

SUPPLEMENTARY MATERIALS

Porous Layered Double Hydroxide/TiO₂ Photocatalysts for the Photocatalytic Degradation of Orange II

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Table SI 1: Comparison of the efficiency of different photocatalyst for OII degradation

Photocatalyst	% of OII photodegradation	Ref
TiO ₂ immobilized on glass slides	9%	[1]
TiO ₂ /SiO ₂	26%	[2]
ZnCr- LDH	10%	[3]
ZnCr- LDH calcined at 600°C	18%	[3]
[MgAl/TiO ₂] _{0.66}	42%	This work
Reduced graphene oxide-TiO ₂	95%	[4]

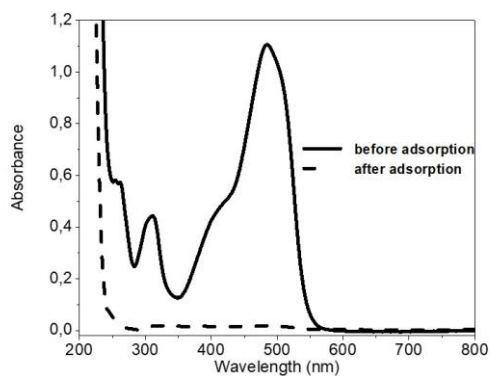


Figure S1. UV-visible spectra of OII (5.10^{-5} M, pH = 9.30) solution in presence of $[\text{Zn}_2\text{AlNO}_3/\text{TiO}_2]_{\text{dry}}$ before and after stirring the suspension in the dark

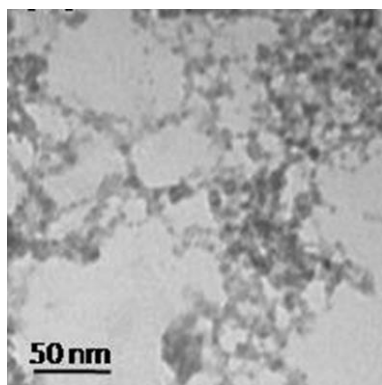


Figure S2. TEM image of TiO_2 nanoparticles aged after titration (pH = 9.30)

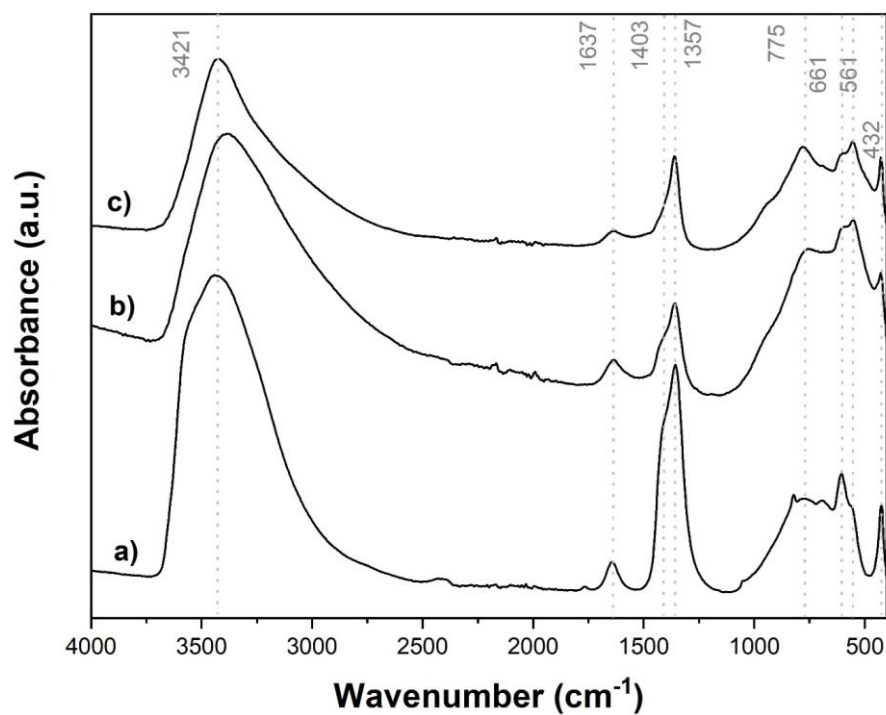


Figure S3. FTIR spectra of compounds obtained by impregnation method a) [Zn₂Al-NO₃] precursor
b) [ZnAl/FreshTiO₂]_{dry} and c) [ZnAl/FreshTiO₂]_{wet}

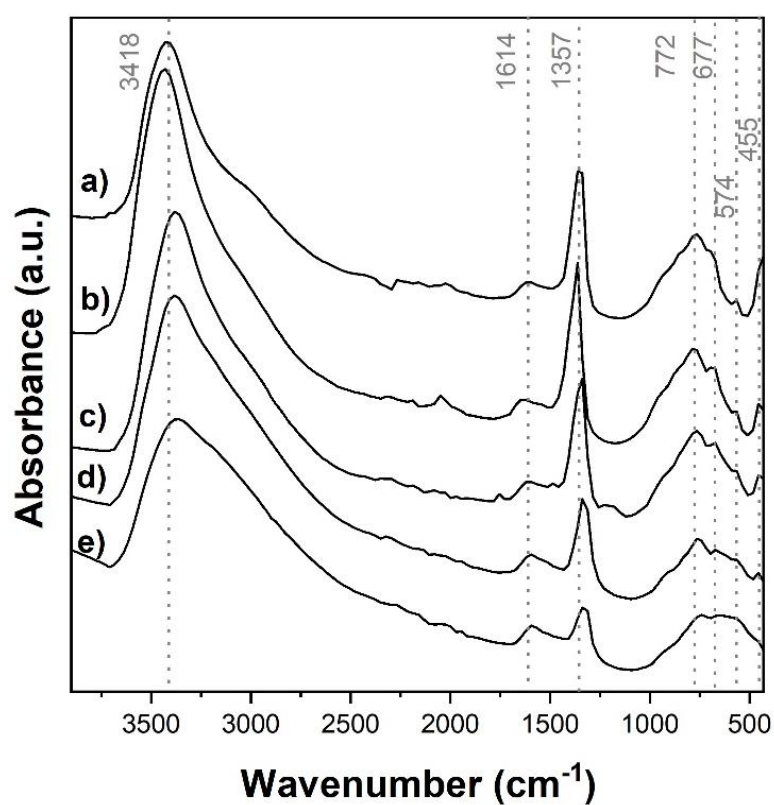


Figure S4. FTIR spectra of [MgAl/TiO₂] nanocomposite obtained by coprecipitation for different MgAl/TiO₂ ratios a) [Mg₂Al-CO₃], b) [MgAl/TiO₂]₈, c) [MgAl/TiO₂]₄, d) [MgAl/TiO₂]₂ and e) [MgAl/TiO₂]_{0.66}

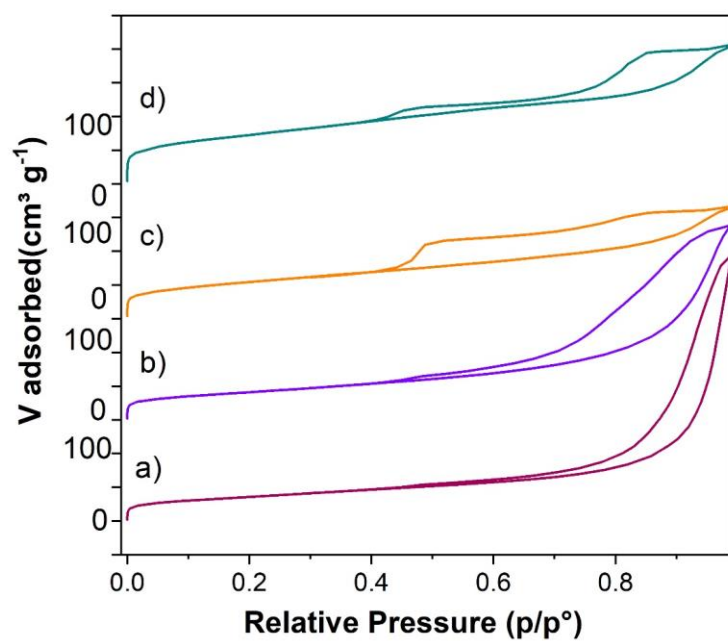


Figure S5. N₂ adsorption-desorption isotherms of a) [MgAl/TiO₂]₈, b) [MgAl/TiO₂]₄, c) [MgAl/TiO₂]₂ and d) [MgAl/TiO₂]_{0.66}

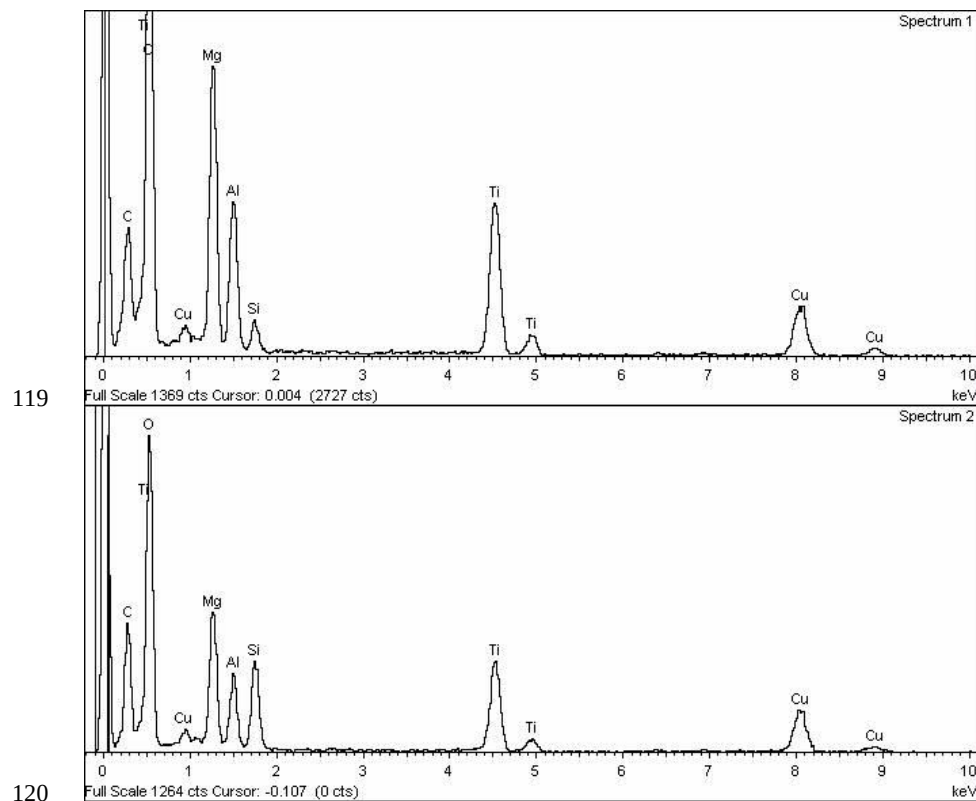


Figure S6. EDX spectra from TEM

References:

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2. Paušová, Š., Krýsa, J., Jirkovský, J., Prevot, V., and Mailhot, G. (2014). Preparation of TiO₂-SiO₂ composite photocatalysts for environmental applications. *J. Chem. Tech. Biotech.* 89, 1129-1135.
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