

Supplementary Information

CeO₂-Blended Cellulose Triacetate Mixed-Matrix Membranes for Selective CO₂ Separation

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1. Results

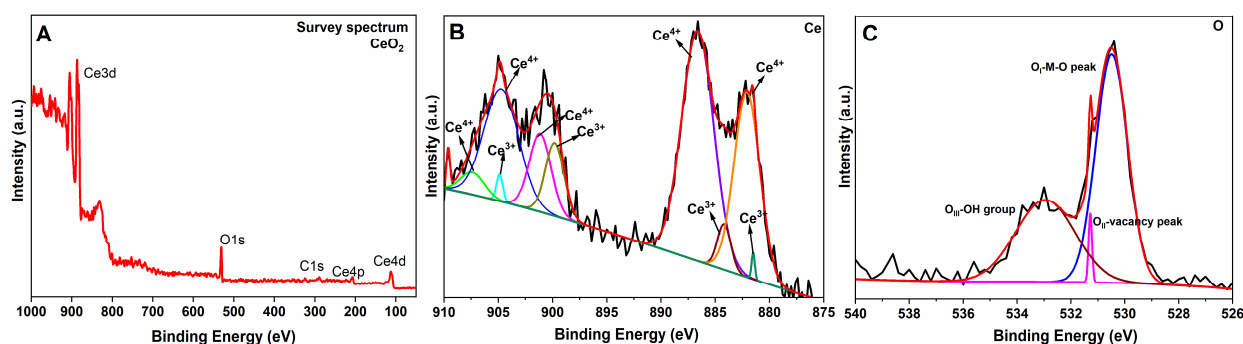


Figure S1. XPS spectrum of the CeO₂ nanoparticles: (A) survey spectrum, (B) Ce3d, and (C) O1s.

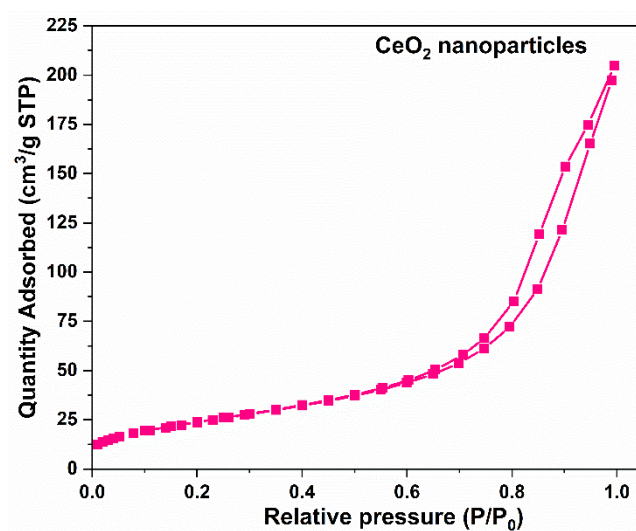


Figure S2 BET surface area of CeO₂ nanoparticles.

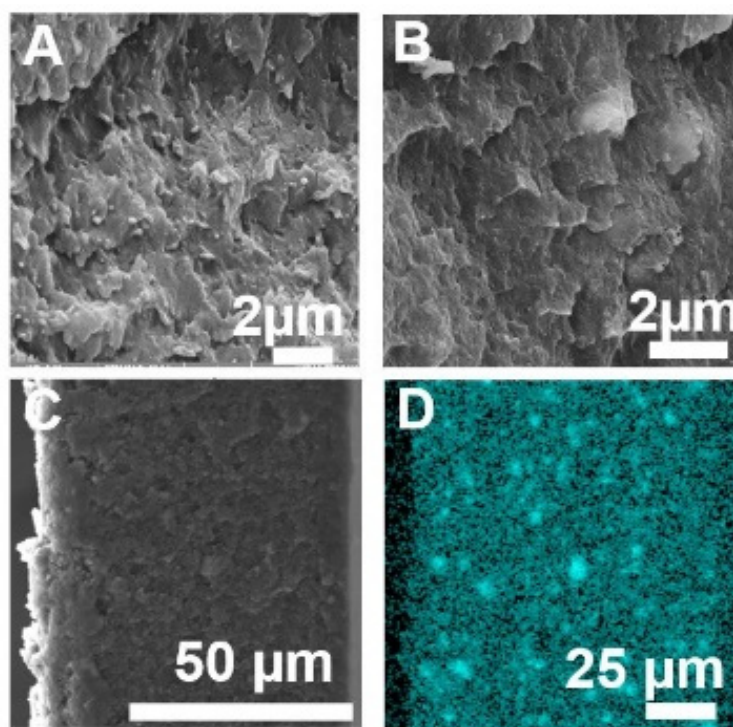


Figure S3 SEM images of the magnified portion of the cross-section of; A) CTA-CeO₂ (0.64) : B) CTA-CeO₂ (0.9) showing the formation of the voids : C) portion of cross-section of CTA-CeO₂ (0.64) for EDS mapping: and D)Element Ce mapping over the cross-section of image in C showing the homogeneous distribution of Ce.

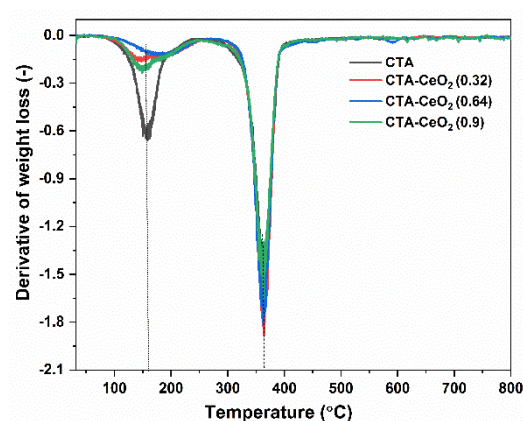


Figure S4. DTG plot for pristine and CeO₂-loaded CTA membranes.

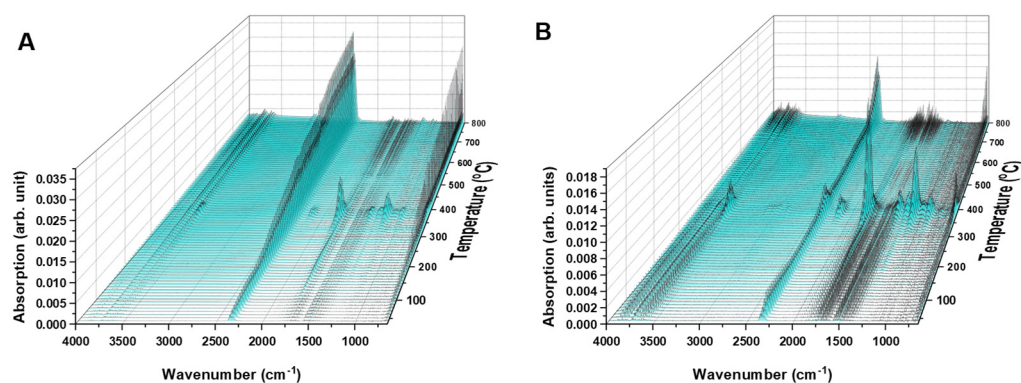


Figure S5. FTIR spectra of the product from samples measured during TGA measurement: (A) CTA, (B) CTA-CeO₂ (0.64).

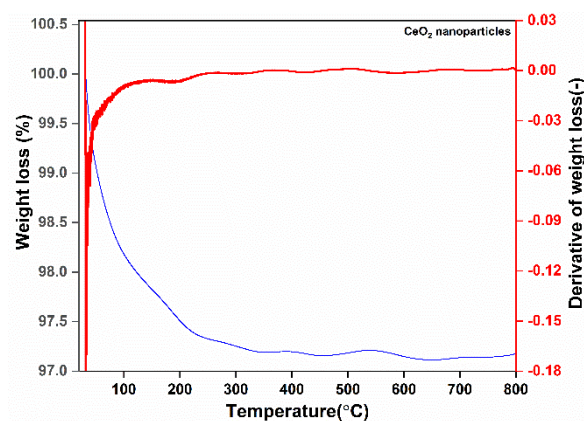


Figure S6. TGA-DTG thermogram for CeO₂ nanoparticles.

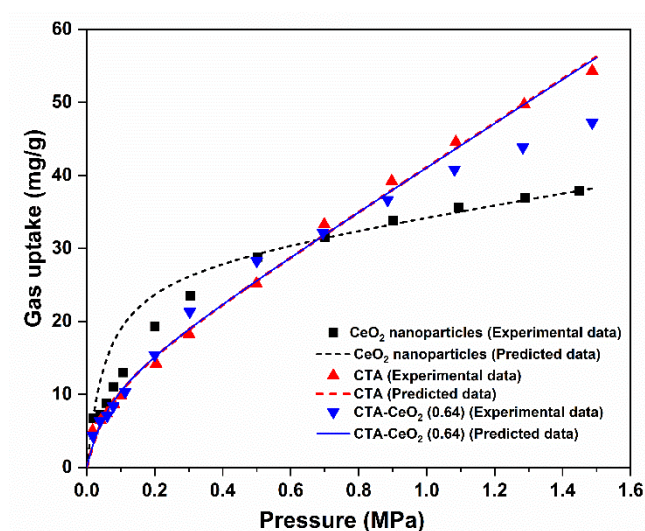


Figure S7. Sorption isotherms of CO₂ in pristine and CTA-CeO₂ MMMs (filled symbol indicates the experimental data, whereas the dashed/solid line indicates the value predicted using the DMS model).

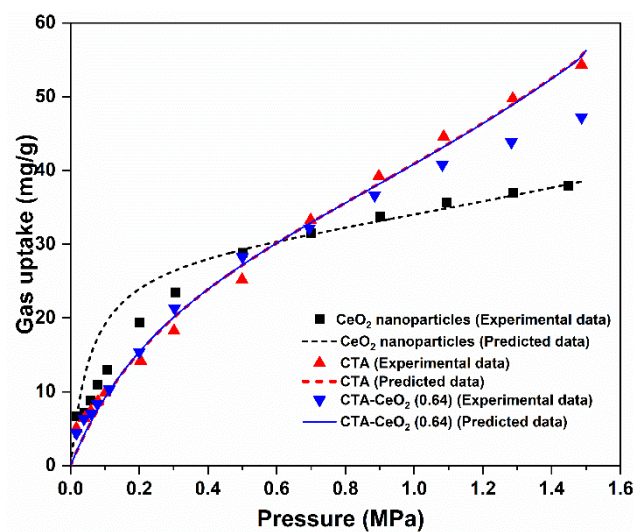


Figure S8. Sorption isotherms of CO₂ in pristine and CTA–CeO₂ MMMs (filled symbol indicates the experimental data, whereas the dashed/solid line indicates the value predicted using the GAB model).

Table S1. Fitting parameters for DMS and GAB models for CO₂ gas sorption.

Samples	Fitting Parameters (DMS Model)				Fitting Parameters (GAB Model)			
	k _D	C _H	b	R ²	<i>v</i> _m	<i>h</i>	P*	R ²
CTA	29.87	11.92	16.17	0.99	35.80	11.23	3.51	0.99
CeO ₂ NP	7.30	28.37	18.03	0.99	29.61	101.14	5.88	0.99
CTA-CeO ₂ (0.64)	13.47	32.02	3.88	0.99	39.35	17.95	5.39	0.99

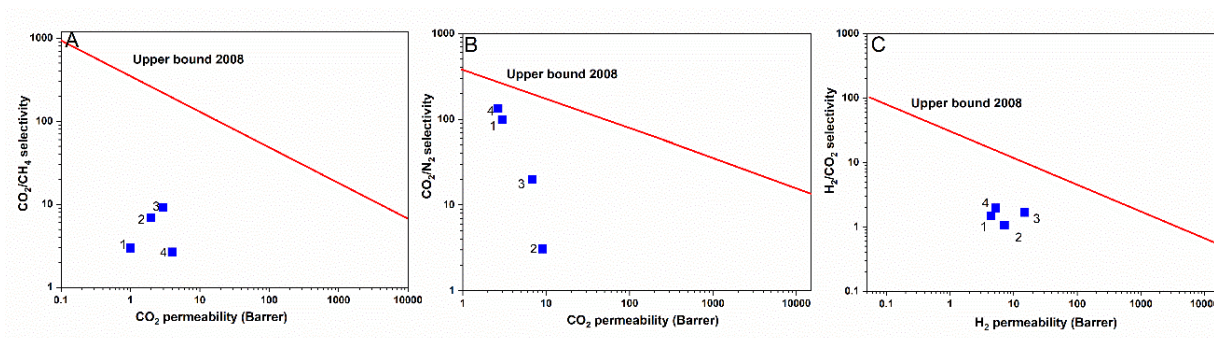


Figure S9. Robeson upper bound comparison for the different gas pairs in this study. The number in each symbol indicate the MMM samples: 1) CTA, 2) CTA–CeO₂ (0.32), 3) CTA–CeO₂ (0.64), and 4) CTA–CeO₂ (0.9).

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