

Supplementary Information: Bimetallic Organic Gel for Effective Methyl orange

Dye Adsorption

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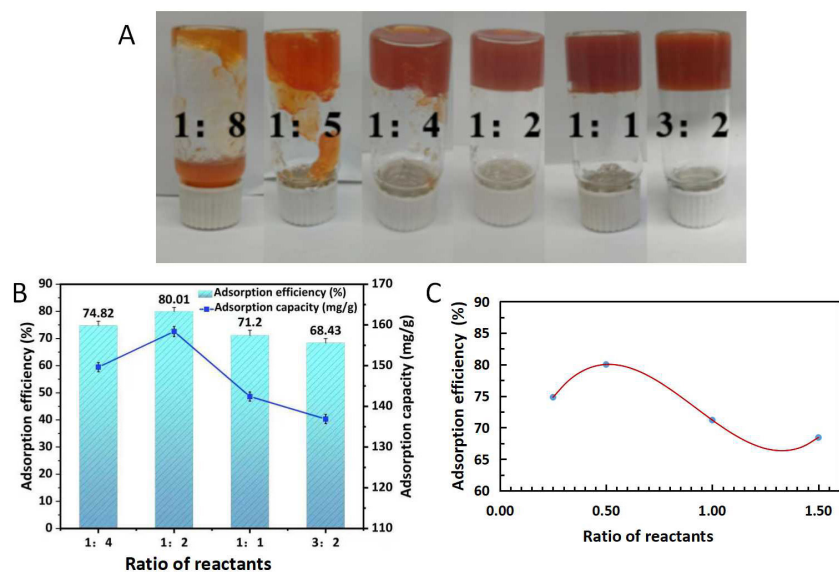


Figure S1. Products synthesized under different reactants ratio (under the same synthesis conditions) (A); Adsorption efficiency of the products (reactant ratios of 1:4, 1:2, 1:1, and 3:2, respectively) (B); Fitting results of the relationship between the reactant ratio and the adsorption efficiency (C) (C_0 (MO): 100 mg/L; adsorbent: 0.5 g/L; pH: deregulation).

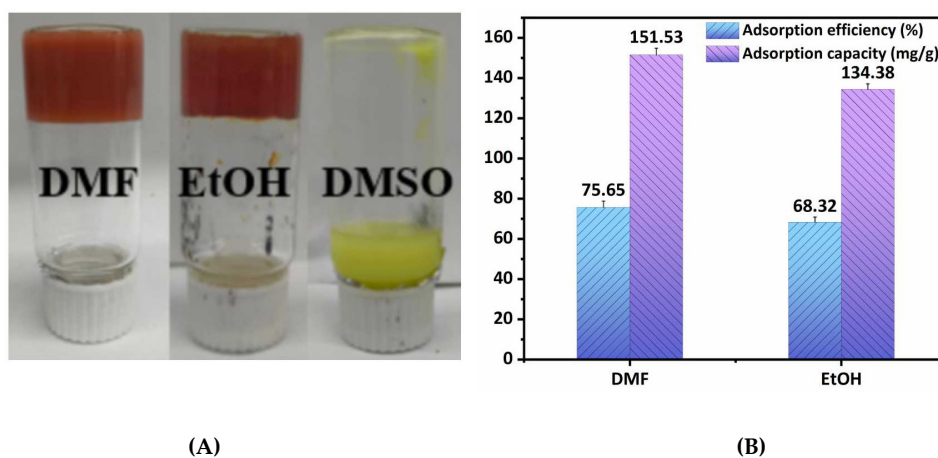


Figure S2. Products synthesized in different solvents (under the same synthesis conditions) (A); Adsorption performance of products from DMF and EtOH (B) (C_0 : 100 mg/L; adsorbent: 0.5 g/L; pH: deregulation).

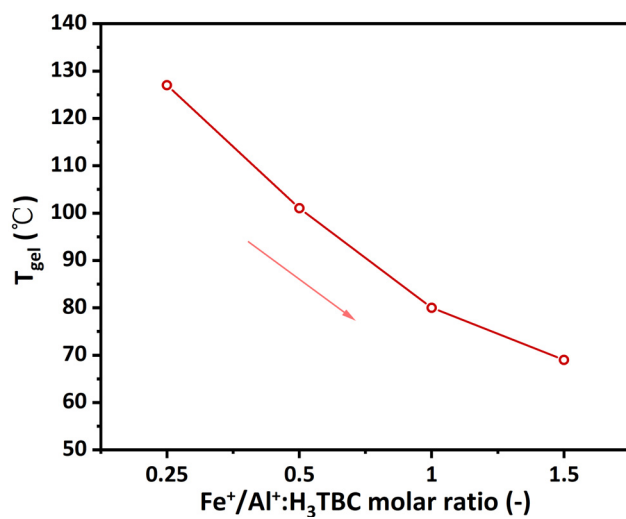


Figure S3. The result of T_{gel} screening (MOG-Fe/Al, reactant ratios:1:4, 1:2, 1:1, and 3:2).

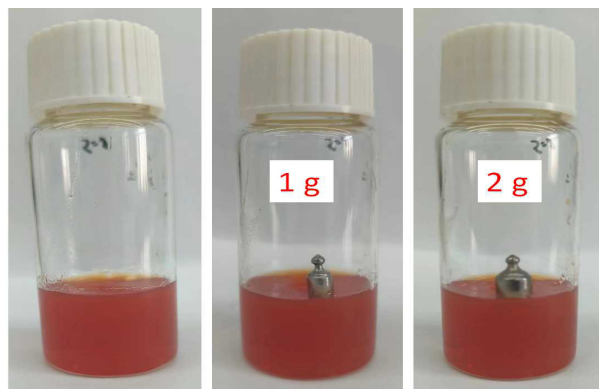


Figure S4. The result of falling ball experiment of MOG-Fe/Al (Observation time: 24 h, 1 g and 2 g weights were used).

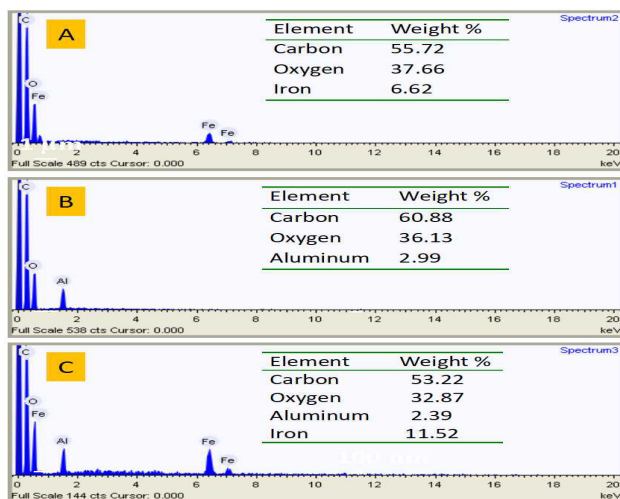


Figure S5. EDS spectra of MOG-Fe (A), MOG-Al (B), MOG-Fe/Al (C).

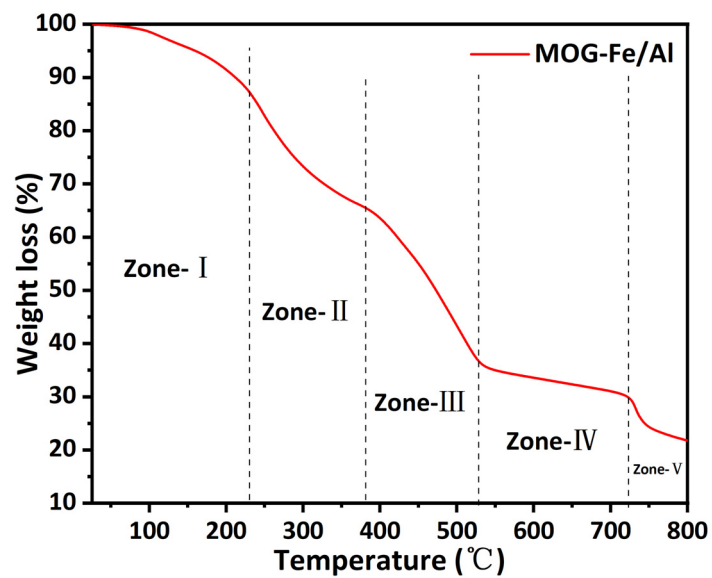


Figure S6. The TGA curve of MOG-Fe/Al.

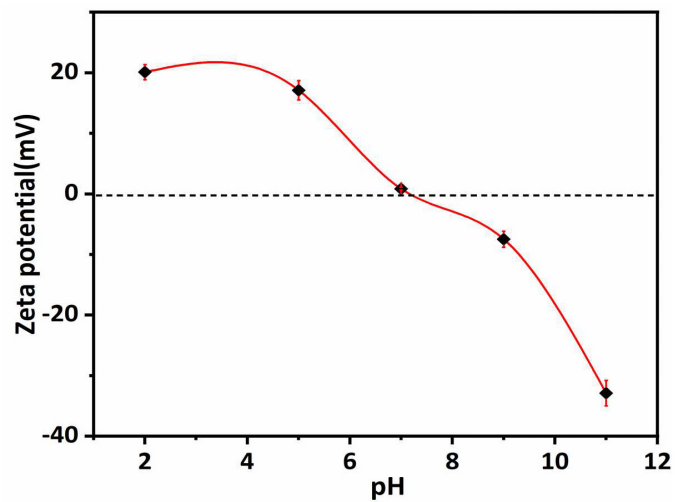


Figure S7. Zeta potential at different pH (adsorbent: 0.5 g/L; pH: deregulation; T: 298 K).

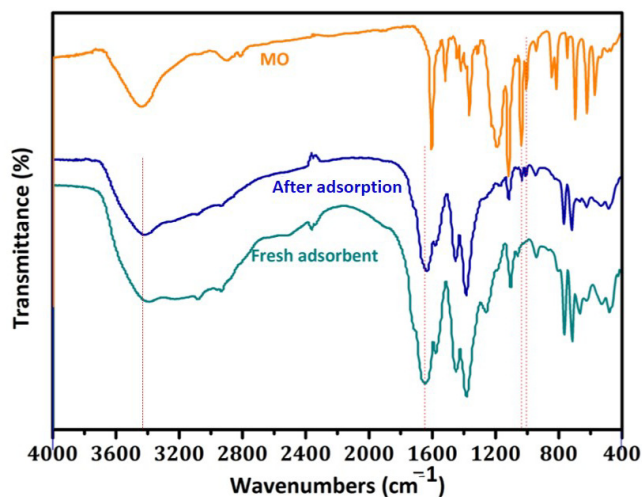


Figure S8. The FT-IR spectra of MOG-Fe/Al before and after MO adsorption.

Table S1 – Homogenous processes by leached iron and aluminum from the MOG-Fe/Al.

Metal element	Fe	Al
Mass of fraction of element ^a (%)	11.52	2.39
Concentration of leached element ^b (mg/L)	0.0512	0.0449
Leached element percentage of total element ^c (%)	0.09	0.37
<p>a: Mass fraction of element: tested by EDS.</p> <p>b: Concentration of leached element: tested by ICP-OES.</p> <p>c: Leached element percentage of total element: calculated with 20 mg adsorbents and by the equation $\text{percentage} = \frac{m_L}{m_T} \times 100 \%$</p> <p>$m_L$ = leached element content (mg), m_T = total element content in the fresh adsorbents (mg).</p>		

Table S2 – Comparison of MO adsorption capacities of with MOG-Fe/Al other reported adsorbents.

Adsorbent	Dosage (g/L)	Pollutant concentration (mg/L)	Adsorption capacity (mg/g)	References
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Zn -MOG	2	50	100	Colloid surface A. 2021 , 628, 127335 [40]
Bimetallic MOG	1	50-200	265	Sci. Rep. 2015 , 5, 10556. [24]
Magnetic porous carbon (MPC)	0.5	12.5-400	182.82	Dalton Transactions, 2016 , 45, 4541-4547 [41].
Three-dimensional graphene aerogel (TCGA)	1	100	56.4	New Journal of Chemistry, 2020 , 44, 16285-16293. [42]
MOG-Fe/Al	0.5	10-500	335.88	This work