

Nano Hard Carbon Anodes for Sodium-Ion Batteries

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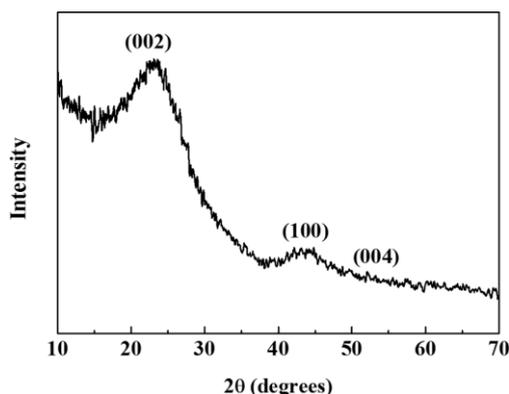


Figure S1. X-ray diffraction patterns of SCB.

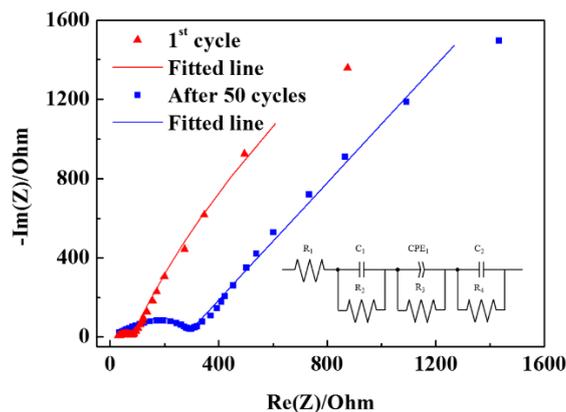


Figure S2. Nyquist plots of SCB (in a 1:1 (v/v) mixture of ethylene carbonate and dimethyl carbonate) after 1st and 50th cycles.

Table S1. The fitting values for the elements of 1st cycle and after 50 cycles.

	R_1 ($\Omega \text{ cm}^{-2}$)	C_1 (F cm^{-2})	R_2 ($\Omega \text{ cm}^{-2}$)	CPE_1 ($\text{S s}^{1/2} \text{ cm}^{-2}$)	R_3 ($\Omega \text{ cm}^{-2}$)	C_2 (F cm^{-2})	R_4 ($\Omega \text{ cm}^{-2}$)
1st cycle	24.7	1.193×10^{-1}	55.45	3.804×10^{-4}	66.45	1.109×10^{-2}	2,672
After 50 cycles	16.05	5.161×10^{-3}	241	7.593×10^{-7}	332.7	8.012×10^{-3}	3,285

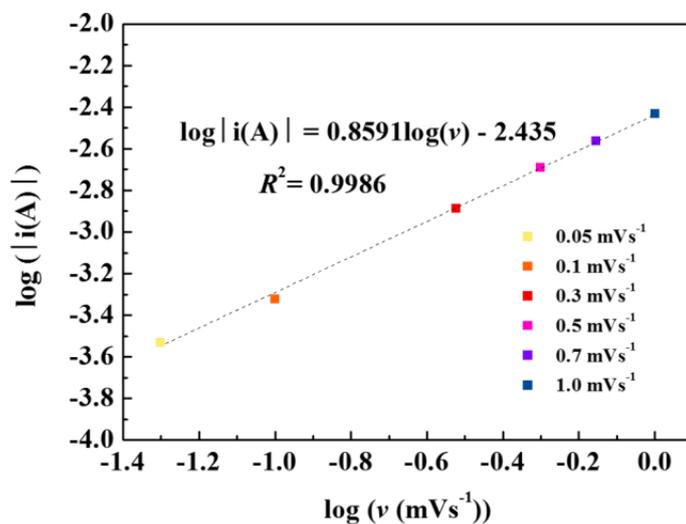


Figure S3. Determination of cathodic peak current values.

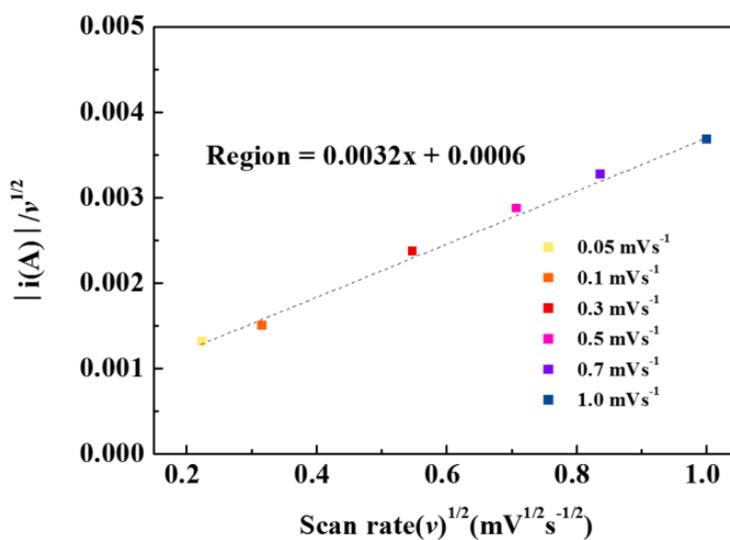


Figure S4. Cathodic peak current dependence on scanning rate, which is used to determine capacitive and intercalation contributions to energy storage.