

Supporting information

TiO₂ Nanobelt@Co₉S₈ Composites as Promising Anode Materials for Lithium and Sodium Ion Batteries

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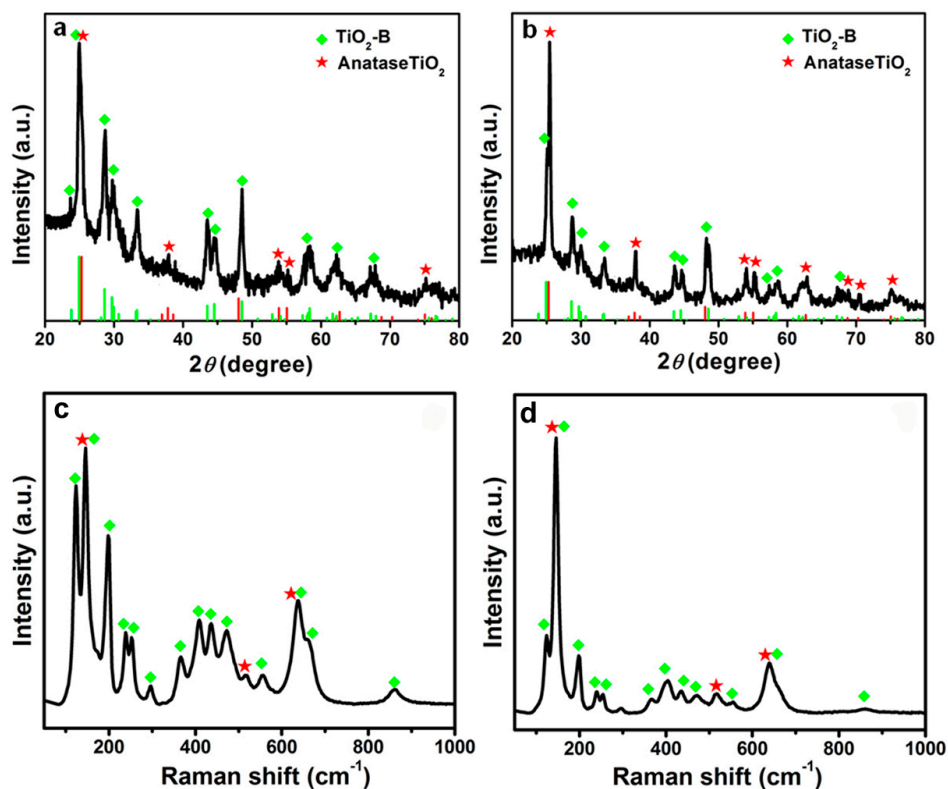


Figure S1. XRD patterns and Raman spectra of (a) and (c) as-prepared TiO₂ nanobelts and (b) and (d) TiO₂ nanobelts obtained by a heat treatment of 650 °C under Ar/H₂ atmosphere, respectively.

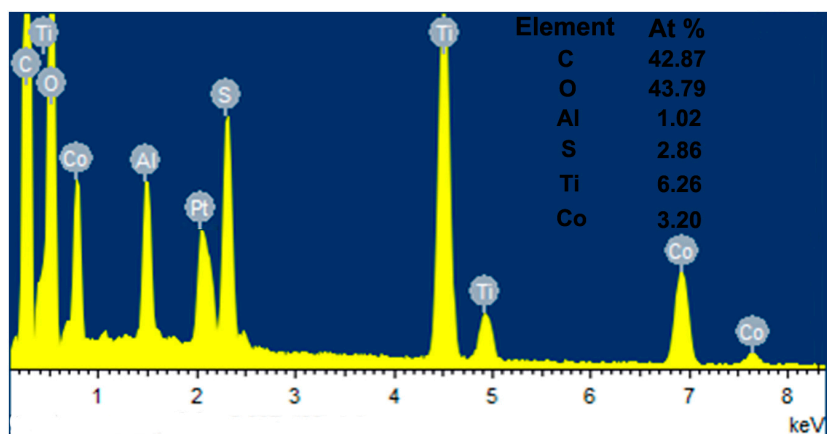


Figure S2. EDS spectrum of the as-prepared TiO₂ nanobelt@Co₉S₈ composites (the inset is the molar ratio of element Ti, Co and S, respectively).

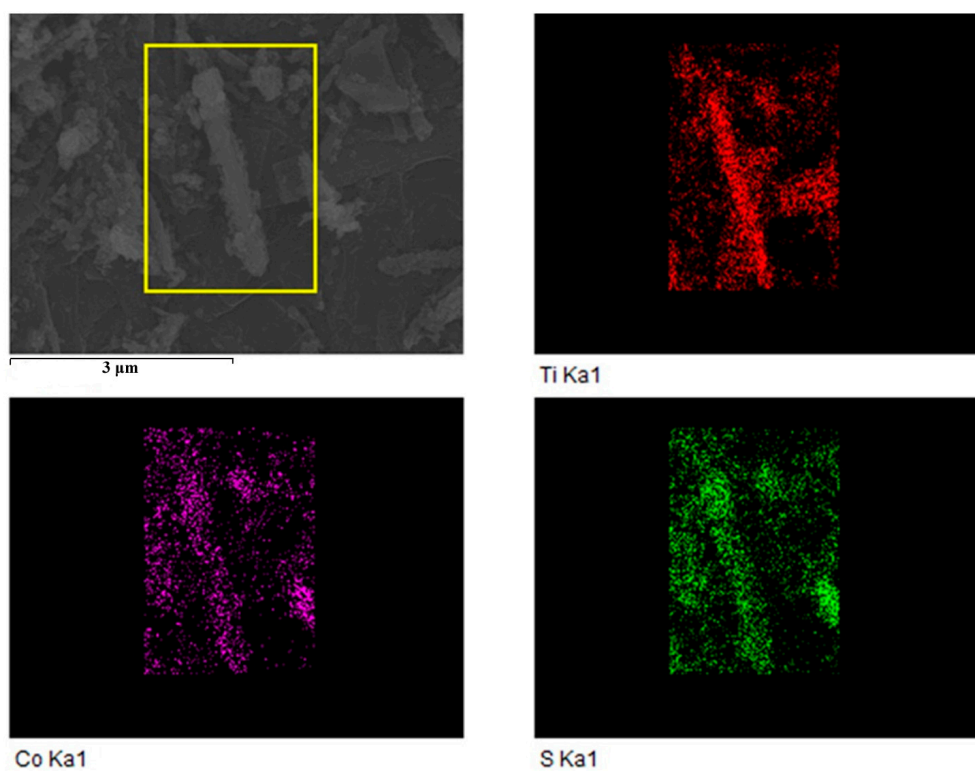


Figure S3. SEM image corresponding to EDS elemental mapping of Ti, Co and S of as-prepared TiO₂ nanobelt@Co₉S₈ composites.

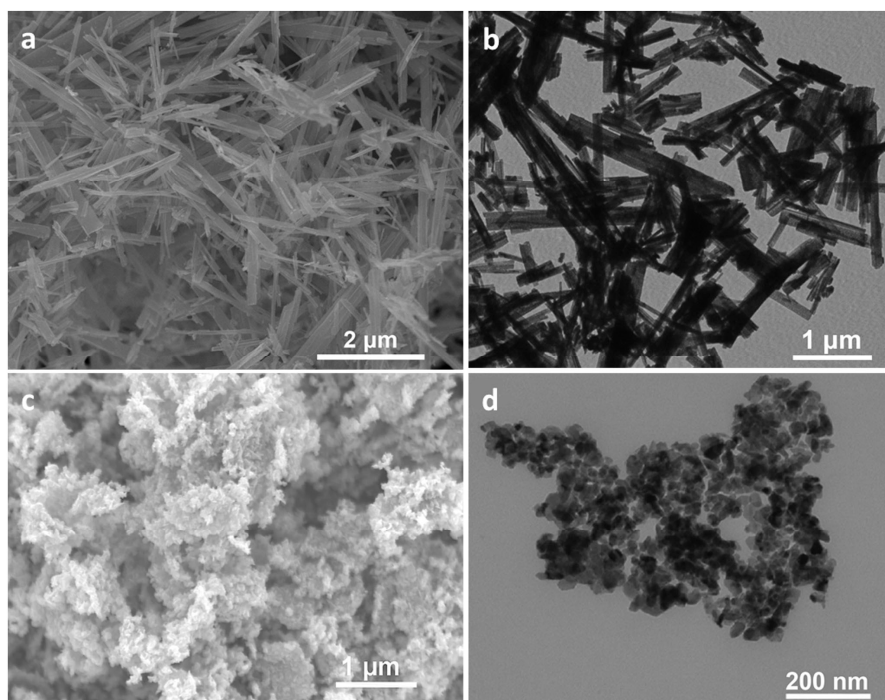


Figure S4. SEM images (a) and (c), TEM images (b) and (d) of TiO₂ nanobelts and Co₉S₈ nanoparticles, respectively.

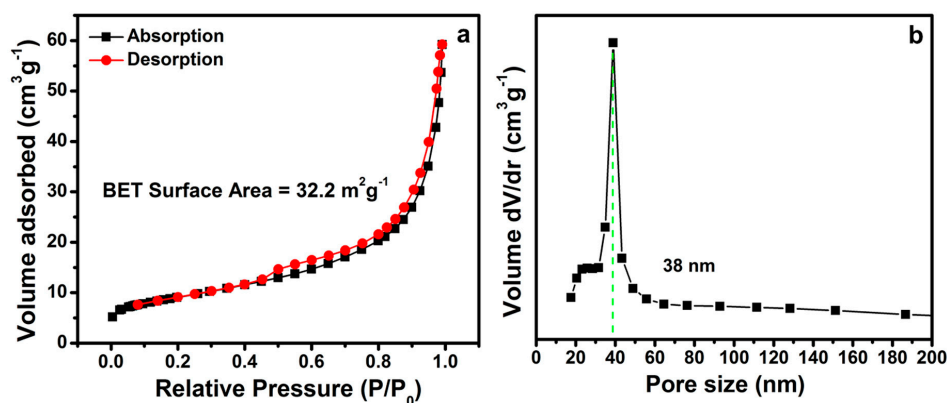


Figure S5. Nitrogen adsorption-desorption isotherm (a) and pore size distribution (b) of TiO₂ nanobelt@Co₉S₈ composites at 77.3 K.

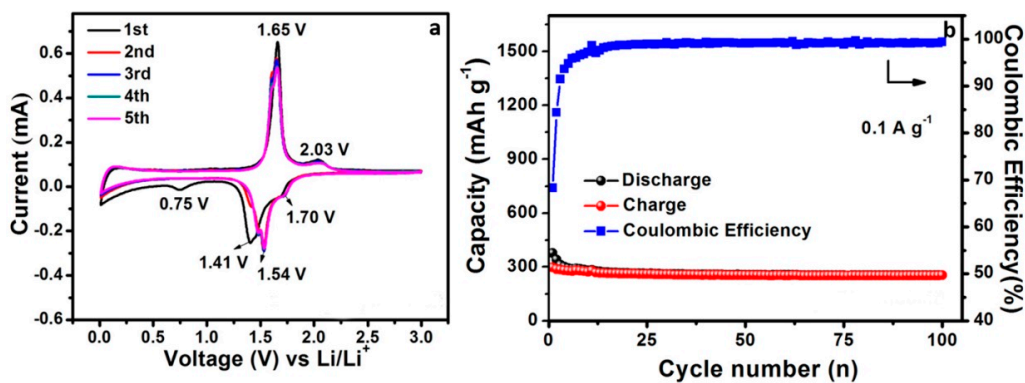


Figure S6. CV curves (a) of TiO₂ nanobelts for the first five cycles at a scan rate of 0.1 mV s⁻¹ and cycling performances (b) of TiO₂ nanobelts at 0.1 A·g⁻¹.

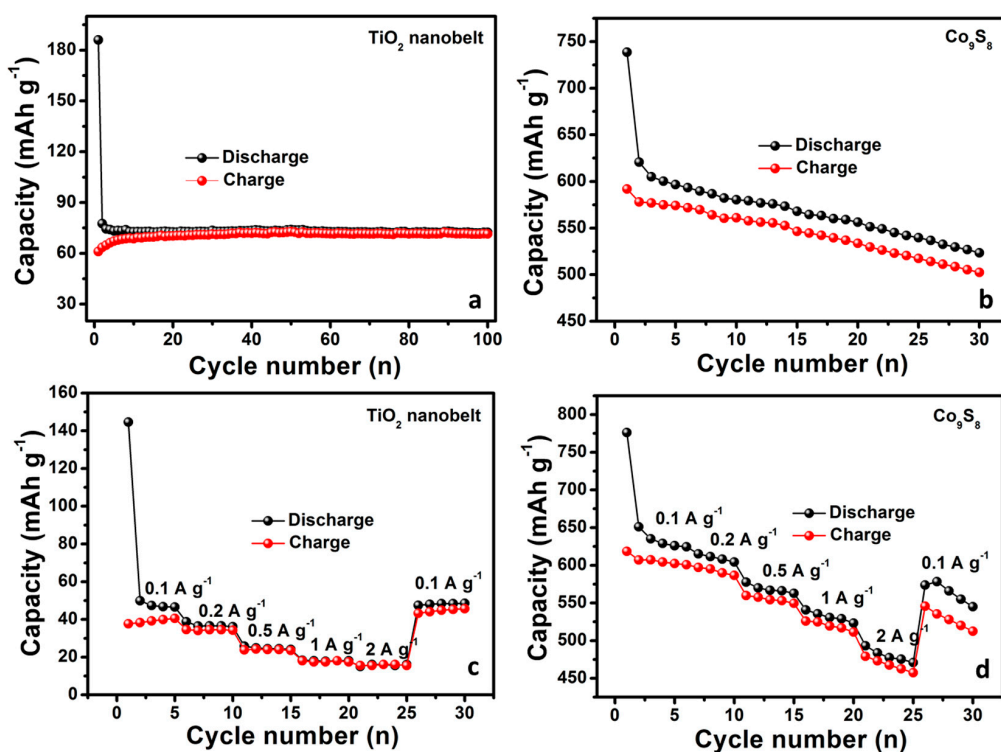


Figure S7. Cycling performances (a) and (b) at 0.1 A g⁻¹, rate capacities (c) and (d) at different current densities of TiO₂ nanobelts and Co₉S₈ nanoparticles for SIBs.

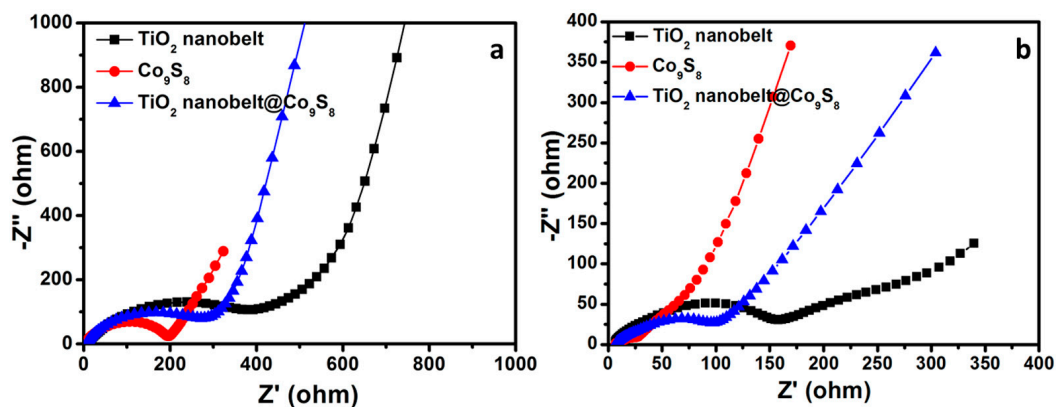


Figure S8. Electrochemical impedance spectra (EIS) of (a) before cycling and (b) after cycling 30 cycles of TiO₂ nanobelts, Co₉S₈ nanoparticles and TiO₂ nanobelt@Co₉S₈ composites.