



# Article The Working Performance and Mechanical Strength of Reactive Powder Concrete with the CO<sub>2</sub> Curing Method on the Surface of Secondary Aluminum Ash

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Abstract: Secondary aluminum ash (SAA) is a common waste that, without reasonable treatment, results in pollution to the environment. A large amount of  $CO_2$  is emitted by human activities every day. If the CO<sub>2</sub> cannot be treated in a timely manner, it will accelerate the greenhouse effect and pollute the environment. The CO<sub>2</sub> curing on the surface of SAA can reduce excess CO<sub>2</sub> emissions while improving the performance of the SAA. The application of CO<sub>2</sub>-cured SAA can simultaneously consume the emitted  $CO_2$  and solidify the SAA. In this article, the effect of  $CO_2$ -cured secondary aluminum ash on the rheological properties, the initial setting time, the flexural strength ( $f_t$ ), the compressive strength ( $f_{cu}$ ) of reactive powder concrete (RPC), and the corresponding dry shrinkage rate (DSR) are investigated. Meanwhile, the capillary water absorption, the chloride ion migration coefficient (CMC), and the carbonization depth of RPC are determined. Scanning electron microscope (SEM) and the X-ray diffraction spectrum curves are selected to reveal the mechanism of the macro performance. Results indicate that CO<sub>2</sub>-cured secondary aluminum ash can increase the fluidity and decrease the plastic viscosity of fresh RPC. The initial setting time is increased by the CO<sub>2</sub> curing.  $CO_2$ -cured secondary aluminum ash can increase the  $f_t$  and  $f_{cu}$  by (0%~26.3% and 0% to 68.7%), respectively. The DSR is increased by adding secondary aluminum ash with an increasing rate of 0% to 91.3%. The capillary water absorption of RPC increases in the form of a linear function. The CMC and the carbonization depth of RPC are decreased by adding the CO<sub>2</sub>-cured secondary aluminum ash with decreasing rates of 0%~46.7% and 0%~45.7%. The CO<sub>2</sub>-cured secondary aluminum ash can make the hydration more compact and increased increase the hydration products (Ca(OH)<sub>2</sub>).

**Keywords:** CO<sub>2</sub> curing; surface; secondary aluminum ash; rheological properties; dry shrinkage rate; initial setting time; X-ray diffraction spectrum

## 1. Introduction

As a kind of solid waste, secondary aluminum ash (SAA) can pollute the environment if reasonable treatment is not carried out [1–3]. The magnetic separation method, plasma technology, the acid leaching method, and fused salt electrolysis are the main processing methods for the treatment of secondary aluminum ash [4,5]. However, these processing methods are not only inefficient but also cost-ineffective. Therefore, some simple, low-cost, and efficient methods should be adopted to handle this solid waste. The resource utilization of solid waste is a relatively ideal treatment method.

The application of SAA in cement materials is a relatively recognized method. SAA possesses some reactive oxygen species, which can accelerate the hydration process of cement, thus increasing the corresponding mechanical strengths and durability. SAA can be used as an active admixture, which can enhance the cementitious activity, further achieving



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**Copyright:** © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). the enhancement of the corresponding mechanical strength and durability. The flexural strength ( $f_t$ ) and the compressive strength ( $f_{cu}$ ) are increased by rates of 23.1% and 33.4% by adding SAA. Hence, the addition of SAA is advantageous to the performance of the cement matrix [6–8].

RPC is short for reactive powder concrete, which is a kind of high-performance concrete. This cement concrete is composed of cement, mineral admixtures, quartz sand, and some reinforced fibers. The influence of waste fly ash, fly ash, and rice husk ash is able to improve the mechanical strengths and the resistance to chloride erosion. Moreover, the corrosion resistance of reinforced RPC can be improved by adding these solid wastes when NaCl erosion is exerted on the specimens [9–15].

CO<sub>2</sub> curing on raw materials has been proven to make the raw materials more compact and further increase the mechanical strengths of the cement matrix. Witoon et al. have reported that CO<sub>2</sub> curing on the waste fly ash, sludge, and fly ash can improve the corresponding mechanical strengths with these cementitious materials. The  $f_t$  and  $f_{cu}$  of RPC can be increased by 14.6% and 16.3% with CO<sub>2</sub>-cured waste fly ash. On the other hand, CO<sub>2</sub> curing on solid waste can solidify the inner toxic elements and reduce their external emissions [16–22]. Xu et al. [23] found that CO<sub>2</sub> curing on waste fly ash can decrease the release of heavy metals by dozens of times. Moreover, some researchers have pointed out that CO<sub>2</sub>-cured steel slag, yellow phosphorus slag, and furnace slag can effectively improve the mechanical strengths of the cement matrix. The flexural strength of cement mortar with CO<sub>2</sub>-cured steel slag, yellow phosphorus slag, and furnace slag is 23.6%, 13.8%, and 24.5% higher than that of cement mortar with raw solid waste. Moreover, the corresponding compressive strengths are 21.4%, 14.6%, and 26.7% higher [24–26].

However, the compactness of ordinary cement-based materials is not high, and their inner harmful substances easily seep out. RPC with high compactness can prevent the heavy metals in SAA from leaching. CO<sub>2</sub> curing on SAA can solidify toxic substances in SAA while increasing the mechanical strengths of RPC. However, this method of handling SAA has not been mentioned.

The effect of CO<sub>2</sub>-cured SAA on the rheological properties of fresh RPC is measured in this article. The initial setting time, the  $f_{t}$ , and the  $f_{cu}$  of RPC are determined. The capillary water absorption of RPC is obtained. The corresponding dry shrinkage rate is measured. The chloride ion migration coefficient (CMC) and carbonization depth are found. The scanning electron microscope (SEM) photos and the X-ray diffraction spectrum curves are obtained to analyze the mechanism of the macro performance. This research can offer an excellent method for better resource utilization of SAA, manufacturing RPC, and dealing with excessive CO<sub>2</sub> emissions. The mechanical strengths and durability of SAA–RPC under various erosive environments are very novel and meaningful, and will be systematically studied in the future.

#### 2. Experimental Section

### 2.1. Raw Materials

In this study, the initial and final setting times of ordinary Portland cement (OPC) are 98.7 min and 318.1 min, respectively, provided by Zhengzhou Shunbao Cement Co., Ltd., Zhengzhou, China. Fly ash (FA) with a specific surface area of 14–25 m<sup>2</sup>/g is selected as a mineral admixture, with a bulk density of 196 kg/m<sup>3</sup> and an average particle size of 0.11–0.17  $\mu$ m. FA was purchased from Shandong Shunke Building Materials Technology Co., Ltd., Longkou, China. The aluminum ash purchased from Anyang Jiacheng Yenai Co., Ltd., Anyang, China, contains 22% to 50% Al<sub>2</sub>O<sub>3</sub> and less than 10% SiO<sub>2</sub>. Level S95 blast furnace slag powder (BFS) with a density of 2.88 g/cm<sup>3</sup>, activity index above 95%, a specific surface area of 437.1 m<sup>2</sup>/g, and a loss on ignition of 2.21% manufactured by Hebei Chuangtian Engineering Materials Co., Ltd., Shijiazhuang, China, is used as another mineral admixture. The quartz sand produced by Xinghongye Calcium Industry Co., Ltd., Changxing, China, has a SiO<sub>2</sub> content of over 99.5% and an apparent density of 2.66 g/cm<sup>3</sup>. The aggregates used in this study are three different particle sizes of quartz

sand, with particle sizes ranging from 3.28 to 1.59 mm, 0.82 to 0.31 mm, and 0.31 to 0.19 mm, respectively. The dosage ratio of the three particle sizes is 1:1.5:1. The particle size and compositions of the cementitious materials are shown in Tables 1 and 2. The flowability of fresh RPC is adjusted by polycarboxylate superplasticizer, whose water-reducing rate is 37.8%.

Trans		Particle Size/µm							
Types	0.3	0.6	1	4	8	64	360		
OPC	0.12	0.31	2.5	15.6	28.2	93.3	100		
BFS	0.045	0.14	3.27	19.36	35.15	98.12	100		
FA	31.26	58.47	82.35	100	100	100	100		
Quartz sand	0	0	0	0	0.039	23	100		
SAA	0.06	0.23	0.62	1.15	3.96	25.7	87.47		

Table 1. The accumulated pass rate of the powder materials (%).

Types	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	Fe <sub>x</sub> O <sub>y</sub>	MgO	CaO	SO <sub>3</sub>	K <sub>2</sub> O	Na <sub>2</sub> O	Ti <sub>2</sub> O	Loss on Ignition
OPC	20.9	5.5	3.9	1.7	62.2	2.7	-	-	-	3.1
BFS	33.9	15.0	0.48	9.82	36.7	0.4	3.7	-	-	-
SF	90.7	0.22	0.61	0.24	0.45	0.1	7.6	-	-	-
Quartz sand	99.4	-	0.6	-	-	-	-	-	-	-
SAA	4.6	79.3	3.9	5.7	1.5	-	-	0.9	-	-

Table 2. Chemical composition of the powder materials (%).

#### 2.2. The Manufacturing Process of Specimens

NJ-160A mixer is used for mixing SAA–RPC with a water–binder ratio of 0.2. The dried material is placed in a mixer and mixed at a rotational speed of  $60 \pm 2 \text{ r/min}$  for 2 min. After mixing the material, the solution of water and water reducer is poured into the mixer, and the rotational speed is then increased to  $80 \pm 2 \text{ r/min}$  for 6 min. The freshly mixed slurry is injected with a size of  $40 \times 40 \times 160 \text{ mm}^3$  and  $50 \times 50 \times 50 \text{ mm}^3$  molds. After demolding, the specimens are cured in the standard curing environment ( $(20 \pm 2)$  °C and relative humidity of 96.2%). The concrete carbonization test box provided by Xianxian Jiantong Test Instrument Sales Department, Cangzhou, China, is used to offer the carbonization environment. The concentration of CO<sub>2</sub> in the concrete carbonization test box is 20%, and the concrete carbonization test box shows a gas pressure of 0.5 MPa. Figure 1 shows the manufacturing process of the RPC samples. The mixing proportions are shown in Table 3.

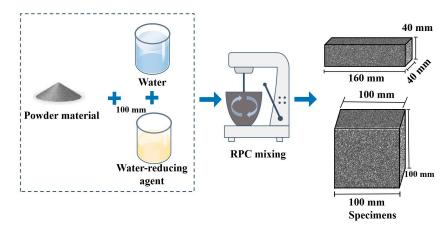


Figure 1. The manufacturing process of RPC with SAA.

Water	OPC	SAA	FA	BFS	Quartz Sand	Water- Reducer
243.42	737.74	0	368.82	110.66	973.99	16.23
243.42	737.74	92.23	276.59	110.66	973.99	16.23
243.42	737.74	184.46	184.46	110.66	973.99	16.23
243.42	737.74	276.59	92.23	110.66	973.99	16.23
243.42	737.74	368.82	0	110.66	973.99	16.23

**Table 3.** The mixing proportions of RPC with SAA  $(kg/m^3)$ .

## 2.3. Measurement of Rheological Properties and Setting Time

An NLD-3CSA mortar dry material fluidity tester is used for the measurement of slump flow of fresh RPC. The yield shear stress of freshly mixed RPC is tested in the experiment; the Huck rotary rheometer used for testing is provided by Shanghai Diguan Industrial Co., Ltd. (Shanghai, China), with a testing speed range of 0 r/min to 30 r/min. The measuring details are described in Sun's research [27]. The measuring process of yield shear stress can be observed in Wang's research [28]. The initial setting time of RPC is measured using a ZKS-100A mortar setting time tester offered by Shanghai Leiyun Testing Instrument Manufacturing Co., Ltd. in Shanghai, China. The measuring process can be found in Chinese standard JGJ70-90 [29]. Figures 2 and 3 show the measuring equipment of rheological properties and setting time.



Slump flow's measurement



Determination of yield shear stress

Figure 2. The measuring process of fresh SAA-RPC's rheological parameters.



Figure 3. The initial setting time of fresh SAA–RPC.

#### 2.4. The Determination of Mechanical Strengths

The mechanical strength measurement process mainly refers to the Chinese standard GB/T17671-1999 [30] and Zhu's research [31]. In the experiment, the flexural and compressive strength of the samples are mainly collected, and a YAW-300C fully automatic bending and compression testing machine is used to measure these two strengths. Specimens with size of  $40 \times 40 \times 160$  mm<sup>3</sup> are used for the determination of flexural and compressive

strengths. The testing speeds for bending strength and compressive strength are 2.4 kN/s and 0.1 kN/s, respectively. The testing process of mechanical strength is shown in Figure 4. Three specimens are used for the measurement of flexural strength. After the specimens are folded into two sections, the six broken samples are moved to the compression clamp to test their compressive strength. After the flexural and compressive experiments, the average values and the error bars' values are obtained to characterize the experiments' accuracy.

The manufacturer



**RPC** specimens

Flexural strength

Compressive strength

Figure 4. The measurement of SAA–RPC's mechanical strength.

## 2.5. The Determination of Drying Shrinkage Rate

A micrometer provided by Shenzhen Lide Xinmao Technology Co., Ltd., Shenzhen, China, is used for drying shrinkage rate (*DSR*) testing. *DSR* testing requires the following steps: first, install the sample on the bracket of the multimeter, and then use a micrometer to test the shrinkage value. *DSR* can be obtained by Equation (1).

$$DSR = \frac{L_1 - L_t}{L_1} \tag{1}$$

In Equation (1),  $L_0$  represents the initial length of the specimen, and L means the length of specimen at different curing ages. By this method, the *DSR* is obtained. The measurement of *DSR* is shown in Figure 5.

## **Device Model**



Figure 5. The testing process of *DSR* for SAA–RPC.

#### 2.6. Capillary Water Absorption

The capillary water absorption (CA) is measured with the specimens of size  $\Phi$ 100 mm × 100 mm. The vacuum oven is used to dry the specimens. The around sides of the specimens are coated with epoxy resin. After the epoxy resin is hardened, the specimens are moved for the measurement of capillary water absorption. The bottom surface is immersed in distilled water, and the mass of specimens is measured every ten minutes.

#### 2.7. The Measurement of CMC

The specimens with size of  $\Phi$ 100 mm × 50 mm are used to measure the CMC. The NELD-CCM540 cement chloride ion diffusion coefficient tester provided by Shanghai Meiyu Instrument Technology Co., Ltd., Shanghai, China, is used for testing the CMC.

#### 2.8. The Measurement of Carbonation Depth

The specimens with size of 100 mm  $\times$  100 mm  $\times$  100 mm are used for the measurement of the carbonation depth (D<sub>c</sub>). The specimens are placed in a carbonization curing box with a CO<sub>2</sub> concentration of 20% for 90 days, and then the carbonation depth is measured. The Chinese standard GB/T 50082-2009 is the reference for the determinations of CMC and carbonation depth [32]. Six specimens are used for the experiments to test the DSR, the CA, the CMC, and the D<sub>c</sub>. After the measurement, the average values are considered as the tested results.

#### 2.9. The Scanning Electron Microscopy and XRD Experiments

The samples are removed from the inner parts of the specimen. The particle size range of the sample is 0.5~3 mm. All samples are dried in a Li Chen vacuum drying chamber provided by China Experimental Instrument Sales Center, Beijing, China. Subsequently, the dried sample is moved to a vacuum spraying chamber for gold spraying. The SEM photos of the samples are observed with Zeiss Scanning electron microscope. The remaining samples are crushed into powder in a mortar. The powder sample is placed in a TD-3500 Diffractometer (purchased from Wuxi Lingen Electromechanical Equipment Co., Ltd., Wuxi, China) to obtain the X-ray diffraction spectrum.

### 3. Results and Discussions

#### 3.1. The Rheological Parameters of SAA–RPC

The slump flow of fresh RPC is shown in Figure 6. In Figure 6, the slump flow increases with the increasing dosages of CO<sub>2</sub>-cured SAA. As described in Li's research, CO<sub>2</sub> curing can promote the reaction between alumina and CO<sub>2</sub>, forming  $Al_2(CO_3)_3$  and decreasing the pores' inner SAA. The decreased pores can decrease the absorption of free water in the fresh RPC [33,34]. Therefore, the slump flow is increased by adding an amount of SAA. The growth rate ranges from 0% to 23.04%. Moreover, the error bars are less than 2.5% of the slump flow, indicating the accuracy of the experiment.

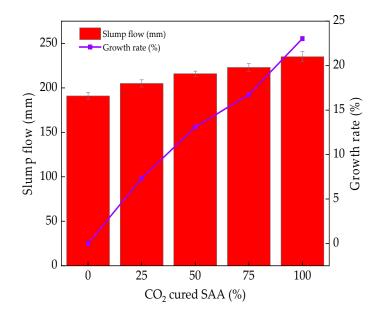


Figure 6. The slump flow of fresh SAA–RPC.

Figure 7 shows the yield shear stress of fresh RPC. As illustrated in Figure 7, the yield shear stress decreases with the increasing dosages of  $CO_2$ -cured SAA. Liu et al. found that  $CO_2$  curing on SAA can decrease its ability to adsorb free water, thus increasing the fluidity of fresh RPC [35]. As described in previous research, the yield shear stress of fresh RPC shows a negative relationship with that of slump flow [36,37]. Therefore, the yield shear stress of fresh RPC is reduced by adding the  $CO_2$ -cured SAA. The reduction rate of the yield shear stress varies from 0% to 31.96%. Meanwhile, the error bars' values are 1.0%~2.2%, showing the accuracy of the measuring results.

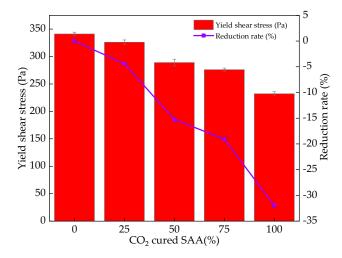


Figure 7. The yield shear stress of fresh SAA–RPC.

## 3.2. The Initial Setting Time of SAA-RPC

The initial setting time of fresh RPC is illustrated in Figure 8. It can be found that the initial setting time of fresh RPC increases by adding  $CO_2$ -cured SAA, due to how the  $CO_2$  curing can increase the amount of  $Al_2(CO_3)_3$  and decrease the pores' inner SAA. The  $Al_2(CO_3)_3$  adsorbs on the surface of hydration products, increasing the initial setting time of RPC [38]. The growth rate of RPC's initial setting time ranges from 0% to 117.3%. Additionally, the error bars' values are lower than 3.1% of the initial setting time. Therefore, the results of this experiment are accurate.

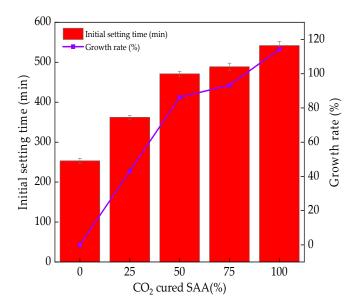
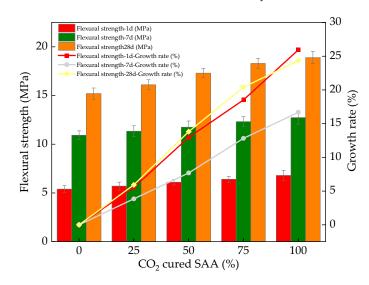


Figure 8. The initial setting time of fresh SAA–RPC.

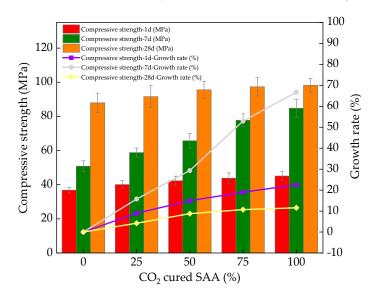
## 3.3. The Mechanical Strengths of SAA–RPC

The flexural strength ( $f_t$ ) of RPC is observed in Figure 9. The  $f_t$  of specimens cured for 1 day, 7 days, and 28 days show an ascending trend when adding CO<sub>2</sub>-cured SAA. CO<sub>2</sub> curing on SAA can increase the amount of Al(OH)<sub>3</sub>, thus increasing the  $f_t$  of RPC. The growth rates of the  $f_t$  are 0%~26.3%. The curing age demonstrates a positive effect on the  $f_t$  of RPC. This is attributed to the increased hydration degree of cement [39–41]. The values of the error bars are less than 2.41% of the  $f_t$ .



**Figure 9.** The  $f_t$  of SAA–RPC.

The compressive strength ( $f_{cu}$ ) of RPC is shown in Figure 10. The  $f_{cu}$  of RPC increases with the increasing dosages of CO<sub>2</sub>-cured SAA. The  $f_{cu}$  of RPC is increased by the increasing curing age. The growth rate of RPC's  $f_{cu}$  ranges from 0% to 68.7%. The error bars' values are less than 3.76% of the  $f_{cu}$ , which indicates the accuracy of the experiment.



**Figure 10.** The *f*<sub>*cu*</sub> of SAA–RPC.

## 3.4. The DSR of SAA-RPC

The *DSR* of RPC is observed in Figure 11. The *DSR* of RPC increases with the increasing dosages of  $CO_2$ -cured SAA. The *DSR* of RPC is increased by the increasing curing age. The hydration of cement is increased by adding  $CO_2$ -cured SAA, as it decreases the free water and increases the *DSR* of RPC. The increasing rate of the *DSR* ranges from 0% to 91.3%.

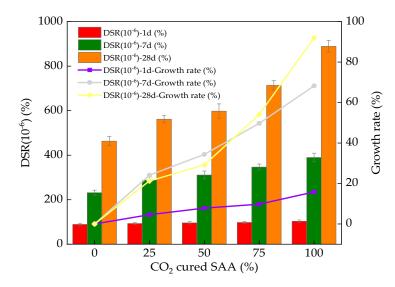


Figure 11. The *DSR* of SAA–RPC.

#### 3.5. Capillary Water Absorption

Figure 12 shows the capillary water absorption of RPC with different dosages of  $CO_2$ cured SAA. The mass of absorbed water decreases in the form of a linear function with the immersion time. The fitting degrees of the functions are higher than 0.95, which confirms the accuracy of the fitting results. As found in Figure 12, the addition of  $CO_2$ -cured SAA can decrease the mass rate of absorbed water that increases with the immersion time, due to the improved compactness caused by the  $CO_2$ -cured SAA.

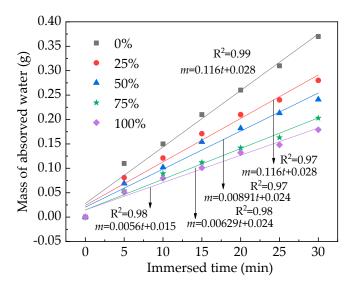


Figure 12. The capillary water absorption of SAA–RPC.

## 3.6. The CMC of RPC

The CMC of RPC is shown in Figure 13. The CMC of RPC ranges from  $1.1 \times 10^{-12}$  to  $2.1 \times 10^{-12}$ . This is ascribed to the fact that open pores exist in the RPC. Therefore, the chloride ions migrate along open pores under the action of an electric field, leading to an increase in the CMC. Moreover, as illustrated in Figure 13, the CMC of RPC decreases with increasing dosages of CO<sub>2</sub>-cured SAA. The decreasing rate of the CMC ranges from 0% to 46.7% due to the fact that the CO<sub>2</sub>-cured SAA can increase the hydration degree of cement, thus improving the compactness of RPC [42–44]. Consequently, the CMC of RPC is decreased by adding the CO<sub>2</sub>-cured SAA. The error bars' values of the CMC are lower than the real values of the CMC, which confirm the accuracy of the CMC's testing results.

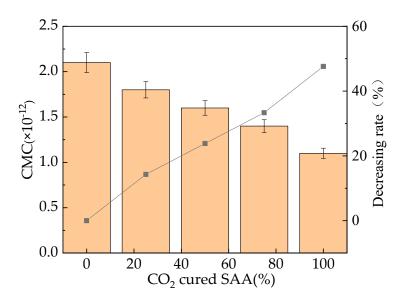


Figure 13. The CMC of SAA-RPC.

## 3.7. The Carbonization Depth of RPC

Figure 14 shows the carbonization depth of RPC. As observed in Figure 14, the carbonization depth decreases with the increasing dosages of CO<sub>2</sub>-cured SAA. This is ascribed to the fact that the CO<sub>2</sub>-cured SAA can improve the compactness of RPC [45,46]. Therefore, the migration rate of CO<sub>2</sub> in RPC becomes slow, leading to a decrease in the carbonization depth of RPC. The decreasing rate of carbonization depth by the CO<sub>2</sub>-cured SAA ranges from 0% to 45.7%, with the mass ratio of CO<sub>2</sub>-cured SAA increasing from 0% to 100%. The error bars are lower than 7.8% of the real values of the carbonization depth, indicating the accuracy of the experimental results.

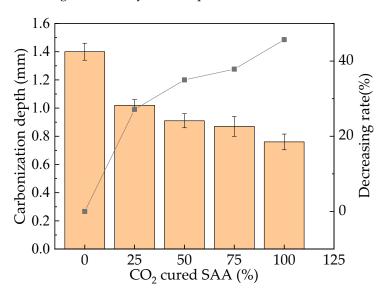
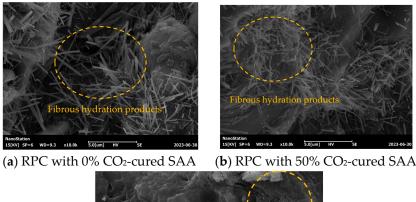


Figure 14. The carbonization depth of SAA-RPC.

### 3.8. Microscopic Analysis

The SEM photos of RPC are shown in Figure 15. The specimens are cured in the standard curing environment for 28 days. As observed in Figure 15, fibrous hydration products are found when the mass ratios of  $CO_2$ -cured SAA are 0% and 50%, respectively. The fibrous hydration products are decreased by the increasing dosages of  $CO_2$ -cured SAA. This indicates the fact that the addition of  $CO_2$ -cured SAA can inhibit the appearance of fibrous hydration products. Moreover, the increased  $CO_2$ -cured SAA can improve the

compactness of the RPC. Therefore, the mechanical strengths are improved by adding the  $CO_2$ -cured SAA.





(c) RPC with 100% CO<sub>2</sub>-cured SAA

Figure 15. The SEM of RPC with CO<sub>2</sub>-cured SAA.

The XRD curves are exhibited in Figure 16. As shown in Figure 16, the crystals' diffraction peaks of calcium silicate ( $C_3S$ ), dicalcium silicate ( $C_2S$ ), silica (SiO<sub>2</sub>), calcium hydroxide (CH), and aluminum oxide (Al<sub>2</sub>O<sub>3</sub>) are observed. As found in Figure 16, the crystals' diffraction peaks of calcium silicate, dicalcium silicate, silica, and aluminum oxide are decreased by adding CO<sub>2</sub>-cured SAA, while the crystals' diffraction peaks of calcium hydroxide are increased. This is attributed to the fact that the addition of CO<sub>2</sub>-cured SAA can improve the hydration degree of cement, thus decreasing the content of calcium silicate, dicalcium silicate silica, and aluminum oxide and increasing the content of calcium hydroxide.

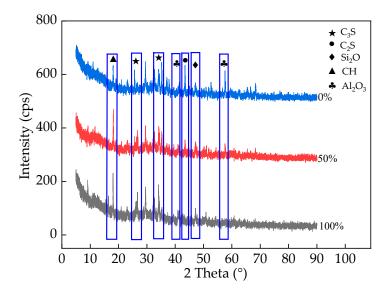


Figure 16. The XRD of RPC with CO<sub>2</sub>-cured SAA.

## 4. Conclusions

The influence of  $CO_2$ -cured SAA on the working properties, the mechanical strengths, and the durability were investigated. The research results are summarized as follows.

Due to the reduction in SAA porosity caused by carbonization, the slump flow of fresh RPC is increased, and the corresponding plastic viscosity is decreased by adding CO<sub>2</sub>-cured SAA. The slump flow increases by 0%~23.04%, and the plastic viscosity decreases by 0%~31.96%. The initial setting time is increased by CO<sub>2</sub>-cured SAA, with an increasing rate of 0%~117.3%.

The addition of SAA can improve the mechanical strengths of RPC by increasing the SAA's activity and reducing the corresponding porosity. The flexural and compressive strengths of RPC are increased with the increasing rates of the CO<sub>2</sub>-cured SAA, with the increasing rates of 0%~26.3% and 0%~68.7%, respectively. The CO<sub>2</sub>-cured SAA can increase the *DSR* of RPC with an increasing rate of 0%~91.3%. The capillary water absorption of RPC increases in the form of a linear function.

Following the research on RPC's durability, it was found that the addition of  $CO_2$ -cured SAA can decrease the CMC and the carbonization depth of RPC by rates of 0%~46.7% and 0%~45.7%, respectively.

As obtained from the SEM results, the  $CO_2$ -cured SAA can decrease the production of needle-shaped hydration products, increasing the compactness of hydration products. The  $Ca(OH)_2$  crystals are increased by the addition of  $CO_2$ -cured SAA.

This study proved that the CO<sub>2</sub>-cured SAA can effectively improve the mechanical strengths of RPC. The application of CO<sub>2</sub>-cured SAA in RPC can increase the strength of RPC while reducing its production cost and alleviating environmental pollution.

The effect of  $CO_2$ -cured SAA on the durability of RPC under various erosive environments needs to be investigated in the future.

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