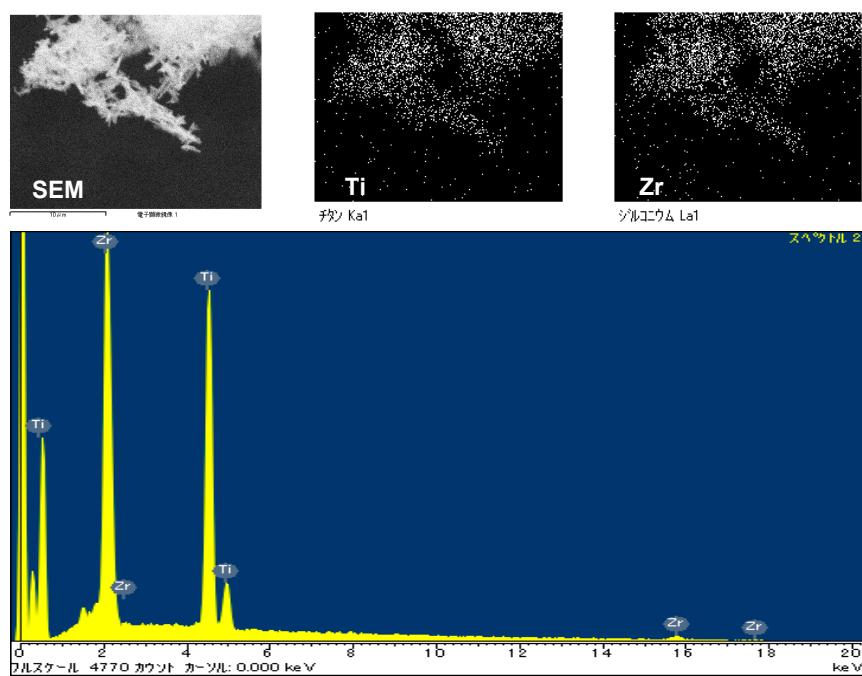
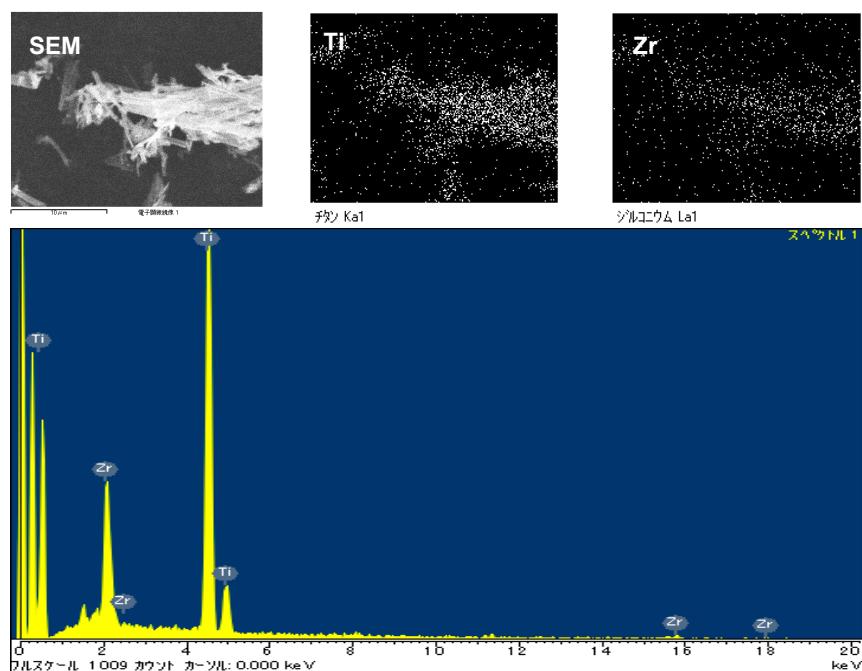
**Figure S5.** EDX spectra and element mappings of  $\text{Ta}_2\text{O}_5/\text{TiO}_2$  hybrid nanotubes. Upper: Ti:Ta = 7:3; lower: Ti:Ta = 6:4. Scale bars are 10  $\mu\text{m}$ .



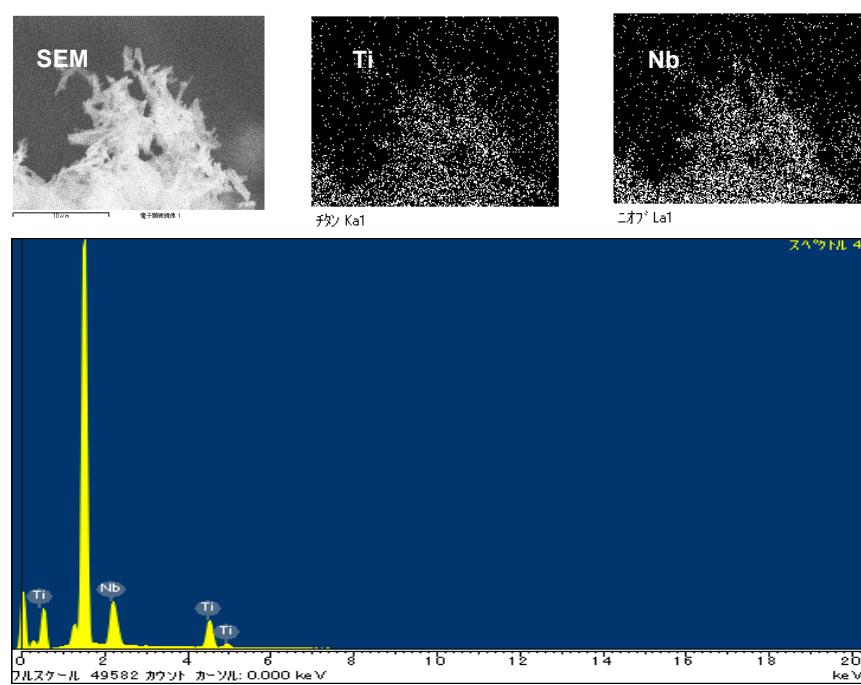
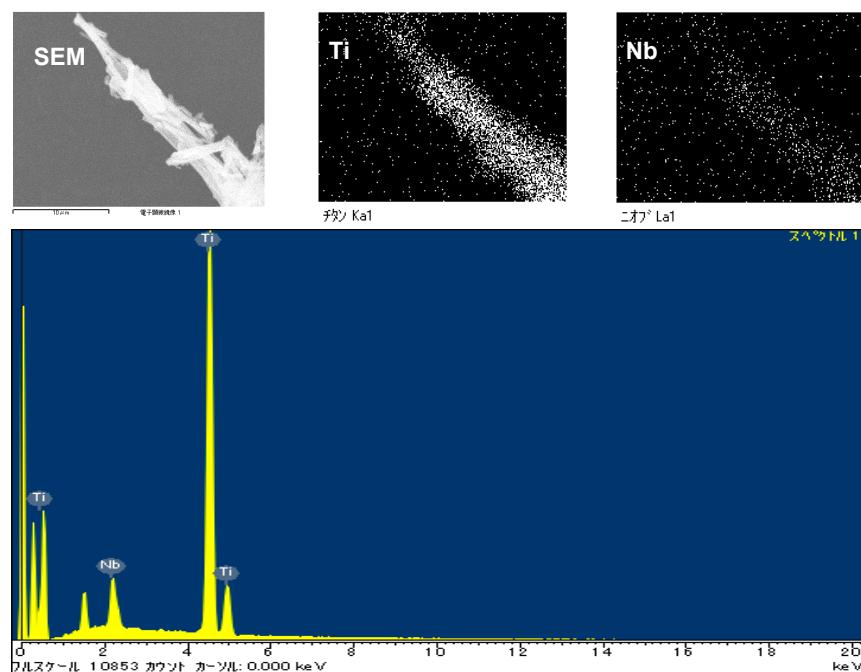
**Figure S6.** XRD patterns of Ta<sub>2</sub>O<sub>5</sub>/TiO<sub>2</sub> hybrid nanotubes fabricated at various calcination temperatures.



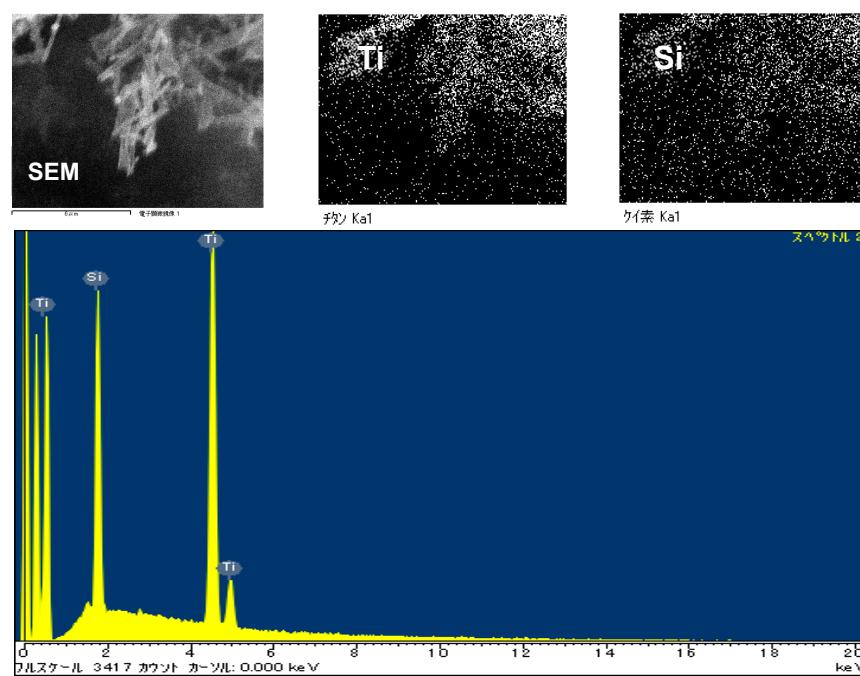
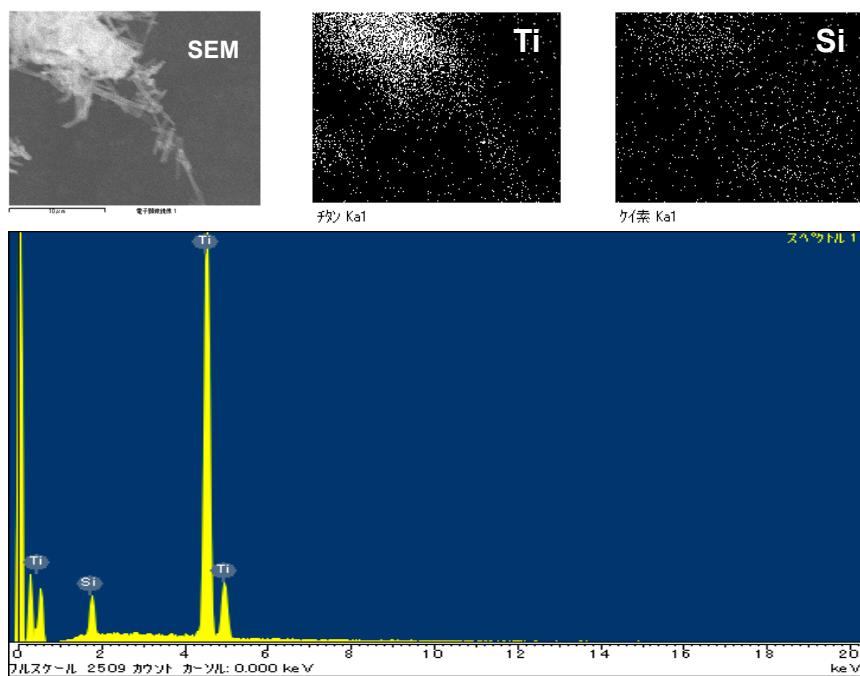
**Figure S7.** XRD patterns of ZrO<sub>2</sub>/TiO<sub>2</sub>, Nb<sub>2</sub>O<sub>5</sub>/TiO<sub>2</sub>, and SiO<sub>2</sub>/TiO<sub>2</sub> hybrid nanotubes.



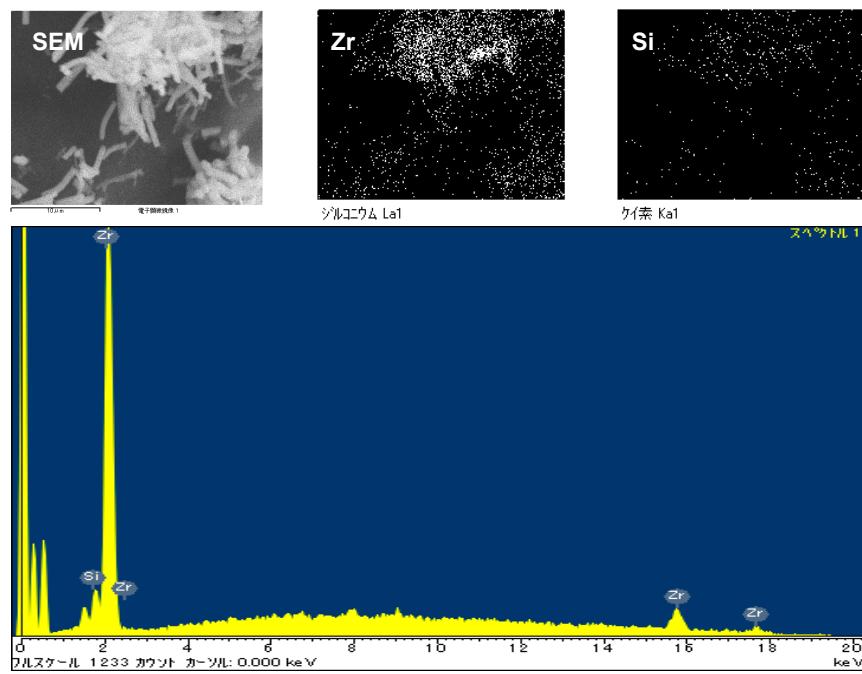
**Figure S8.** EDX spectra and element mappings of  $\text{ZrO}_2/\text{TiO}_2$  hybrid nanotubes. Upper:  $\text{Ti}:\text{Zr} = 9:1$ ; lower:  $\text{Ti}:\text{Zr} = 5:5$ . Scale bars are  $10 \mu\text{m}$ .



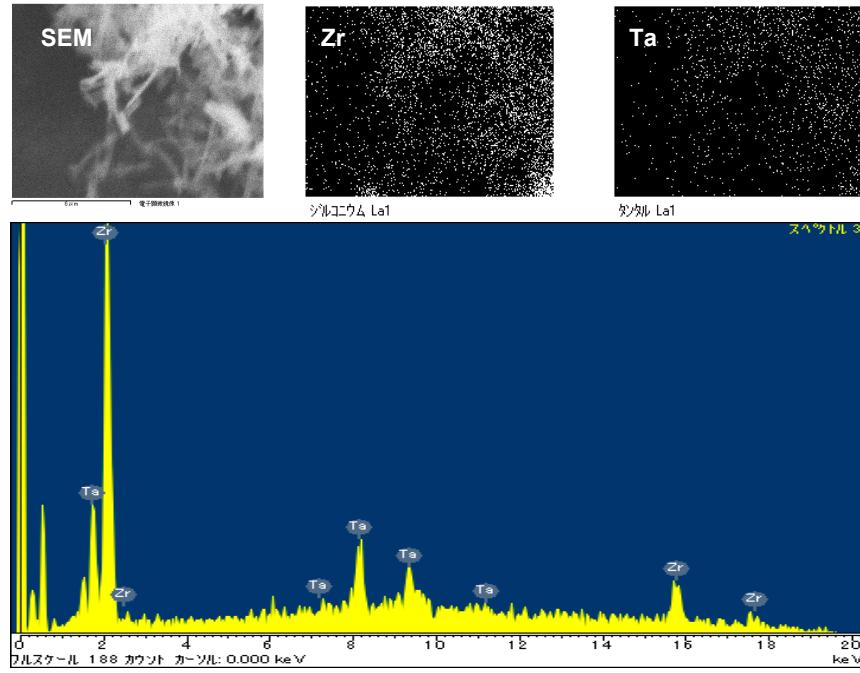
**Figure S9.** EDX spectra and element mappings of  $\text{Nb}_2\text{O}_5/\text{TiO}_2$  hybrid nanotubes. Upper: Ti:Nb = 9:1; lower: Ti:Nb = 5:5. Scale bars are 10  $\mu\text{m}$ .



**Figure S10.** EDX spectra and element mappings of  $\text{SiO}_2/\text{TiO}_2$  hybrid nanotubes. Upper: Ti:Si = 9:1; lower: Ti:Si = 6:4. Scale bars are 10  $\mu\text{m}$ .



**Figure S11.** EDX spectra and element mappings of ZrO<sub>2</sub>/SiO<sub>2</sub> hybrid nanotube (Zr:Si = 9:1). Scale bars are 10 μm.



**Figure S12.** EDX spectra and element mappings of ZrO<sub>2</sub>/Ta<sub>2</sub>O<sub>5</sub> hybrid nanotube (Zr:Ta = 9:1). Scale bars are 10 μm.