

Article

## Microstructural Evolution of (Ti,W,Cr)B<sub>2</sub> Coatings Deposited on Steel Substrates during Annealing

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**Abstract:** The topic of the present experiments are transition metal diboride coatings of composition (Ti<sub>0.49</sub>W<sub>0.51</sub>)B<sub>2</sub> and (Ti<sub>0.44</sub>W<sub>0.30</sub>Cr<sub>0.26</sub>)B<sub>2</sub>. The coatings were deposited on steel substrates using dc magnetron sputtering. We investigated how annealing in argon at elevated temperatures modifies microstructure. The as-deposited films are amorphous. Annealing between 700 and 1100 °C results in the formation of nano-crystalline precipitates with average grain diameters of about 10–50 nm. A TiC phase (Fm-3m;  $a \approx 4.3$  Å) is observed as the dominating precipitate phase. In addition, small amounts (10%–20%) of a Cr<sub>23</sub>C<sub>6</sub> phase (Fm-3m;  $a \approx 10.6$  Å) are observed. In contrast to literature data on the same coatings deposited on silicon substrates, the formation of boride precipitate phases is strongly suppressed here. From investigations with X-ray diffractometry, electron microscopy and secondary ion mass spectrometry we conclude that the nanostructure of the coatings is formed by reactive phase formation of the boride coating with the carbon containing steel substrate.

**Keywords:** transition metal boride coatings; phase formation; magnetron sputtering

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## 1. Introduction

Titanium diboride ( $\text{TiB}_2$ ) ceramics exhibit various attractive properties which are interesting for technical applications: a high melting point ( $\sim 3225\text{ }^\circ\text{C}$ ), high thermal conductivity ( $37\text{--}122\text{ W}\cdot\text{m}^{-1}\cdot\text{K}^{-1}$  at 300 K), low electric resistivity ( $\sim 0.13\text{ }\mu\Omega\cdot\text{m}$ ), a relatively low density ( $4.5\text{ g/cm}^3$ ), excellent chemical stability, and a high hardness ( $20\text{--}30\text{ GPa}$ ) [1–7]. Materials based on titanium diboride are in use as cutting tools, wear resistant parts, protective coatings, diffusion barriers, and for a reinforcement of carbide ceramics [1–10]. Appropriate deposition techniques like magnetron sputtering, dynamic ion mixing, electron beam evaporation or chemical vapour deposition allow producing  $\text{TiB}_2$  thin films or coatings as hard materials [11–20]. Superhard coatings composed of over-stoichiometric nanocolumnar  $\text{TiB}_2$  and synthesized by DC magnetron sputtering show hardness values of 60 GPa [11]. In order to improve toughness and creep resistance of bulk titanium diboride ceramics, tungsten rich platelets were precipitated in-situ in a supersaturated  $(\text{Ti,W,Cr})\text{B}_2$  solid solution [21–25]. These precipitates lead to crack deflection and branching [26]. Concerning thin films, it was recently shown [27] that it is possible to deposit  $(\text{Ti,W,Cr})\text{B}_2$  coatings with a thickness of about  $1\text{ }\mu\text{m}$  on silicon substrates by DC magnetron sputtering of compound targets. X-ray diffractometry and electron microscopy characterizations show the presence of a columnar amorphous structure after deposition. Annealing the coatings at temperatures up to  $1300\text{ }^\circ\text{C}$  results in the formation of nano-crystalline precipitations composed of  $(\text{Ti,W,Cr})\text{B}_2$  (P6mmm),  $\beta\text{-(W,Ti,Cr)B}$  (Cmcm) and  $\text{W}_2\text{B}_4$  (P6mmc) phases.

The aim of the present paper is to compare the thermally induced microstructural modifications in coatings of composition  $(\text{Ti}_{0.49}\text{W}_{0.51})\text{B}_2$  and  $(\text{Ti}_{0.44}\text{W}_{0.30}\text{Cr}_{0.26})\text{B}_2$  which are deposited on steel substrates to those sputtered on silicon substrates as given in [27]. Heat treatments are applied in order to induce formation of a nano-crystalline structure and to tailor grain diameters. Further, it should be investigated in which way precipitation and phase formation are different for both type of substrates. The ability of these coatings as a candidate for hard coatings, re-enforced by nano-structuring is explored.

## 2. Experimental Section

The coatings were deposited by dc magnetron sputtering using sputter targets of composition  $(\text{Ti}_{0.49}\text{W}_{0.51})\text{B}_{1.97}$  and  $(\text{Ti}_{0.44}\text{W}_{0.30}\text{Cr}_{0.26})\text{B}_{1.95}$  [27]. The chemical composition of the targets was measured by inductively coupled plasma-optical emission spectroscopy. The targets (about 50 mm diameter) were produced at RWTH Aachen by reaction sintering of  $\text{TiB}_2$ ,  $\text{WB}_2$  and  $\text{CrB}_2$  powders (Japan New Metals Co., LTD., Osaka, Japan). The powders, which contain less than 1 wt% of impurities (C, Fe, N, O) were first mixed with isopropanol and were afterwards high-energy milled with WC/Co hard balls in order to improve sinter activity. Reaction sintering was done in argon in a uni-axial graphite hot press in boron nitride coated graphite crucibles at 50 MPa and  $2100\text{ }^\circ\text{C}$  for 8 h. In a second process step, the specimens were homogenized at  $2000\text{ }^\circ\text{C}$  for 8 h at ambient argon pressure and afterwards ground and polished.

About  $1\text{ }\mu\text{m}$  thick films were sputtered on steel substrates (1.2210 115CrV3), containing about 1.2 wt% C, 0.7 wt% Cr, and 0.1 wt% V. After cleaning ultrasonically with ethanol, the substrates were fixed on a rotating sample holder with a target to substrate distance of about 13 cm. The target was tilted by about  $17^\circ$  from the direction perpendicular to the sample surface. All experiments were done in argon gas at 0.67 Pa (base pressure:  $1.3 \times 10^{-3}\text{ Pa}$ ) at a sputtering power between 0.2 and 0.4 kW. The

temperature of the coating during sputtering was detected by a type K thermocouple to be about 110 °C. As demonstrated in [27], the chemical composition of the target is nearly identical to that of the coating, as proven by Rutherford backscattering spectrometry (RBS). Consequently, the chemical composition of the coatings is assumed to be  $(\text{Ti}_{0.49}\text{W}_{0.51})\text{B}_2$  and  $(\text{Ti}_{0.44}\text{W}_{0.30}\text{Cr}_{0.26})\text{B}_2$ . The possible slight under-stoichiometry of boron is neglected.

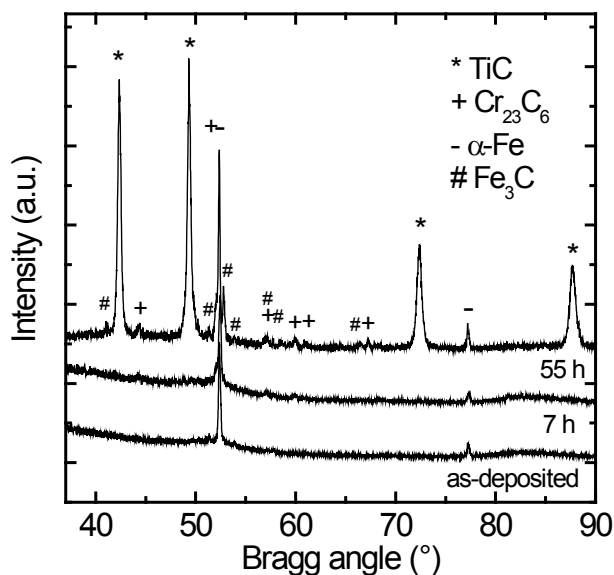
The coatings were isothermally annealed at temperatures between 700 °C and 1100 °C at ambient argon pressure and quenched afterwards. In order to reduce a possible formation of oxide layers during annealing, two samples were placed on each other with the coated surface, were embedded in fine powder of  $\text{TiB}_2$  and were placed in a crucible.

Grazing incidence X-ray diffraction (GI-XRD) was carried out by a Siemens D5000 diffractometer in the asymmetric  $\alpha/2\theta$  modus using  $\text{CoK}\alpha$  radiation (40 kV, 40 mA). Phase analysis was done using the program Powdercell for Windows 2.4 [28]. The cross-sectional morphology of the coatings was examined by a field emission scanning electron microscope (FESEM) of the type FP 2067/32 Helios Nanolab (FEI), operating at 2–5 keV. Element depth profiles were measured by secondary ion mass spectrometry (SIMS) using a CAMECA IMS-3F machine and an O- primary ion beam (15 keV). This analysis gives a plot: intensity of secondary ions vs. sputter time for each element. The sputtered area was about  $250 \mu\text{m} \times 250 \mu\text{m}$ . Depth calibration according to a standard procedure was done by first measuring the resulting SIMS crater depth with a mechanical profilometer (Alphastep 500, KLA-Tencor, Milpitas, CA, USA). Afterwards the sputter time is converted into the actual sputtered depth by assuming that the crater depth corresponds to the maximum sputter time and that a linear relationship exists.

### 3. Results and Discussion

In Figure 1 X-ray diffractograms of the Cr free coating  $(\text{Ti}_{0.49}\text{W}_{0.51})\text{B}_2$  are shown after deposition and after annealing at 700 °C for different times. As obvious, the coatings are X-ray amorphous after deposition (the same is true for the Cr containing coatings). The pattern only shows Bragg peaks which can be attributed to a  $\alpha$ -Fe phase and a  $\text{Fe}_3\text{C}$  phase (low peaks) resulting from the steel substrate. Annealing of the as-deposited coatings at a temperature of 700 °C for some hours does not significantly modify the diffraction pattern as shown in Figure 1. Only long term anneals for several tens of hours leads to the formation of crystalline precipitates. However, no boride phases are found as expected from preliminary work, where coatings of the same composition and with the same sputter parameters were deposited on silicon substrates [27]. Instead, strong reflections of a cubic TiC phase (Fm-3m) [29] with a lattice parameter of  $a = (4.283 \pm 0.005) \text{ \AA}$  occur. Lattice parameters are determined by fitting the X-ray diffractogram by the program Powder Cell for Windows 2.4 [28]. An average grain size is assessed from the XRD pattern (width of the Bragg peak at  $42.4^\circ$ ) annealed for 55 h using the Scherrer equation [30]. A grain size of about 30 nm is found, indicating the formation of nano-crystalline precipitations during annealing. This grain size might be slightly under-estimated because effects of microstrain are neglected. However, for a rough estimate this approach is justified. In addition, small amounts (about 10%) of a cubic  $\text{Cr}_{23}\text{C}_6$  phase (Fm-3m) [29] with a lattice parameter of  $a = 10.57 \text{ \AA}$  are also found. It is very likely that crystallization is not complete and a residual amorphous matrix phase is also present.

**Figure 1.** Grazing incidence X-ray diffraction (GI-XRD) patterns of  $(\text{Ti}_{0.49}\text{W}_{0.51})\text{B}_2$  coatings in the as-deposited state and after annealing at 700 °C for different annealing times. The precipitated phases of type TiC and  $\text{Cr}_{23}\text{C}_6$  are indicated. The phases  $\alpha\text{-Fe}$  and  $\text{Fe}_3\text{C}$  result from the steel substrate.

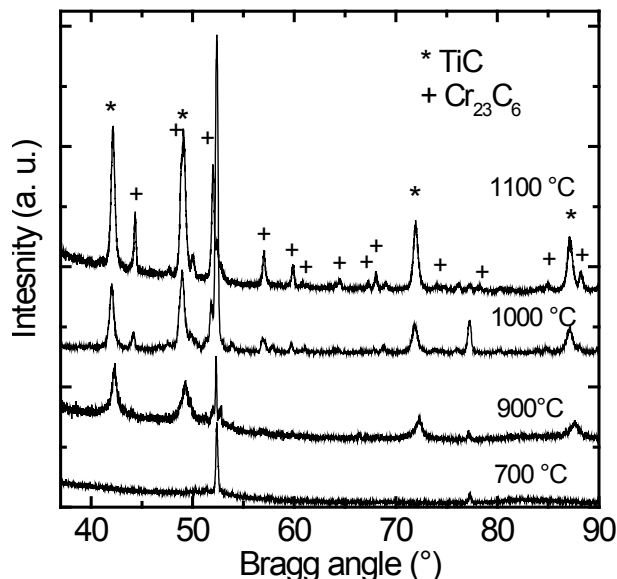


Similar behavior is observed for annealing the as-deposited coatings at higher temperatures of 900 °C, 1000 °C and 1100 °C, respectively. Here, precipitation kinetics is accelerated and the formation of TiC and  $\text{Cr}_{23}\text{C}_6$  phases occurs already for an annealing time of only 0.5 h (shortest annealing time applied). In Figure 2, X-ray patterns for annealing times of 0.5 h are plotted for these temperatures. It can be clearly seen, that first the TiC phase crystallizes at 900 °C and afterwards the  $\text{Cr}_{23}\text{C}_6$  phase forms at 1000 °C and higher temperatures. The amount of the crystallized  $\text{Cr}_{23}\text{C}_6$  phase increases up to 25% at higher annealing temperatures. The formation of crystalline precipitates starts at an about 200 K lower temperature as observed for silicon substrates [27]. Grain sizes of the TiC and  $\text{Cr}_{23}\text{C}_6$  phases obtained from the XRD patterns using the Scherrer formula (see above) are given in Table 1. The grain size of the TiC phase ranges from 20 nm at low temperatures and short annealing times to 35 nm at high temperatures. This indicates that the crystallite size can be tailored by choosing adequate time-temperature annealing profiles. Similar results are found for the  $\text{Cr}_{23}\text{C}_6$  phase (Table 1). Here, the crystallite size varies between 35 and 60 nm. A further peculiarity of the results given in Figure 2 and Table 1 is that no significant amounts of boride compounds of type  $\text{TiB}_2$   $\beta\text{-WB}$  or  $\text{W}_2\text{B}_4$  are found, also at temperatures as high as 1100 °C. This is in strong contrast to the results on silicon substrates [27], where exclusively boride compounds are formed.

In Figure 3 characteristic diffractograms of samples with and without chromium [ $(\text{Ti}_{0.49}\text{W}_{0.51})\text{B}_2$  and  $(\text{Ti}_{0.44}\text{W}_{0.30}\text{Cr}_{0.26})\text{B}_2$ ] are compared. The samples were annealed at 900 °C for 0.5 h (Figure 3a) and at 1100 °C for 0.5 h (Figure 3b), respectively. No significant differences between the patterns of the two different types of samples can be observed concerning type and amount of crystallized phases. Consequently, the precipitation kinetics is very similar for the two coatings. This is again in strong contrast to the findings for the same coatings on silicon substrates [27], where the precipitation kinetics is strongly accelerated for the Cr containing coatings. The assessed crystallite size for the two phases in

$(\text{Ti}_{0.44}\text{W}_{0.30}\text{Cr}_{0.26})\text{B}_2$  is given in Table 2. It varies between 10 and 40 nm for TiC and is about 50 nm for  $\text{Cr}_{23}\text{C}_6$ , respectively.

**Figure 2.** GI-XRD patterns of  $(\text{Ti}_{0.49}\text{W}_{0.51})\text{B}_2$  coatings after annealing for 0.5 h at 900 °C, 1000 °C, and 1100 °C, respectively. The precipitated phases of type TiC and  $\text{Cr}_{23}\text{C}_6$  are indicated. Peaks referring to the steel substrate are unmarked.



**Table 1.** Lattice constants and grain size for  $(\text{Ti}_{0.49}\text{W}_{0.51})\text{B}_2$  coatings determined from XRD measurements.

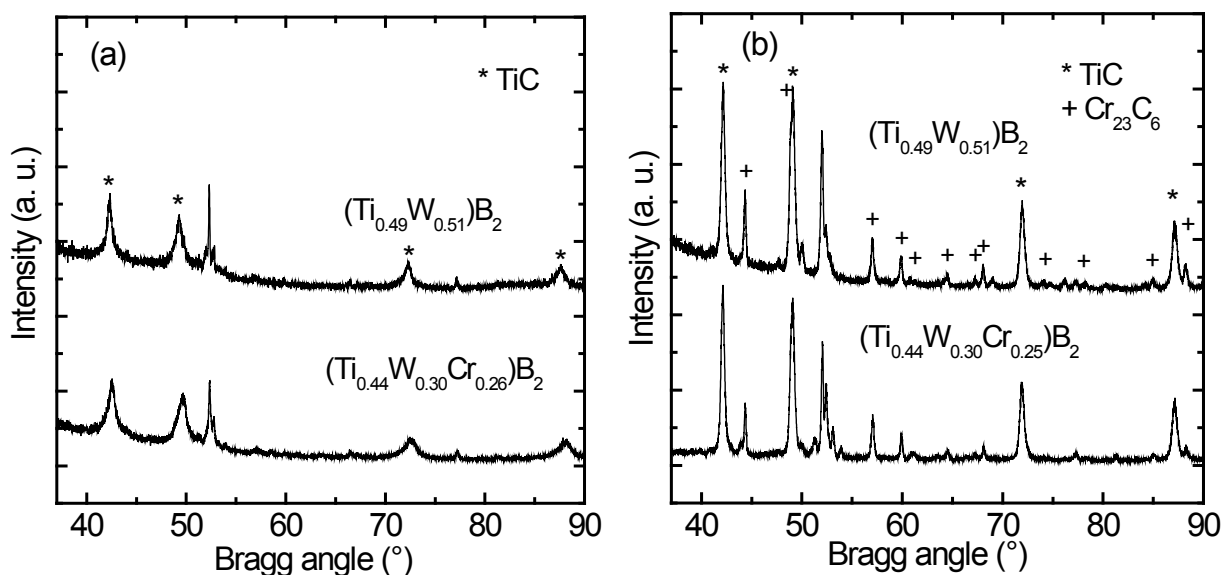
| Conditions         |      | TiC (Fm-3m)                                    |                                        | $\text{Cr}_{23}\text{C}_6$ phase (Fm-3m)      |                                        |
|--------------------|------|------------------------------------------------|----------------------------------------|-----------------------------------------------|----------------------------------------|
| Annealing time (h) |      | Lattice constant (Å)<br>(error: $\pm 0.005$ Å) | Grain size (nm)<br>(error: $\pm 5$ nm) | Lattice constant (Å)<br>(error: $\pm 0.01$ Å) | Grain size (nm)<br>(error: $\pm 5$ nm) |
| 700 °C             | 55   | 4.283                                          | 30                                     | 10.57                                         | –                                      |
| 900 °C             | 0.5  | 4.281                                          | 20                                     | 10.58                                         | –                                      |
|                    | 1    | 4.286                                          | 25                                     | 10.59                                         | –                                      |
|                    | 7.5  | 4.291                                          | 30                                     | 10.58                                         | 55                                     |
|                    | 31.5 | 4.291                                          | 35                                     | 10.57                                         | 55                                     |
| 1000 °C            | 0.5  | 4.301                                          | 25                                     | 10.60                                         | 35                                     |
|                    | 1    | 4.302                                          | 25                                     | 10.60                                         | 45                                     |
|                    | 0.5  | 4.302                                          | 30                                     | 10.59                                         | 55                                     |
| 1100 °C            | 1    | 4.303                                          | 30                                     | 10.59                                         | 50                                     |
|                    | 1.5  | 4.304                                          | 35                                     | 10.60                                         | 60                                     |
|                    | 2    | 4.303                                          | 35                                     | 10.60                                         | 55                                     |
|                    | 4    | 4.301                                          | 35                                     | 10.60                                         | –                                      |

More insight into the precipitation behavior gives experiments with secondary ion mass spectrometry (SIMS) which are shown in Figure 4. Experimentally derived depth profiles (SIMS intensity vs. depth) of B, Ti, Cr, W, C, and Fe species are displayed for  $(\text{Ti}_{0.44}\text{W}_{0.30}\text{Cr}_{0.26})\text{B}_2$  coatings. For the as-deposited sample in Figure 4a an about 1  $\mu\text{m}$  thick film with sharp interfaces is clearly visible. After annealing at 1100 °C for 1 h (Figure 4b) a strong intermixing of the elements takes place across the substrate/coating.

Especially interesting are the strong diffusion of B and Cr from the coating into the substrate and the diffusion of Fe in the reverse way. This result indicates, that crystalline phase formation during annealing is not governed by the precipitation of crystalline boride compounds in an inert amorphous matrix of the coating, as found for silicon substrates [27], but it is controlled by strong reactive diffusion processes between substrate and film. Since during annealing at elevated temperatures preferably carbide phases are formed, there has to be also a strong diffusion flux from the steel substrate (containing 1.2 wt% C) into the boride coating. The fact that the relatively low amount of carbon present in the substrate may lead to the formation of carbide compounds in the coating can be explained with a high atomic mobility of carbon at the temperatures under investigation and the much larger thickness of the substrate (1 mm) compared to the coating (1  $\mu\text{m}$ ). In order to introduce carbon into the coating in the ten percent range, the average carbon concentration in the substrate has to be lowered by only  $10^{-2}$  percent if carbon can quickly diffuse across the whole substrate thickness. According to ref. [31] the carbon diffusivity in steel at 900 °C for a carbon content of about 1.2 wt% is in the order of  $10^{-11}$  m<sup>2</sup>/s. This gives a diffusion length in the 100  $\mu\text{m}$  to mm range, justifying the argumentation given above. Based on these results, in principle the formation of FeO precipitations cannot be excluded due to the diffusion of Fe into the coating. The FeO phase has the same space group (Fm-3m) and similar lattice parameters than the TiC phase. However, its existence is rather unlikely because the coatings were annealed in high purity argon and for the same boride coatings deposited on silicon [27], which were annealed in an identical way, no oxide phases were found.

The fact that SIMS detects a lower amount of C inside the steel substrate than in the film can be explained by a lower ionization probability of C<sup>+</sup> ions in steel than in the boride compound (matrix effect). The high atomic mobility of Cr will also explain the formation of the Cr<sub>23</sub>C<sub>6</sub> phase in Cr free coatings of composition (Ti<sub>0.49</sub>W<sub>0.51</sub>)B<sub>2</sub> by the diffusion of Cr from the steel substrate into the boride coating.

**Figure 3.** GI-XRD patterns of (Ti<sub>0.49</sub>W<sub>0.51</sub>)B<sub>2</sub> and (Ti<sub>0.44</sub>W<sub>0.30</sub>Cr<sub>0.26</sub>)B<sub>2</sub> coatings after annealing for 0.5 h at (a) 900 °C and (b) 1100 °C. The precipitated phases of type TiC and Cr<sub>23</sub>C<sub>6</sub> are indicated. Peaks referring to the steel substrate are unmarked.



**Table 2** Lattice constants and grain size for  $(\text{Ti}_{0.44}\text{W}_{0.30}\text{Cr}_{0.26})\text{B}_2$  coatings determined from XRD measurements.

| Conditions         |      | TiC (Fm-3m)                                    |                                        | $\text{Cr}_{23}\text{C}_6$ phase (Fm-3m)      |                                        |
|--------------------|------|------------------------------------------------|----------------------------------------|-----------------------------------------------|----------------------------------------|
| Annealing time (h) |      | Lattice constant (Å)<br>(error: $\pm 0.005$ Å) | Grain size (nm)<br>(error: $\pm 5$ nm) | Lattice constant (Å)<br>(error: $\pm 0.01$ Å) | Grain size (nm)<br>(error: $\pm 5$ nm) |
| 700 °C             | 55   | 4.257                                          | 25                                     | 10.60                                         | –                                      |
| 900 °C             | 0.5  | 4.271                                          | 10                                     | -                                             | –                                      |
|                    | 1    | 4.276                                          | 15                                     | 10.55                                         | –                                      |
|                    | 7.5  | 4.283                                          | 20                                     | 10.58                                         | 45                                     |
|                    | 31.5 | 4.286                                          | 25                                     | 10.56                                         | 50                                     |
| 1000 °C            | 0.5  | 4.297                                          | 15                                     | 10.60                                         | 50                                     |
|                    | 1    | 4.297                                          | 20                                     | 10.57                                         | 50                                     |
| 1100 °C            | 0.5  | 4.302                                          | 30                                     | 10.58                                         | 50                                     |
|                    | 1    | 4.304                                          | 35                                     | 10.58                                         | 50                                     |
|                    | 1.5  | 4.301                                          | 40                                     | 10.58                                         | 50                                     |
|                    | 2    | 4.306                                          | 40                                     | 10.59                                         | –                                      |
|                    | 4    | 4.301                                          | 40                                     | 10.59                                         | –                                      |

**Figure 4.** Secondary ion mass spectrometry depth profiles of  $(\text{Ti}_{0.44}\text{W}_{0.30}\text{Cr}_{0.26})\text{B}_2$  coatings in (a) the as-deposited state and (b) after annealing for 1 h at 1100 °C.

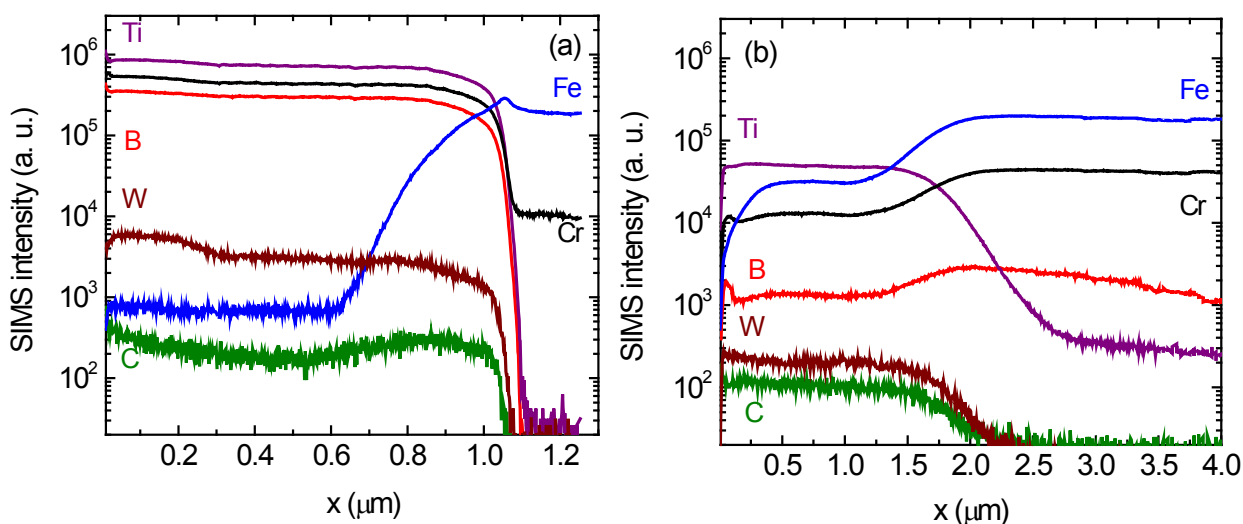
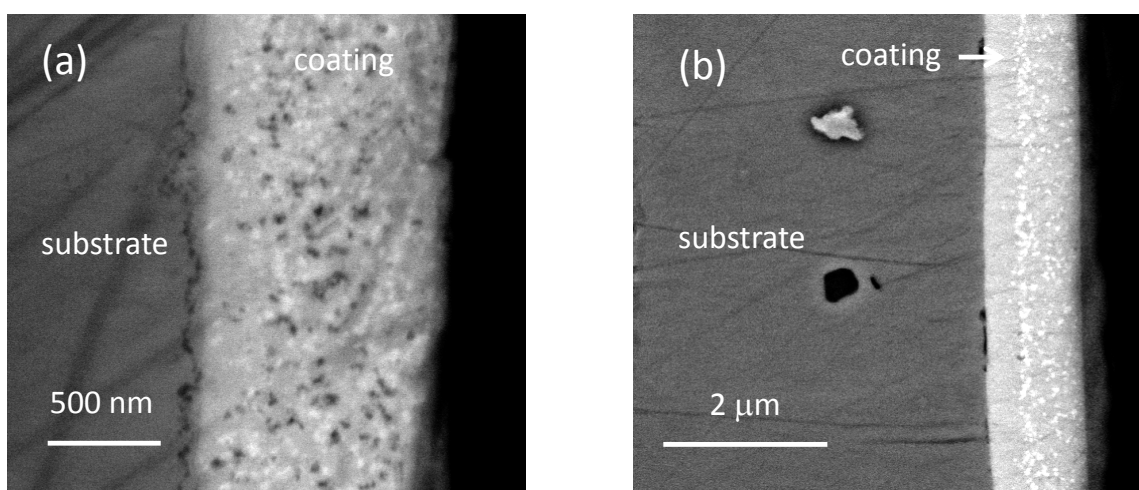


Figure 5 shows characteristic electron microscopy images of the microstructure of  $(\text{Ti}_{0.49}\text{W}_{0.51})\text{B}_2$  and  $(\text{Ti}_{0.44}\text{W}_{0.30}\text{Cr}_{0.26})\text{B}_2$  coatings annealed at 1000 °C in cross-sectional geometry. The actual boride coating and the steel substrate are indicated. The nanosized crystalline precipitations formed during annealing with dimensions of about 10–50 nm can be clearly seen as white spots. This indicates that heavy elements like W are present in the precipitates. Very likely we have solid solutions of  $(\text{Ti,W,Cr})\text{C}$  and  $(\text{Ti,W,Cr})_{23}\text{C}_6$  phases as precipitates, where a mutual substitution of the transition metal is taking place. This is also indicated by the lattice constants of the crystalline phases as derived from the XRD patterns, which are also given in Tables 1 and 2. For the TiC phase values of 4.26 Å (low temperatures) to 4.30 Å (high temperatures) are found which differ considerably from the values of pure TiC of 4.33 Å [29].

Such (Ti,W,Cr)C solid solution compounds in bulk form are described in literature [32,33]. Compounds containing up to 75% W and 30% Cr were reported [26]. According to Ref. [32,34], the lower lattice parameters in comparison to the pure TiC phase indicate that significant amounts of W might be incorporated into the Ti matrix and also an under-stoichiometry  $(\text{Ti,W,Cr})\text{C}_x$  ( $x < 1$ ). Significant differences between the Cr free and Cr containing coatings were not found, indicating that Cr (originally present in the coating) does not play an important role for the mutual substitution process. A similar effect, however less pronounced, is found for the  $\text{Cr}_{23}\text{C}_6$  phase. Lattice parameters of 10.55–10.60 Å are derived, slightly lower than the value of the pure phase of 10.65 Å [29]. Also a partial incorporation of Fe into the nano-precipitates cannot be excluded.

**Figure 5.** FESEM backscattering image in cross-sectional geometry for (a)  $(\text{Ti}_{0.49}\text{W}_{0.51})\text{B}_2$  and (b)  $(\text{Ti}_{0.44}\text{W}_{0.30}\text{Cr}_{0.26})\text{B}_2$  coatings annealed at 1000 °C for 1 h.



As further obvious from Figure 5, nano-sized dark areas are visible. These can tentatively be identified as pores, which result from the observed formation process of crystalline phases and the associated volume changes. Further, it is interesting that the formation of nano-crystals does not take place homogeneously in the coating, but seems to start from surface near regions and is less pronounced close to the interdiffusion zone between substrate and coating. In that range, the local chemical composition seems to be more resistant against precipitation and crystallization.

The present results reveal that the thermally induced phase formation of boride coatings deposited on steel substrates results in nano-crystalline precipitations as also found for silicon substrates [27]. The crystallite size is in the order of 10–50 nm for annealing up to 1100 °C. However, the nano-crystalline precipitates are composed of carbide phases and not of boride phases. This difference can be explained by the fact that for steel substrates the microstructural evolution and formation of crystalline phases is dominated by reactive interdiffusion and not by precipitation in a supersaturated matrix. This also explains the similar precipitations kinetics of different coatings on steel substrates. In this context, the observed suppression of the formation of crystalline boride compounds is attributed to the concurrent formation of carbide phases, which are formed already at lower temperatures. During precipitation of the carbide phases, the amorphous matrix is modified in such a way that an additional crystallization of boride compound is not possible. In addition, the strong diffusion flux of boron from the coating into the substrate leads to a depletion of the element in the coating.



#### 4. Conclusions

In the present study the annealing behavior in argon of micrometer thick  $(\text{Ti}_{0.49}\text{W}_{0.51})\text{B}_2$  and  $(\text{Ti}_{0.44}\text{W}_{0.30}\text{Cr}_{0.26})\text{B}_2$  films deposited by dc magnetron sputtering on steel substrates is studied. After deposition the films are amorphous. Annealing for temperatures up to 1100 °C results in the formation of nano-crystalline (10–50 nm) precipitates of  $(\text{Ti,W,Cr})\text{C}$  (Fm-3m) and  $(\text{Ti,W,Cr})_{23}\text{C}_6$  phases (Fm-3m). In contrast to recently studied coatings on silicon substrates [27], the formation of boride phases is strongly suppressed and the precipitation kinetics of Cr free and the Cr containing samples is not significantly different. In addition, the formation of crystalline precipitates starts at an about 200 K lower temperature for the present coatings on steel. An explanation is that precipitate formation is not governed by crystallization in a supersaturated host matrix but by reactive interdiffusion of the species of the steel substrate and the boride coating.

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#### Author Contributions

Aleksandra Newirkowez: film deposition, XRD and SIMS analysis, data interpretation; Benjamin Cappi: synthesis of sputter targets; Rainer Telle: data interpretation; Harald Schmidt: data interpretation, manuscript preparation.

#### Conflicts of Interest

The authors declare no conflict of interest.

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