Phosphate Leaching from Green Roof Substrates—Can Green Roofs Pollute Urban Water Bodies?

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Abstract: Green roofs are an effective stormwater measure due to high water retention capacity and the ability of delaying stormwater runoff. However, low importance is still given to the pollutant leaching potential of substrates used in green roof construction. The aim of the study is to estimate the concentrations and loads of P-PO$_4^{3-}$ in runoff from extensive and intensive substrates. To achieve this goal, several commonly-used fresh substrates were analyzed for P-PO$_4^{3-}$ leaching potential in different scale experiments, from laboratory batch tests, leaching column experiments, and long-term monitoring of open air green roof containers. The results of the study confirmed that fresh green roof substrates contain phosphorus in significant amounts of 17–145 mg·P-PO$_4^{3-}$/kg and, thus, can contribute to eutrophication of freshwater ecosystems. High correlation between phosphate content estimated by HCl extraction and cumulative load in leachate tests suggests that the batch HCl extraction test can be recommended for the comparison and selection of substrates with low potential P leaching. Volume-weighted mean concentrations and UALs of P-PO$_4^{3-}$ leaching from fresh substrates were higher in cases of intensive substrates, but there was no clear relationship between substrate type and the observed P-PO$_4^{3-}$ concentration range. To avoid increasing eutrophication of urban receivers the implementation of P reduction measures is strongly recommended.

Keywords: phosphorus; green roofs; substrates; eutrophication; urban water bodies

1. Introduction

Green roof substrates are an artificial mixture of compounds designed to provide proper conditions for plant growth and rainwater retention. The discussion if they are a source or sink of runoff pollutants is still active [1]. Among other pollutants, phosphorus (P) is almost universally found in higher concentrations in green roof leachate than in runoff from conventional roofs [2,3]. Widespread application of green roof technology could, therefore, lead to increases in P levels entering urban waterways [2].

Most green roof substrates are a mixtures of mineral and organic compounds. The most popular mineral components are: clay, sand, volcanic materials, crushed brick, and expanded lightweight materials, and among organic components, the most popular are: compost and peat [4]. Compost typically has very high P content. Buffam and Mitchell [5] obtained the value of leachable phosphate of above 200 mg·P-PO$_4^{3-}$/kg in deionized water extract. Toland et al. [6] reported that SRP concentrations in runoff from green roofs with compost were more than 10 times greater than SRP concentrations without compost added. Organic components of the substrate have often been suggested as a source of P-PO$_4^{3-}$ to green roof runoff (e.g., [7]). Growing media with 50% to 60% organic matter had the highest leachate concentrations of phosphorus [8]. Organic matter included in the substrate is beneficial for
plant growth, but when it decomposes it may leach nutrients [9]. It was stated that the addition of 10% organic matter is optimal for the substrate and plants, and organic matter exceeding 25% is not recommended for extensive green roofs [10]. Often practice in green roof substrate preparation is the integration of slow-release inorganic fertilizers [8,11]. Malcolm et al. [12] suggested to decrease fertilizer use during installation in order to reduce the negative impacts on water quality during the first few years. Organic matter content and fertilizers are well described sources of phosphorus, however, mineral components, e.g., commonly used crushed red brick, sand, or gravel can also be a significant sources of phosphorus in green roof runoff [4].

Phosphorus in vegetated roof runoff appears mainly in phosphate form [13,14], and P-PO$_{4}^{3-}$ concentrations vary from low, e.g., 0.003–0.079 mg/P-PO$_{4}^{3-}$/L [15] or 0.006–0.012 mg/P-PO$_{4}^{3-}$/L [16] to medium, e.g., 0.27–0.37 mg/P-PO$_{4}^{3-}$/L [17] or high (above 1 mg/P-PO$_{4}^{3-}$/L) (e.g., [18–21]). Due to increasing problems with stormwater management in urban areas, the total area of green roofs is going to rise in the future [22,23]. Extensive green roofs have thin substrate layers and need little maintenance, while intensive green roofs have thicker substrates and require more maintenance [24]. Intensive green roofs can accommodate various plant types, while on extensive green roofs plant selection is limited [25]. Substrates are usually designed to retain rain water and support plant growth, whereas nutrient retention is still of low importance. Most of substrates used in green roof construction are potential sources of P in runoff, and the difference can lay in the amount of P released. Substrate composition, depth, age, fertilizer use, plant species and coverage, local air quality, and the intensity of precipitation events are the factors shaping P runoff from green roofs [4,14].

The goals of this study were: (i) to confirm that green roof substrates are a significant source of P-PO$_{4}^{3-}$ in runoff; (ii) to estimate the concentrations and loads of P-PO$_{4}^{3-}$ leaching from extensive and intensive substrates; and (iii) to demonstrate the necessity of implementation of P-protection measures to limit green roofs’ contributions to eutrophication of urban water bodies.

2. Materials and Methods

2.1. Materials

Five green roof substrates collected from a local market were named as S, followed by a number (S1–S5). S1–S3 (intensive type) and S5 (extensive type) are fresh substrates sampled from the newly-constructed green roofs or from large bags before implementation. S1 is an intensive substrate predicted for lawns or small shrubs. It consists of lightweight aggregates, mineral aggregates, sand, compost, low peat, and it was fertilized before application. S4 is a growing medium sampled from the fresh prefabricated Sedum mate (Xelo Flor moss-sedum-herbs XF317). Substrates S1–S3 are mixtures of mineral and organic compounds, while S5 is a 100% mineral mixture of crushed red brick, gravel, lime, and sand. For the substrates S2–S3 we were not able to obtain specifications, we only obtained the manufacturer’s warranty that they are FLL (Forschungsgesellschaft Landschaftsentwicklung Landschaftsbau) [26] or DAFA (Stowarzyszenie Wykonawców Dachów Płaskich i Fasad) [27] compliant. Substrate S6 used in long-term experiments consisted of washed sand, chalcedonite, clay, low peat, and compost. It is an intensive type of substrate for use in multi-layered green roof systems. The characteristics of the tested substrates are set in Table 1.
Table 1. Characteristics of substrates used in the study.

<table>
<thead>
<tr>
<th>Substrate</th>
<th>S1</th>
<th>S2</th>
<th>S3</th>
<th>S4</th>
<th>S5</th>
<th>S6 (2013)</th>
<th>S6 (2017)</th>
</tr>
</thead>
<tbody>
<tr>
<td>type</td>
<td>intensive</td>
<td>intensive</td>
<td>intensive</td>
<td>extensive</td>
<td>extensive</td>
<td>intensive</td>
<td>intensive</td>
</tr>
<tr>
<td>age</td>
<td>fresh</td>
<td>fresh</td>
<td>fresh</td>
<td>fresh</td>
<td>fresh</td>
<td>fresh</td>
<td>fresh</td>
</tr>
<tr>
<td>composition</td>
<td>mineral-organic</td>
<td>mineral-organic</td>
<td>mineral-organic</td>
<td>no data</td>
<td>mineral</td>
<td>mineral-organic</td>
<td>mineral-organic</td>
</tr>
<tr>
<td>pH</td>
<td>7.31</td>
<td>7.19</td>
<td>7.60</td>
<td>8.03</td>
<td>7.74</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>OM content [%]</td>
<td>10.4</td>
<td>7.0</td>
<td>7.4</td>
<td>7.2</td>
<td>0</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>bulk density [kg/m³]</td>
<td>1054.8</td>
<td>1051.1</td>
<td>983.4</td>
<td>1145.6</td>
<td>1498.7</td>
<td>1430.0–1570.0</td>
<td>1560.0–1755.0</td>
</tr>
</tbody>
</table>

2.2. Methods

2.2.1. Extraction (Batch Test)

Leachable phosphates: triplicate samples of dried substrates were placed in an extractor vessel made of inert material and distilled water, and were agitated for 24 h at 135 rpm at room temperature. Phosphorus content: samples of dried substrates were placed in a vessel and a quantity of 1nHCl equal to two times the weight of the solid phase was added. Agitation time and environment were similar to water extraction.

For pH, PN-ISO standard was used [28], which specifies an instrumental method for the routine determination of pH using a glass electrode in a 1:5 (v/v) suspension of soil in water. The pH was measured by a Volcraft PH-212 meter. Organic matter (OM) content (%) was determined using the loss of ignition at 550 °C.

2.2.2. Leaching (Column) Experiment

Five columns of a diameter of 144 mm were filled with substrates S1–S5, as described in Section 2.1. The thickness of each substrate was 4 cm, regardless the type of the substrate. The use of different thicknesses for intensive and extensive substrates would change the retention capacity and thus, reduce the volume of the outflow, and make the results of the experiment difficult to compare. All columns were watered before the experiment. Then the substrates were irrigated with tap water for 90 days, in doses and a schedule developed on the basis of the precipitation observed at a nearby meteorological station (52°16′07″ N, 21°04′59″ E) in 2013. Within the time of the leaching test, 470 mm of water was supplied to each column, which is the value similar to natural rainfall observed in the vegetation period in the open air long-term experiment. The volume of leachate was measured manually after each simulated precipitation. The cumulative load of the P-PO₄³⁻ supplied with the tap water and discharged from the substrates was calculated based on volumes of irrigation rates, leachate volumes, and P-PO₄³⁻ concentrations.

2.2.3. Long-Term Green Roof Monitoring (Open Air Experiment)

The green roof substrate (intensive type, S6) was placed in plastic container of dimensions of 0.5 × 0.3 × 0.3 m (length, width, height) in early spring, 2013. The substrate had a thickness of 17 cm and was underlined with the drainage layer made of washed gravel with a thickness of 8 cm and Polyfelt TS 20 filtration geomembrane. From April 2013 to May 2015, a bare substrate was tested, but due to the occurrence of colonizers, from June 2015 Sedum plants obtained from the dismantled full-scale nine year-old green roof were introduced to the container. No fertilizers were applied to the green roof during construction, nor the monitoring period. Plant cover in years 2015–2017 were not well developed due to the lack of irrigation and fertilization (Figure 1), therefore, the role of plants was not discussed in this study.
The precipitation data for monitoring period were obtained from a nearby meteorological station (52°16′07.16″ N, 21°04′89.84″ E). Leachates from the model were collected in tanks and sampled after each precipitation event. The measurement of the volumes of the leachates and sampling were made manually. In this study the results from the first, second, and fifth year of monitoring of P-PO$_4$ leaching from green roof substrates are presented. The third year of the experiment (2015) was very dry. No precipitation occurred in April–May, and the total precipitation in the vegetation period was lower than 190 mm. That was the reason why leachates were not collected and can explain poor plant development on the substrate. The monitoring period of the green roof container in 2016 covered the months from August to December and, thus, could not be included in the comparison of green roof performance in the vegetation periods.

All extracts, samples of tap water, rainwater, and leachates collected from the column experiment and open air models were filtered and analyzed for P-PO$_4$$^{3−}$ on a FiaStar analyzer by the ammonium molybdate method in the range of 0.005–1 mg/L. All P-PO$_4$$^{3−}$ loads obtained from the above tests were recalculated to unit loads and expressed in mg per m$^2$ as unit area load (UAL) to make the results of this study comparable with other studies. As in some samples of rainwater and leachates no phosphates were detected, the volume-weighted mean concentration ($C_{vw}$) was used to characterize rainfall and leachates. The volume-weighted mean concentration was calculated using:

$$C_{vw} = \frac{\sum_{i=1}^{n} (V_i \cdot C_i)}{\sum_{i=1}^{n} V_i}$$

where $V_i$ = total volume of runoff measured for event $i$ (L); $C_i$ = concentration of the pollutant for event $i$ (mg/L).

The unit area load (UAL) is the total load divided by the green roof area (mg/m$^2$), and was calculated using the following equation:

$$UAL = \frac{\sum_{i=1}^{n} (V_i \cdot C_i)}{A}$$

where $A$ = area of green roof (m$^2$); $V_i$ and $C_i$ as described above. For statistical analyzes Statgraphics Centurion XVI software was used.

3. Results

3.1. Extraction

Leachable phosphate content of all tested substrates were low and varied between 0.1 (S6) and 6.7 (S2) mg·P-PO$_4$$^{3−}$/kg (Figure 2). However, total phosphate content in substrates was significant. In general, higher phosphate content was noted for mixtures of mineral and organic compounds (S1–S3...
and S6 fresh) than in case of mineral substrate (S5) and Sedum mat growing medium (S4). Intensive substrates (S1–S3 and S6 fresh) were characterized by higher phosphorus content than extensive substrates. Fresh substrate S6 used in open air experiments had four times higher P content than the same substrate after five years of being used, which was the result of phosphate leaching due to precipitation. There was no significant correlation found between total P-PO$_4^{3-}$ content (mg/kg) extracted by using hydrochloric acid (HCl) and leachable P-PO$_4^{3-}$ extracted by using distilled water (H$_2$O) ($p < 0.005$).

Figure 2. Total P-PO$_4^{3-}$ content (mg/kg) in substrates extracted by using hydrochloric acid (HCl) and leachable P-PO$_4^{3-}$ extracted by using distilled water (H$_2$O) in the batch test (mean value and standard deviation).

3.2. Leaching Test (Column Experiment)

Within 90 days of the leaching test, 470 mm of tap water was supplied in the form of simulated precipitation to each column (Table 2), which was the value similar to the natural rainfall observed in the vegetation period in the open air long-term experiment (see Section 3.3). Within this period, 44 precipitation events were simulated of which 11 were polluted with phosphates. Phosphates occurred also in tap water, as they are added as corrosion inhibitors in some water supply networks and its concentration is not limited in drinking water [23,29–31]. The volume-weighted mean concentration of P-PO$_4^{3-}$ in the simulated precipitation was 0.035 mg P-PO$_4^{3-}$/L (Table 2). In most cases observed phosphate concentrations were low, with the exceptions on day 21, 34, and 38, when the concentrations of 0.393, 0.943, and 0.207 mg P-PO$_4^{3-}$/L were noted, respectively (Figure 3).

Table 2. Experimental data of column leaching experiment.

<table>
<thead>
<tr>
<th>Rainfall</th>
<th>Precipitation [mm]</th>
<th>470.0</th>
</tr>
</thead>
<tbody>
<tr>
<td>Concentration of P-PO$_4^{3-}$ [mg/L] in rainwater samples: mean * (min-max)</td>
<td>0.035 (0–0.943)</td>
<td></td>
</tr>
<tr>
<td>UAL of P-PO$_4^{3-}$ [mg/m$^2$] in precipitation</td>
<td>16.644</td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Runoff</th>
<th>Substrate</th>
<th>S1</th>
<th>S2</th>
<th>S3</th>
<th>S4</th>
<th>S5</th>
</tr>
</thead>
<tbody>
<tr>
<td>Volume of leachate [mm]</td>
<td>330.7</td>
<td>308.3</td>
<td>323.9</td>
<td>308.8</td>
<td>346.5</td>
<td></td>
</tr>
<tr>
<td>Concentration of P-PO$_4^{3-}$ [mg/L] in leachate samples: mean * (min-max)</td>
<td>0.116 (0–0.769)</td>
<td>0.242 (0–0.791)</td>
<td>0.286 (0–3.169)</td>
<td>0.050 (0–0.961)</td>
<td>0.060 (0–0.974)</td>
<td></td>
</tr>
<tr>
<td>UAL of P-PO$_4^{3-}$ [mg/m$^2$ of substrate]</td>
<td>37.6</td>
<td>73.2</td>
<td>90.8</td>
<td>15.1</td>
<td>20.4</td>
<td></td>
</tr>
<tr>
<td>Unit load of P-PO$_4^{3-}$ [mg/kg of substrate]</td>
<td>0.908</td>
<td>1.773</td>
<td>2.351</td>
<td>0.337</td>
<td>0.346</td>
<td></td>
</tr>
</tbody>
</table>

Note: * Volume-weighted mean concentration.
3.2. Leaching Test (Column Experiment)

Within 90 days of the leaching test, 470 mm of tap water was supplied in the form of simulated precipitation to each column (Table 2), which was the value similar to the natural rainfall observed in the vegetation period in the open air long-term experiment (see Section 3.3). Within this period, 44 precipitation events were simulated of which 11 were polluted with phosphates. Phosphates occurred also in tap water, as they are added as corrosion inhibitors in some water supply networks and its concentration is not limited in drinking water [23,29–31]. The volume-weighted mean concentration of P-PO$_4^{3-}$ in the simulated precipitation was 0.035 mg·P-PO$_4^{3-}$/L (Table 2). In most cases observed phosphate concentrations were low, with the exceptions on day 21, 34, and 38, when the concentrations of 0.339, 0.943, and 0.207 mg·P-PO$_4^{3-}$/L were noted, respectively (Figure 3).

Phosphates were detected in most leachates from all tested substrates (Figure 3). The volume-weighted mean concentration of P-PO$_4^{3-}$ in leachates ranged from 0.050 mg·P-PO$_4^{3-}$/L (S4) to 0.286 mg·P-PO$_4^{3-}$/L (S3) (Table 2). Phosphate concentrations in leachates from most of the columns were not significantly correlated with the volumes of precipitation, volumes of leachate, or the concentration of phosphates in simulated precipitation (Table 3).

The UAL of phosphate supplied to tested substrates with simulated precipitation amounted 16.644 mg/m$^2$ and was lower than the UALs in leachates from all substrates except, Sedum plant-growing medium (S4). That means that both intensive- and extensive-type green roof substrates were the source of phosphorus and could potentially contribute to eutrophication of freshwater ecosystems. Cumulative load of phosphorus leached from the substrates during the time of experiment varied from 0.337 to 2.351 mg/kg (Table 2). Intensive substrates (S1–3) released more phosphates than extensive (mineral) substrate (S5), which was not a significant source of P-PO$_4^{3-}$ if compared with the amount of P-PO$_4^{3-}$ supplied.
Table 3. Results of correlation analyses between phosphate concentrations in leachate S1–S5 and the volumes of precipitation, volumes of leachate, and concentration of phosphates, presented as Pearson correlation coefficient $r$ and probability $p$ (in brackets). $p$-values below 0.05 indicate statistically significant non-zero correlations at the 95.0% confidence level.

<table>
<thead>
<tr>
<th>Concentration of Phosphates in Leachate</th>
<th>Volume of Precipitation</th>
<th>Volume of Leachate</th>
<th>Concentration of Phosphates in Simulated Precipitation</th>
</tr>
</thead>
<tbody>
<tr>
<td>S1</td>
<td>0.3218 (0.0332)</td>
<td>0.2935 (0.0531)</td>
<td>0.1034 (0.5041)</td>
</tr>
<tr>
<td>S2</td>
<td>−0.0643 (0.6784)</td>
<td>0.0139 (0.9286)</td>
<td>0.1742 (0.2580)</td>
</tr>
<tr>
<td>S3</td>
<td>0.0484 (0.7551)</td>
<td>0.0737 (0.6347)</td>
<td>0.7847 (0.0000)</td>
</tr>
<tr>
<td>S4</td>
<td>−0.1059 (0.4938)</td>
<td>−0.1004 (0.5168)</td>
<td>0.0068 (0.9651)</td>
</tr>
<tr>
<td>S5</td>
<td>0.0519 (0.7441)</td>
<td>0.0238 (0.8810)</td>
<td>0.9543 (0.0000)</td>
</tr>
</tbody>
</table>

3.3. Long-Term Open Air Green Roof Container Monitoring

Over the entire monitoring period, covering vegetation seasons 2013 (first year), 2014 (second year), and 2017 (fifth year), the green roof container received rainfall from 338.1 mm to 539.6 mm, while runoff volumes ranged from 106.8 to 158.7 mm (Table 4). The green roof model retained 64.6% to 80.2% of the precipitation. The volume-weighted mean concentration of P-PO$_4^{3-}$ in the precipitation in the first two years of monitoring were lower than the concentrations observed in the green roof runoff, but it was opposite in the fifth year of the observation period. The range of the observed P-PO$_4^{3-}$ concentrations in precipitation and the runoff from the green roof model over the entire monitoring period was similar (Table 4).

Table 4. Experimental data of long-term open air green roof monitoring.

<table>
<thead>
<tr>
<th></th>
<th>2013 (1st Year)</th>
<th>2014 (2nd Year)</th>
<th>2017 (5th Year)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Observation period</td>
<td>IV–IX</td>
<td>IV–VIII</td>
<td>IV–IX</td>
</tr>
<tr>
<td>Precipitation [mm]</td>
<td>448.9 **</td>
<td>338.1 **</td>
<td>539.6</td>
</tr>
<tr>
<td>Concentration of P-PO$_4^{3-}$ [mg/L] in rainwater samples: mean * (min–max)</td>
<td>0.018 (0–0.274)</td>
<td>0.012 (–0.249)</td>
<td>0.043 (–0.213)</td>
</tr>
<tr>
<td>UAL of P-PO$_4^{3-}$ [mg/m$^2$] in precipitation</td>
<td>8.27</td>
<td>4.00</td>
<td>23.47</td>
</tr>
<tr>
<td>Volume of leachate [mm]</td>
<td>158.7</td>
<td>119.0</td>
<td>106.8</td>
</tr>
<tr>
<td>Concentration of P-PO$_4^{3-}$ [mg/L] in leachate samples: mean * (min–max)</td>
<td>0.075 (–0.229)</td>
<td>0.090 (–0.281)</td>
<td>0.026 (–0.223)</td>
</tr>
<tr>
<td>UAL of P-PO$_4^{3-}$ [mg/m$^2$ of substrate]</td>
<td>12.00</td>
<td>10.73</td>
<td>2.80</td>
</tr>
</tbody>
</table>

Notes: * Volume-weighted mean concentration. ** Calculated from the time of the first leachate collected to the time of the last leachate collected.

Phosphates were detected in most leachates from tested substrate in all monitored vegetation seasons (Figure 4). The range of observed P-PO$_4^{3-}$ concentrations was similar in first, second, and fifth years of observation (Table 4). The volume-weighted mean concentration in runoff was higher in the second year of operation than in the first year, but in the fifth year of operation the volume-weighted mean concentration in runoff was three times lower than observed in the first year. Despite the highest load of P-PO$_4^{3-}$ supplied to the model with precipitation in the fifth year of the experiment, the cumulated load of P-PO$_4^{3-}$ released from the substrate was four times lower than values noted in the first and second years of model monitoring. Correlation analyses were applied over all data combined to assess the relationships between phosphate concentration in leachates and the volume of precipitation, volume of leachate, and concentration of phosphates in precipitation. Only in the
second year of the experiment, phosphate concentrations in green roof leachate were correlated with
the volume of precipitation and the volume of leachate (Table 5).

Figure 4. Concentration of P-PO$_4^{3-}$ in precipitation and leachates from substrate tested in long-term
monitoring open air experiment.
Table 5. Results of correlation analyses between phosphate concentration in leachates from the open air experiment in 2013, 2014, and 2017, and volumes of precipitation, volumes of leachate, and concentration of phosphates in precipitation, presented as Pearson correlation coefficient $r$ and probability $p$ (in brackets). $p$-values below 0.05 indicate statistically significant non-zero correlations at the 95.0% confidence level.

<table>
<thead>
<tr>
<th>Phosphate Concentrations in Leachate</th>
<th>Volume of Precipitation</th>
<th>Volume of Leachate</th>
<th>Concentration of Phosphates in Precipitation</th>
</tr>
</thead>
<tbody>
<tr>
<td>2013 (1st year of experiment)</td>
<td>0.1011 (0.6157)</td>
<td>0.3428 (0.0801)</td>
<td></td>
</tr>
<tr>
<td>2014 (2nd year of experiment)</td>
<td>0.5138 (0.0172)</td>
<td>0.2507 (0.2729)</td>
<td></td>
</tr>
<tr>
<td>2017 (5th year of experiment)</td>
<td>0.0456 (0.8486)</td>
<td>0.3622 (0.1166)</td>
<td></td>
</tr>
</tbody>
</table>

4. Discussion

Cumulative P-$\text{PO}_4^{3-}$ load released from fresh substrates tested in leaching column experiment, simulating the first vegetation period of green roof use (Table 2), accounted for only 1.74–2.41% of P content in substrate estimated in HCl extraction test (Figure 2). However, good correlation ($r = 0.9930$, $p = 0.002$) was observed between those values for all five tested fresh substrates. That suggests, that batch HCl extraction may be useful and recommended test for selection of substrates based on potential phosphorus leaching. However, it should be confirmed on the larger population samples of fresh green roof substrates. In this study, the correlation between leachable phosphates estimated by distilled water extraction and cumulative P loads in leachate was not significant (at $p < 0.005$).

The green roof runoff volume-weighted concentrations of P-$\text{PO}_4^{3-}$ observed in this study were lower in case of open air green roof model than in column leaching experiment. A similar situation was also observed by [12]. They explained, that stormwater on full scale roofs come in contact with more of the green roof surface prior to runoff. Comparing the results from column leaching experiment only, intensive substrates released P in higher concentrations than extensive substrates. Concentrations were higher than those reported in some studies [15,16], but lower than those reported by [17]. Concentrations from open-air experiments were generally very low, comparable more to extensive substrates than intensive ones. The differences among this study and studies reported by other authors may be attributed to a variety of factors, including the composition of growing media, the type and extent of vegetative cover, fertilizer use, and the age of the roofs. The impact of the substrate age has also been confirmed in this study in the case of volume-weighted mean concentrations and UALs, but not in the observed P-$\text{PO}_4^{3-}$ concentrations range (Table 3). In the fifth year of the experiment the green roof acted as a net sink of phosphorus (mean runoff concentration < atmospheric deposition). Roof age, followed by the summer and winter seasonal dynamics, had a greatest effect on phosphate concentrations in runoff from green roofs tested by [14]. Based on the created model, they predicted that the time, within which the green roof can act as a source of P may be as long as 11 years. Malcolm et al. [12] observed elevated P concentrations in the runoff more than three years after construction. However, still not enough long-term green roof studies have been carried out.

In terms of the impacts of green roof runoff on water bodies downstream, loads are more important than concentrations [32]. Unit area loadings (UAL), even if useful in the estimation of P-pollution from specific green roofs, are difficult to compare between studies, as the construction of green roofs (thickness of the substrate) vary in different cases. For example, [33] irrigated unplanted Arkalyte mix and GAF (Gardenscapes) substrates got similar to this study’s amount of rainwater (400–500 mm). Substrates released of about 500 and 800 mg P/m$^2$, respectively. The information of substrate depth is not included in the cited study, but the maximum depth of Green Roof Blocks™ is 10.16 cm, so the thickness of the substrate itself was less than 10 cm. In both studies fresh substrates were tested. Initial P content assessed for Arkalyte mix and GAF substrates was 60 mg/kg and 219 mg/kg, respectively, so they do not differ much from the values noted in our study (17–145 mg/kg). Gregoire and Clausen [15] calculated unit area loading from extensive GreenGrid®
media to 0.21 kg P-PO$_4^{2-}$/ha/year (21 mg P-PO$_4^{2-}$/m$^2$ year), what is the value close to obtained for extensive green roof substrate for the vegetation period in this study. UALs comparable to intensive substrates tested in this study were also reported by [8] for the substrate composed of crushed volcanic rock, compost, blonde peat, cooked clay, and washed sand.

Urban water bodies are already rich in phosphorus [34] because urban runoff and erosion are escalated by increased impervious cover. Mallin et al. [35] found mean orthophosphate concentrations of 0.0698 mg/L in creeks draining a highly-urbanized catchment, while in suburbanized catchment and rural catchment creeks orthophosphate concentrations were 0.0540 and 0.0486 mg/L, respectively. Taebi and Droste [36] reported unit loads of TP in urban stormwater runoff in areas with small, mild, and high precipitation of 0.2, 1.0, and 2.0 kg/ha/year, respectively. The load of total phosphorus discharged to the receiver from an urban stormwater system in Poland was estimated to be about 1.5 kg/ha of sealed area [37]. The role of green roofs is to solve environmental problems, e.g., by delaying stormwater runoff, improving air quality, or mitigating the urban heat island effect, not creating new problems by supplying urban waters with nutrients. Therefore, the study of green roof components for potential phosphorus outflow is so important.

Runoff from the rooftops is transported through a sewer system or is discharged directly into a receiving water body. Phosphorus export from green roofs may be problematic in this context, as P is commonly the main governing factor for eutrophication of freshwater [38,39]. The loss of P to urban water reservoirs can result in increased productivity of the system, and what follows, the rapid growth of algae and loss of aesthetic values. The relationship between phosphorus levels in lake water and lake clarity is well established. As the total amount of phosphorus in the water increases, the amount of algae in the water increases and the clarity of the water decreases [40]. Moreover, most algae are able to take in more phosphorus than needed during times when phosphorus levels are high and store it for later use when phosphorus levels in the water column are low [41]. An effect of phosphorus leaching from green roof substrates on urban water bodies may be attributed to a type of receiver. Reservoirs and rivers may be impacted by excess phosphorus loads, however, their response, in terms of algal blooms, may vary because of various physical factors, such as flow. A combination of a green roof with a pond, acting as a rainwater retention reservoir, is the common solution in newly-constructed residential areas. Shallow and small in volume, urban rainwater retention ponds are the most sensitive type of receiver, and may respond dramatically even to small inputs (loads) of phosphorus, resulting in excessive algae and plant growth. Phosphorus concentrations as low as 0.02–0.03 mg/L can already begin to promote blooms of algae. Significant and frequent blooms are noted because of various physical factors, such as flow. 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To keep urban water bodies in high environmental and aesthetic value, limiting P inflow from all potential sources is advisable. One of measure to prevent excessive P leaching from green roofs to urban receivers can be the implementation of specially-prepared low P-emission substrate [4]. Camm [43] integrated an adsorptive filter media for P removal to a green roof system. The evaluated media, reduced SRP loadings by 82.4% and TP loadings by 86.6%, however, the effluent concentrations still exceeded values of 0.03 mg/L for TP. Beck et al. [19] obtained 43% phosphate reduction in leachate by adding 7% biochar to bare green roof substrate. Malcolm et al. [12] implemented alum and Ultra_Phos Filter® to mitigate effects of nutrient pollution. The chosen method of application of Ultra_Phos Filter® did not reduce P in runoff, but alum reduced P by approximately 20% in runoff. Another option to mitigate runoff of phosphorus is to underline the substrate with P-reactive drainage material [44–46]. Construction of green roofs without any P release protection measure can strongly contribute to eutrophication of urban water bodies. To make green roof technology more sustainable at least one P reduction measures should be implemented. Those measures are: (i) selection of the substrate
mix with limited P content and release potential [4]; (ii) underlying the substrate with the P reactive material [44–46]; or (iii) implementation of a P-reactive filtration system to treat green roof runoff [47].

5. Conclusions

Results of performed experiments confirmed, that green roof substrates are a significant source of phosphorus in runoff. Batch HCl extraction test is recommended for the comparison and selection of substrates with low potential phosphorus leaching. Volume-weighted mean concentrations and UALs of P-PO$_4^{3-}$ in leachates from fresh substrates were higher in case of intensive substrates, but there was no clear relationship between the substrate type and the observed P-PO$_4^{3-}$ concentration range. This study also confirmed that substrate age influenced the volume-weighted mean P-PO$_4^{3-}$ concentrations and UAL was calculated based on the data from the vegetation period, but did not influence P-PO$_4^{3-}$ concentrations range found in the outflow.

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