Supporting Information: Synthesis, characterization and application of four novel electrochromic materials employing nitrotriphenylamine unit as the acceptor and different thiophene derivatives as the donor

Shuai Li¹, Guoliang Liu², Xiuping Ju³, Yan Zhang¹, Jinsheng Zhao^{1,4*}

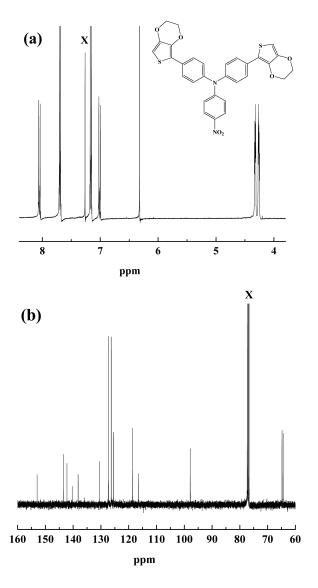


Fig.S1. ¹H NMR spectrum of NETPA in CDCl₃ (a). Solvent peak is at δ =7.26 ppm. ¹³C NMR spectrum of NETPA in CDCl₃ (b). Solvent peak is at δ = 77.3 ppm.

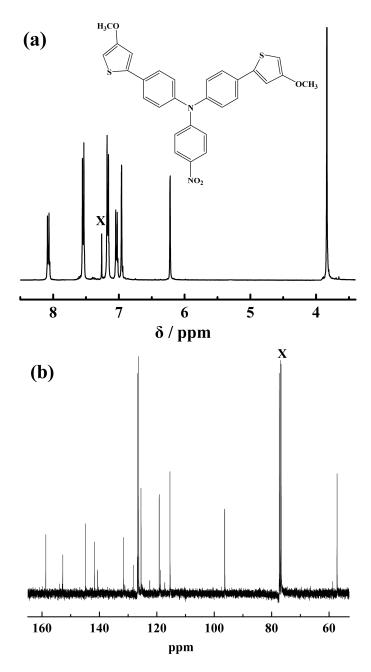


Fig.S2. 1 H NMR spectrum (a) and 13 C NMR spectrum (b) of NMOTPA in CDCl₃. The solvent peak in 1 H NMR is at δ = 7.26 ppm, and the solvent peak in 13 C NMR spectrum is at δ = 77.3 ppm.

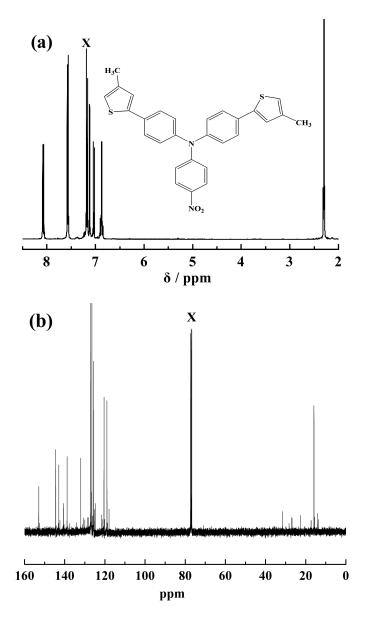


Fig.S3. ¹H NMR spectrum (a) and ¹³C NMR spectrum (b) of **NMTPA** in CDCl₃. The solvent peak in ¹H NMR spectrum is at δ = 7.26 ppm and the solvent peak in ¹³C NMR spectrum peak is at δ = 77.3 ppm.

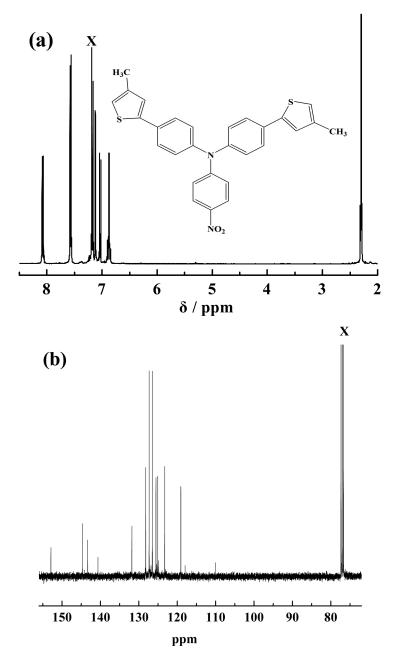


Fig. S4. ¹H NMR spectrum(a) and ¹³C NMR spectrum (b) of **NTTPA** in CDCl₃. The solvent peak for ¹H NMR spectrum is at δ = 7.26 ppm. The solvent peak in ¹H NMR spectrum is at δ = 7.26 ppm and the solvent peak in ¹³C NMR spectrum peak is at δ = 77.3 ppm.

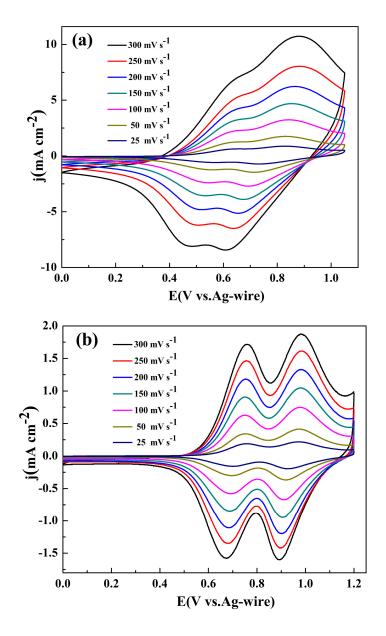


Fig. S5. CV curves of the NETPA (a), NMOTPA(b), NMTPA (c) and NTTPA (d) film at different scan rates between 25 mV s⁻¹ and 300 mV s⁻¹ in the monomer-free 0.2 M NaClO₄/ACN/DCM solution.