

Supplementary Materials: Local Oxidation Nanolithography on Metallic Transition Metal Dichalcogenides Surfaces

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S1. Crystal growth and characterization of TMDCs

(1) TaS₂

2H-TaS₂ crystals were synthesized from the elemental components in a two-step process by chemical vapor transport (CVT). Firstly, a polycrystalline sample was obtained by ceramic combination of stoichiometric ratios of the metal and chalcogen as described elsewhere [1]. Powder starting materials were intimately mixed, placed inside an evacuated quartz ampoule and reacted at 900 °C during 9 days. The resulting free-flowing glittery grey microcrystals were then transformed into large single-crystals by the CVT methodology. 1g of TaS₂ polycrystalline material together with 275 mg of I₂ were loaded into a 500 mm long quartz ampoule (OD: 18 mm, wall-thickness: 1.5 mm). The mixture was thoroughly stashed at one end of the ampoule and the latter was exhaustively evacuated and flame-sealed. The quartz tube was finally placed inside a three-zone split muffle where a gradient of 50 °C was established between the leftmost load (800 °C) and central growth (750 °C) zones. A gradient of 25 °C was also set between the rightmost and central regions. The temperature gradient was maintained constant during 15 days and the muffle was eventually switched off and left to cool down at ambient conditions. Millimetric TaS₂ crystals were recovered from the ampoule's central zone, exhaustively rinsed with diethyl ether and stored under a N₂ atmosphere.

Phase purity was confirmed by fitting the XRPD pattern (Figure S1.1). The fit yield to a single crystallographic phase with $a = b = 3.315$ (1) Å, $c = 12.083$ (4) Å, $\alpha = \beta = 90^\circ$, $\gamma = 120^\circ$, in accordance with the values reported in the literature [2].

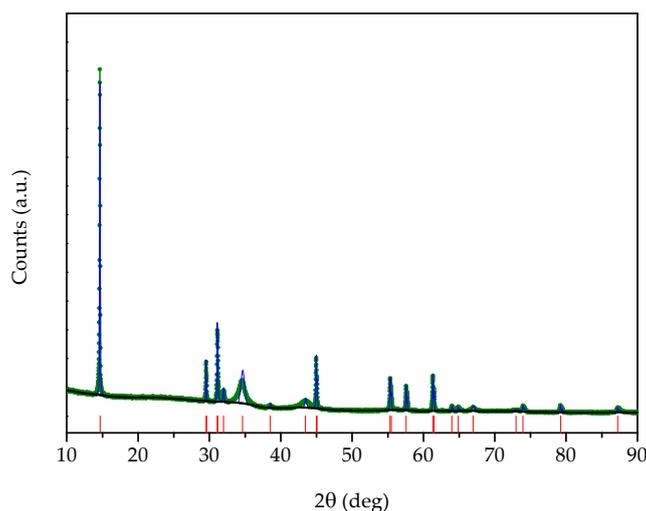


Figure S1.1. 2H-TaS₂ XRPD pattern (green) and unit cell refinement (blue). Peaks positions are marked with vertical red lines and the background with a black line.

(2) TaSe₂

1g of polycrystalline 2H-TaSe₂ was processed using the same CVT procedure and the same experimental conditions used for the growth of 2H-TaS₂. This resulted in the growth of very homogeneous samples of hexagonal flake-like crystals of 2H-TaSe₂. These crystals were recovered

from the growth zone of the ampoule and thoroughly rinsed with diethyl ether and then stored in a N₂ atmosphere.

Phase purity was confirmed by fitting the XRPD pattern (Figure S1.2) to a single crystallographic phase: $a = b = 3.4391$ (3) Å, $c = 12.707$ (2) Å, $\alpha = \beta = 90^\circ$, $\gamma = 120^\circ$, in accordance with the values reported in the literature [2].

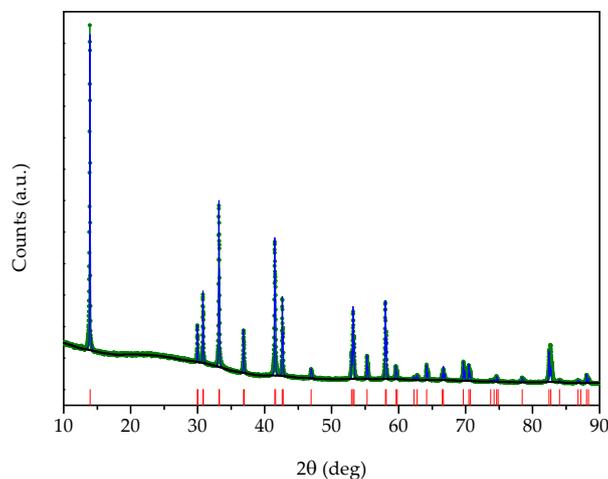


Figure S1.2. 2H-TaSe₂ XRPD pattern (green) and unit cell refinement (blue). Peaks positions are marked with vertical red lines and the background with a black line.

(3) NbS₂

1g of polycrystalline NbS₂ was processed using the same CVT procedure and the same experimental conditions used for the growth of 2H-TaS₂. After the CVT process, millimetre long hexagonal prisms could be observed in the growth zone of the fused silica vessel. Single-crystalline specimens of 2H-NbS₂ were thoroughly rinsed with diethyl and finally stored in a N₂ atmosphere.

Phase purity was checked by fitting the XRPD pattern (Figure S1.3). The fit yield to a single crystallographic phase with $a = b = 3.321$ (1) Å, $c = 11.945$ (5) Å, $\alpha = \beta = 90^\circ$, $\gamma = 120^\circ$, in accordance with the values reported in the literature [2].

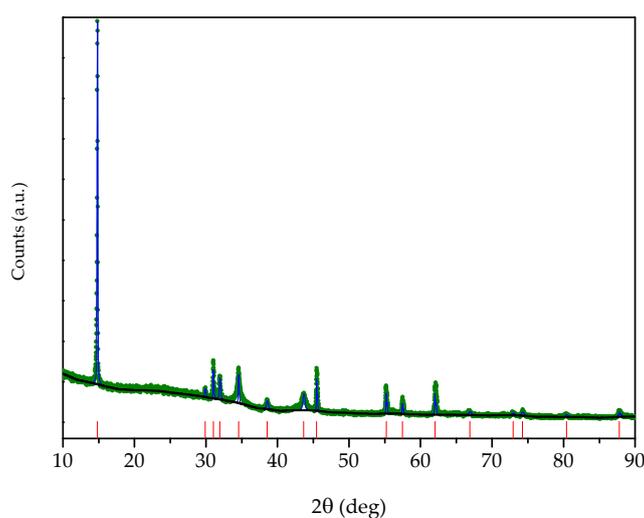


Figure S1.3. 2H-NbS₂ XRPD pattern (green) and unit cell refinement (blue). Inset: Photography of a 2H-NbS₂ crystal. Peaks positions are marked with vertical red lines and the background with a black line.

During the exfoliation of NbS₂ crystals, several needles were observed next to the platelet like flakes (characteristic of the 2H phase). The study of the Transition Electron Microscopy (TEM) images and EDX analysis on one single needle (Figure S1.4) gives rise to the following atomic composition (%): S 67.7 and Nb 32.3, in good accordance with the expected one for NbS₂. The Fast Fourier Transform (FFT) image indicates a rhombohedral structure. Thus, needles can be assessed as the 3R polytype of NbS₂.

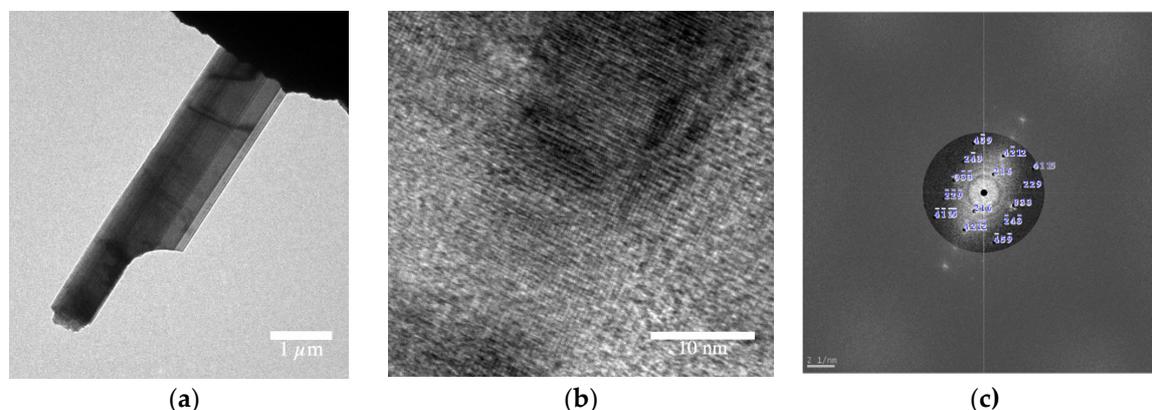


Figure S1.4. TEM characterization of 3R-NbS₂ needles. (a) TEM image of the inspected needle; (b) detail with atomic scale area; and (c) TEM FFT diffraction pattern of (b).

(4) NbSe₂

In an analogous way as described for the crystal growth of 2H-TaS₂, 1g of polycrystalline NbSe₂ was processed using the same CVT procedure. As a result, homogeneous hexagonal-like crystals of 2H-NbSe₂ with a high aspect ratio and several millimetres in length appeared in the growth zone. Crystals of 2H-NbSe₂ were thoroughly rinsed with diethyl ether before storing in a N₂ atmosphere.

Phase purity was confirmed by fitting the XRPD pattern (Figure S1.5). The fit yield to a single crystallographic phase with $a = b = 3.4460$ (4) Å, $c = 12.549$ (2) Å, $\alpha = \beta = 90^\circ$, $\gamma = 120$, in accordance with the values reported in the literature [2].

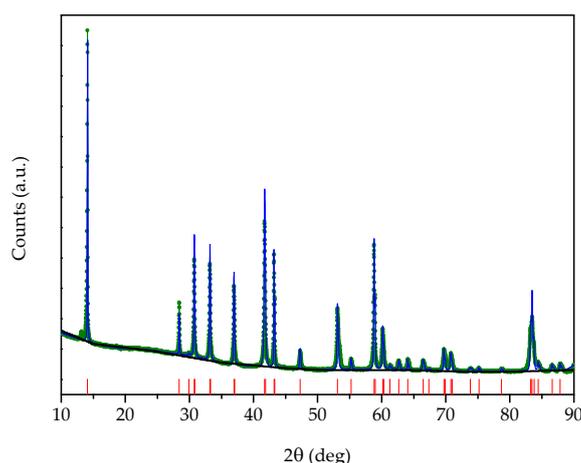


Figure S1.5. 2H-NbSe₂ XRPD pattern (green) and unit cell refinement (blue). Peaks positions are marked with vertical red lines and the background with a black line.

S2. Schematic figure of Dynamic-tip LON vs. Static-tip LON

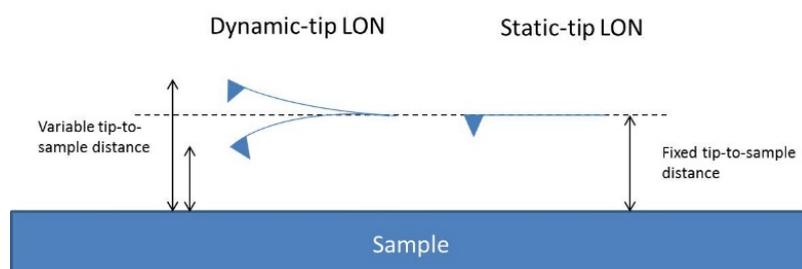


Figure S2. Sketch describing AFM tip movement during dynamic-tip LON and static-tip LON processes.

S3. Etching in thin flakes

Etching in thin flakes

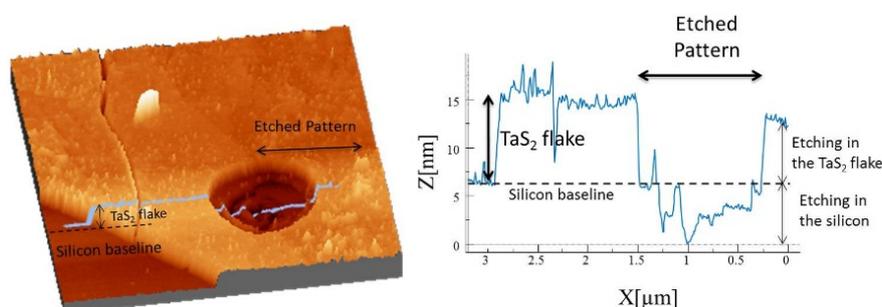


Figure S3. Etching on a thin TaS₂ layer. 3D topography image of an HF etched oxide dot (left) and the corresponding height profile (right).

S4. LON on NbSe₂

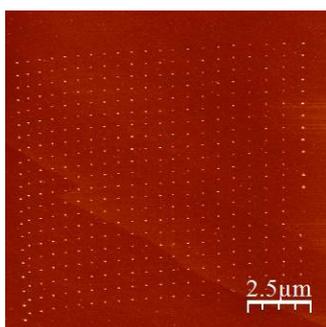


Figure S4. AFM topography image of a matrix of 20 × 20 oxide dots onto the NbSe₂ surface by static-tip LON. Image size: 12.5 μm × 12.5 μm.

References

1. Gamble, F.; DiSalvo, F.; Klemm, R.A.; Geballe, T. Superconductivity in layered structure organometallic crystals. *Science* **1970**, *168*, 568–570.
2. Wilson, J.A.; Yoffe, A.D. The transition metal dichalcogenides discussion and interpretation of the observed optical, electrical and structural properties. *Adv. Phys.* **1969**, *18*, 193–335.