

SUPPORTING INFORMATION

SURFACE-CATALYZED ZINC OXIDE NANORODS AND INTERCONNECTED TETRAPODS AS EFFICIENT METHANE GAS SENSING PLATFORMS

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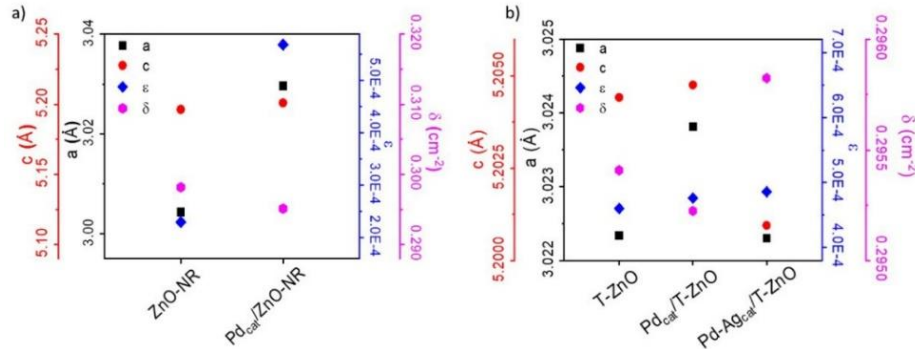


Figure S1. Calculations from the obtained XRD spectra in Fig. 1: (a) Lattice constants, c and a , lattice strain, ϵ , and dislocation density, δ , of ZnO-NR and Pd_{cat}/ZnO-NR; (b) Lattice constants, c and a , lattice strain, ϵ , and dislocation density, δ , of T-ZnO, Pd_{cat}/T-ZnO, and Pd-Ag_{cat}/T-ZnO.

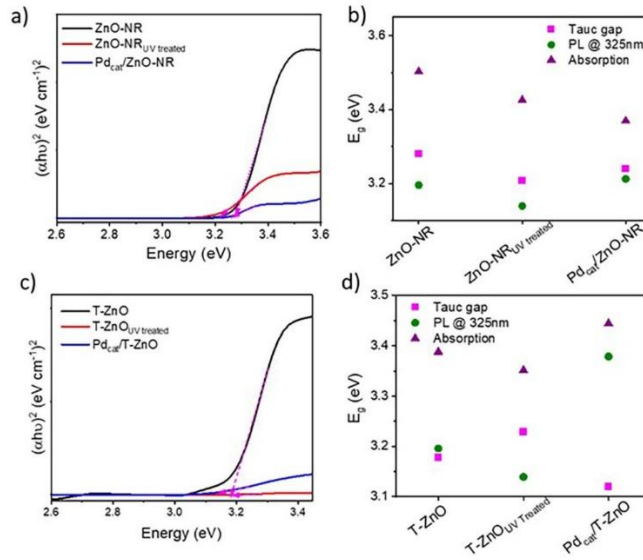


Figure S2. (a) Tauc plots of ZnO-NR, ZnO-NR_{UV treated}, and Pd_{cat}/ZnO-NR; (b) Comparison of bandgap values, E_g , determined from UV-visible spectra, photoluminescence spectra, and Tauc plot of ZnO-NR, ZnO-NR_{UV treated}, and Pd_{cat}/ZnO-NR; (c) Tauc plots of T-ZnO, T-ZnO_{UV treated}, and Pd_{cat}/T-ZnO; (d) Comparison of bandgap values, E_g , determined from UV-visible spectra, photoluminescence spectra, and Tauc plot of T-ZnO, T-ZnO_{UV treated}, and Pd_{cat}/T-ZnO.

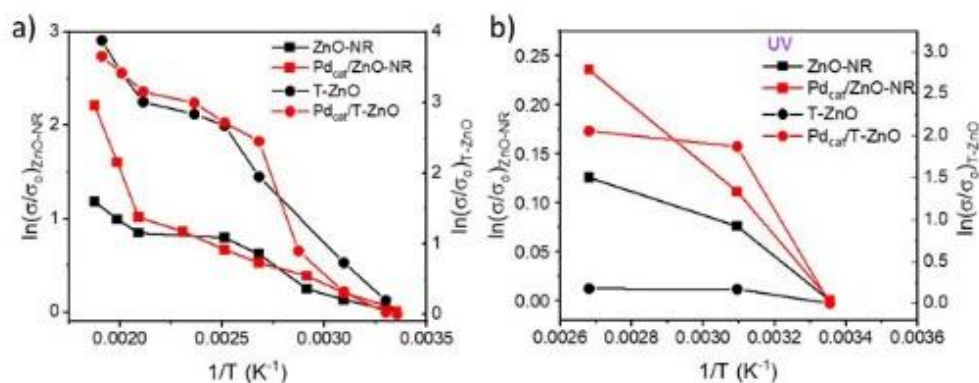


Figure S3. Conductivity measurements as a function of temperature in air for ZnO-NR and T-ZnO with and without Pd catalyst: (a) Conductivity in dark chamber; (b) Conductivity under UV excitation in chamber ($\lambda = 365$ nm).

Table S1. Activation energy determined from conductivity measurements as a function of temperature in air for ZnO-NR and T-ZnO with and without Pd catalyst in the dark and under UV excitation ($\lambda = 365$ nm).

Sample	E_A (eV) UV OFF	E_A (eV) UV ON
ZnO-NR	0.31	0.024
Pd _{cat} /ZnO-NR	0.24	0.06
T-ZnO	0.51	0.10
Pd _{cat} /T-ZnO	0.54	0.09

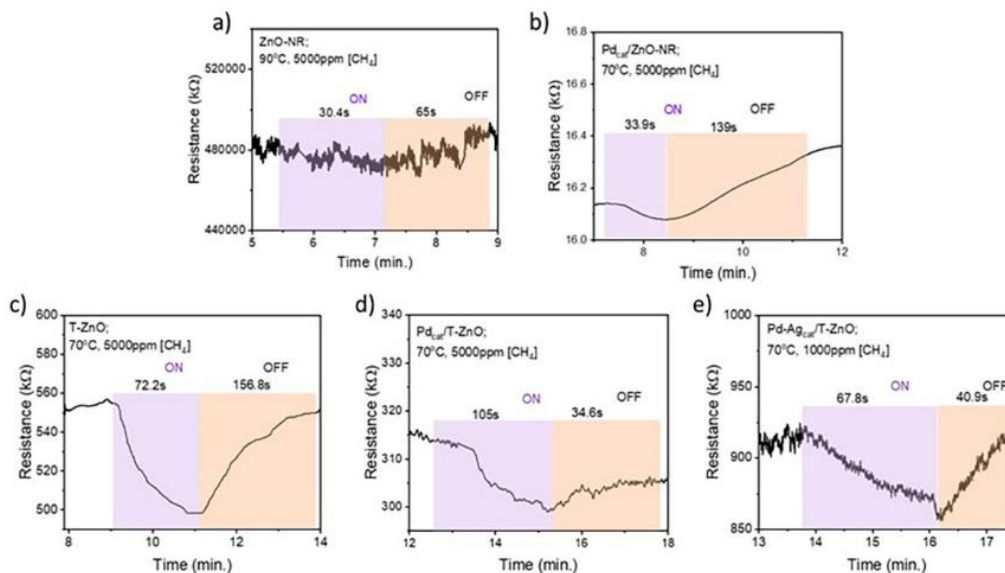


Figure S4. Low temperature dynamic resistance curves for response/recovery times when exposed to CH₄ for (a) ZnO-NR; (b) Pd_{cat}/ZnO-NR; (c) T-ZnO; (d) Pd_{cat}/T-ZnO; (e) Pd-Ag_{cat}/T-ZnO films.

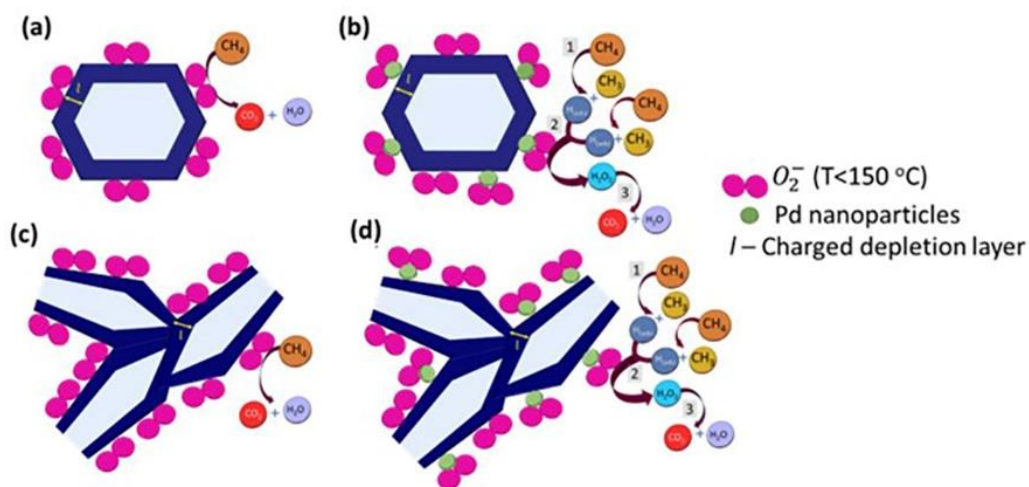


Figure S5. Proposed mechanistic chemical pathways for low temperature (<150 °C) thermal excitation for unmodified and Pd-decorated films: (a) ZnO-NR, b) Pd_{cat}/ZnO-NR, c) T-ZnO, d) Pd_{cat}/T-ZnO sensors.