

Review

# Challenges Related to the Processing of Fines in the Recovery of Platinum Group Minerals (PGMs)

Kirsten C. Corin <sup>1,\*</sup>, Belinda J. McFadzean <sup>1</sup>, Natalie J. Shackleton <sup>2</sup> and Cyril T. O'Connor <sup>1</sup>

<sup>1</sup> Centre for Minerals Research, Chemical Engineering Department, University of Cape Town, P Bag X3, Rondebosch, Cape Town 7700, South Africa; belinda.mcfadzean@uct.ac.za (B.J.M.); cyril.oconnor@uct.ac.za (C.T.O.)

<sup>2</sup> Minerals Expertise Tech Pty Ltd., Germiston 2007, South Africa; natalieshackleton0@gmail.com

\* Correspondence: Kirsten.Corin@uct.ac.za

**Abstract:** In order to increase the recovery of PGMs by flotation, it is necessary to optimise the liberation of the key minerals in which the platinum group elements (PGEs) are contained which include sulphides, arsenides, tellurides, and ferroalloys among others, while at the same time ensuring the optimal depression of gangue minerals. In order to achieve this, comminution circuits usually consist of two or three stages of milling, in which the first stage is autogeneous, followed by ball milling. Further liberation is achieved in subsequent stages using ultra-fine grinding. Each comminution stage is followed by flotation in the so-called MF2 or MF3 circuits. While this staged process increases overall recoveries, overgrinding may occur, hence creating problems associated with fine particle flotation. This paper presents an overview of the mineralogy of most of the more significant PGM ores processed in South Africa and the various technologies used in comminution circuits. The paper then summarises the methodology used in flotation circuits to optimise recovery of fine particles in terms of the collectors, depressants, and frothers used. The effect of entrainment, slimes coating, changes in rheology caused by the presence of a significant amount of fines and of chromite recovery is addressed.

**Keywords:** fines; PGMs; mineralogy; comminution; flotation; rheology



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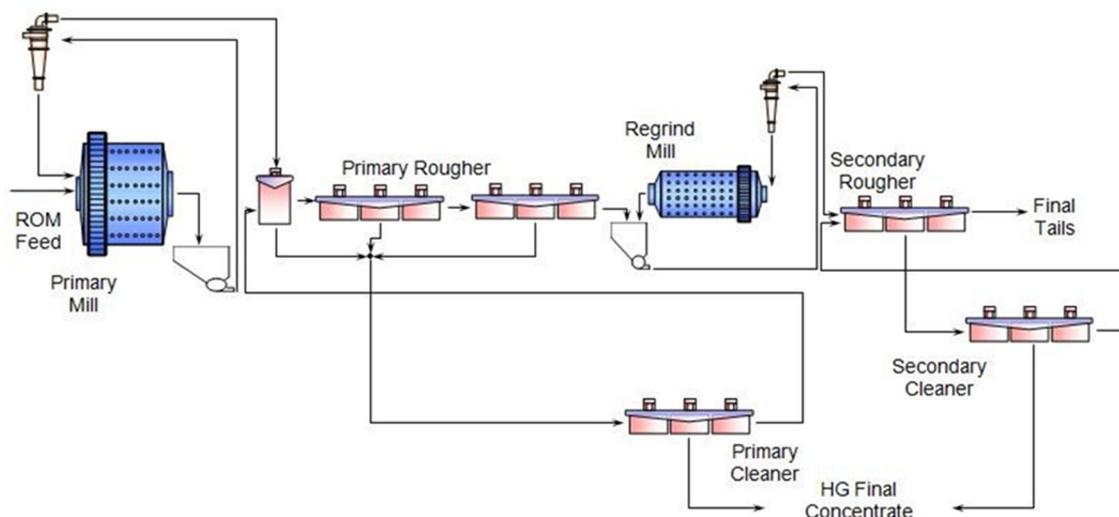
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## 1. Introduction

The issues related to fine particle flotation have been extensively studied (e.g., [1,2]). It is generally accepted that many of these problems are owing to the low mass and high surface area of fine particles and a number of contributory factors that are usually interrelated [3]. All of these factors, which will be discussed below, have the potential to affect the flotation recovery and grade of the valuable product.

Gangue slimes can also create problems by coating fresh mineral surfaces, and this may be detrimental, in particular, for fines flotation [4]. Surface coatings of value bearing minerals by hydrophilic species reduces flotation recovery in all size fractions, but particularly for fine particles. Surface coatings affect fine particles more than coarse articles most likely due to the relative surface areas—fine particles are rendered hydrophilic by a lesser degree of surface coating.

It has however been shown that it is possible that the problems associated with fine flotation may be mitigated by improved circuit design (e.g., [5,6]). In the processing of PGMs, these include the widespread use of flow sheets which take into account multiple liberation/beneficiations and are referred to as MF (mill-float) 2 or MF3 flow sheets. Figure 1 shows a typical MF2 flowsheet. The advantages of such circuits are that they reduce the propensity of over-grinding.



**Figure 1.** Typical MF2 circuit used in treatment of PGMs in Merensky, Upper Group Chromitite No.2 (UG2), and Platreef ores [7].

These improved processes are aimed at managing the problem of increasing liberation while not excessively grinding finer. As has been shown by Jameson [8], large particles with a high % surface liberation are able to float as fast as small particles with a lower % surface liberation.

In a study of the effect of milling circuits for a UG2 ore, it has been shown that on a size-by-size basis, the inclusion of additional mill-float stages results in increased PGM recovery for all size classes, and interestingly, the highest recoveries were achieved in the finest size fraction [9]. This study also showed that liberation of PGM particles in the very fine –10  $\mu\text{m}$  fraction, rather than overgrinding, was the dominant factor at this very fine grind and that fine, liberated particles are more floatable than locked or partially liberated particles in the typically optimum size range (e.g., 38–75  $\mu\text{m}$ ). Such observations emphasize the critical importance of identifying the degree of liberation in various stages of the circuit rather than the particle size. It was proposed that liberation of valuable minerals was more efficient under attrition breakage than impact breakage in the finest size fraction and thus stirred milling proved more efficient at liberating valuable minerals in the finest size fraction but still had the potential to result in over-grinding of liberated particles.

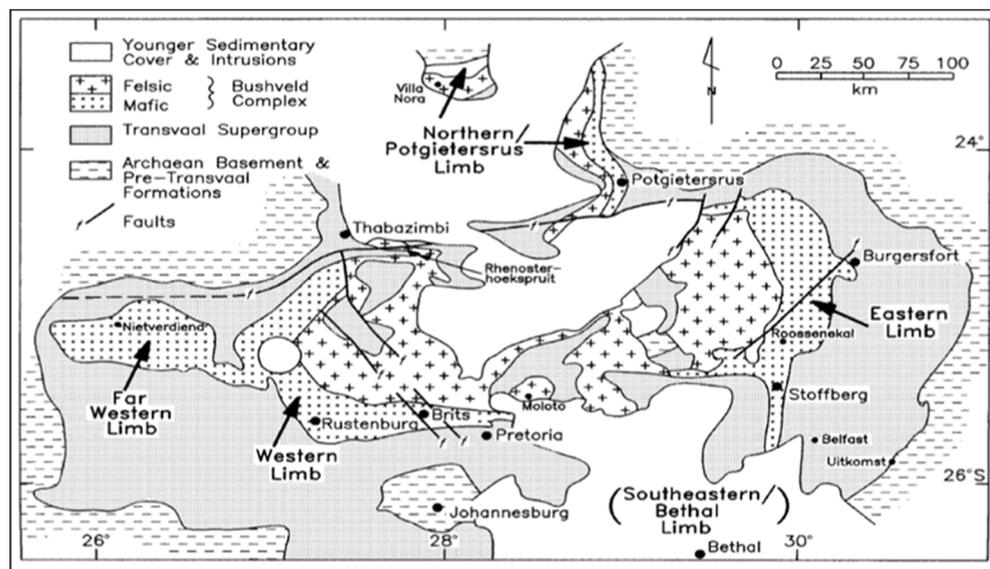
Excessively fine grinding may lead to higher chromite recoveries, which is known to be via entrainment. This is problematic since it is necessary to ensure that the chromite concentration in the concentrate is less than  $\approx 3\%$  so that it does not negatively impact the downstream smelting process. It has been observed that increasing both solids percentage and frother dosage resulted in a non-selective increase in solids recovery owing to entrainment which is a non-selective mechanism based on the particle size and density as well as the water recovery to the concentrate [10].

A further problem when considering fine particle flotation is that the large specific surface area of fine particles increases the amount of reagent that can adsorb at the surface. Thus, a large proportion of the reagent is consumed by the very fine particles. When present in limited amount, there may not be enough reagent available to float the larger particles, which results in decreased recovery. Moreover, especially in the case of sulphide minerals such as those associated with PGMs, rapid oxidation of the mineral might occur, rendering the mineral non-floatable under the conditions used for their flotation.

The present paper reviews the challenges arising from the presence of fines in the processing of PGM ores. These relate to the importance of optimising liberation in comminution circuits as indicated by the mineralogy of the flotation feed and then the management of the gangue minerals including siliceous minerals and chromite in these circuits by using various circuit configurations as well as reagent suites. Finally, the problems created by fines in terms of rheological effects and slime coatings are considered.

## 2. Mineralogy

In South Africa, platinum group elements (PGEs) are concentrated within the Bushveld Igneous Complex (BIC) (Figure 2).



**Figure 2.** Diagrammatic map of the Bushveld Igneous Complex giving the geographical locations of the western, eastern, and northern limbs [11].

The three main PGE reefs in the BIC are Merensky, UG2, and Platreef. Currently, UG2 and Platreef have become the primary source of PGE production [9,12]. The Merensky, UG2, and Platreef of the BIC contain about 75% and 50% of the world's reserves of platinum (Pt) and palladium (Pd), respectively [13,14].

The Merensky Reef is located in the western limb of the BIC. The rock-forming silicate gangue mineralogy of the Merensky Reef is predominantly orthopyroxene, plagioclase feldspar, pyroxene, phlogopite, and occasionally olivine. Secondary minerals present include talc, serpentine, chlorite, and magnetite. The base metal sulphides (BMS) are predominantly pyrrhotite, pentlandite, chalcopyrite with trace amounts of millerite, troilite, pyrite, and cubanite. The primary platinum group minerals (PGM) in the Merensky Reef are cooperite (PtS), braggite ((Pt,Pd)NiS), sperrylite (PtAs<sub>2</sub>), and PGE alloys. While minerals such as laurite (RuS<sub>2</sub>) can also be abundant. Typical PGE grades in the Merensky Reef are about 3–8 g/t [15,16].

The UG2 Reef is composed of the gangue mineral chromite with lesser silicate minerals, i.e., pyroxene and plagioclase [17]. A platiniferous chromitite layer exists, which varies from 30 to 55%, depending on the location within the BIC and the Cr<sub>2</sub>O<sub>3</sub> content of the reef. Other minor minerals include silicates (phlogopite and biotite), oxides (ilmenite, rutile, and magnetite), and BMS. Secondary minerals are constituted of quartz, serpentine, and talc [15]. PGE grades range between 4 and 7 g/t on average and vary by location. The PGMs in the UG2 Reef are generally characterised by the abundant PGE sulphides; laurite (RuS<sub>2</sub>), cooperite (PtS), braggite ((Pt,Pd)NiS), and an unnamed PtRhCuS [15,16].

The Platreef, north of Mokopane, is located in the northern limb of the BIC. The reef is a complex assemblage of different rock types with pyroxenites, serpentinites, and calc-silicates being the most abundant. Base metal mineralisation and platinum group element (PGE) concentrations are found to be highly irregular, both in value as well as in distribution [15]. The predominant PGMs in the mined area of the Platreef are the PGE tellurides, alloys, arsenides, and sulphides. Typical PGE grades are 2–5 g/t. The Pt and Pd tellurides contribute between 30% and 40% of the PGMs present in the Platreef ore followed by the alloys (26%), arsenides (21%), and sulphides (19%) [15,16,18]. Typical

compositions and mineralisation of the ore types found in the Bushveld-type rocks are shown in Tables 1–3 [19,20].

**Table 1.** Typical bulk composition of Merensky, UG2, and Platreef ores.

Mineral Name	Merensky Reef	UG2 Reef	Platreef
	Volume %		
Pyroxene	55–60	15–30	30–40
Feldspar	30–40	3–9	18
Chromite	6	50–75	-
Talc	<1	<1	<1
Serpentine	2–3	1	5
Amphibole	1–2	<1	4
Chlorite	1–2	<1	4
Mica	<1	<1	1
BMS	<1	<1	2
Other *	1–2	<1	5

\* Other minerals are mainly other silicates, Fe-oxides, and carbonates.

**Table 2.** Base Metal Sulphides (BMS) Mineralisation of Merensky, UG2, and Platreef ores.

Minerals	Merensky Reef	UG2 Reef	Platreef
	Volume %		
Pentlandite	35	44–52	27
Pyrrhotite	46	26–35	52
Chalcopyrite	20	21	19

**Table 3.** Platinum Group Metal (PGM) Mineralisation of Merensky, UG2, and Platreef ores.

Class	Minerals	Merensky Reef	UG2 Reef	Platreef
		Volume %		
PGM Alloys	Ferroplatinum Pt Alloy Pd Alloy	40	40	11–30
	Electrum (Au)	2	0.2	3
Arsenides	Pt-arsenides Pd-arsenides	4	0.1–1	1–20
	PGE-sulphurarsenides	3	0.8–7	16–35
PGM sulphides	PtPd-sulphide Pt-sulphide PtRh-sulphide	16	40–60	1–7
Tellurides	Pt-tellurides Pd-tellurides	35	0.5–5	20–50

Prior to separation by flotation, the liberation of the economic minerals from the waste minerals is necessary to ensure selective separation. This is achieved by a series of comminution stages, crushing, grinding, and classification [21]. Both the UG2 and Platreef ores contain a larger proportion of fine grained PGMs in association with gangue minerals. For example, PGM grain size in UG2 is typically <10 µm. Flotation is less effective if the value minerals are not fully liberated, while flotation may be impossible if the valuable minerals are completely enclosed in gangue silicates, unliberated [12]. Pease et al. [5,22] have shown that PGM grain size is ultra-fine (<10 µm) in UG2 ores. The PGMs are associated with the silicates and sulphides, as opposed to the chromite which has particles that are larger (200 µm) and denser. Middling streams had a median overall grain size of

20  $\mu\text{m}$  and PGM grain size of 3  $\mu\text{m}$ . The PGMs were found to be mostly locked in silicate minerals or on the grain boundaries of silicates and BMS.

Following liberation by comminution, froth flotation is the primary process for mineral separation. Typically, a 1–5% reduction in mass of the original ore flotation feed is attained in this step [12]. UG2 and Platreef are metallurgically more complex and difficult to process than Merensky, owing to their mineralogy, finer grinds, and more complex circuit designs required to achieve suitable extractions and product grades to be smelted [12]. The key to energy efficiency and improved extraction metallurgy lies in the exploitation of the mineralogy of the ores. Reducing the size of the ore particles to liberate the valuable minerals is highly energy intensive [23], and therefore complete liberation of the valuable minerals from the gangue is not an economically viable practice. An important contributor to the overall recovery achieved during flotation is the recovery of the unliberated (locked) and partially liberated (composite) particles.

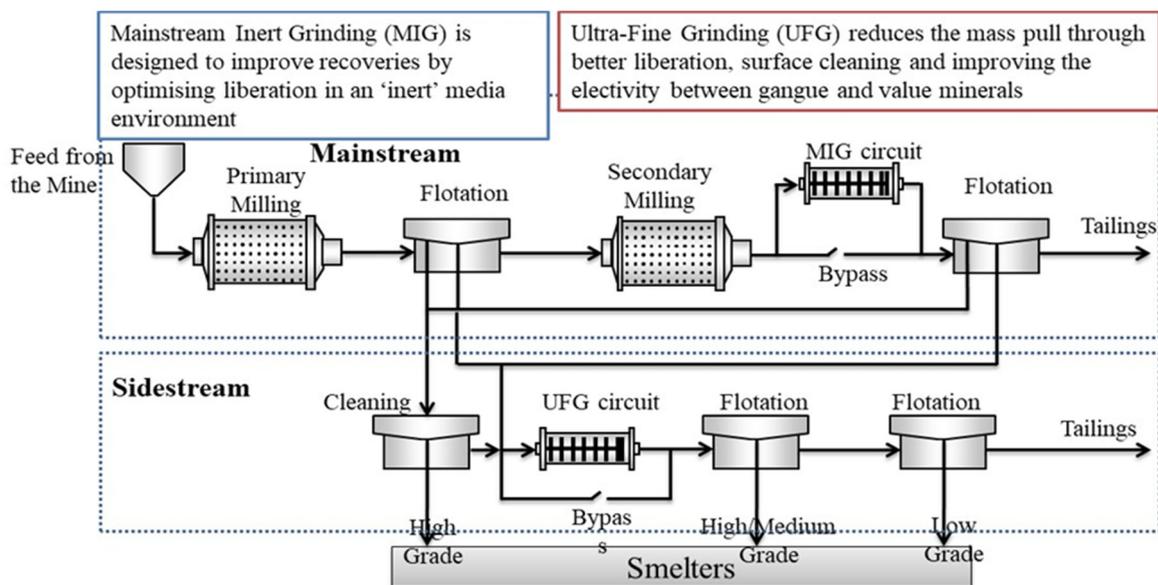
### 3. Comminution

As indicated in the introductory section, the comminution technologies and circuits used in mineral processing play an important role in the management of the recovery of fines and these must address the optimisation of liberation without a further increase in the amount of ultra-fines in the flotation circuit feed. Comminution circuits in the PGM industry have generally comprised a MF2 (mill–float–mill–float) circuit as referred to above. The MF2 configuration is comprised of two rougher flotation stages and a regrind mill between the rougher banks. This stage-wise collection of concentrate targets the relatively coarse BMS and fine platinum sulphide minerals in separate circuits. Thus, the over-grinding of both the valuable minerals and chromite/gangue is minimised [9]. However, both the UG2 and Platreef ores have “more difficult” complex mineralogy with higher proportions of fine grained PGMs (<10  $\mu\text{m}$ ) and flotation is less effective when the valuable minerals are only partially liberated and flotation may be impossible if the valuable minerals are locked in the silicate gangue [12,24,25]. The metallurgical effect on recovery for these “difficult” ores is therefore sensitive to the target grind. Further, 80% passing 75  $\mu\text{m}$  for Platreef is the defined operating parameter at the Anglo American Platinum. However, work has been in progress in recent years, utilizing ultra-fine grinding technology to target a grind of 80% passing 53  $\mu\text{m}$  has shown improvements which are economically beneficial. Ore zones of poor metallurgical performance appear to contain higher amounts of very fine composite silicate/PGM particles, which therefore require finer grinding to liberate for recovery by flotation [26].

Typical primary milling circuits employ Run-of-Mine (ROM) mills. ROM mills include ball, semi-autogenous (SAG), and fully autogenous (FAG) mills [12]. In recent years, an alternative in the primary comminution circuit, the high-pressure grinding roll (HPGR) technology has found its place. With consideration for the global rise in energy costs, HPGR installations have become more prominent, especially in applications where the ore types were harder, such as the Platreef. Innovative flowsheets were proposed by Rule et al. [12] and by Morley [27] which have resulted in significantly increased energy savings. Chapman et al. [28] discussed the results obtained on the impact of different comminution procedures on the flotation of PGMs. The PGE ore showed preferential liberation of platinum when the HPGR was used, compared to the equivalent cone-crushed material. Significant amounts of the value mineral were reported to the finer size fraction. These results indicated, however, that despite greater liberation at equivalent grain size, HPGR did not improve flotation performance, perhaps due to a slime coating problem which could be addressed by sending the fine material straight to the flotation circuit and by passing any further milling. The advantage of HPGR technology compared to classical routes is reduced energy consumption. HPGRs do not need steel grinding media and throughput is less impacted by changes in ore hardness. Therefore, HPGR could prove economically beneficial when considering the variability in ore hardness [29]. Micro-cracking is a further advantage of using an HPGR. Daniel [30] showed that micro-cracking was

present when ore was ground in the HPGR. Micro-cracks reduce the Bond ball mill work index of 10–25%, when compared with the product from a tertiary cone crusher [29,31,32] and that the HPGR is likely to result in weaker coarser particles, owing to microcracking, and intact fines.

In the past, the secondary milling of UG2 ore was conducted with ball mills. However, in order to improve the liberation of the valuable minerals, IsaMills™ (Glencore, Brisbane, Australia) have been introduced to the secondary milling circuit [12]. The IsaMill™ is a horizontal stirred mill, with a high energy density and fine, inert grinding media, specifically intended for the effective liberation of the very fine, valuable PGMs [5]. Isamills had been operating at Mt ISA in Australia for several years and after IP issues had been resolved, AngloPlatinum decided to integrate this technology into their PGM circuits as they offered significant energy savings over ball milling. Overgrinding is a consideration when using the IsaMill™ owing to the high energy density, this is particularly so for the UG2 chromite. Chromite can accumulate in the IsaMill™, impacting grinding efficiency and may result in unplanned stoppages. Rule [12] found that by allowing a chromite rich stream from bypassing the IsaMill™, the overgrinding of chromite could be avoided. Stirred mills have found their place within the UG2 cleaning circuits. According to Rule [12], the benefits of fine grinding in the cleaning circuit are the inert grinding environment, high intensity attrition and improved liberation and these result in improved recoveries and grades. Pease et al. [22] also reported that the use of electrochemically inert grinding media at plant scale increased the recovery of fine sulphide mineral particles. Since the addition of mainstream regrind circuits or mainstream inert grinding (MIG) stirred milling applications have been implemented, Anglo American Platinum Concentrator plants, UG2 recoveries have increased by more than 5% at the two main production units, viz. the Rustenburg and Amandelbult mining sites. This is one of the largest step change increases in the operations' history. Further flow-sheet modifications and improvements well guided by process mineralogical information are expected in the next decade. Figure 3 shows two applications for IsaMill™; MIG and ultra-fine grinding (UFG) on a typical PGM circuit [24,33,34].



**Figure 3.** The application of IsaMill™ technology in the concentrator flowsheet [33,34].

It is also important to note the critical role of efficient classification in comminution circuits since this can play a major role in reducing overgrinding and limiting the production of fine particles smaller than the required liberation size [35]. Rule and Schouwstra [25] have discussed the future outlook for PGM concentrators as well as the obstacles facing

the industry. They noted that the PGM industry had adopted significant new technologies in the concentrator flow sheet design in the last decade. The main drivers for this were as a result of a major switch to UG2 and Platreef ore types, declining plant feed grades, and PGM recovery which has become more difficult to achieve with known technology due to more complex and difficult mineralogy. Finer grinding for better liberation and progressively higher power costs are being fueled by the PGM market dynamics.

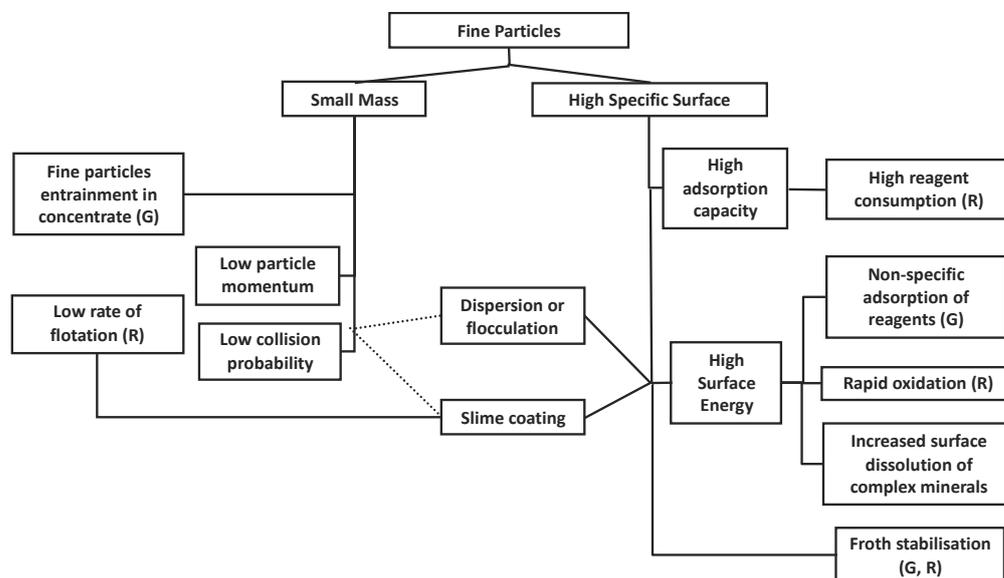
#### 4. Flotation

Flotation, a well-known selective separation process, is applied to the separation of different minerals from each other. The process takes advantage of the differences in physico-chemical surface properties of minerals within an ore. Decreasing ore reserves necessitates the increased exploitation of low-grade ores and ores with fine intergrowths. In general, these ores are more difficult to process and they often require fine grinding to liberate the valuable minerals so as to increase the valuable minerals recovery as well as the concentrate grade. In this context, the interest in technological steps to improve the flotation recovery of fine particles has significantly increased. The role that particle size plays in froth flotation is complex. Literature suggests that there is an optimum particle size for flotation which is a function of size and degree of liberation [5,22,36,37]. As referred to above, Jameson [8] has shown that large particles with a high % surface liberation are able to float as fast as small particles with a lower % surface liberation, emphasizing again the importance of careful management of the comminution circuit and importance of using modern mineralogical techniques such as QEMSCAN (Quantitative Evaluation of Materials by Scanning Electron Microscopy) to investigate the degree of liberation of material entering the various stages of the flotation circuit. While increasing particle size and increasing hydrophobicity/liberation improves the probability of attachment, increasing particle size also increases the probability of detachment as liberation decreases [38–40].

In a froth flotation system, particles may enter the froth phase by either true flotation or by mechanical means, such as entrainment or entrapment. While true flotation is selective and involves the attachment of a hydrophobic particle to an air bubble, the other mechanisms are non-selective and mostly depend on particle size and the hydrodynamics of the flotation cell [41,42]. Gaudin et al. [43] demonstrated that different sized particles have different flotation character. The influence of particle size on the recovery rate of minerals has been extensively investigated (e.g., [1,2,44–49]). The reasons why the introduction fine particles are problematic in terms of recovery by flotation are mainly due to their low mass and large surface area which results in factors such as low particle momentum, low probability of bubble–particle collision, heterocoagulation, high consumption of reagent, collector non-selectivity, excessive froth stability, entrainment of fines, high dissolution rate in water, high pulp viscosity, and slimes coating all of which affect the recovery and grade of the valuable minerals. A schematic diagram of the parameters affecting fine particle flotation is shown in Figure 4. Floatability varies depending on the size range of the mineral. Floatability is also dependent upon the type of flotation cell, e.g., laboratory batch flotation or industrial flotation [5,22].

To overcome the difficulties in flotation of fine particles, several flotation technologies, which intend to increase bubble–particle collision efficiency by either decreasing bubble size or increasing particle size, have been developed. However, many of the processes are non-selective and selectivity is essential in the minerals processing industry. A laboratory flotation study [3] carried out on a barren Norite sample from the Merensky Reef and a synthetic mineral, moncheite (PtTe<sub>2</sub>), showed that the moncheite particle size has a significant impact on flotation. Coarse particles (−53 + 25 μm) showed the highest Pt flotation kinetics and overall recoveries while the −5 μm size fraction of moncheite gave the lowest flotation rate and recoveries. These results correlate with other findings reported in the literature (e.g., [50]). In addition, the data showed that an increased collector dosage enhanced the flotation rate of the −5 μm size fraction of moncheite when compared to a lower collector dosage. The good Pt recovery results obtained for the −5 μm moncheite

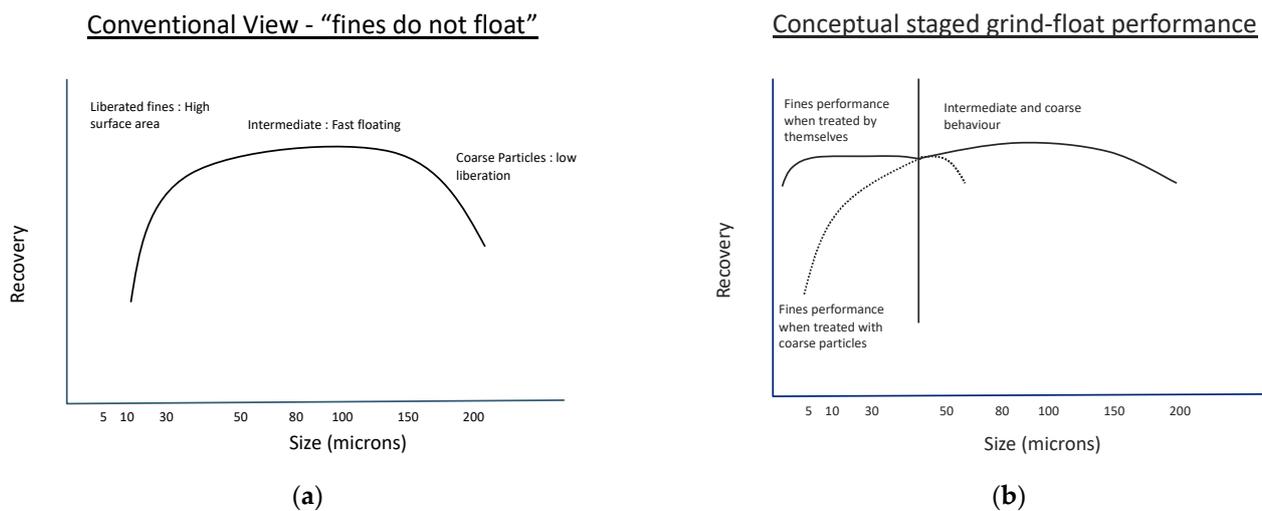
size fraction at the higher collector dosage correlated with the findings in literature where fine particles require higher reagent dosages due to their higher surface area per unit mass [37,50,51].



**Figure 4.** Factors affecting fine particle flotation [5,22].

Higher energy input during laboratory flotation also increased the rate of flotation of the  $-5\ \mu\text{m}$  size range, at a lower concentrate grade, when compared to a lower impeller speed. It was reported [52] that more favourable Pt kinetics flotation was observed at higher impeller speeds when floating ultrafine PGEs ( $-5\ \mu\text{m}$  moncheite) and this was attributed to high power inputs, or highly turbulent contacting, improving the rate of flotation of fine particles. At constant air flow, the bubble size decreases when power intensity is increased, and this will also lead to improved particle–bubble attachment, thus improving the fines recovery. However, as will be noted below, it is necessary to maintain a careful balance between power input and bubble size. Matis et al. [53] showed that a smaller bubble size is required when floating fines to maintain a suitable probability of collision. Feng and Aldrich [49] noted that fine particles reported to the concentrate mainly by entrainment and this phenomenon is dealt with below. High bubble loading and high froth stability was observed for fine particles compared to the coarse and medium fractions. However, medium sized particles gave better flotation results. Small and stable bubbles were observed when floating medium sized particles. It was concluded that good results obtained were also influenced by the froth structure.

Design principles for good fines flotation according to Pease et al. [5,22] incorporate a staged grind-float performance. It is important to note that Figure 5a assumes that a wide range of hydrophobic particle sizes are being floated, while the same recovery profile remains and the optimum size fraction shifts depending on the size range of the particles present in the system (Figure 5b) [5,22,54]. It is therefore possible to control the flotation of fine particles by controlling the particle size distribution being floated. If the size range is narrowed towards the fine particle range, more fines will be recovered. This is the basis for the use of MF2 or MF3 circuits, referred to above, which enable the stagewise recovery of different classes of particle sizes.



**Figure 5.** (a) Recovery as a function of particle size, showing three defined regions. (b) Recovery of fines when treated separately from the other size fractions [5,22,54].

With respect to reagents when treating fines, Corin et al. [55] showed that each collector performs a different action and therefore may target different sized particles. It was shown that when dithiophosphate (DTP) is used as a co-collector with sodium isobutyl xanthate (SIBX), the recovery of  $<10\ \mu\text{m}$  particles can be increased. Shackleton et al. [3] also noted that as collector dosage is increased, an increase in recovery of particles in the  $<5\ \mu\text{m}$  size fraction is seen. These observations are of particular importance in the flotation of PGMs since a significant amount of the valuable minerals such as Pt and Pd report to such size fractions. The use of DTP increased the recovery of Pt and more so Pd. A case study conducted on a Platreef ore [20], in which collectors containing [SIBX + DTP mixtures] and [SIBX + DTC mixtures], each of varying ratios, were evaluated, showed that [SIBX + DTP mixtures] specifically targeted the PGM sulph-antimonides and sulphur-arsenides while [SIBX + DTC mixtures] targeted the platinum sulphide and telluride minerals. More importantly for the middling's material, the [SIBX + DTP mixtures] reagent regime targeted the PGE silicates while the [SIBX + DTC mixtures] targeted the PGE sulphides. Similar trends were observed for Merensky and UG2 ores. These results have shown a strong correlation between PGE mineral type and collector type.

Froth stability is significantly impacted by particle size. Fine hydrophilic particles have been noted to render bubbles less likely to coalesce and thereby promoting a stable froth, leading to an increase in fine particle recovery [56,57]. Finely sized particles can become entrapped between two adjacent bubbles which can either enhance or reduce froth stability dependent upon the particle properties: shape, roughness, concentration, hydrophobicity, and the thin film formation rate [58]. The relationship between froth stability and entrainment is discussed in the next section.

The use of novel cell technology to enhance fines recovery has also recently enjoyed considerable interest. In a recent study using a pilot-scale oscillating grid column (OGC) investigating PGM flotation in a Platreef ore, it was concluded that when using normal flotation-size bubbles, increasing power input increases the rate of flotation of finer particles ( $-25\ \mu\text{m}$ ) and an optimum rate of flotation of intermediate ( $+25\text{--}53\ \mu\text{m}$ ) and coarse particles ( $+53\text{--}75\ \mu\text{m}$ ). However, for microbubbles, increasing power input leads to a decrease in the rate of flotation for all particle sizes. It was found that optimum conditions for fine particle PGM flotation require microbubbles at low power inputs or normal bubbles at high power inputs [59].

## 5. Entrainment

The degree of entrainment of mineral particles, defined as the ratio of the recovery by entrainment to the recovery of water, is well known to increase exponentially with

decreasing particle size [42]. The size at which particles become entrained is dependent on their density and shape, but as a rule occurs at about  $<50\ \mu\text{m}$  [60]. Since PGM flotation requires that particles are ground very fine in order to liberate them, particularly in the secondary circuit, entrainment is problematic in PGM flotation. Particles feeding the secondary roughers may be anywhere between 40% to 80% less than  $50\ \mu\text{m}$ . This means that up to 80% of particles could theoretically be entrained to some degree. Lima et al. [61] showed the dramatic effect that particle size has on the degree of entrainment. For example, the hematite degree of entrainment of particles increased by about eight times when the mean particle size decreased from  $45$  to  $25\ \mu\text{m}$ . Considering that PGM grain sizes average about  $7\ \mu\text{m}$ , since liberation requires a very fine grind, it is likely that this will result in significant degree of entrainment. Obviously, entrainment recovery dilutes the grade of the concentrate and should be controlled in order to improve grade and reduce mass pull. However, it is especially important in the processing of UG2 ores, where about 70% (by mass) of the feed consists of chromite which is problematic since in the downstream smelting of PGM ores, chromite needs to be kept below about 2.5% in the feed to the furnace [62].

The amount of material recovered by entrainment to the concentrate depends not only on the particle size but on the operation of the cells and the circuit. Higher air flow rates, frother concentrations, pulp solids concentration, and shallow froth depths all contribute to higher recovery by entrainment. This was illustrated in a study of the interactive effects of dosage of frother and depressant, air flow rate, and froth height on a UG2 ore [63]. This showed that the superficial gas velocity had the greatest effect on recovery by entrainment of the chromite, with froth height and frother concentration being the other two significant factors. The study was done at froth heights ranging between 180 and 300 mm and superficial gas velocities between 0.5 and 1.5 cm/s in order to mimic standard froth residence times under plant conditions. Results from tests conducted in a modified, continuous laboratory flotation cell with 150 mm froth depth have shown that between 50% and 70% of total gangue recovered to the concentrate did so as a result of entrainment. This is even though much of the pyroxene is recovered by true flotation due to finely disseminated talc within the pyroxene [64].

Hay and Roy [62] illustrated the effects of the head grade (or solids concentration) on recovery by entrainment by showing significant differences in chromite recovery between different head grades.  $\text{Cr}_2\text{O}_3$  head grades of between 25–31% produced recoveries of up to 18%, whereas head grades of between 14–20% only produced recoveries of up to about 6%. This was also well illustrated by Valenta [10] who showed that solids recovery by entrainment increased fourfold with a threefold increase in pulp solids concentration. This corresponded with 2.5 times reduction in PGM grade, which was attributed to increased gangue entrainment. This shows the benefits of operating at lower percent solids, which is also carried through to improvements in valuable mineral recovery due to improved rheological behaviour [65,66]. Obviously, this comes with a loss in residence time or throughput and these different requirements need to be balanced.

Circuit design and residence time is critical in limiting recovery by entrainment of fine particles. Many plants employ multiple stages of cleaning [10], while others bypass cleaning altogether and send rougher concentrates directly to final concentrate. This is unusual and detrimental to grade, as there is usually at least one stage of cleaning to allow for removal of entrained gangue. Another method to deal with the removal of fines by entrainment is froth washing [5,22]. However, this is seldom used in the PGM industry due to the potential for loss of PGMs arising from such treatment. Another method of limiting entrainment, which is utilised for different ore types, is the use of column flotation cells. These cells have deep froths which allow for long froth drainage times, thus limiting recovery by entrainment. In laboratory scale column flotation tests, it was found that froth height had little effect on the water or solids recoveries or on the grade of chromite in the concentrate possibly due to the density of chromite particles [67]. However, columns are seldom used in the PGM industry presumably due to the low energy input which may limit

bubble particle collision efficiency for the very fine particle sizes produced during PGM liberation and the inability to maintain the deep froths required for column cells. There are a few examples of column cells within the PGM industry, one of which is discussed by Hay [68]. Two column cells were installed as final cleaners on a PGM plant treating UG2 ore and this reduced the  $\text{Cr}_2\text{O}_3$  grade from 4.0% to 2.2%. There was, apparently, no difficulty in maintaining the deep froths required. This is, perhaps, not surprising because, although there are very little floating valuable minerals in UG2 ores, there are relatively large amounts of floatable gangue with which to stabilise the froth. The external sparger on the column cell was driven by a recirculation pump, imparting energy for improved bubble–particle collision, which improved PGM recovery by 6%.

Residence time is another critical factor in limiting recovery by entrainment in fine particle PGM flotation. It is crucial to match the residence time of each bank to the kinetics of the valuable mineral flotation. The residence time of the bank should be just long enough to achieve the desired valuable mineral recovery and not so long as to allow prolonged recovery of entrained and slow floating gangue species. This is usually achieved in the rougher banks but may be very poor in cleaner banks. This is well illustrated by Lima et al. [61] in a discussion of matching the residence time of banks to the relative kinetics of quartz and hematite, which is analogous to the PGM and chromite recovery by true flotation (quartz and PGMs) and entrainment (hematite and chromite). It is worth noting that there are options such as the use of gravity concentration techniques to manage chromite recovery [7,69].

As mentioned above, the abundance of fine particles in PGM flotation processing has an impact on the froth structure. Achaye et al. [70] showed that there is an exponential increase in froth stability for feed particle sizes below about 50  $\mu\text{m}$ . This was driven, in large part, by the increase in particle surface area recovered to the froth by entrainment, which resulted in a more viscous and more stable froth. This will be true for PGM processing where much of the feed, particularly after secondary and tertiary grinding, is less than 50  $\mu\text{m}$  in size. This drives a circular dependence, where more fines lead to more entrainment, which produces a more stable froth that, in turn, leads to more entrainment.

Reagents, of course, influence both froth stability and entrainment. McFadzean et al. [71] showed that mixtures of alcohol and polypropylene glycol frothers could improve both recovery and grade compared to the sum of their constituents in a UG2 ore. The froth effects were decoupled from the pulp effects to explain this. It was found that the frother blends could improve froth stability while also reducing entrainment. This resulted in improved recovery, due to improved froth stability and improved grade due to less entrainment. Another study by Nyabeze and McFadzean [72] showed that the use of  $\text{CuSO}_4$  had a similar effect on the froth and not on the pulp phase recovery, as expected. This showed that the addition of  $\text{CuSO}_4$  destabilised or improved the drainage effect on both Merensky and UG2 ores. At full scale plant performance this translated to improved PGM grades with very slight decreases in recovery. The difference in degree of entrainment was clear, with lower degrees of entrainment observed in the absence of  $\text{CuSO}_4$  compared to when  $\text{CuSO}_4$  was used.

## 6. Rheology and Slimes Coatings

Rheological and slimes coatings problems in PGM flotation are generated from similar causes, mainly through the production of ultrafine phyllosilicate mineral particles during grinding. Although they have similar causes, the mechanisms of their impact in the concentrator circuit are different and they will be dealt with separately below. The methods whereby these detrimental effects can be mitigated does, however, overlap to some extent and will be dealt with in Section 6.3.

### 6.1. Rheology

Rheology influences the flow of slurry through mills, the separation efficiency of hydrocyclones, the hydrodynamics of flotation cells, the stability/mobility of flotation

froths, and the dewatering and pumping of slurries [73]. The presence of ultrafine particles has a dominating impact on rheology, particularly at high volumetric solids concentrations. In general, slurries display non-Newtonian behaviour, characterised by a yield stress and a viscosity, both of which increase exponentially with decreasing particle size. However, despite the important influence of particle size and solids concentration on rheology, factors such as particle morphology, mineralogy, and pulp chemistry also have a significant influence. For pulp chemistry, the pH has a strong influence as it affects the mineral surface charge distribution and the associated aggregation or dispersion of particles. Particle morphology and mineralogy can also have an important influence on slurry rheology. Studies on particle mineralogy and rheology have noted phyllosilicate gangue minerals, particularly swelling clays and serpentine minerals, as major contributors [74,75]. Studies on various platinum ores have shown that rheologically complex behaviour is dependent on the amount of phyllosilicate alteration minerals, with ores such as Platreef and Great Dyke being potentially problematic [76,77].

Rheology has an influence on most of the processing steps on platinum concentrators. For tumbling mills, rheological complexity affects the retardation of grinding within the ultrafine regime [78] and slurry transport through the mill [79]. Rheology is also one of the most important considerations for the performance of ultrafine grinding in stirred mills, such as IsaMills [73]. Rheology has a significant influence on hydrocyclone performance as here the flow behaviour is extremely complex [80]. For flotation, rheological complexity influences flotation cell hydrodynamics which may result in poor gas dispersion [81,82], increased turbulence damping, and lower flotation rates [83]. However, a higher pulp viscosity can improve the stability of bubble–particle aggregates [84]. Rheology also affects the behaviour of flotation froths, such as froth mobility and froth stability. Froth mobility refers to the general motion of the froth while froth stability includes processes occurring within the froth such as bubble bursting and drainage of water and solids. While several studies have attempted to model froth recovery in flotation cells [85–88], only a few have focused on the effect of rheology on froth behaviour [89–93]. Results have indicated that froth properties are correlated with rheological measurements, but the underlying mechanisms are not clear due to the complexity of froth behaviour. Rheological complexity also affects downstream processes such as dewatering, tailings disposal, and storage [73]. Tailings dams are sensitive to rheology as it affects both the rate of settling and the slurry compressibility, potentially resulting in instability [94].

### 6.2. Slimes Coatings

Slimes coatings are the ultrafine (less than 10  $\mu\text{m}$ ) gangue mineral particles that coat the surface of valuable minerals. These slimes are generated from the same minerals that create a problematic response in viscosity, the phyllosilicate minerals, which are soft minerals and prone to overgrinding. Slimes are attracted to the surface of the valuable minerals by means of an electrostatic attraction. Shackleton et al. [95,96] measured the zeta potential of sperrylite and moncheite, two common PGM minerals, and found them to be  $-17$  and  $-22$  mV, respectively. Serpentine, talc, and chlorite, three common altered phyllosilicate minerals present in PGM ores, are known to have both positively and negatively charged planes, which can interact electrostatically with the negatively charged PGMs [97,98]. This results in poor collector adsorption and increased hydrophilic behaviour of the valuable PGMs.

### 6.3. Mitigation of Rheological and Slimes Effects in PGM Flotation

Both physical and chemical methods can be used to mitigate the effects of complex rheology and slime coatings due to ultrafine particles. While much work has been done on these subjects in other commodities, little work has been done in the PGM industry, which is perhaps surprising, considering the altered phyllosilicates present in many PGM ores.

For rheology, the simplest method of mitigating rheological complexity is to reduce the ultrafine particles generated by more efficient comminution and classification. Operating

at lower volumetric solids concentrations is also possible in areas such as flotation, but generally not in milling and tailings disposal. Effective gangue management can also be considered by on-going monitoring of the concentrator feed for phyllosilicate mineral content as platinum ores show considerable local geological variation and alteration [77]. If these physical methods are not viable, then chemical methods may be considered, such as changing the pulp pH or using chemical dispersants and rheology modifiers. These methods generally rely on reducing the interparticle interaction force to promote dispersion [99]. Changing the pulp pH, to promote electrostatic repulsion, is generally not viable on platinum concentrators. The use of grinding aids, such as polyelectrolytes, to improve rheology in mills has been studied for many years due to the high solids concentrations [100,101]. However, research on rheology modifiers within the context of flotation is not as extensive but has increased in recent years with the complexity of ores. In general, the use of dispersants and rheology modifiers can aid in the dispersion of the slimes, improve rheology, and thereby improve process efficiency [84,97,102].

For slimes coatings, both physical and chemical investigations have been conducted to mitigate the effects on PGM flotation. Ross et al. [103] investigated the use of a high shear hydrodynamic cavitation device to physically remove slimes from valuable mineral particles. They found that the use of this device significantly improved the grade and recovery of rougher, cleaner, and recleaner feed material for a UG2 tailings retreatment plant. Molifie et al. [97] found that the use of sodium silicate was effective in improving recovery and grade of a PGM ore containing large quantities of serpentine and talc. This study decoupled the various effects of the sodium silicate and found that it had a beneficial effect on reversing the surface charge of the serpentine mineral, thus reducing the electrostatic attraction, as well as reducing viscosity and depressing the naturally floatable talc. There is scope for further investigation of the effects of slimes mitigating factors in PGM flotation.

## 7. Conclusions

This study has reviewed the problems and challenges associated with the presence of fines in the processing of PGMs. In the treatment of such ores, the need to grind to fine particle sizes is probably more important than in the treatment of any other ores due to economic value of the minerals being recovered and hence the need to ensure adequate liberation to optimize recovery. The application of sophisticated mineralogy to quantify the liberation throughout the circuit is of paramount importance. It is critical to ensure that there is a careful design of flotation circuits as well as the need to use novel comminution and flotation devices as well as appropriate reagent suites. Finally, the need to manage the effect of the generation of fines on factors such as entrainment, rheology, and slimes coatings is of critical importance. The paper addresses all these issues and indicates how they are being handled on modern concentrators.

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