

Supplementary Materials

X-ray Photoelectron Spectroscopy (XPS) Study of the Products Formed on Sulfide Minerals Upon the Interaction with Aqueous Platinum (IV) Chloride Complexes

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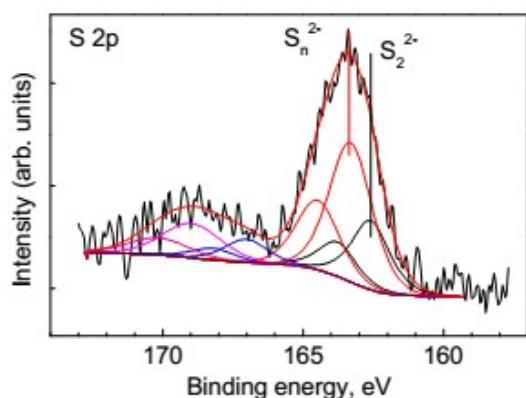


Figure S1. XPS the S 2p spectra from a droplet of sol formed in the reaction $\text{H}_2\text{PtCl}_6 + \text{H}_2\text{S}$ dried on pyrolytic graphite.

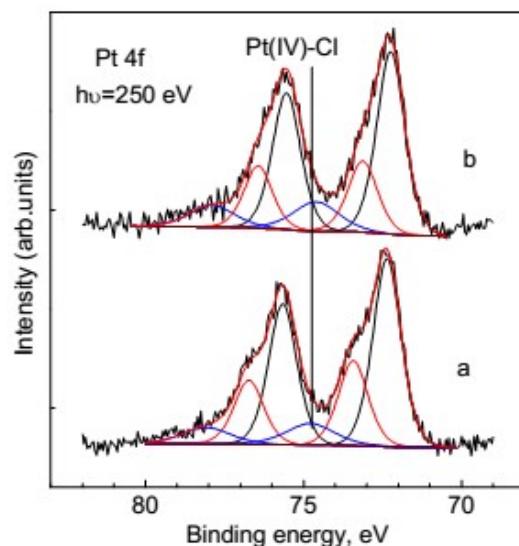


Figure S2. SR photoelectron Pt 4f spectra of the surfaces of pyrrhotite (a) and pyrite (b) after the interaction with solution of 1 mmol/L $\text{H}_2\text{PtCl}_6 + 0.03$ mol/L HCl for 12 h.

Table S1. Atomic concentrations of the elements on the surfaces of different sulfide minerals after 2 h interaction with 1 mmol/L M H₂PtCl₆.

Mineral	Pt	S	Cl	Fe	O	Cu	Others
	at. %						
Galena	2.1	37.4	5.0	-	24.8	-	30.7 (Pb)
Valleriite	2.0	8.2	0.9	1.5	63.7	1.3	22.4 (Mg,Si,Al)
Chalcopyrite	1.6	45.5	2.7	5.6	38.9	5.7	-
Pyrrhotite	0.9	29.0	3.7	3.2	63.2	0	-
Pyrite	0.7	30.8	1.5	7.4	59.6	0	-