

Supplementary Materials

Speciation of Ruthenium(III) Chloro Complexes in Hydrochloric Acid Solutions and Their Extraction Characteristics With an Amide-containing Amine Compound

Tomoya Suzuki ^{1,*}, Takeshi Ogata ¹, Mikiya Tanaka ¹, Tohru Kobayashi ², Hideaki Shiwaku ², Tsuyoshi Yaita ², and Hirokazu Narita ¹

¹ Environmental Management Research Institute, National Institute of Advanced Industrial Science and Technology (AIST), 16-1 Onogawa, Tsukuba, Ibaraki 305-8569, Japan; t-ogata@aist.go.jp (T.O.); mky-tanaka@aist.go.jp (M.T.); hirokazu-narita@aist.go.jp (H.N.)

² Materials Sciences Research Center, Japan Atomic Energy Agency (JAEA), 1-1-1 Koto, Sayo, Hyogo 679-5148, Japan; tohru-k@spring8.or.jp (T.K.); shiwaku@spring8.or.jp (H.S.); yaita@spring8.or.jp (T.Y.)

* Correspondence: suzuki.tomoya@aist.go.jp; Tel.: +81-29-861-8260

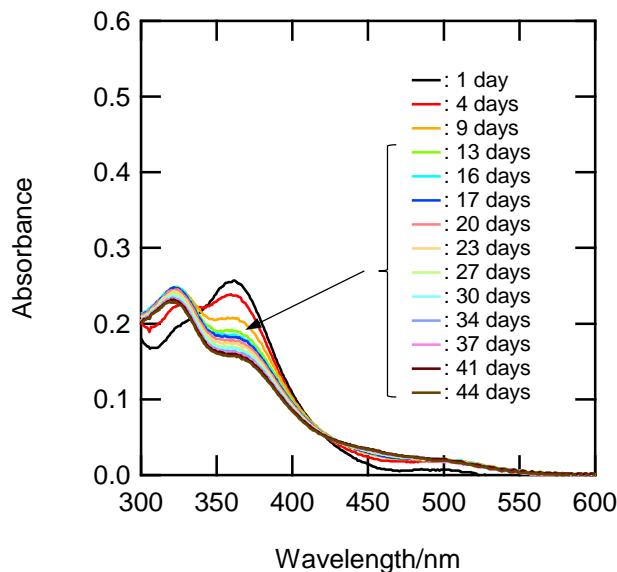


Figure S1. UV-Vis spectra for the 0.5 M HCl–Ru solution at various standing times. [Ru] = 0.9 mM; standing times: 1, 4, 9, 13, 16, 17, 20, 23, 27, 30, 34, 37, 41, and 44 days; and optical path length: 0.20 cm.

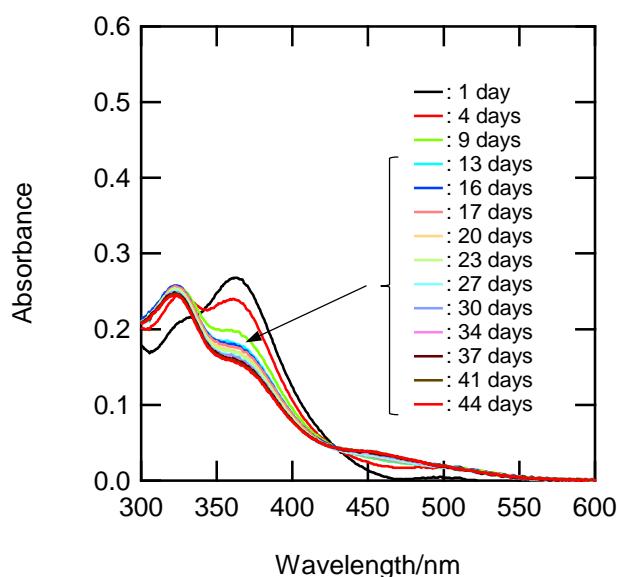


Figure S2. UV-Vis spectra for the 0.7 M HCl–Ru solution at various standing times. [Ru] = 0.9 mM; standing times: 1, 4, 9, 13, 16, 17, 20, 23, 27, 30, 34, 37, 41, and 44 days; and optical path length: 0.20 cm.

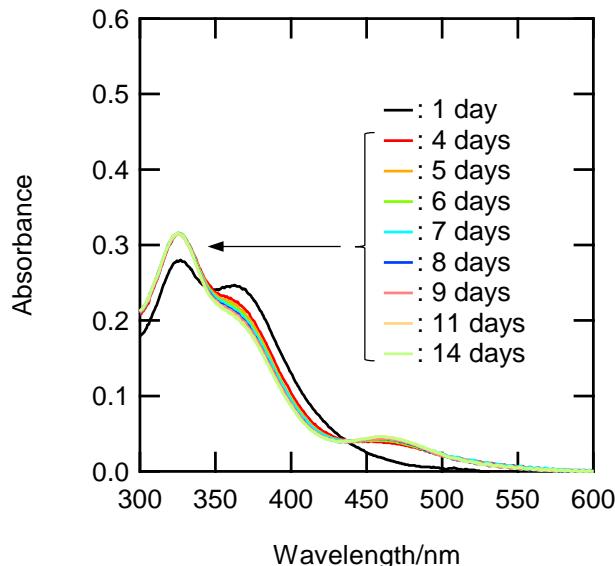


Figure S3. UV-Vis spectra for the 2 M HCl–Ru solution at various standing times. [Ru] = 0.9 mM; standing times: 1, 4, 5, 6, 7, 8, 9, 11, and 14 days; and optical path length: 0.20 cm.

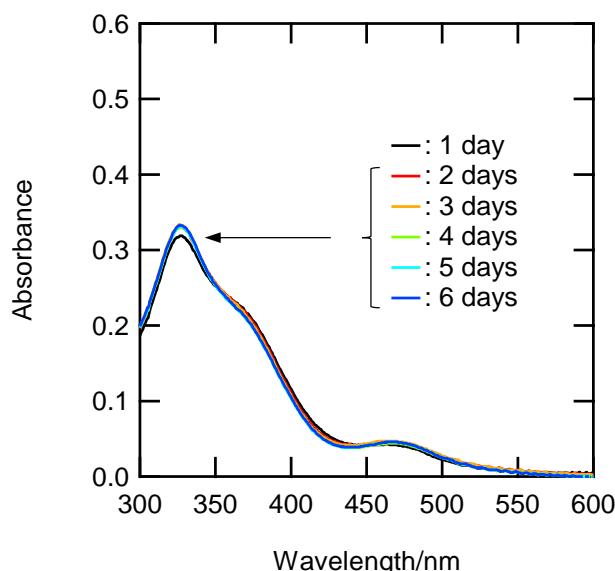


Figure S4. UV-Vis spectra for the 3 M HCl–Ru solution at various standing times. [Ru] = 0.9 mM; standing times: 1, 2, 3, 4, 5, and 6 days; and optical path length: 0.20 cm.

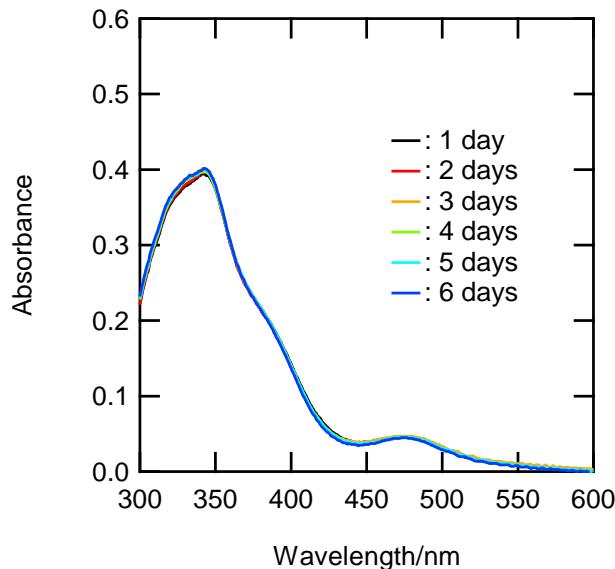


Figure S5. UV-Vis spectra for the 5 M HCl–Ru solution at various standing times. [Ru] = 0.9 mM; standing times: 1, 2, 3, 4, 5, and 6 days; and optical path length: 0.20 cm.

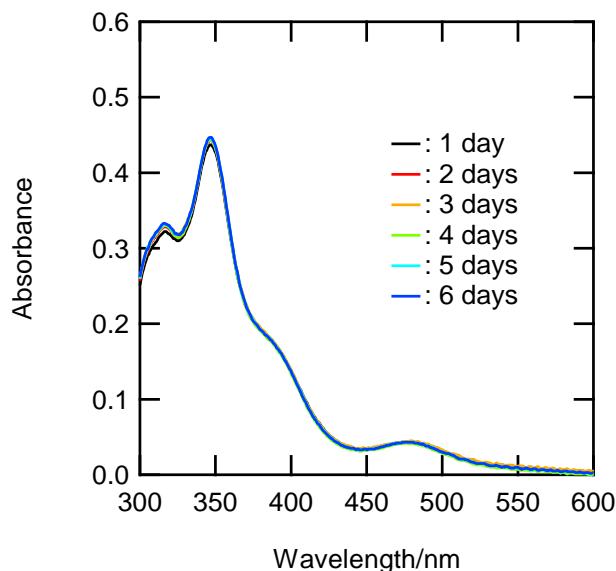


Figure S6. UV-Vis spectra for the 7 M HCl–Ru solution at various standing times. [Ru] = 0.9 mM; standing times: 1, 2, 3, 4, 5, and 6 days; and optical path length: 0.20 cm.

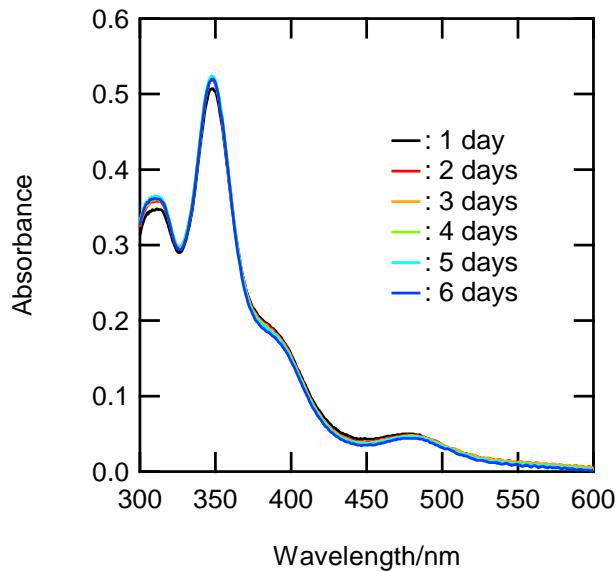


Figure S7. UV-Vis spectra for the 10 M HCl–Ru solution at various standing times. $[Ru] = 0.9\text{ mM}$; standing times: 1, 2, 3, 4, 5, and 6 days; and optical path length: 0.20 cm.

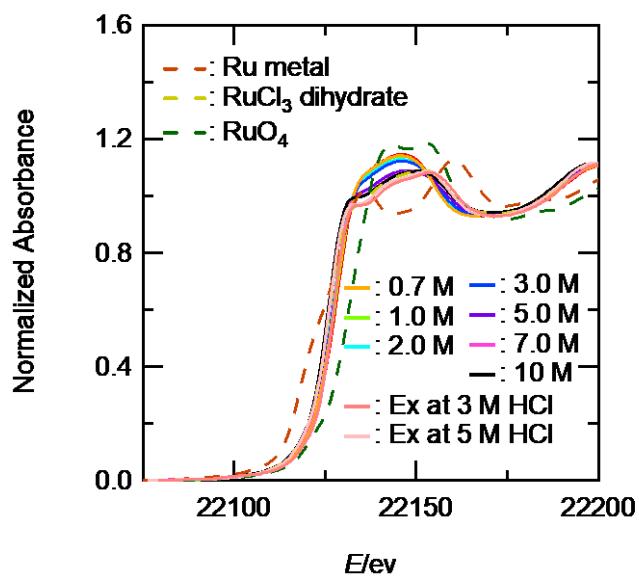


Figure S8. Comparison of XANES spectra for the Ru complexes in aqueous and organic solutions with those of Ru metal, RuCl_3 dihydrate, and RuO_4 .

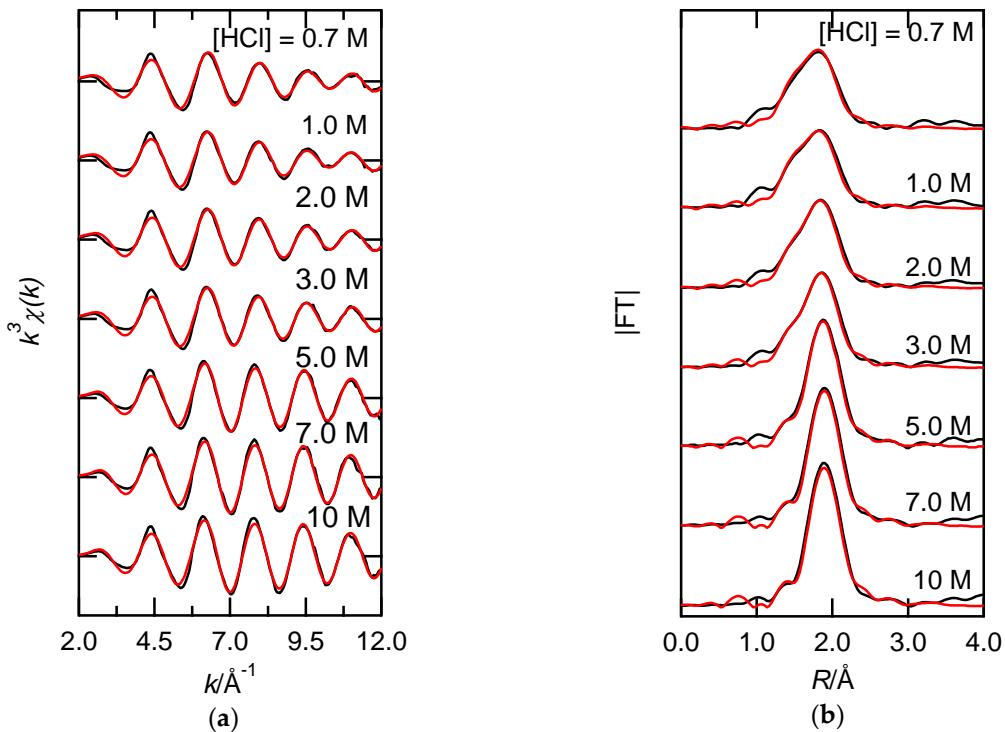


Figure S9. (a) The Ru K-edge k^3 -weighted EXAFS spectra and (b) the corresponding Fourier transforms for the Ru(III) in 0.7–10 M HCl solutions. Experimental data (black lines) and theoretical fit (red lines) are shown. The phase shifts are not corrected.

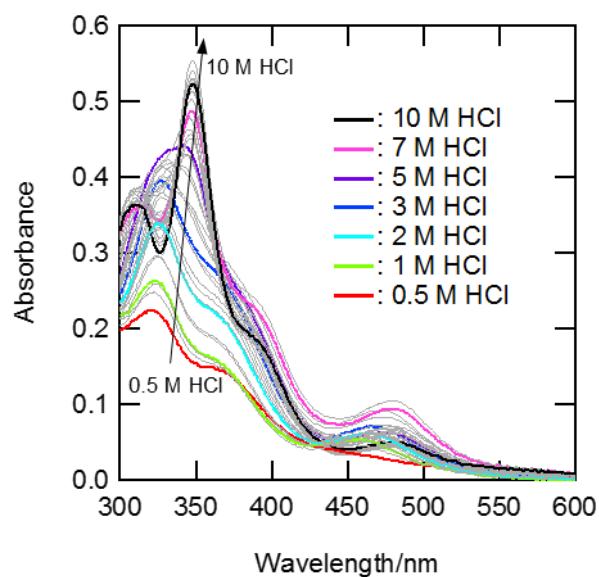


Figure S10. The UV-Vis spectra of HCl solutions (0.5–10 M) containing 0.9 mM Ru(III).

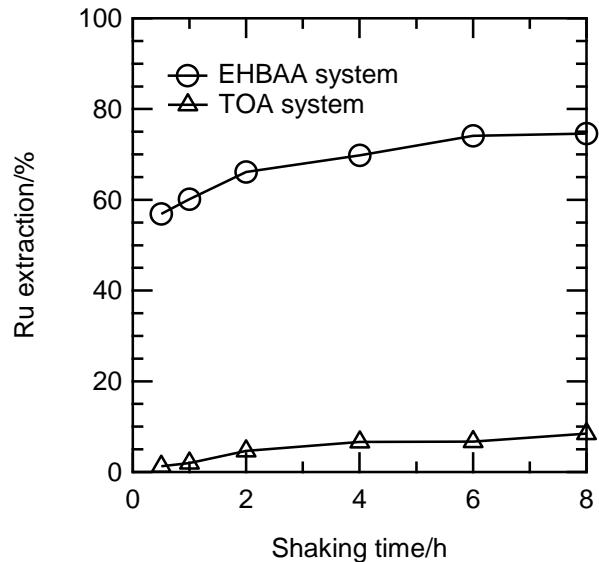


Figure S11. Extraction percentages of Ru with 0.5 M EHBAA and TOA in CHCl_3 versus shaking time. $[\text{Ru}] = 1 \text{ mM}$; $[\text{HCl}] = 3 \text{ M}$.

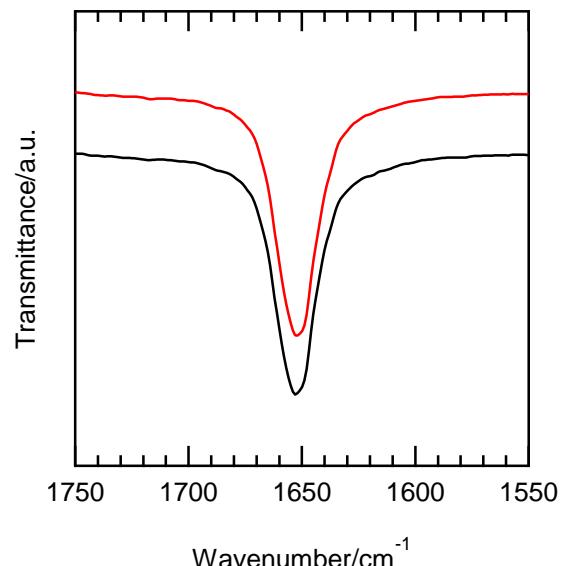


Figure S12. The FT-IR spectra of dodecane containing 0.5 M EHBA with pre-equilibrium by HCl (black line) and that medium posterior to extraction of Ru(III) in 5 M HCl (red line).