

# ***Ranunculus sceleratus* as a Model Species to Decrypt the Role of Ethylene in Plant Adaptation to Salinity**

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**Supplementary Table S1.** Results of ANOVA analysis of experimental data.

Experiment	Parameter	DF	F	p
Experiment 1	Petiole length	5	42.6	< 0.0001
	Na <sup>+</sup> concentration on dry mass basis	11	265.4	< 0.0001
	Na <sup>+</sup> concentration on dry water basis	11	38.6	< 0.0001
	K <sup>+</sup> concentration on dry mass basis	11	34.0	< 0.0001
	K <sup>+</sup> concentration on dry water basis	11	3.1	0.011
Calibration of the experimental system	Petiole length (silica xerogel)	4	5.0	0.004
	Petiole length (ethylene)	6	57.4	< 0.0001
	Petiole water content (ethylene)	6	68.8	< 0.0001
Experiment 2	Petiole length	9	26.0	< 0.0001
	Petiole dry mass	9	6.1	< 0.0001
	Blade dry mass	9	5.8	< 0.0001
	Petiole water content	9	17.8	< 0.0001
	Blade water content	9	10.8	< 0.0001
	Stem water content	9	122.7	< 0.0001
	Na <sup>+</sup> concentration in petiole	9	162.6	< 0.0001
	Na <sup>+</sup> concentration in blade	9	96.2	< 0.0001
	Na <sup>+</sup> concentration in stem	9	28.7	< 0.0001
	K <sup>+</sup> concentration in petiole	9	53.7	< 0.0001
	K <sup>+</sup> concentration in blade	9	16.7	< 0.0001
	K <sup>+</sup> concentration in stem	9	125.3	< 0.0001
	Electrical conductivity in petiole	9	33.3	< 0.0001
	Electrical conductivity in blade	9	40.4	< 0.0001
	Electrical conductivity in stem	9	47.2	< 0.0001

**Supplemmentary Figure S1.** Characteristic morphology of *Ranunculus sceleratus* plants cultivated at high humidity (A) and normal humidity (B). Plants were cultivated in closed 48 L (high humidity) and 24 L (normal humidity, with desiccation) containers for 6 days. Containers were ventilated for 1 h and new C<sub>2</sub>H<sub>4</sub> and 1-methylcyclopropene (MCP) releasing solutions were replaced every day.



A, at high humidity. From left to right: control, NaCl, NaCl + C<sub>2</sub>H<sub>4</sub>, NaCl + MCP, NaCl + C<sub>2</sub>H<sub>4</sub> + MCP.



B, at normal humidity. From left to right: control, NaCl, NaCl + C<sub>2</sub>H<sub>4</sub>, NaCl + MCP, NaCl + C<sub>2</sub>H<sub>4</sub> + MCP.