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Effect of TiN/C Microstructure Composite Layer on the Adhesion of FDLC Film onto Silicon Substrate

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Abstract: Deposition techniques of direct current and radio-frequency magnetron sputtering were used to separately prepare TiN/C microstructural composite layer and fluorinated diamond-like carbon (FDLC) film on monocrystalline silicon. The aim was to investigate the effects of microstructural composite layers on the adhesion property of FDLC film. The results indicated that the TiN/C microstructural composite layer can distinguish from the substrate and the FDLC film. After adding the composite layers, the film–substrate binding force significantly increased. When the composite film were prepared at a partial pressure of 0.25, the binding force reached 30.5 N, which was greater than the value of 22.6 N for the sample without composite layers. TiN/C in the composite layers intensified the integration with silicon substrate because the C–C bond acted as a bridge linking the FDLC film to the silicon substrate. Furthermore, the bulges on the surface of the composite layers strengthened the physical bonding of the film with silicon substrates.

Keywords: FDLC film; magnetron sputtering; TiN/C composite layers; adhesion

1. Introduction

Diamond-like carbon (DLC) film is characterized by high hardness, low friction coefficient, good optical transparency, excellent electrical insulating property, and great corrosion resistance. Hence, DLC film is widely applied in mechanical engineering, electronic engineering, military equipment, aerospace, and biomedicine [1–5]. Fluorinated diamond-like carbon (FDLC) film is modified materials that are based on the DLC film. FDLC film possess not only the excellent properties of DLC film but also strong absorption of ultraviolet light, low dielectric constant, small friction coefficient, and biocompatibility [6–11]. Accordingly, FDLC film has been a research focus in material science. However, the unsatisfactory binding force between the film and substrate hinders its industrialization.

To solve the low adhesive force of the DLC film and substrate (film–substrate), a number of studies have been conducted to actively investigate various ways to increase the film–substrate binding force. Some methods that are mentioned in these studies include the following: (1) insertion of a transition layer such as SiC layer [12–14], chromium nitride film [15,16], and WC–Co layer [17] between the film and substrate (most commonly used method); (2) addition of a buffer layer on the surface finish of the substrate (mostly monometal layers) [18,19]; (3) doping of a third element in the film (or implantation of an ion, such as Si [20,21], N [22–24], Ti [25], and Al [21]) to improve the film–substrate binding; and, (4) combination of a buffer layer and ion implantation to increase the film–substrate binding force [9,26]. These techniques decrease the unconformability of the film–substrate boundaries or the stress that is caused by different thermal expansion factors, thereby greatly improving the film–substrate binding force. However, some specific practical requirements are

still not met, including the requirements for use as bioelectrode coatings [27], as protective coatings for automobile components [28,29], and as low-k materials in ultra-large-scale integrated circuit (ULSI) chips [30]. Hence, other methods are continually being explored. Nevertheless, reports on the binding force of FDLC film and substrate are few. Some researchers have deposited a thin SiC layer onto steel substrate before depositing FDLC film. The resulting improvement in the binding force of FDLC film and substrate is certain, but is still unsatisfactory [31,32].

In this work, we increased the film–substrate binding force through direct-current (DC) magnetron sputtering by adding microstructural composite layers between film and substrate. The microstructural layer of TiN and carbon (TiN/C) was initially deposited onto substrate, followed by FDLC film. Afterwards, the effects of composite layers on the film–substrate bonding structure were analyzed to determine the factors that were influencing the adhesion between film and substrate.

2. Materials and Methods

TiN/C and FDLC film were developed using a JGP-450a radio frequency magnetron sputtering (RFMS) system from Sky Technology Development Co., Ltd. Chinese Academy of Sciences (Shenyang, China). The substrate, which had monocrystalline silicon (100) indices, was ultrasonically cleaned for 10 min by immersing in acetone, alcohol, and deionized water to remove the contamination on the surface. Afterwards, the substrate was dried in an oven and placed in a vacuum chamber to prevent dust contamination. During each experiment, the background pressure of the vacuum was 1.5×10^{-3} Pa. Before deposition, the target material was bombarded on the surface at 100 W of power for 10 min in an Ar environment, and the surface was cleaned further. During deposition, the substrate was not heated.

The TiN/C microstructural composite layer was prepared by DC magnetron sputtering. The target material was highly pure Ti (99.99%). DC sputtering power remained at 20 W. The sputtering gas was a mixture of CH₄, Ar, and N₂. The total flow was 50 sccm, and the Ar flow rate was fixed at 10 sccm. The partial pressure of the mixed gas R ($R = [\text{CH}_4]/([\text{CH}_4] + [\text{N}_2] + [\text{Ar}])$) was 25%, 50%, and 75%, respectively, and the corresponding film samples were marked as TiN/C25%, TiN/C50%, and TiN/C75%. The deposition gas pressure was maintained at 1.0 Pa. The use of CH₄ during gas preparation induced the formation of C–C bonds in the prepared composite layer; these bonds intensified the bonding of the composite layers and FDLC film. The deposition time was 3.0 min.

The FDLC film was prepared using RFMS. Highly pure carbon (99.99%) was the target material, and CH₄ and CF₄ were the source gases, at a flow of 40 sccm. Ar was the working gas at a flow of 10 sccm. The deposition gas pressure was 1.0 Pa. RF power was 150 W. The deposition time was 30 min. The FDLC film samples that were prepared on Si substrate, TiN/C25%, TiN/C50%, and TiN/C75%, were marked as FDLC, TiN/C25%-FDLC, TiN/C50%-FDLC, and TiN/C75%-FDLC, respectively.

The chemical composition of the film surface was analyzed using an X-ray photoelectron spectroscopy (XPS; ESCALAB 250Xi model, Boston, MA, USA, using Mg K α X-ray anode operating at 10 kV and 15 mA). An X-ray diffractometer (XRD; Lambda 750 model, Waltham, MA, USA, using Cu K α radiation, and X-ray optical wavelength = 0.15406 nm) was used to test the chemical bond structure of the microstructural composite layers. Scanning electron microscopy (SEM; SU5000 model, Tokyo, Japan) and atomic force microscopy (AFM; Bruker Dimension Icon model, Lyrique, MA, USA) were used to observe the surface morphology and cross-sections of the film. A micro-scratch tester (WS-2005 model, Lanzhou, China) was used to examine the wearing features of film friction on silicon substrate (diamond indenter = 100 μm radius of curvature; 120° taper angle; and, linear step loading), and then the binding force between film and substrate was analyzed in detail. The morphology of the microstructure composite layer after being subjected to tribological test was observed with an optical metallographic microscope (Olympus pmg3 model, Tokyo, Japan).

3. Results and Discussion

3.1. Test Results and Analysis of the Sample Structures

The surface morphology of the samples is shown in the SEM images in Figure 1a–d, and the cross-section of the TiN/C composite layer and FDLC film prepared at different R values are shown in Figure 1e,f. Evidently, R and the surface roughness of TiN/C composite layer increased. Although the massif-like bulge on the surface is distinct, the film is still compact. This bulge may be due to the matrix's impurities on the substrate or preferentially grown grains on the film, as shown in Figure 1a. The SEM surface morphology image of FDLC film showed that the prepared samples had a smooth compact surface and uniform size without pinholes, and only a few flaws, as shown in Figure 1d. Figure 2 shows the surface morphology AFM images of the TiN/C composite layer prepared at $R = 25\%$ and 75% . The root-mean-square (RMS) surface-roughness values of the composite layer prepared at $R = 25\%$ and 75% were 1.302 and 3.379 nm, respectively. In our experimental conditions, a smaller RMS was obtained at a smaller R .

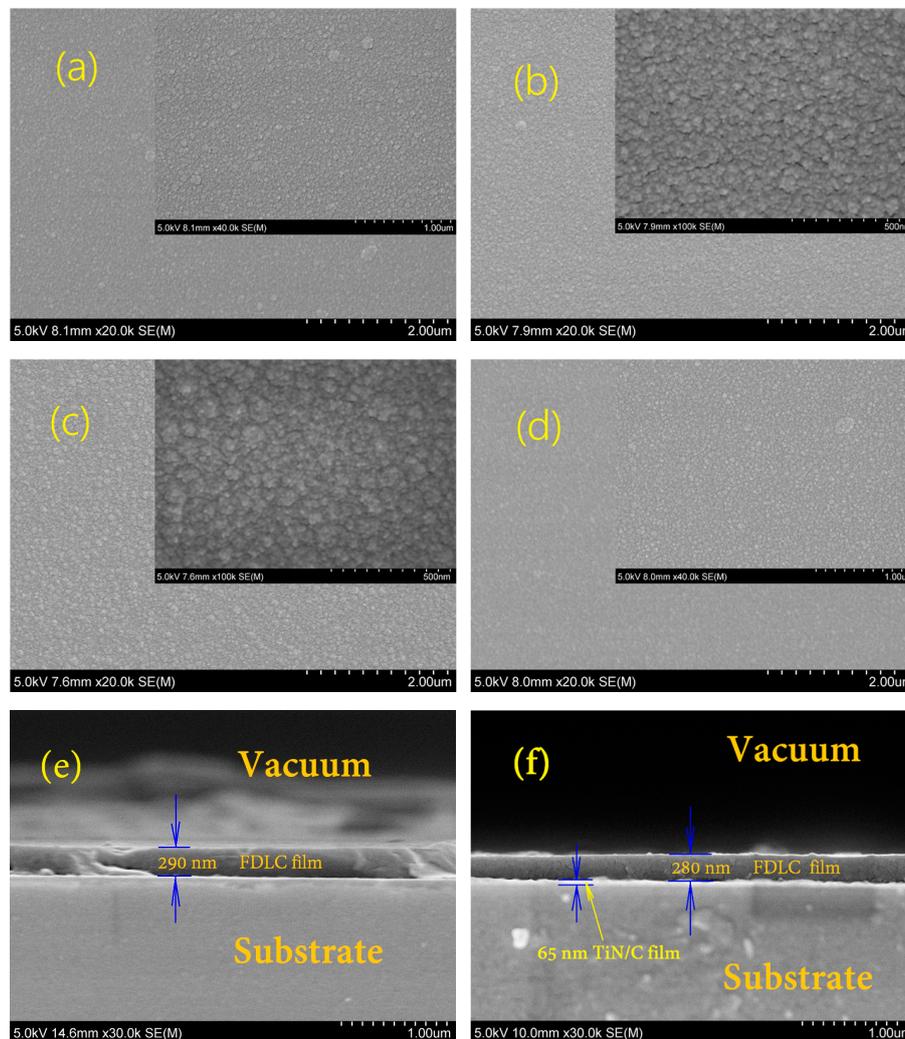


Figure 1. SEM image of the surface morphology and cross-sections of the TiN/C microstructural composite layer and fluorinated diamond-like carbon (FDLC) film: (a) TiN/C25%; (b) TiN/C50%; (c) TiN/C75%; (d) FDLC is 20,000 times surface morphology of the film, the illustration is 40,000 times surface morphology; and (e) FDLC and (f) TiN/C25%-FDLC is the cross-section of the film.

Figure 3 shows the XRD spectrum of the TiN/C composite layer. Two significant diffraction peaks (111) and (200) corresponded to TiN (PDF 38-1420), whereas the weaker diffraction peak (002) corresponded to Ti (PDF 01-1197) [33,34]. This result indicated that a small R induced minimal changes in the diffraction peak of TiN (200), but $R = 75\%$ caused a significantly intensified and broadened peak. Conversely, TiN (111) and Ti (002) remained almost unchanged. No diffraction peak of the combination of Ti and C in XRD spectrum was found. This finding demonstrated that the major element of the composite film was TiN under the present preparation conditions and parameters.

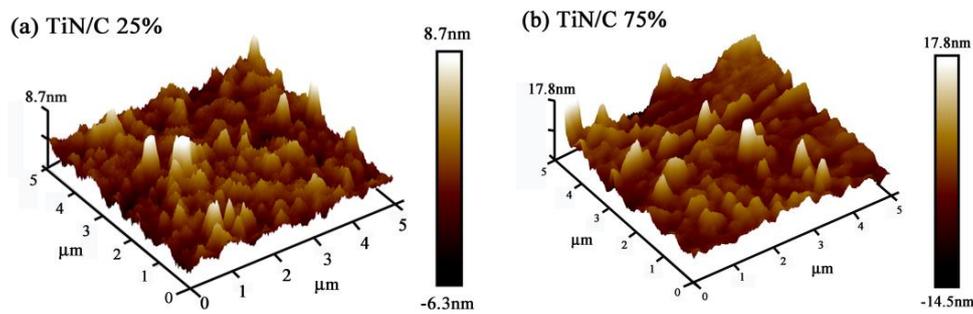


Figure 2. Atomic force microscopy (AFM) three-dimensional image ($5.0 \mu\text{m} \times 5.0 \mu\text{m} \times Z$) of the surface morphology of the TiN/C microstructural composite layer: (a) TiN/C25%; and (b) TiN/C75%.

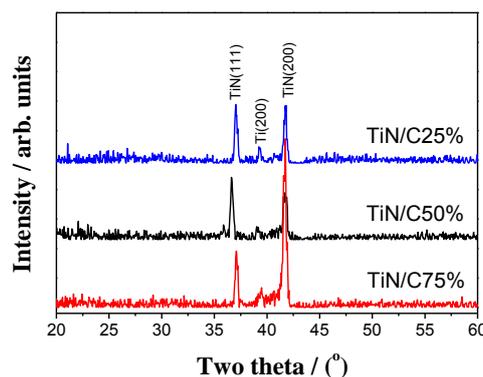


Figure 3. X-ray diffractometer (XRD) spectrum of the TiN/C microstructural composite layer prepared at different R .

The lattice constant of TiN crystalline is calculated using Bragg's Equation (1) and cubic system interplanar distance Equation (2) [35]:

$$\lambda = 2d_{hkl} \sin \theta \quad (1)$$

$$a = d_{hkl} \sqrt{h^2 + k^2 + l^2} \quad (2)$$

where a is the lattice constant; λ is the X-ray wavelength; θ is the diffraction angle; h , k , and l refer to crystal indices; and, d_{hkl} is the interplanar distance of the crystal plane (hkl). The lattice constant of TiN under $R = 25\%$, 50% , and 75% is 0.4205 nm , 0.4247 nm , and 0.4195 nm , respectively. When compared with the diamond lattice constant of similar face-centered cubic structure ($\sim 0.3570 \text{ nm}$) and the intergranular lattice constant of graphite ($\sim 0.340 \text{ nm}$), the lattice constant of TiN is slightly greater. When compared with monocrystalline silicon (0.5431 nm), the difference is less; hence, they match comparatively.

The XPS of the FDLC film and composite layers are shown in Figure 4. In the XPS of the TiN/C composite layer, an evident substrate $\text{Si}2p$ peak appears. At the same time, one strong O peak was observed, and this peak was caused by the residual oxygen in the vacuum chamber or the samples that were absorbing oxygen from air before being tested. From the XPS results, the F/C ratio on

the surface of TiN/C25%-FDLC, TiN/C75%-FDLC, and FDLC samples is 9.06%, 9.31%, and 9.92%, respectively. This indicated that the composition of FDLC film deposited on the composite layer and silicon substrate has little change.

Through the method of dividing overlapping peaks, Gaussian fitting is conducted on peak C1s of the XPS of FDLC film, as shown in Figure 5. According to the reference [36–39], peak C1s can be divided into three peaks. The two peak position at about 283.7 and 284.4 eV corresponds to the Csp² and Csp³ hybrid-bonding structures, respectively, which indicate that graphite and diamond phases existed in the composite layer or FDLC film. The peak positions at 285.5 eV corresponded to C–O bond. The peak position of the sample moved to the low energy direction, which may be caused by different preparation conditions or by the calibration of the instrument wrong. A comparison of Figure 5a,b demonstrated that the ratio sp²/sp³ in the composite layer increased with an increased R, whereas the proportion of aromatic nucleus structure increased in the film, and that of chained (olefin) structures decreased [40]. When compared with the peak C1s in Figure 4c,d, the difference was minimal and was probably caused by the carbon in the FDLC film, but obvious changes were observed in Figure 4c,d when compared with Figure 4b. Meanwhile, a comparison of Figure 5a,c,d revealed little differences among the fitting peaks of the three, indicating that the bonding structure that was formed by carbon was almost the same. This finding provided a basis for the formation of a firmer bond between carbon in FDLC film and that in TiN/C microstructure composite layer.

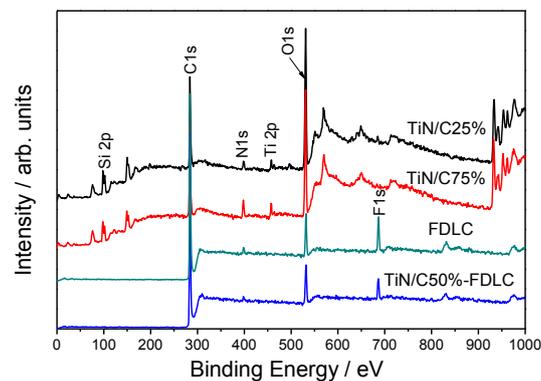


Figure 4. X-ray photoelectron spectroscopy (XPS) spectrum of the TiN/C25% and TiN/C75% microstructural composite layer, TiN/C25%-FDLC film, and FDLC film.

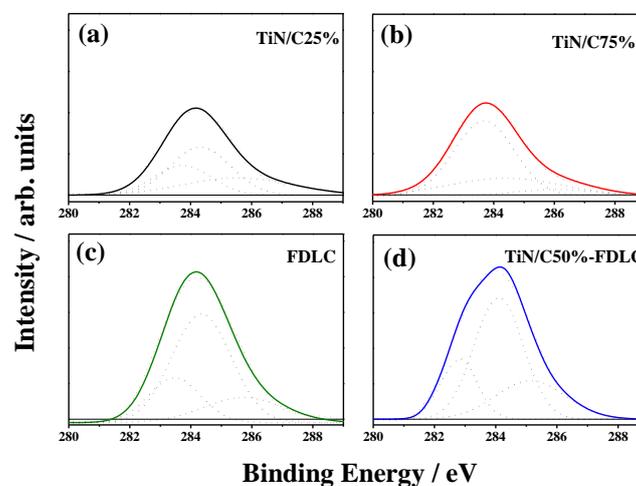


Figure 5. Gaussian fitting of XPS C1s peak of the TiN/C microstructural composite layer and FDLC film. (a) TiN/C25%; (b) TiN/C75%; (c) FDLC; and (d) TiN/C25%-FDLC.

Figure 6a shows the Gaussian fitting of peak Ti2p of the TiN/C composite layer, which can be divided into two doublets. The peak positions at 455.7 and 461.5 eV corresponded to Ti2p_{3/2} and Ti2p_{1/2} of TiN, whereas the peak positions at 457.6 and 464.2 eV corresponded to Ti2p_{3/2} and Ti2p_{1/2} of TiO [41,42]. N1s can be divided into two peaks, i.e., at 397.9 and 399.3 eV, which corresponded to TiN and N-oxide [41], as shown in Figure 6b. A comparison of Figure 6a,b indicated that an increased R led to decreased Ti–O bonding in the composite layer, whereas N–O increased bonding and slightly increased the Ti–N content.

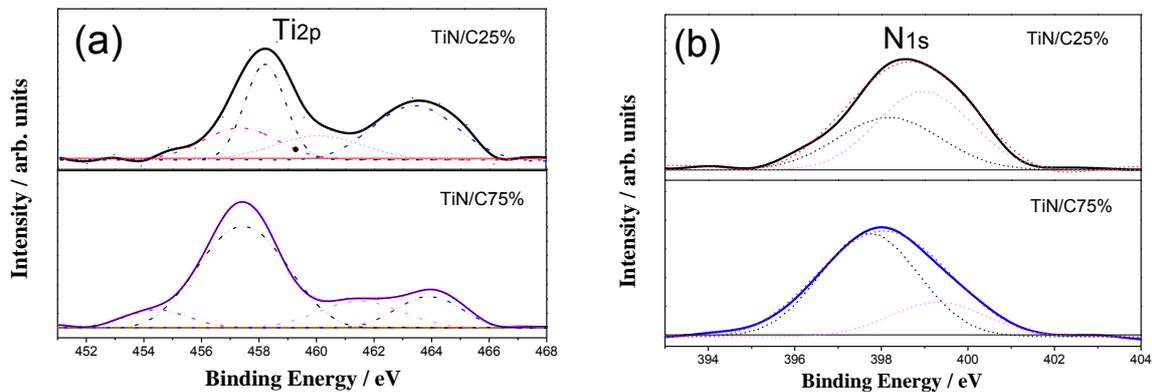


Figure 6. (a) Gaussian fitting of peak Ti2p of TiN; (b) Gaussian fitting of peak N1s of TiN.

3.2. Results and Analysis of the Film–Substrate Binding Force

During the scratch tests, completely identifying the critical load of the film was difficult by simply examining the changes in acoustic emission signals because the internal cohesive failures (flaws and large particles) of the film cause abnormal acoustic emission signals. Therefore, the acoustic emission signal and scratch morphology were combined to eliminate interference signals, and thus estimate the film–substrate binding force. Figure 7 shows the micro-scratch test results of FDLC film before and after the insertion of the composite layer, and Figure 8 shows the metallographic microscope images of the failure zones of the film in the scratch-adhesion tests. From Figure 7a, it can be seen that the acoustic signal has changed at about 22.6 N, and that the scratches on the film are damaged. Before inserting the composite layer between the film and substrate, the binding force of the film was small. After the insertion, the binding force is ~30 N. At $R = 25\%$, the force reached 30.5 N, indicating that the composite layer significantly increased the adhesive force of the film. Obviously, the adhesive force in Figure 7c,d only slightly change. Only the sound signal has a greater change.

The remarkable increase of the film–substrate binding force after inserting TiN/C composite layer was possibly because TiN and C atom inlaid with each other in the microstructural composite layer. On one interface of the composite layer (bottom surface), the lattice constant of TiN was closer to that of monocrystalline silicon than that of diamond. Matching lattices benefit the mutual binding of crystalline and intensify the binding of the composite layer and substrate. At the other interface of the composite layer (top surface), C on the surface of composite layer bonded with C in FDLC film. This bond strengthens the bonding of the composite layer and film layer. In addition, the bulges on the surface of the composite layer become embedded into FDLC film like rivets after the deposition of the FDLC film. Consequently, the contact area between the composite layer and FDLC film increased, which increased the possibilities of C–C bonding, and the physical bonding of the composite layer and FDLC film intensified. Accordingly, after the insertion of TiN/C composite layer, the film–substrate binding force evidently increased.

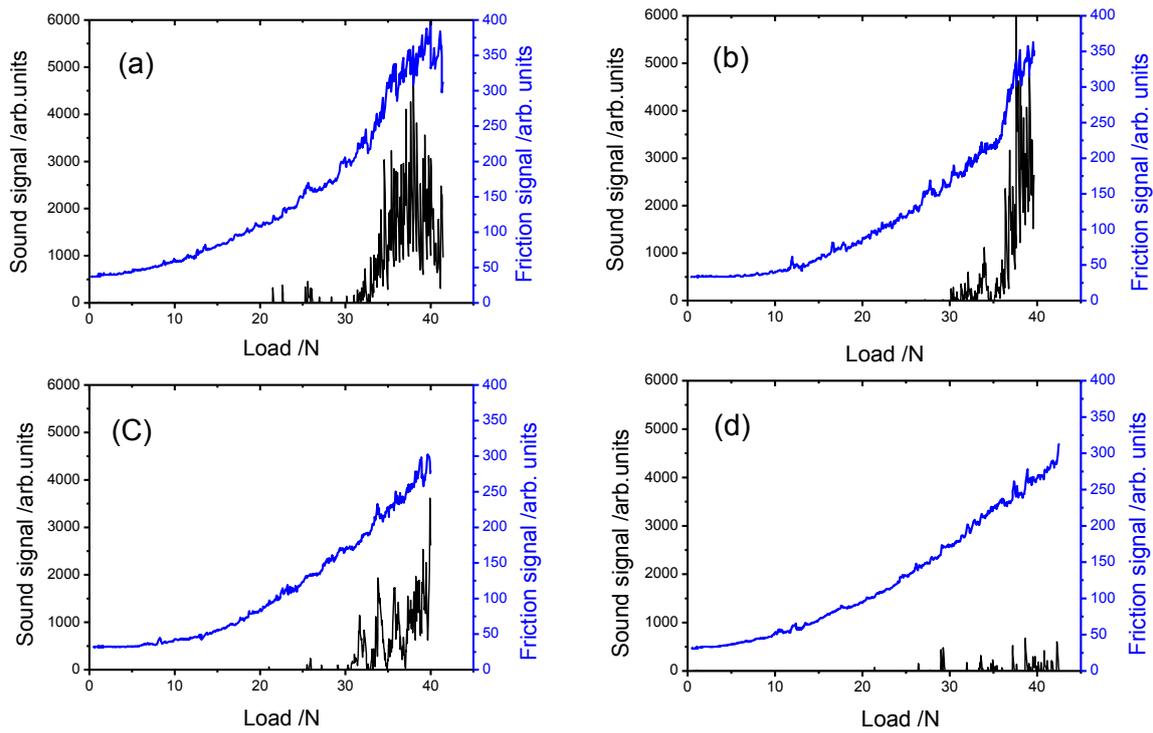


Figure 7. Results of the film–substrate binding force of the FDLC film prepared on different microstructural composite layers: (a) FDLC; (b) TiN/C25%-FDLC; (c) TiN/C50%-FDLC; and (d) TiN/C75%-FDLC.

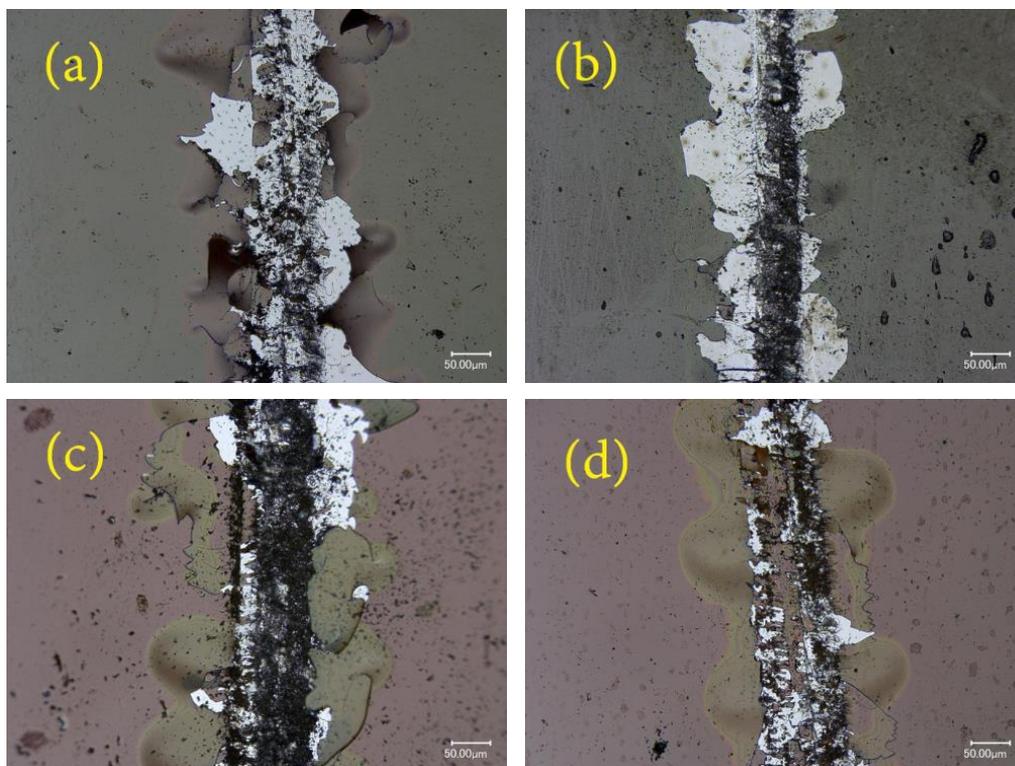


Figure 8. Metallographic microscope images of the failure zones of films in scratch-adhesion test: (a) FDLC; (b) TiN/C25%-FDLC; (c) TiN/C50%-FDLC; and (d) TiN/C75%-FDLC.

The film–substrate binding force changed with R possibly variations in R consequently induced changes in the TiN lattice constant and sp³C hybrid structure in the composite layer. The intensified bonding and resulting structure increased the film–substrate binding force. Meanwhile, at $R = 75\%$, the XRD diffraction peak broadened, grain size decreased, which resulted in a decreased number of pores (or increased film density) in the structure of the composite layer. This phenomenon further led to increased hardness [43] and a subsequent increase in film brittleness. Therefore, when considering that acoustic signals are discontinuous discrete signals, the film–substrate binding force slightly declined when this signal was relatively weak.

4. Conclusions

The TiN/C microstructural composite layer on monocrystalline silicon substrate is sputtered before sputtering FDLC film. This method did not change the primarily diamond-like structure of FDLC film. The film–substrate binding force significantly increased from 22.6 N to 30 N. R influenced the surface roughness, lattice constant, and particle size of TiN/C composite layer to some degree. A certain preparation condition can improve the bonding structure of the composite layer and consequently optimize the ultimate film–substrate binding force. The increase of film–substrate binding force was largely due to the fact that the inlaid TiN and C–C bonds of the composite layer are, respectively, the bonds that are linked the silicon substrate and FDLC film. These bonds are the key to augmenting the film–substrate binding force. The massif-like bulges on the surface of the composite layer enabled the formation of more C–C structures on the interface of the composite layer and FDLC film. The bulges also functioned as rivets, which increased the film–substrate chemical bonding and physical binding force—which is another reason for the increase of binding force.

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Author Contributions: Aihua Jiang conceived and designed the experiments; Chenyang Gong and Meng Qi performed the experiments; Jiafeng Ma, Ming Li and Zhiyong Wang analyzed the data; Jianrong Xiao wrote the paper; all authors participated and discussed this work.

Conflicts of Interest: The authors declare no conflict of interest.

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