

Supplementary Materials

Tuning Metal-Support Interactions on Ni/Al₂O₃ Catalysts to Improve Catalytic Activity and Stability for Dry Reforming of Methane

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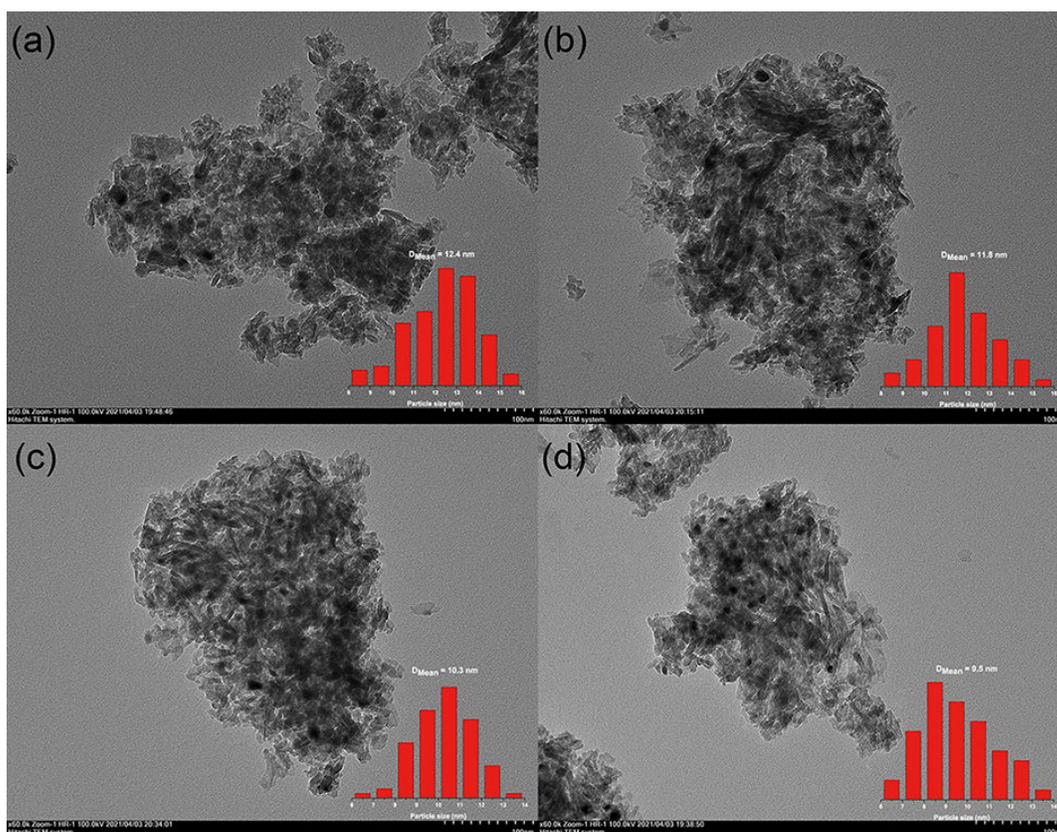


Figure S1. TEM images and Ni particle size distributions of the reduced catalysts: (a) Ni/Al₂O₃-450, (b) Ni/Al₂O₃-550, (c) Ni/Al₂O₃-650, and (d) Ni/Al₂O₃-750.

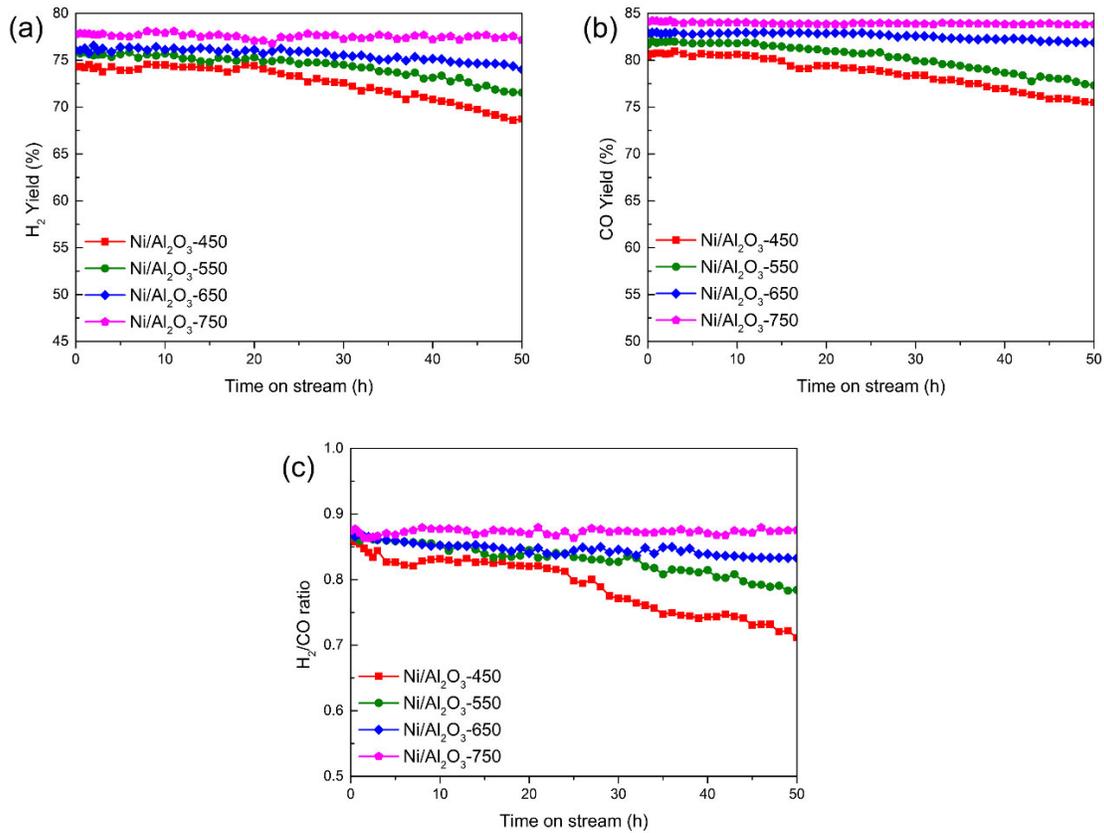


Figure S2. (a) H₂ yield, (b) CO yield and (c) H₂/CO ratio as the function of time for stream of Ni/Al₂O₃-x catalysts (50 h test). Reaction conditions: CH₄:CO₂:Ar =1:1:3, GHSV= 24000 mL g⁻¹ h⁻¹, 700 °C, atmospheric pressure.

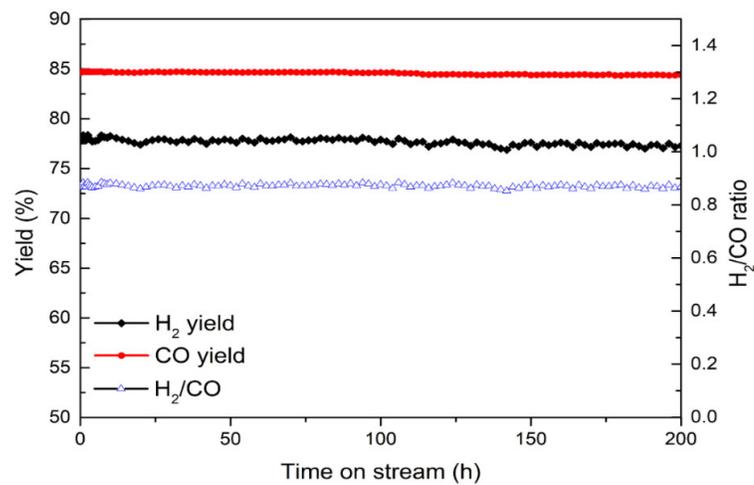


Figure S3. H₂ yield, CO yield and H₂/CO ratio of Ni/Al₂O₃-750 catalyst in the long-term stability (200 h) test for dry reforming of methane reaction. Reaction conditions: CH₄:CO₂:Ar =1:1:3, GHSV= 24000 mL g⁻¹ h⁻¹, 700 °C, atmospheric pressure.

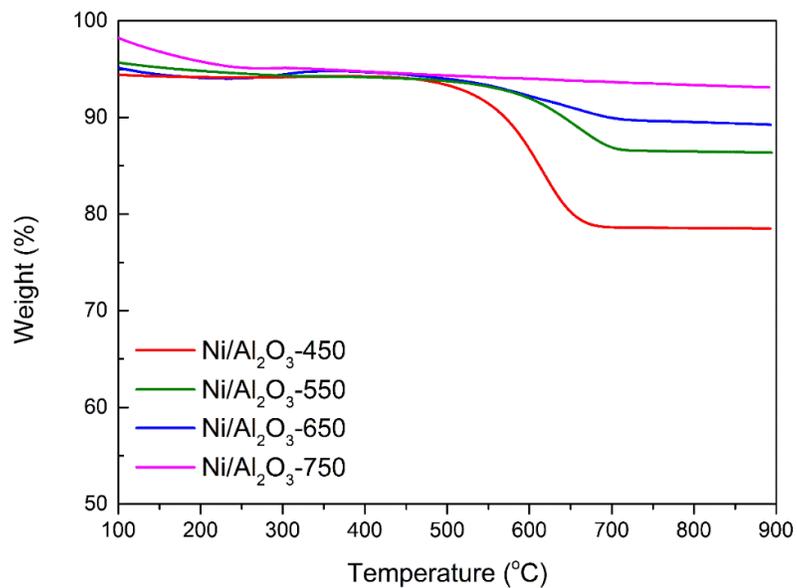


Figure S4. TG analysis of spent Ni/Al₂O₃-x catalysts after the 50 h reaction.